

N-Containing Biomass for the Sustainable Synthesis of *N*-Heterocycles *via* Cyclization Reactions

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Abstract

The *N*-containing compounds from biomass, including chitin, chitosan and D-glucosamine, are one of the largest sustainable native biobased materials on earth. Human processing of the sustainable chitin/chitosan/D-glucosamine-containing raw materials exhibits only a very minor fraction of all the chitin produced annually in nature, while the major fraction maintains intact. In order to reduce the high reliance on non-renewable feedstocks, chitin/chitosan/D-glucosamine as a substitute has been increasingly utilized for value-added functional materials and important chemical feedstocks through diverse routes, mainly including chemical modifications, transformations, as ligands or catalysts in organic synthetic pathways. Despite the tremendous progress on the usage of the *N*-containing biomass, it is still a great challenge to directly utilize it to prepare important *N*-heterocyclic compounds through cyclization reactions.

The targeted cleavage of C–N bonds of alkyl primary amines in chitin, chitosan and D-glucosamine *via* a metal-free pathway and the conjunction of nitrogen in the synthesis of imidazo[1,5-*a*]pyridines are still highly challenging. In publication 1, we reported an anomeric stereoauxiliary approach for the synthesis of a wide range of imidazo[1,5-*a*]pyridines after cleaving the C–N bond of D-glucosamine (α -2° amine) from biobased resources. This new approach expands the scope of readily accessible imidazo[1,5-*a*]pyridines relative to existing state-of-the-art methods. A key strategic advantage of this approach is that the α -anomer of D-glucosamine is capable of C–N bond cleavage through a seven-membered ring transition state. Using this novel method, a series of imidazo[1,5-*a*]pyridine derivatives (more than 80 examples) were synthesized from pyridine ketones (including para-dipyridine ketone) and aldehydes (including para-dialdehyde). Moreover, imidazo[1,5-*a*]pyridines derivatives containing diverse important deuterated C(sp²)–H and C(sp³)–H bonds were also efficiently achieved.

In publication 2, we described a facile and efficient one-pot methodology that enables nitrogen interception directly from chitosan/chitin for the synthesis of a broad range of important *N*-heterocycles imidazo[1,5-*a*]pyridines (52 examples). This strategy is featured by directly synthesizing tridentate ligands and important deuterated imidazo[1,5-*a*]pyridines. In particular, an extended range of various functional moieties on imidazo[1,5-*a*]pyridine backbone are tolerated with high efficiency under these mild and catalyst-free conditions.

Diverse aminocatalysis modes have been discovered over the last decades, while sustainable aminocatalyst with native chiral skeleton from biomass for the regioselective annulations reaction are highly desirable but not realized yet. In publication 3, a stereoauxiliary aminocatalysis strategy from β -anomeric glucosamine was achieved through the regioselective annulation of pyridine ketone with α,β -

unsaturated aldehyde for the construction of trisubstituted indolizine-2-carbaldehydes. Using our strategy with native chiral skeleton for the regioselective control, a highly expanded range of commercially available but oxidatively sensitive α,β -unsaturated aldehydes can act as coupling partners for the efficient preparation of readily accessible trisubstituted indolizine-2-carbaldehydes *via* one-pot pathway.

The thesis is a cumulative work with 3 publications. Two of them are published in peer-reviewed journals and the third one is in preparation. The background, the objective of the studies, results, discussions, general conclusions, perspectives and experimental section are shown in sections 1-5.

Zusammenfassung

Die N-haltigen Verbindungen aus Biomasse, darunter Chitin, Chitosan und D-Glucosamin, sind eines der größten nachhaltigen nativen biobasierten Materialien der Erde. Die menschliche Verarbeitung der nachhaltigen Chitin/Chitosan/D-Glucosamin-haltigen Rohstoffe weist nur einen sehr geringen Teil des jährlich in der Natur produzierten Chitins auf, während der Hauptteil intakt bleibt. Um die hohe Abhängigkeit von nicht erneuerbaren Rohstoffen zu verringern, wurde Chitin/Chitosan/D-Glucosamin als Ersatz zunehmend für wertsteigernde funktionelle Materialien und wichtige chemische Rohstoffe auf verschiedenen Wegen verwendet, hauptsächlich einschließlich chemischer Modifikationen, Transformationen, als Liganden oder Katalysatoren in organischen Synthesewegen. Trotz der enormen Fortschritte bei der Nutzung der N-haltigen Biomasse ist es immer noch eine große Herausforderung, sie direkt zu nutzen, um wichtige N-heterocyclische Verbindungen durch Cyclisierungsreaktionen herzustellen.

Die gezielte Spaltung von C-N-Bindungen von primären Alkylaminen in Chitin, Chitosan und D-Glucosamin über einen metallfreien Weg und die Bindung von Stickstoff in der Synthese von Imidazo[1,5-a]pyridinen stellen nach wie vor eine Aufgabe für die aktuelle Forschung dar. In Publikation 1 berichteten wir über einen anomeren Stereoauxiliar-Ansatz für die Synthese einer breiten Palette von Imidazo[1,5-a]pyridinen nach Spaltung der C-N-Bindung von D-Glucosamin (α -2°-Amin) aus biobasierten Ressourcen. Dieser neue Ansatz erweitert den Anwendungsbereich leicht zugänglicher Imidazo[1,5-a]pyridine im Vergleich zu bestehenden Methoden nach dem Stand der Technik. Ein entscheidender strategischer Vorteil dieses Ansatzes besteht darin, dass das α -Anomer von D-Glucosamin über einen siebengliedrigen Ringübergangszustand zur C-N-Bindungsspaltung fähig ist. Mit dieser neuartigen Methode wurde eine Reihe von Imidazo[1,5-a]pyridin-Derivaten (mehr als 80 Beispiele) aus Pyridinketonen (einschließlich para-Dipyridinketon) und Aldehyden (einschließlich para-Dialdehyd)

synthetisiert. Darüber hinaus wurden auch Imidazo[1,5-a]pyridine-Derivate mit verschiedenen wichtigen deuterierten C(sp²)-H- und C(sp³)-H-Bindungen effizient erhalten.

In Veröffentlichung 2 haben wir eine einfache und effiziente Eintopfmethode beschrieben, die das Abfangen von Stickstoff direkt aus Chitosan/Chitin für die Synthese einer breiten Palette wichtiger N-Heterocyclen Imidazo[1,5-a]pyridine (52 Beispiele) ermöglicht. Diese Strategie zeichnet sich durch die direkte Synthese von dreizähligen Liganden und wichtigen deuterierten Imidazo[1,5-a]pyridinen aus. Insbesondere wird unter diesen milden und katalysatorfreien Bedingungen ein breiter Bereich verschiedener funktioneller Einheiten am Imidazo[1,5-a]pyridin-Rückgrat mit hoher Effizienz toleriert.

In den letzten Jahrzehnten wurden verschiedene Arten der Aminokatalyse entdeckt, während nachhaltige Aminokatalysatoren mit nativem chiralem Gerüst aus Biomasse für die regioselektive Anellierungsreaktion sehr wünschenswert, aber noch nicht realisiert sind. In Publikation 3 wurde eine stereoauxiliäre Aminokatalysestrategie ausgehend von β -anomerem Glucosamin durch die regioselektive Anellierung von Pyridinketon mit α,β -ungesättigtem Aldehyd zum Aufbau trisubstituierter Indolizin-2-carbaldehyde erreicht. Unter Verwendung unserer Strategie mit nativem chiralem Gerüst zur regioselektiven Steuerung kann eine stark erweiterte Palette kommerziell erhältlicher, aber oxidativ empfindlicher α,β -ungesättigter Aldehyde als Kupplungspartner für die effiziente Herstellung von leicht zugänglichen trisubstituierten Indolizin-2-carbaldehyden im Eintopfverfahren fungieren Weg.

Die Dissertation ist eine kumulative Arbeit mit 3 Publikationen. Zwei davon werden in Fachzeitschriften mit Peer-Review veröffentlicht und das dritte ist in Vorbereitung. Der Hintergrund, das Ziel der Studien, Ergebnisse, Diskussionen, allgemeine Schlussfolgerungen, Perspektiven und der experimentelle Teil werden in den Abschnitten 1-5 dargestellt.

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List of Publications

Publication 1

Anomeric Stereoauxiliary Cleavage of the C–N Bond of D-glucosamine for the Preparation of Imidazo[1,5-a]pyridines

Kui Zeng, Jin Ye, Xintong Meng, Sebastian Dechert, Martin Simon, Shuaiyu Gong, Ricardo A. Mata, Kai Zhang*

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Publication 2

Direct Nitrogen Interception from Chitin/chitosan for the Preparation of Imidazo[1,5-a]pyridines

Kui Zeng, Ruhuai Mei, Xizhou Cecily Zhang, Loren B Andreas, Kai Zhang*

Chem. Commun. **2022**, 58, 6068-6071.

Publication 3

Anomeric Stereoauxiliary-organocatalyzed One-pot Site-selective C₂-aldehylation for Trisubstituted Indolizine-2-carbaldehydes

Kui Zeng, Jonathan J. Wong, Sebastian Dechert, Franc Meyer, K. N. Houk,* Kai Zhang*

In preparation **2022**.

Abbreviations and explanations

Ac	acetyl	HBTU	hexafluorophosphate benzotriazole tetramethyl uronium
Ac ₂ O	acetic anhydride	HATU	hexafluorophosphate azabenzotriazole tetramethyl uronium
Alk	alkyl	HSQC	heteronuclear single quantum coherence
ADGF	2-acetamido-3,6-anhydro-2-deoxyglucofuranose	HRMS	high resolution mass spectrometry
[A]	amidation process	<i>J</i>	coupling constant
Ar	aryl	L	ligand
ADMF	2-acetamido-3,6-anhydro-2-deoxymannofuranose	M	metal
3A5AF	3-acetamido-5-acetylfuran	Me	methyl
Bn	benzyl	<i>m</i>	<i>meta</i>
Bu	butyl	M.p.	melting point
BSA	<i>N,O</i> -bis(trimethylsilyl)acetamide	MHET	mono(2-hydroxyethyl) terephthalate
BH reaction	Baylis–Hillman reaction	NBS	<i>N</i> -bromosuccinimide
Boc	<i>tert</i> -butoxycarbonyl	NCS	<i>N</i> -chlorosuccinimide
Cy	cyclohexyl	NBE-CO ₂ Me	2-carbomethoxynorbornene
CDC	cross-dehydrogenative coupling	NMR	nuclear magnetic resonance
<i>Cis</i>	functional groups (substituents) are on the same side of some plane	<i>o</i>	<i>ortho</i>
CmCDA	cyclobacterium marinum chitobiose deacetylase	[O]	oxidation process
DCE	1,2-dichloroethane	<i>p</i>	<i>para</i>
DMA	<i>N,N</i> -dimethylformamide	PA	propionic acid
DIEA	<i>N,N</i> -diisopropylethylamine	Pd(dba) ₂	bis(dibenzylideneacetone)palladium
DMAP	4-(dimethylamino)pyridin	PdCl ₂ (dtbpf)	1,1' -Bis-(di- <i>tert</i> .-butylphosphino-

)ferrocen-palladiumdichlorid
DMSO	dimethylsulfoxide	Ph	phenyl
DNA	deoxyribonucleic acid	Piv	2,2-dimethylpropanoyl
DCAA	dichloroacetic acid	pKa	logarithmic acid dissociation constant
DFT	density functional theory	Pr	propyl
DIBAH	diisobutylaluminum hydride	PET	poly(ethylene terephthalate)
EDG	electron-donating group	py	pyridine
EI	electron ionization	Pe	pentanyl
equiv.	equivalents	PMP	para-methoxyphenyl
<i>E</i> isomer	on opposite sides of the double bond	Pin	pinacol
ESI	electrospray ionization	δ	chemical shift
<i>ee</i>	enantiomeric excess	SET	single electron transfer
Et	ethyl	S _N 1	first-order nucleophilic substitution
Enz	enzyme	S _N 2	second-order nucleophilic substitution
EWG	electron-withdrawing group	<i>Si</i> face	originate from the Latin sinister (left)
FG	functional group	TMS	trimethylsilyl
FTIR	fourier transform infrared spectroscopy	<i>Trans</i>	functional groups (substituents) are on the opposing side of some plane
FAD	flavin adenine dinucleotide	TEA	triethylamine
FMF	5-(formyloxymethyl)furfural	<i>t</i> -Am	2-methylbut-2-yl
GlcN	D-glucosamine hydrochloride salt	TEMPO	2,2,6,6-tetramethylpiperidinyloxy
GlcNAc	<i>N</i> -acetyl-D-glucosamine hydrochloride salt	Tf	trifluoromethanesulfonyl
GC-MS	gas chromatography-mass spectrometry	TFE	2,2,2-trifluoroethanol
Hept	heptyl	THF	tetrahydrofuran
5HMF	5-hydroxymethylfurfural	TM	transition metal

HFIP

1,1,1,3,3,3-hexafluoro-2-
propanol

Tert

tertiary

1. Introduction

Biomass is renewable organic material derived from plants, bacteria, fungi, protists, animals and viruses (**Figure 1**).¹ The biomass resource can be considered as organic matter in which solar energy is stored chemically.² In the past decade, research on biomass valorization has mainly focused on the (bio)chemical conversion of C-, H- and O-containing fractions (*viz.* DNA, RNA, carbohydrates, lignin, triglycerides and proteins),²⁻¹⁸ whereas the N-containing fractions (*viz.* chitin, chitosan and D-glucosamine) have received little attention so far.^{12,19,20} The abundant N-containing biomass, such as chitin and its derivatives, plays an irreplaceable role in organic organization.^{18,21,22} In the following we will mainly discuss chitin and its derivatives.

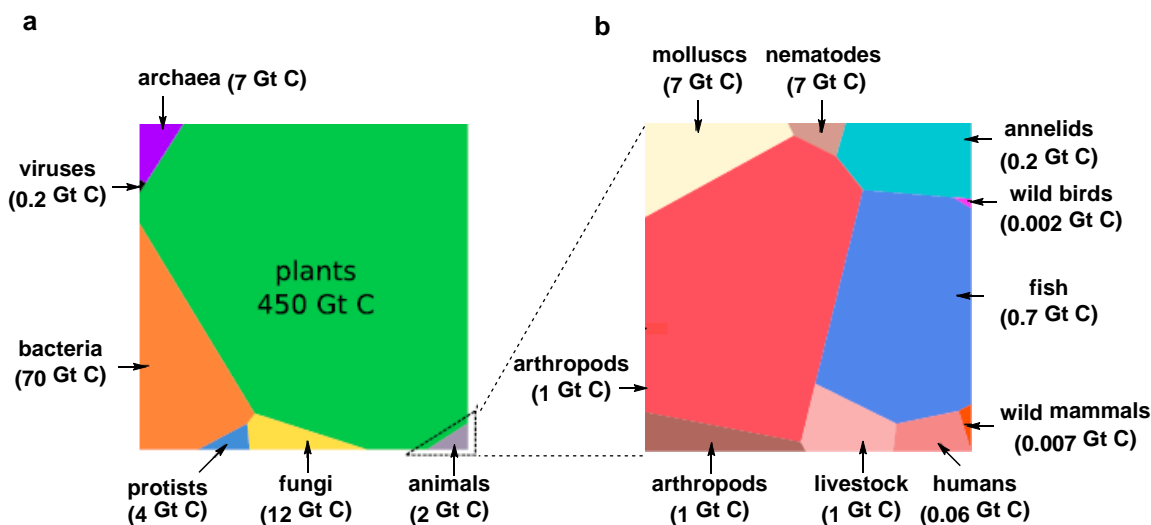


Figure 1 Graphical representation of the global biomass distribution by taxa: **(a)** Absolute biomasses of different taxa are represented using a Voronoi diagram, with the area of each cell being proportional to that taxa global biomass. Values are based on the estimates presented in ref.¹. A visual depiction without components with very slow metabolic activity, such as plant stems and tree trunks, is shown in SI Appendix, Fig. S1 of ref.¹ **(b)** Absolute biomass of different animal taxa. Related groups such as vertebrates are located next to each other. We estimate that the contribution of reptiles and amphibians to the total animal biomass is negligible, as we discuss in the SI Appendix of ref.¹. Visualization performed using the online tool at bionic-vis.biologie.uni-greifswald.de/. 1 Gt C = 1015 g of carbon. (Adapted with permission from ref.¹, Copyright from National Academy of Sciences.).

1.1 Chitin, chitosan and D-glucosamine

1.1.1 Origin, structures and characterization

Chitin, the second largest sustainable native polymer, can be found as a support material for many terrestrial organisms, marine organisms, microorganisms and crustacean shells (**Figure 2**).¹⁸ For example, the organic matrix of honeybees is composed of 23%-32% chitin, 35%-45% proteins, 30%-40% melanin and 3% minerals.²³ Although chitin occurs in fungi, diatoms, nematodes, arthropods, and many other animals and plants, development has focused on extraction on a limited number of chitin species due to inconsistent physicochemical characteristics of products from aquatic organisms on an industrial scale.^{24,25} Shrimps, crabs, lobsters, krill and squid waste from the marine processing industry has become a major resource in use today.²⁶

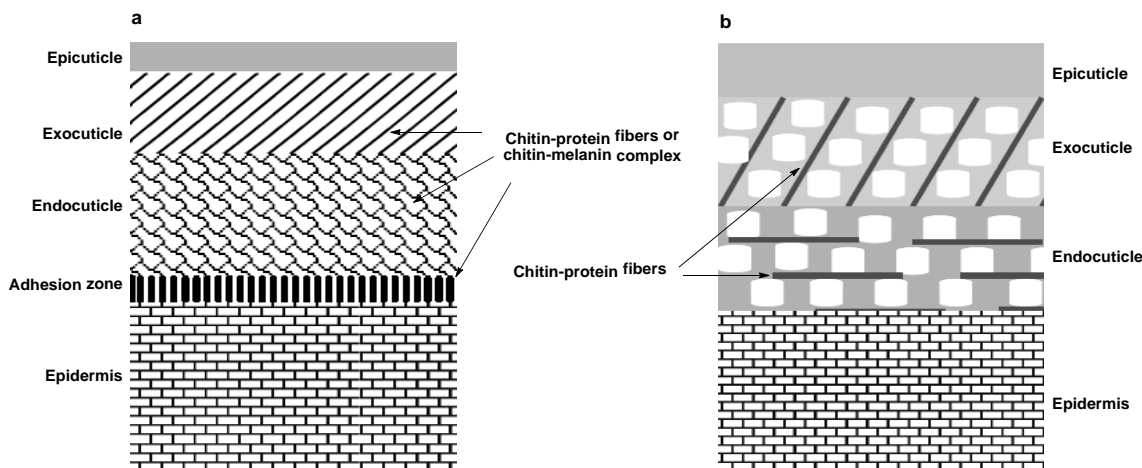


Figure 2 Schematic interpretation of organic matrix: (a) insect cuticle (rod and sheet types of chitin); (b) terrestrial crustacean (rod type of chitin) (Not drawn to scale. Adapted with permission from Ref.¹⁸, Copyright from Taylor and Francis Group LLC (Books) US).

The main production processes for chitin and its derivatives are based on chemical,^{27,28} enzymatic^{29,30} and fermentative methods.³¹⁻³³ The general procedures for the extraction of chitin, chitosan and D-glucosamine from organic organisms include demineralization, deproteination, decolorization, deacetylation, hydrolysis and crystallization (**Figure 3**).³⁴⁻³⁶

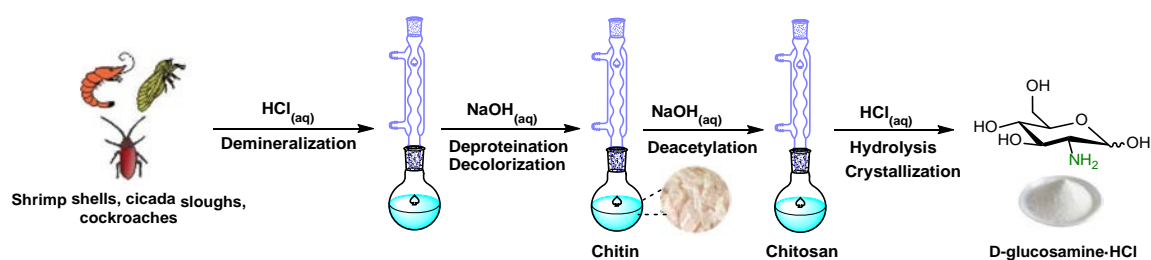


Figure 3 Schematic interpretation of the general procedure to obtain chitin, chitosan and D-glucosamine (Adapted with permission from Ref,³⁶ Copyright from WILEY-VCH Verlag GmbH & Co.).

Chitin is an abundant native linear copolymer composed of *N*-acetyl-D-glucosamine and D-glucosamine units linked with β -(1-4) glycosidic bonds (**Figure 4**).¹⁸ Natural chitin samples may contain broadly different amounts of *N*-acetyl groups, depending on their origin and isolation procedure. The quantities of *N*-acetyl-D-glucosamine units are generally more than 50% in chitin backbone, while chitosan contains less than 50% *N*-acetyl-D-glucosamine.

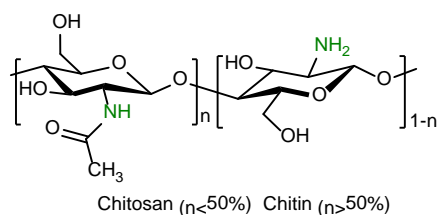


Figure 4 Schematic interpretation of chitin and chitosan structure.

Depending on the source, chitin in its natural state has three anhydrous crystalline polymorphs, including α -, β -, γ -chitin (**Table 1**).^{37,38} β - and γ -chitin irreversibly convert to α -chitin, therefore, the α -chitin form is the majority of the three anhydrous crystalline polymorphs.¹⁸ α -Chitin exhibits two antiparallel molecules per unit cell, while β -chitin exhibits a parallel arrangement (**Figure 5**). α -Chitin has strong inter-sheet and intra-sheet hydrogen bonding.³⁹ β -Chitin has strong intermolecular hydrogen bonding (C=O and NH) between the chains along the *a*-axis⁴⁰ and weak hydrogen bonding by intra-sheets.⁴¹ γ -Chitin has not been completely identified. Different from chitin as anhydrous crystalline polymorphs, chitosan is primarily a semicrystalline polymer in the solid state.¹⁸

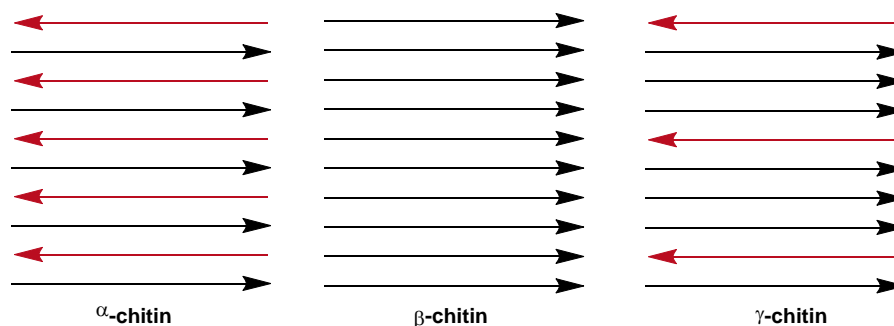


Figure 5 Antiparallel and parallel arrangements of different allomorphs of chitin.

Table 1 Sources of chitin and chitosan

Types	Sources
α -chitin	Insects (cuticle, ⁴² ovipositors ⁴³); Crustaceans (Crab and shrimp shell); Centric Diatoms (Algae); Fungi (Mucor rouxi, Aspergillus nidulans ⁴²)
β -chitin	Squid (Ommastrephes pen ⁴⁴); Centric Diatoms (Thalassiosira fluviatilis ⁴⁵)
γ -chitin	Insects (Beetle cocoon ³⁸); Squid (Loligo stomach wall ⁴⁶)

In amino sugars, the hydroxyl group of the sugar is replaced by an amino group. Theoretically any hydroxyl group of a sugar can be substituted. Therefore, there should be many different varieties of amino sugars. In nature, amino sugars are limited in number, including D-glucosamine, D-mannosamine, D-galactosamine, D/L-rhamnosamine, D/L-fucosamine and their *N*-acetyl derivatives (**Table 2**).⁴⁷ Sugar molecules with amino groups in unnatural positions can be artificially synthesized.⁴⁷

D-glucosamine with four free hydroxyl groups and one free amino group is usually formulated as hydrochloride or glucosamine sulfate, which is hydrolyzed from chitin and chitosan. The hydroxyl group in the anomeric carbon C₁ of D-glucosamine contains α and β -anomer. Both forms are stable in the solid state, but interconvert in aqueous solutions to achieve equilibrium distributions through a mutarotation process (**Figure 6**).^{36,48} The predominant anomeric form of unprotonated glucosamine is the β -form, while protonated glucosamine takes foremost the α -form.

Table 2 Representation of native amino sugars

	D-glucosamine	D-mannosamine	D-galactosamine	D-rhamnosamine	L-rhamnosamine	D-fucosamine	L-fucosamine
Fischer projection	$\begin{array}{c} \text{CHO} \\ \\ \text{H} - \text{C} - \text{NH}_2 \\ \\ \text{HO} - \text{C} - \text{H} \\ \\ \text{H} - \text{C} - \text{OH} \\ \\ \text{H} - \text{C} - \text{OH} \\ \\ \text{CH}_2\text{OH} \end{array}$	$\begin{array}{c} \text{CHO} \\ \\ \text{H}_2\text{N} - \text{C} - \text{H} \\ \\ \text{HO} - \text{C} - \text{H} \\ \\ \text{H} - \text{C} - \text{OH} \\ \\ \text{H} - \text{C} - \text{OH} \\ \\ \text{CH}_2\text{OH} \end{array}$	$\begin{array}{c} \text{CHO} \\ \\ \text{H} - \text{C} - \text{NH}_2 \\ \\ \text{HO} - \text{C} - \text{H} \\ \\ \text{HO} - \text{C} - \text{H} \\ \\ \text{H} - \text{C} - \text{OH} \\ \\ \text{CH}_2\text{OH} \end{array}$	$\begin{array}{c} \text{CHO} \\ \\ \text{H}_2\text{N} - \text{C} - \text{H} \\ \\ \text{HO} - \text{C} - \text{H} \\ \\ \text{H} - \text{C} - \text{OH} \\ \\ \text{H} - \text{C} - \text{OH} \\ \\ \text{CH}_3 \end{array}$	$\begin{array}{c} \text{CHO} \\ \\ \text{H} - \text{C} - \text{NH}_2 \\ \\ \text{H} - \text{C} - \text{OH} \\ \\ \text{HO} - \text{C} - \text{H} \\ \\ \text{HO} - \text{C} - \text{H} \\ \\ \text{CH}_3 \end{array}$	$\begin{array}{c} \text{CHO} \\ \\ \text{H} - \text{C} - \text{NH}_2 \\ \\ \text{HO} - \text{C} - \text{H} \\ \\ \text{HO} - \text{C} - \text{H} \\ \\ \text{H} - \text{C} - \text{OH} \\ \\ \text{CH}_3 \end{array}$	$\begin{array}{c} \text{CHO} \\ \\ \text{H}_2\text{N} - \text{C} - \text{H} \\ \\ \text{H} - \text{C} - \text{OH} \\ \\ \text{H} - \text{C} - \text{OH} \\ \\ \text{HO} - \text{C} - \text{H} \\ \\ \text{CH}_3 \end{array}$
Chair confirmation							
Haworth projection							

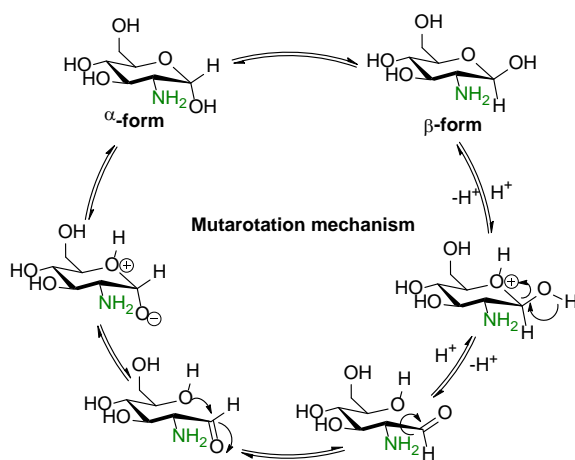


Figure 6 The mutarotation mechanism of D-glucosamine under aqueous conditions.⁴⁷

1.1.2 Physical and chemical properties

Chitin as a highly crystalline material has specific solvent behavior, which is influenced by $-\text{NH}_2$ and $-\text{OH}$ groups. Chitin is generally hydrophobic, soluble in hexafluoroisopropanol, chloroalcohols, and hexafluoroacetone mixed with an aqueous mineral acid solution, soluble in dimethylacetamide containing 5% lithium chloride, but insoluble in water and organic solvents at room temperature.⁴⁹ Due to the high degree of deacetylation, chitosan is generally dissolved in some organic acids (methanoic acid and acetic

acid *etc.*) and inorganic acids (1% hydrochloric acid, dilute nitric acid and lactic acid *etc.*) by protonation of amino groups. In contrast, chitosan is insoluble in alkaline and neutral aqueous solutions. The most frequently used solvent for chitosan is 1% acetic acid aqueous solution at pH 4.0. In addition, depolymerization of chitosan occurs at high temperatures in concentrated acetic acid.⁵⁰ D-glucosamine hydrochloride (GlcN) and *N*-acetyl-D-glucosamine hydrochloride (GlcNAc) are obtained by hydrolysis of chitin and chitosan. Generally, it is a white and sweet powder that decays at 221°C.⁵¹ The solubility of GlcNAc in water is about 25%. Chitin, chitosan and D-glucosamine have attracted great interest in developing diverse applications due to the beneficial properties such as excellent biocompatibility, antimicrobial activity and nontoxicity.⁵⁰⁻⁵⁶

1.1.3 Modifications

Chitin, chitosan and D-glucosamine have natural active sites, including hydroxyl and amino groups, that can be easily further modified. The modified derivatives can show better solubility and interesting physicochemical properties for various applications.

There are many reviews on the modification of chitin and chitosan.^{66,52} Therefore, we briefly summarize the four reaction types (**Table 3**). Sulfonation of chitin and chitosan is a general route to endow them with glycosaminoglycans-analogue biological activity.⁵² In this respect, advances from diverse sulphating agents including H₂SO₄, SO₃, SO₃·pyridine complex, ClSO₃H/H₂SO₄, ClSO₃H/formic acid, Me₃N·SO₃ complex, Oleum/DMF, ClSO₃H/formamide/dichloroacetic acid (DCAA) have widely expanded the chemical repertoire of accessible reaction sites (**Table 3**).^{52,57,58} Phosphorylated chitin and chitosan have attracted considerable interest due to their anti-inflammatory properties, ability to form metal complexes, blood compatibility, and formation of anionic polyelectrolyte hydrogels. Therefore, tremendous synthetic strategies have been reported in the past decade. Representative phosphorylating agents are P₂O₅/CH₃SO₃H; H₃PO₄/urea; H₃PO₄/P₂O₅/Et₃PO₄; ClP(O)(OEt)₂; H₃PO₄/HCHO; ClC₂H₄P(O)(OH)₂; PhP(O)(OH)₂/HCHO (**Table 3**).⁵⁹⁻⁶³ The most common method to increase the positive charge density of chitin and chitosan is to introduce quaternary ammonium salts (**Table 3**). The general direct method is a nucleophilic substitution reaction between the amino group of chitosan and an alkyl halide under alkaline conditions.^{64,65} In another approach, the amino group of chitosan is first reacted with an aldehyde to form an imine, which is subsequently reduced by NaBH₄, and the reduced product is reacted with an haloalkane *via* a nucleophilic substitution reaction.^{66,67} Indirect methods to introduce external quaternary ammonium groups broadly expand the types of reactants to include *N*-(3-chloro-2-hydroxypropyl) trimethyl ammonium chloride, glycidyltrimethyl ammonium chloride, (3-bromopropyl)

trimethylammonium bromide, (5-bromopentyl) trimethylammonium bromide *etc.*⁶⁸ The carboxyalkylation reaction of chitin and chitosan has attracted great attention in application fields such as biosensor wound healing, food industry, and bio-imaging.^{69,70} To date, various synthetic agents have been explored, such as mono- or dicarboxylic acids (phthalic, succinic, maleic), acetic, propionic, butyric, valeric and hexanoic acids anhydride.^{62,71-73} In addition to these four most prominent types of modification, there are several more interesting methods to mention here: thiolation, fluorination, *N*-pathaloylation and other cross-linking methods.⁶²

Table 3. Modification of chitin and chitosan

Reaction types	Reaction agents	Ref.
Sulphonation	H ₂ SO ₄ ; SO ₃ ; SO ₃ ·pyridine; ClSO ₃ H/H ₂ SO ₄ ; ClSO ₃ /formic acid; Me ₃ N·SO ₃ ; Oleum/DMF; ClSO ₃ H/formanide/DCAA	52,57,58
Phosphorylation	P ₂ O ₅ /CH ₃ SO ₃ H; H ₃ PO ₄ /urea; H ₃ PO ₄ /P ₂ O ₅ /Et ₃ PO ₄ ; ClP(O)(OEt) ₂ ; H ₃ PO ₄ /HCHO; ClC ₂ H ₄ P(O)(OH) ₂ ; PhP(O)(OH) ₂ /HCHO	59-63
Quaternary ammonium	Direct quaternization: Halogenated alkane; imine reduction; Indirect quaternization: Introduction of external quaternary ammonium groups <i>via</i> click reaction, nucleophilic substitution reaction	68
Carboxyalkylation	Mono- or dicarboxylic acids (phthalic, succinic, maleic), acetic, propionic, butanoic, valeric and hexanoic acids anhydride	62,71-73

D-glucosamine hydrochloride can generally be modified using organic synthesis strategies. There are many strategies for amino modification of D-glucosamine. The amino group of D-glucosamine is easier to be modified than hydroxyl group due to their higher reactive activity. These mainly include Schiff's bases reaction (**Figure 7a**)⁷⁴⁻⁷⁸ and amidation reaction (**Figure 7b**).⁷⁹⁻⁸¹ In 2019, Voglmeir *et al.* reported that the enzyme is capable of a series of *N*-acylation and *N*-transacylation reactions to prepare *N*-

acylglucosamines (**Figure 7c**).⁸² This strategy overcomes the challenge of chemoselective acylation of glucosamine derivatives and is used for the total synthesis of sialosides.

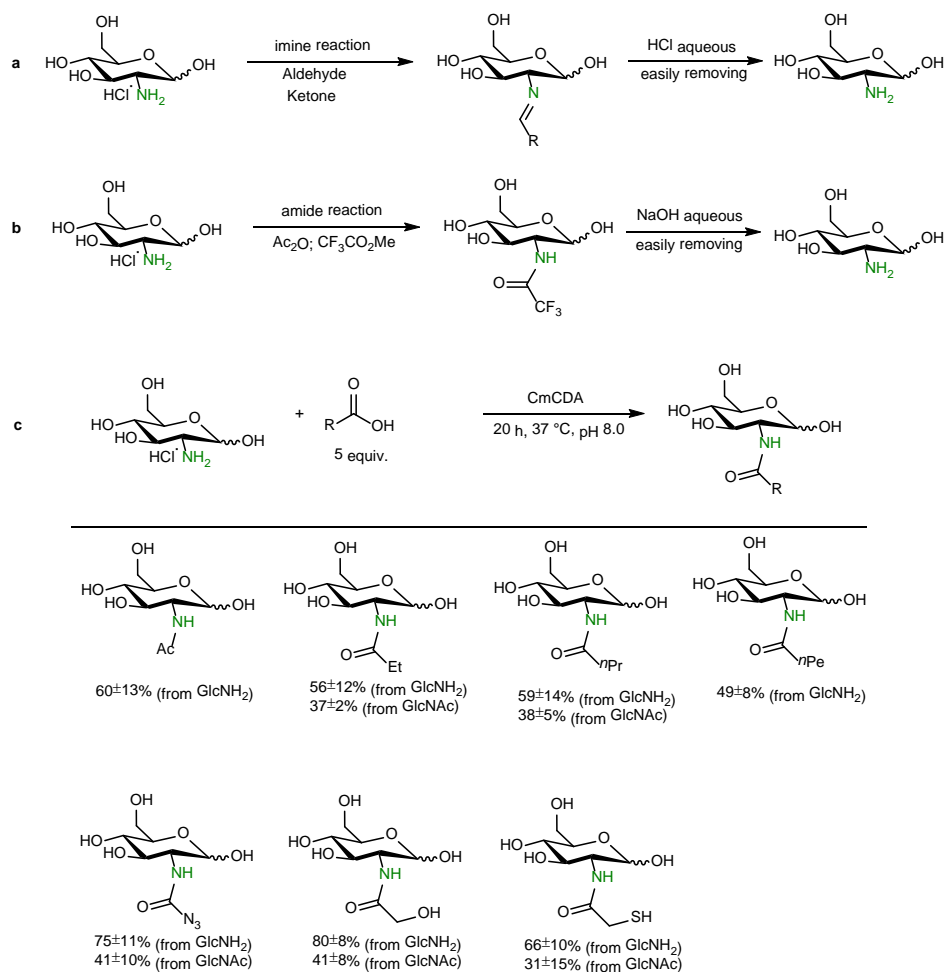


Figure 7 (a)-(b) The representative strategies of modifications of amino group of D-glucosamine. **(c)** Biocatalysis of D-glucosamine for *N*-acylglucosamines. CmCDA= *Cyclobacterium marinum* chitobiose deacetylase

Because these modified groups can be easily removed, for example, the imine group can be removed by hydrolysis with aqueous HCl (**Figure 7a**),⁷⁴⁻⁷⁸ and the trifluoroacetyl group can be removed by hydrolysis with aqueous NaOH (**Figure 7b**)⁷⁹⁻⁸¹. Therefore, chemists usually utilize them as representative block agents for site-selective modification of glucosamine. *O*-site-selective modification of glucosamine is possible by Schiff base reaction using aldehydes and ketones as block agent. The first imines derived from

2-amino-2-deoxyaldoses and salicylaldehyde were reported by Irvine and Hynd in 1913.^{75,76} The method uses H₂O as the solvent and is limited to substrates with a hydroxyl group in ortho position. Costamagna *et al.* improved this strategy by using methanol as the solvent and applied it to the 2-hydroxynaphthylaldehyde substrate.⁷⁷ Nguyen *et al.* reported an efficient method for the preparation of imines *via* α -D-glucosamine and diverse aromatic aldehydes (**Figure 8a**).⁷⁸ After optimizing diverse base and solvent, NaOH and methanol are the best conditions. They discovered that those substrates with a hydroxyl group in the ortho position of the aromatic aldehyde gave more stable products. In 1950, Micheel and Wulff used unprotected glycosylazide with acetone to synthesize imines under acidic conditions. However, imines failed to obtain per-*O*-acetylated imines by the Ac₂O agent (**Figure 8b**).⁸³ Bertho, Wolfrom, Perez *et al.* reported and improved this reaction pathway from D-glucosamine in three steps (**Figure 8c**).⁸⁴⁻⁸⁶ First, per-*O*-acetylated glycosylbromo was synthesized with AcBr, and then the compound was further modified with AgN₃ to give per-*O*-acetylated glycosylazide. Finally, the desired imine product was obtained. Perez *et al.* also expanded the scope of ketone and stereochemistry by DFT and 2D-NMR analysis, providing more details.⁸⁵

Per-*O*-acetylated *N*-TFA-glucosamine can be synthesized by amidation reaction using CF₃CO₂Me as a block agent and is widely used as a precursor of *O*-,^{87,88} *S*-^{89,90} and *N*-glycosides^{91,92} (**Figure 9a**). For example, in 2004, Walsh *et al.* utilized this strategy to prepare glycopeptide variants of the antibiotic tyrocidine (**Figure 9b**).⁹¹

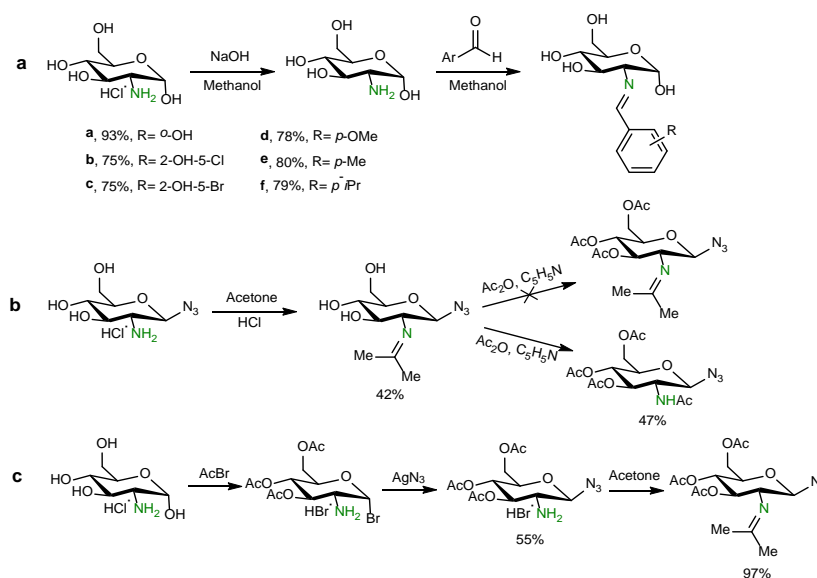


Figure 8 (a) Synthesis of imine *via* α -D-glucosamine with diverse aromatic aldehydes; (b), (c) Synthesis of imine *via* D-glucosamine with ketone.

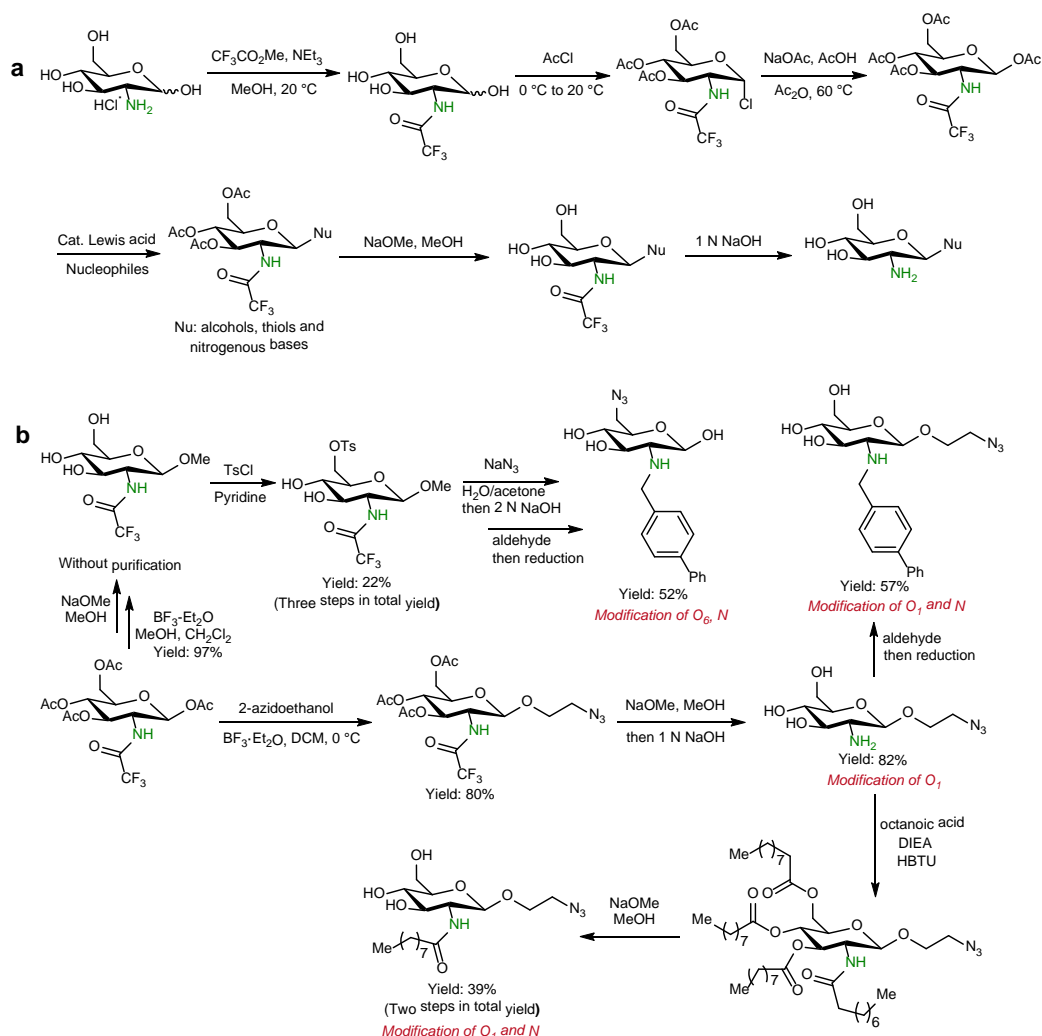


Figure 9 (a) Trifluoroacetyl as protected agents for diverse late-stage modifications. (b) one representative application example (here briefly show the synthesis parts of the ref.⁸⁸). DIEA= *N,N*-Diisopropylethylamine. HBTU= hexafluorophosphate benzotriazole tetramethyl uronium.

1.1.4 Transformations

The transformation of chitin and chitosan reported in the literature mainly involves hydrolysis to produce monomeric GlcNH₂ and GlcNAc.⁹³⁻⁹⁶ Here we do not mention the oligosaccharides of chitin and chitosan.^{97,98} Monomers are further modified into value-added products, such as hydrogenation to prepare alcohols,⁹⁹⁻¹⁰¹ dehydration to synthesize nitrogen-containing-cyclic compounds,¹⁰²⁻¹¹³ oxidation to obtain carboxylic acid compounds,¹¹⁴⁻¹¹⁶ dehydration-deamidation to prepare nitrogen-free aromatics,¹¹⁷⁻¹²¹

enzymatic/fermentative methods for the preparation of amino acid derivatives¹³ and hydrogenation/selective deoxygenation for the preparation of nitrogen-containing chemicals¹²² (**Figure 10**). The mechanism of each type reaction was reviewed by Fukuoka *et al.*^{12,123}

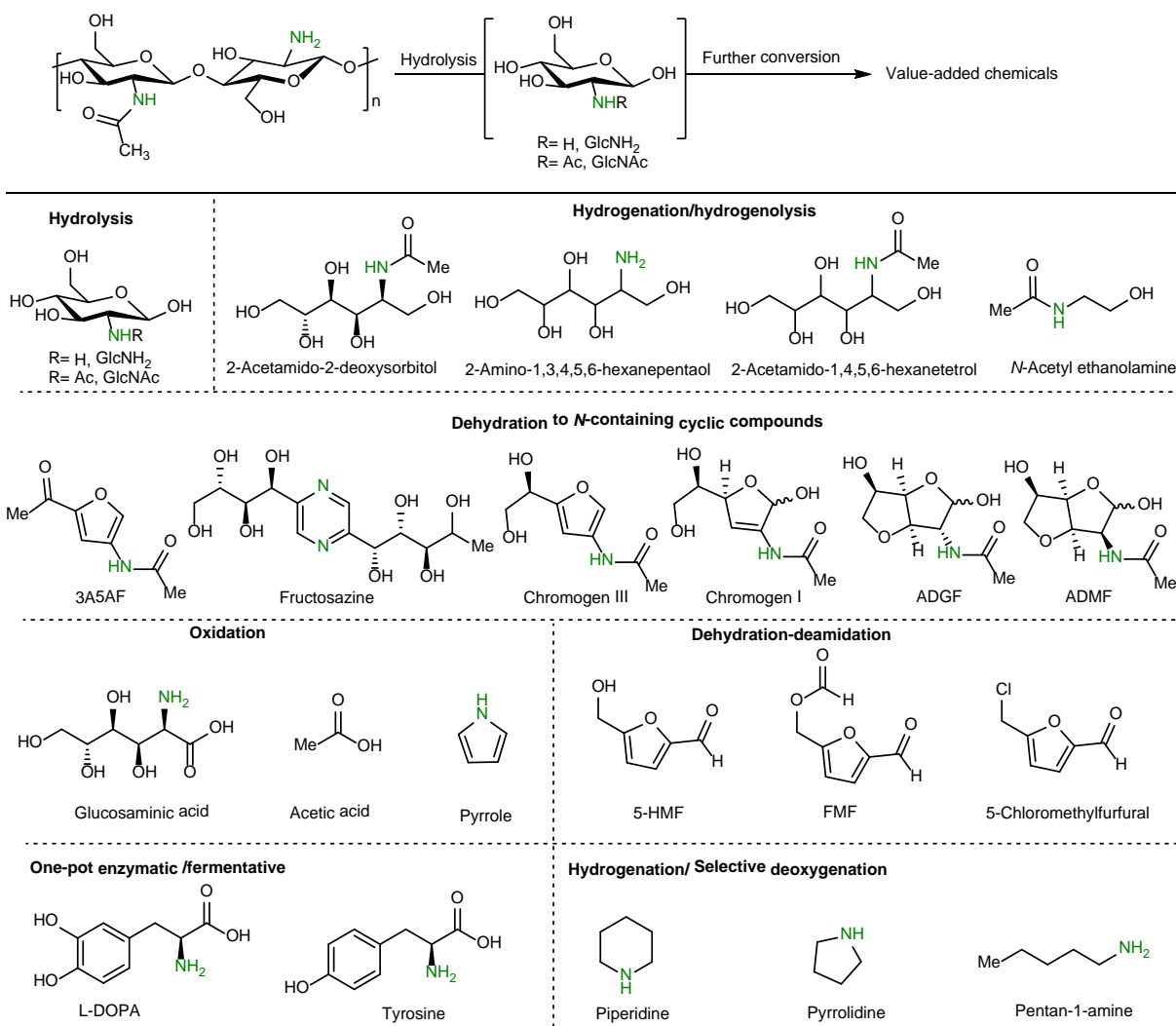


Figure 10 Synthesis of diverse value-added chemicals from chitin, chitosan and D-glucosamine. ADGF= 2-acetamido-3,6-anhydro-2-deoxyglucofuranose; ADMF= 2-acetamido-3,6-anhydro-2-deoxymannofuranose; 3A5AF= 3-acetamido-5-acetylfuran; 5HMF= 5-Hydroxymethylfurfural; FMF= 5-(formyloxymethyl)furfural; L-DOPA= L-3,4-dihydroxyphenylalanine.

For example, in 2020, Yan and Zhou *et al.* reported a biorefinery process to upgrade shell waste-derived chitin to tyrosine and L-DOPA through an integrated process (**Figure 11**).¹³ The process includes

pretreatment of chitin-containing shell waste and an enzymatic/fermentative bioprocess using metabolically engineered *Escherichia coli*.

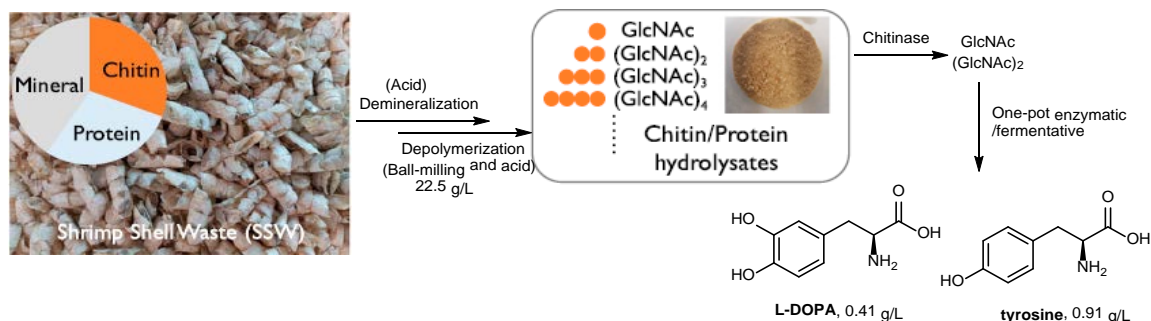


Figure 11 Synthesis of chiral aromatic amino acid from shrimp shell waste with biorefinery process. (Adapted with permission from Ref¹³, Copyright from National Academy of Sciences.)

In 2021, Li *et al.* discovered an electrocatalytic method for the preparation of acetic acid and green hydrogen *via* chitin conversion (**Figure 12**).¹¹⁶ Chitin was electrooxidized to acetate in hybrid electrolysis with yields exceeding 90%. This method also needs to obtain GlcNAc first, and then undergo glucosamine oxidation to acetic acid and H₂.

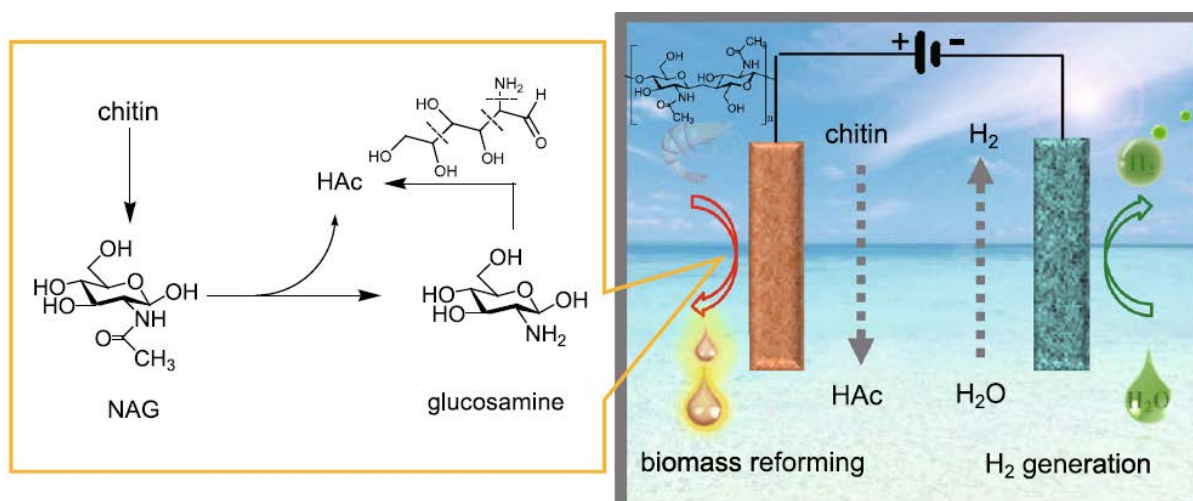


Figure 12 Synthesis of acetic acid from chitin with electrocatalysis. (Adapted with permission from Ref,¹¹⁶ Copyright © 2021, The Author(s))

In 2021, Chen *et al.* reported the selective switchable conversion of chitin-derived *N*-acetyl-D-glucosamine to organic acids at room temperature (**Figure 13**).¹²⁴ In this method, acetic acid and glyceric acid are obtained by using the oxidant O₂, and formic acid is obtained by using H₂O₂ as oxidant in the dilute NaOH solution. Compared with previous methods at high temperature and pressure, this method is economical, efficient and safety risks.

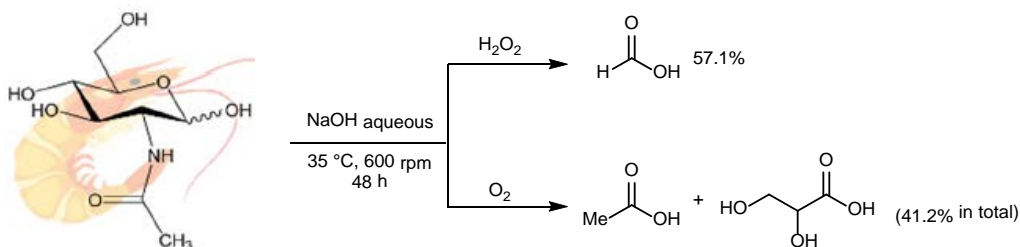


Figure 13 Selectivity-switchable conversion of *N*-acetyl-D-glucosamine into commodity organic acids (Adapted with permission from Ref,¹²⁴ Copyright from 2021 American Chemical Society)

1.1.5 As ligands in organic synthesis

Enantiomerically pure compounds play important roles in pharmaceuticals, agrochemicals and flavors.^{125,126} It can be obtained by organometallic, enzymatic and organocatalytic pathways. Among them, enantioselective homogeneous metal catalysis is an attractive strategy to obtain enantiomerically pure compounds. In 2001, W. S. Knowles, R. Noyori and K. B. Sharpless were awarded the Nobel Prize for their great achievements in this field.¹²⁷⁻¹²⁹ In the process of catalytic enantioselectivity, ligands with stereostructures can cooperate with metal catalysts to form asymmetri chiral space. This facilitates the conversion of of prochiral substrates to chiral products. Therefore, the selection of suitable designed chiral ligands is the most critical step before the reaction. Until today, a series of ligands have been designed that greatly expand the scope of accessible structures and introduce their chiral properties into products.¹³⁰⁻¹³³

Carbohydrates are the most abundant and renewable biomass with a natural chiral backbone and are mainly used as carbohydrate-derived ligands for enantioselective reactions in organic synthesis.¹³⁴⁻¹³⁶ It is shown to be economical and does not require pre-installation of chiral structures through multiple steps like artificial ligands. Many reviews cover ligand-derived sugar,¹³⁴⁻¹⁴³ which are used as phosphine

ligands, phosphinite ligands, phosphite ligands for asymmetric hydrogenation, asymmetric hydroformylation, asymmetric allylic substitution, asymmetric 1,4-addition, asymmetric Heck reaction, asymmetric hydroboration, asymmetric hydrosilylation and asymmetric cyclopropanation. Here we mainly review chitin, chitosan and their monomers as ligand precursors in organic synthesis (**Figure 14**).

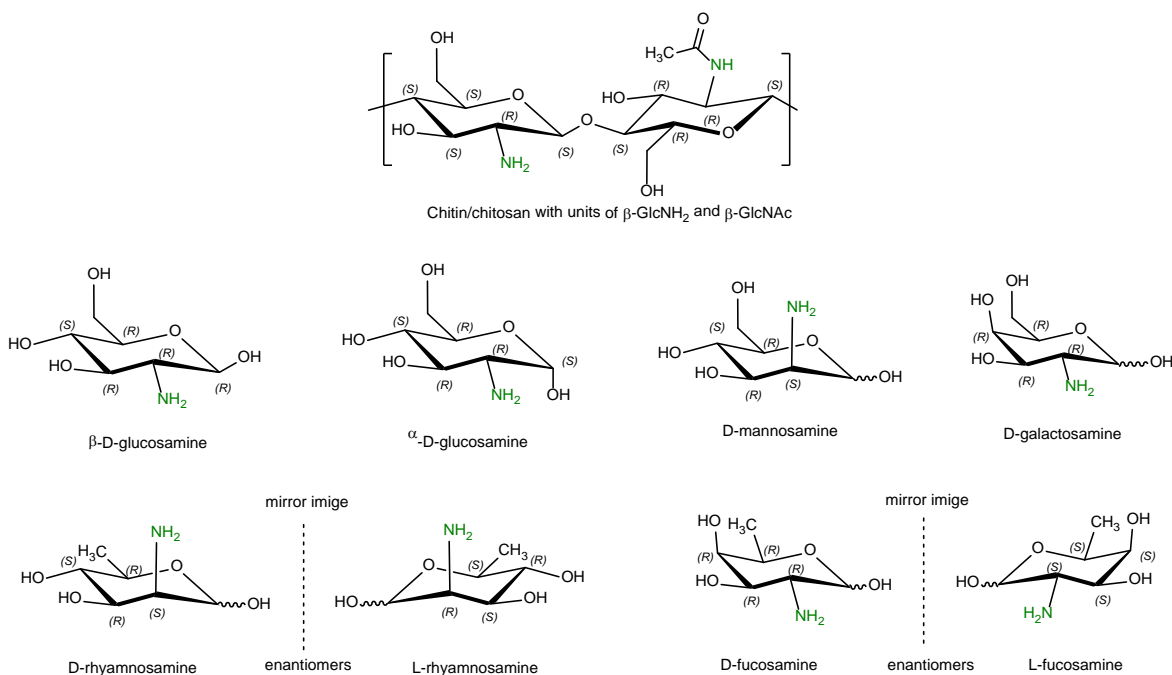
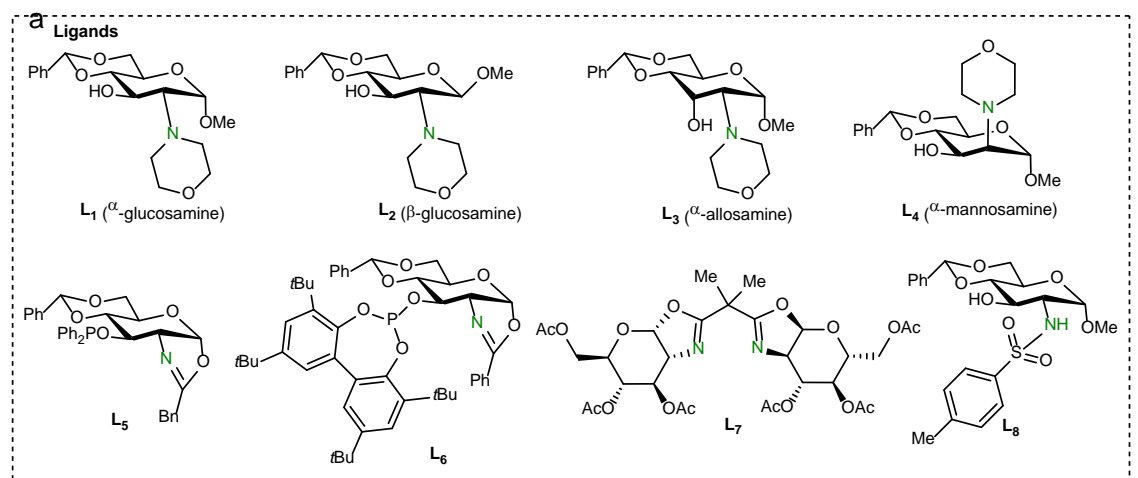


Figure 14 Stereochemistry of native chitin, chitosan and amino sugars.

The natural chitin, chitosan, and amino sugars with amino groups endow them higher reactive activity than those carbohydrates without amino groups. *N*-containing carbohydrates can be easily modified to desired ligands to improve its reactivity in organic synthesis applications (**Figure 15a**).^{144,145} Kunz *et al.* in 1998, first time utilized D-glucosamine for phosphine oxazoline (phox) ligand.¹⁴⁶ Uemura *et al.* improved this ligand with a diphenyl phosphinite group for asymmetric allylic substitution reactions.^{147,148} Ligand **L**₅ is used as a phosphinite-oxazoline ligand for the asymmetric Heck reaction and for the enantioselective arylation of 2,3-dihydrofuran with aryl triflate (**Figure 15b**).¹⁴⁹ Boysen *et al.* introduced a C₂-symmetrical bis-(oxazoline) ligand **L**₇ derived from D-glucosamine. This ligand is used for copper-catalyzed cyclopropanations of styrene with diazoacetate (**Figure 15d**).¹⁵⁰ The ligand **L**₈ designed by Bauer *et al.* was used to the addition of diethylzinc to aldehydes with high enantiomeric excess (*ee*) and yield (**Figure 15e**).¹⁵¹ In 2005, Dieguez *et al.* designed and synthesized a new family of readily available

phosphite-oxazoline ligands **L**₆ for Pd-catalyzed asymmetric allylic substitution reactions (**Figure 15c**).¹⁵² In 2006, a group of amino sugars (**L**₁-**L**₄) were used as chiral ligand additives for the addition of diethyl zinc to aldehydes (**Figure 15f**).¹⁵³ Both **L**₁ with the α -anomer and **L**₂ with the β -anomer can achieve excellent yields and *ee*. However, low yields and *ee* can be achieved with α -allosamine (**L**₃), while high yields with low *ee* can be achieved with α -mannosamine (**L**₄). It was concluded that the α/β anomers do not affect the yield and *ee* of these reactions. The hydroxyl group on C₃ plays an important role in the yield of the reaction, while the amino group on C₂ helps to control the *ee*. Based on these results, they proposed a mechanism for the formation of a five-membered ring between the hydroxyl group on C₂, amino group and zinc.



Applications

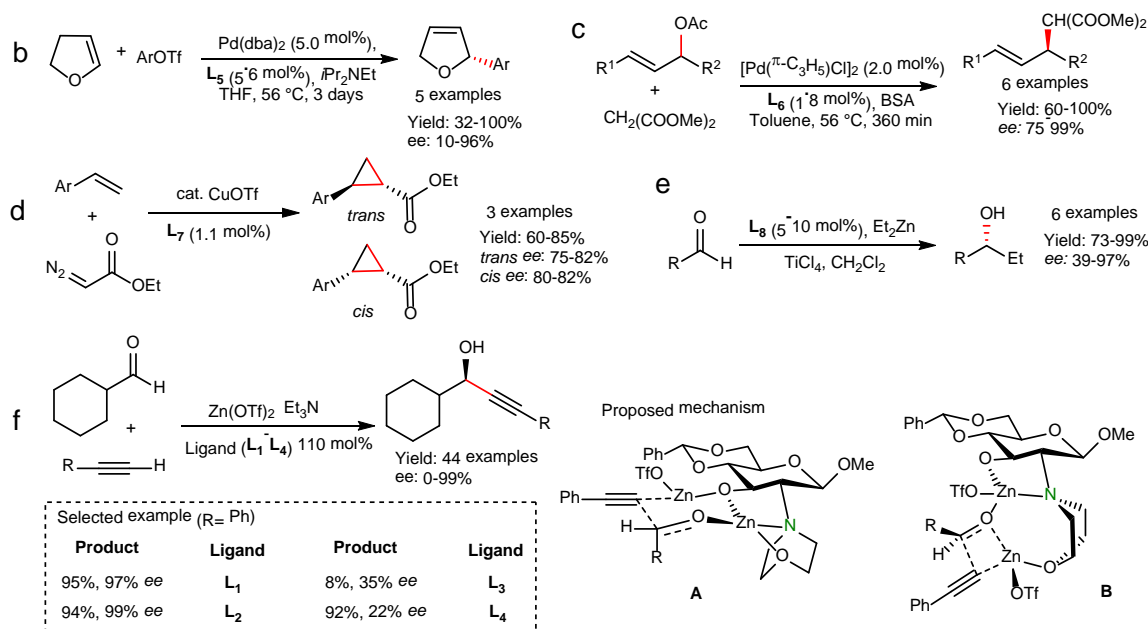
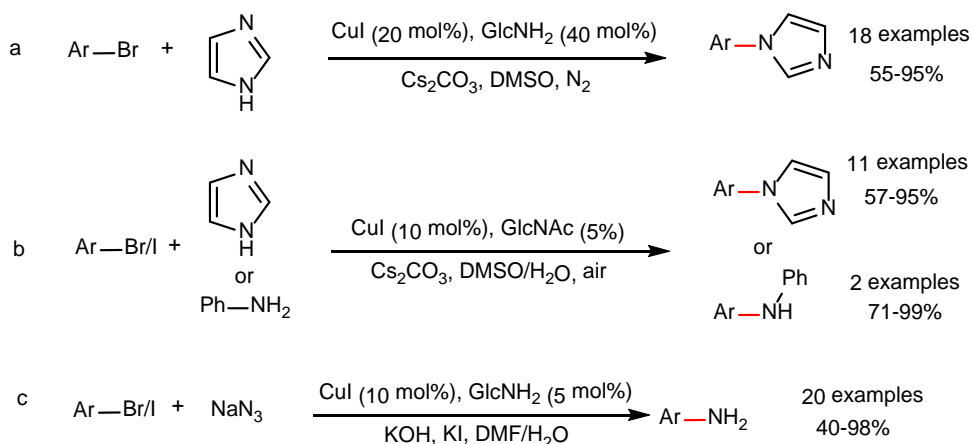


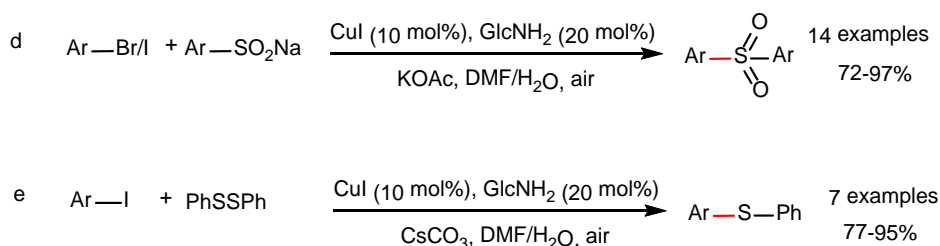
Figure 15 Representatives examples of ligands derived amino sugar and their applications. Pd(dba)₂= Bis(dibenzylideneacetone)palladium; BSA= *N,O*-bis(trimethylsilyl)acetamide; *ee*= enantiomeric excess.

In addition to asymmetric catalytic organic synthesis, in 2008, Bao *et al.* reported D-glucosamine as a ligand for the Ullmann reaction-type *N*-arylation of imidazoles with aryl and heteroaryl bromides (**Figure 16a**).¹⁵⁴ In 2016, Zhou *et al.* not only improved this reaction not only in an air atmosphere by utilizing GlcNAc, but also expanded the substrates from imidazoles to aromatic amines (**Figure 16b**).¹⁵⁵ The role of GlcNAc was demonstrated by theoretical studies, suggesting that the hydroxyl groups at C₃, C₄, and C₆ positions may have a significant influence on the catalytic process. However, the role of the hydroxyl group on C₁ and the role of the amino group on C₂ have not been investigated. In 2011, Sekar *et al.* reported D-glucosamine as an efficient ligand for the copper-catalyzed selective synthesis of anilines from aryl halides and NaN₃ (**Figure 16c**).¹⁵⁶ In 2014, Zhang *et al.* discovered D-glucosamine used as a ligand for copper-catalysed synthesis of aryl sulfones from aryl halides and sodium sulfonates (**Figure 16d**).¹⁵⁷ Zhang *et al.* extended the strategy for the cross-coupling reaction of diphenyl disulfides and aryl iodides in the presence of CuI and glucosamine (**Figure 16e**).¹⁵⁸ D-glucosamine has also been used in the palladium-catalyzed Mizoroki-Heck reaction of aryl halides (**Figure 16f**)¹⁵⁹ and the iron-catalyzed Grignard cross-coupling reaction of vinylic and allylic bromides (**Figure 16g**).¹⁶⁰

C-N bond formations



C-S bond formations



C-C bond formations

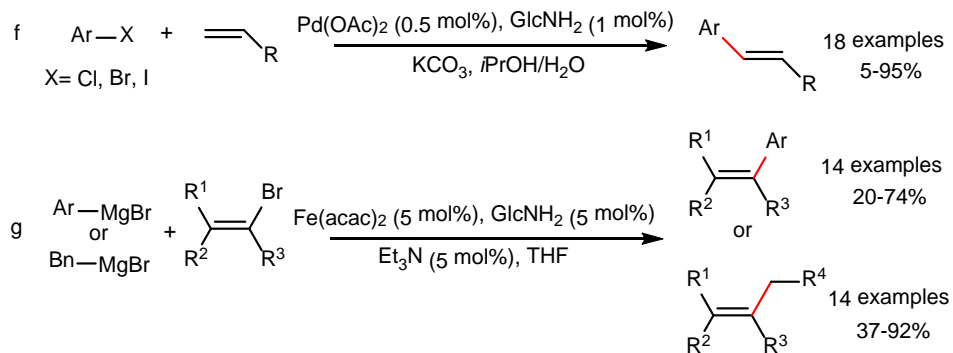


Figure 16 Construction of C–N, C–S and C–C bonds *via* the metal-catalyzed with ligand D-glucosamine. $\text{Fe}(\text{acac})_2$ = iron(II) acetylacetonate.

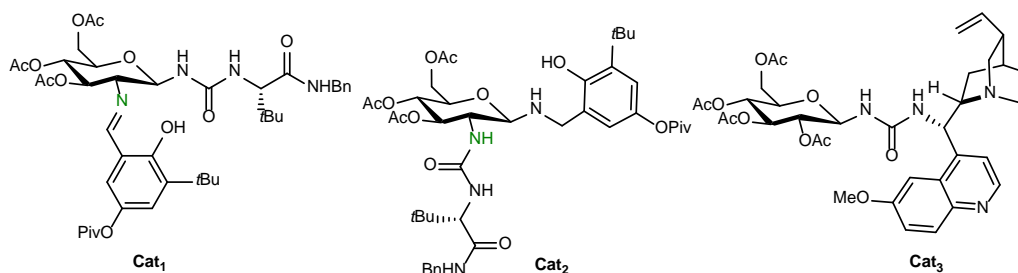
1.1.6 As organocatalysts in organic synthesis

Organocatalysis is one of the most thriving research fields in contemporary organic synthesis.¹⁶¹⁻¹⁶⁶ A variety of classical organocatalyst types have been reported, which highly expand their range of applications from value-added fine chemical production to natural product synthesis and drug discovery.¹⁶⁶⁻¹⁷⁷ Since the chiral backbone of aminocatalysts has tunable steroauxiliary groups, which help to form strong steric shielding, in 2000, Benjamin List *et al.* first reported the use of proline as a catalyst to directly realize asymmetric aldol reactions through enamine process.¹⁷⁸ In the same year, David W. C. Macmillan *et al.* first demonstrated the highly enantioselective organocatalytic Diels-Alder reaction through the iminium-catalyzed pathway.¹⁷⁹ Benjamin List and David W. C. Macmillan were awarded Nobel Prize in 2021. Until today, iminium and enamine catalysis methods based on covalent mechanisms have drawn much attention.¹⁸⁰⁻¹⁸⁶ Despite the enormous advance in these aminocatalyst modes, a sustainable and cheaper organocatalyst with natural chiral skeleton derived from biomass for the enantioselective synthesis are still highly desirable.

In 2007, inspired by bifunctional urea Schiff base organocatalysts,^{187,188} the first example of a highly efficient organocatalyst for enantioselective Strecker and Mannich reactions was constructed from glucosamine as a readily available chiral scaffold (**Figure 17b**).¹⁸⁹ In the Strecker reaction, catalyst **Cat₁** shows an excellent 95% *ee*, while catalyst **Cat₂** only achieves 15% *ee*. Compared with the Strecker reaction, the catalyst **Cat₁** exhibited poor *ee* in the Mannich reaction (**Figure 17c**). In 2012, Miao *et al.* developed sugar-derived thiourea organocatalyst for the catalytic asymmetric addition reaction of α -ketophosphonates and TMS-CN with excellent *ee* and high yield (**Figure 17d**).¹⁹⁰⁻¹⁹²

In 2006, Wong *et al.* demonstrated that the sugar moiety of a glycopeptide modified with a thiol handle at the C₂ position can facilitate the ligation of cysteine-free glycopeptide to thioester peptide (**Figure 18**).¹⁹³⁻²⁰⁰ In this work, they propose that the sugar moiety affects the proximity of the *N*-terminal amine to the thioester. This provides acyl transfer to obtain the final product. But they did not explain the configuration of the anomeric center and the *N*-linked sugars, their effect on the fate of ligation. Two years later, they expanded the sugar types with more elaborate sugars for glycopeptide ligation.¹⁹⁴ In 2016, Liu *et al.* reported a practical approach for the synthesis of *N*-glycopeptide using an auxiliary-mediated dual native chemical ligation.²⁰¹

a Catalysts



Applications

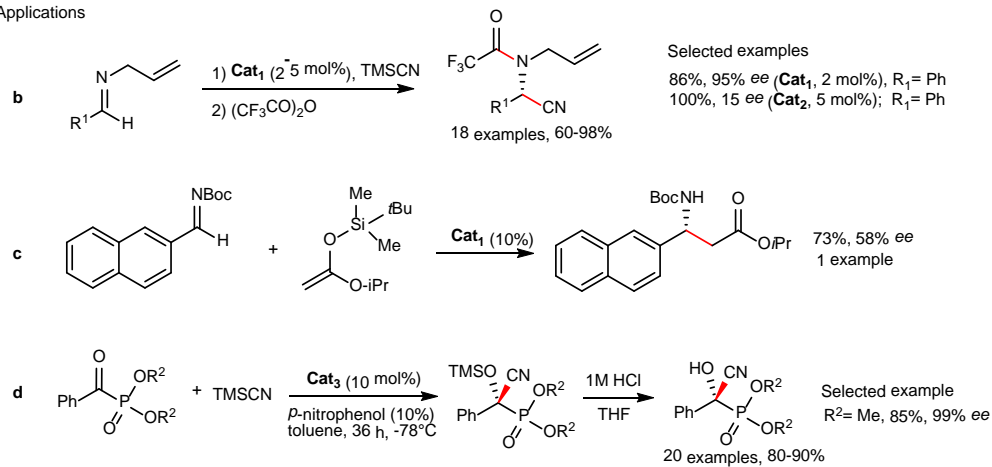


Figure 17 The sugar-derived thiourea organocatalysis and applications.

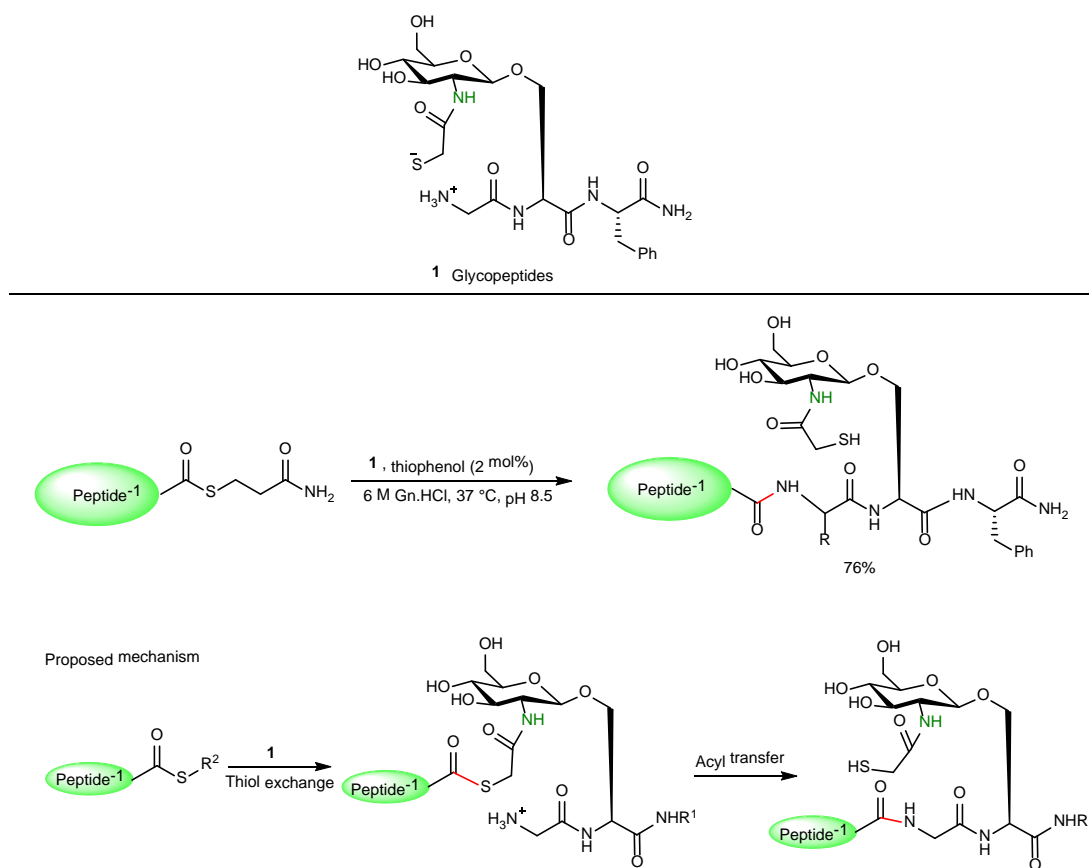


Figure 18 Sugar-assisted glycopeptide ligation.

2. Objectives

1) Targeted cleavage of the C–N bond of D-glucosamine *via* a metal-free pathway and the conjunction of nitrogen in the synthesis of imidazo[1,5-a]pyridines remains a great challenge. Despite tremendous progress in the synthesis of imidazo[1,5-a]pyridines over the past decade, many of them still cannot be efficiently prepared. Therefore, an novel anomeric stereoauxiliary approach for the synthesis of a wide range of imidazo[1,5-a]pyridines after cleavage of the C–N bond of D-glucosamine (α -2° amine) from biobased sources is highly desirable (**Figure 19**).

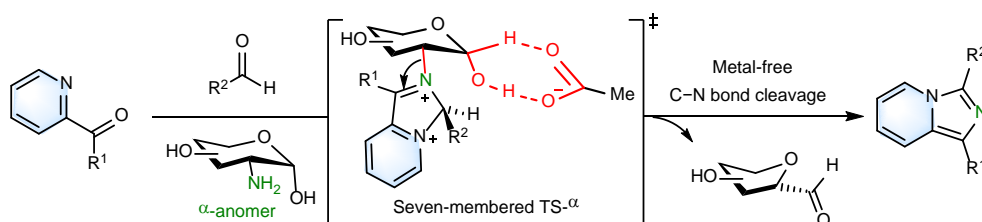


Figure 19 Anomeric stereoauxiliary strategy enables efficient synthesis of wide-ranging imidazo[1,5-a]pyridines.

2) Nitrogen-containing chemicals (NCCs) play a pivotal role in modern life, but direct nitrogen interception from the renewable nitrogen-containing polysaccharide chitosan/chitin for synthesizing value-added NCCs is still a big challenge. Herein, a facile and efficient one-pot methodology that enables direct nitrogen interception of chitosan/chitin for the synthesis of a broad range of important *N*-heterocycles imidazo[1,5-a]pyridines are highly attractive (**Figure 20**).

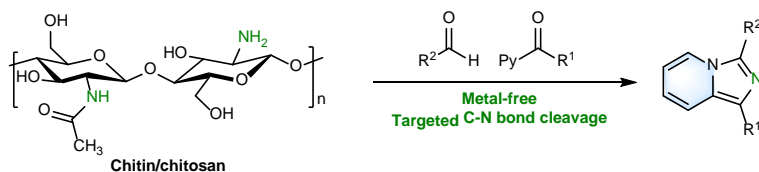


Figure 20 Simultaneous C–N cleavage and incorporation of nitrogen from chitin/chitosan for imidazo[1,5-a]pyridines.

3) Today, there are still no examples of glucosamine being used only as aminocatalysis. Therefore, sustainable aminocatalysts with native chiral skeletons derived from *N*-containing biomass for the regioselective cyclization are highly attractive, but not yet realized. On the other hand, indolizine-carbaldehyde is an important synthetic target. The easily modified carbaldehyde group makes them versatile precursors for diverse indolizines. However, the chemical construction of trisubstituted indolizine-2-carbaldehydes with efficient one-pot direct site-selective C₂-aldehylation represents a long-standing challenge for synthetic chemists.^{202,203} Herein, a novel anomeric stereoauxiliary aminocatalysis approach is highly desirable for the efficient one-pot preparation of trisubstituted indolizine-2-carbaldehydes with highly site-selective C₂-aldehylation *via* dehydration [3+2] annulations (**Figure 21**).

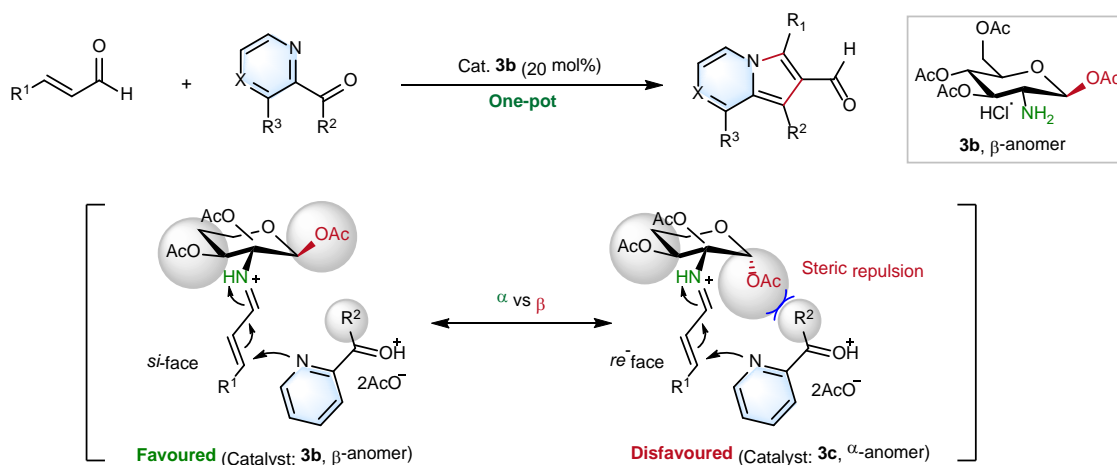


Figure 21 Anomeric stereoauxiliary-organocatalyzed one-pot site-selective C₂-aldehylation for trisubstituted indolizine-2-carbaldehydes.

3. Results and Discussion

3.1 Publication 1: Anomeric Stereoauxiliary Cleavage of the C–N Bond of D-glucosamine for the Preparation of Imidazo[1,5-a]pyridines

3.1.1 Background

The C–N bonds are ubiquitous in many biobased organic molecules, especially in many biomolecules of living organisms.²⁰⁴⁻²⁰⁶ The utilization and transformation of C–N bonds are among the central topics in organic chemistry, biochemistry and material science,²⁰⁷⁻²¹⁰ while only a few examples of the C–N cleavage of alkyl primary amines were reported recently with the assistance of metal-catalysts. Up to now, most protocols are limited to primary amines with benzylic, allylic and pyridine.²¹⁰⁻²¹⁵ It is still highly attractive to develop efficient strategies for selective C–N bonds cleavage in alkyl primary amines, including α -1°, α -2° and α -3° alkyl primary amines. In 2017, the Watson group cleaved C–N bonds of α -1° and α -2° primary alkyl amines *via* single electron transfer to Katritzky pyridinium salts intermediates (**Figure 22a**).²¹⁶⁻²¹⁹ Later in 2020, the Rovis group demonstrated a visible light photoredox approach to cleave C–N bonds of α -3° primary alkyl amines *via* a key imidoyl radical intermediate (**Figure 22a**).²²⁰ Moreover, the Milstein group reported a catalytic oxidative deamination protocol with ruthenium pincer complex (**Figure 22a**).²²¹ Until now, all the reported methods for C–N bonds cleavage in alkyl primary amines require transition metal catalysts. Unlike these previous strategies towards the cleavage of C–N bond via metal catalysts, we present here a metal-free anomeric stereoauxiliary strategy to cleave the C–N bond of native D-glucosamine (α -2° amine) from biobased resources, which offers a readily accessible and sustainable route for the synthesis of a broad range of imidazo[1,5-a]pyridines (**Figure 22b**).

Imidazo[1,5-a]pyridines, as one of the most important *N*-heterocyclic compounds, play pivotal roles in various areas from pharmaceuticals over chemical synthesis to materials science.²²²⁻²²⁹ For example, they can be precursors of *N*-heterocyclic carbenes,²²²⁻²²⁴ ligands in coordination chemistry,^{225,226} and inhibitors of biologically active agents.²²⁷⁻²²⁹ In particular, due to the unique photophysical properties with quantum yields up to 50%, large Stokes shift (up to 100-150 nm) and good stability,²³⁰ imidazo[1,5-a]pyridines are attractive optical materials for many applications²³¹⁻²³³. In this respect, advances from metal-free to metal-catalyzed cyclization strategies of *N*-heterocyclic substrates have increased the number of accessible structures. A representative metal-free method with NH₄OAc as nitrogen source for imidazo[1,5-a]pyridines was reported in 2005.²³⁴ This method was continually improved and widely used in synthetic chemistry and for optical materials,²³⁵ even though the challenges still need to be addressed. For instance,

many imidazo[1,5-a]pyridines including 1-alkylimidazo[1,5-a]pyridines and 3-alkylimidazo[1,5-a]pyridines are inaccessible by this method.

In 2007, Gevorgyan reported the first efficient rhodium-catalyzed metal carbene approach for the preparation of imidazo[1,5-a]pyridines *via* transannulation of pyridotriazoles process.²³⁶ 3-alkylimidazo[1,5-a]pyridine products (6 examples) were unprecedentedly accessed with this metal-catalyzed method. It was shown that an activating group (Cl, Br, or OMe substituents) at C₇, as well as electron-withdrawing groups at C₃, were necessary for an efficient formation of imidazo[1,5-a]pyridines. To overcome these limitations, a general strategy of rhodium-catalyzed NH insertion of pyridyl carbenes for imidazo[1,5-a]pyridines was developed in 2014.²³⁷ Although this strategy widely expanded the library of accessible structures, the study showed only 6 examples in total and the required precious metal catalyst with active triazole substrates hinders its widespread applications. In 2014 and 2016, inexpensive copper catalysts were applied for synthesizing imidazo[1,5-a]pyridines, but these methods are limited to 3-monosubstituted imidazo[1,5-a]pyridines, while many imidazo[1,5-a]pyridines maintain inaccessible, such as 1-alkylimidazo[1,5-a]pyridines.^{238,239} The challenges in accessing 1-alkylimidazo[1,5-a]pyridines *via* ketone activation of alkyl(pyridine-2-yl)methanone substrates lies in the lower reactivity of alkyl(pyridine-2-yl)methanone compared to aryl(pyridin-2-yl)methanone.²³⁸ Additionally, alkyl(pyridine-2-yl)methanone is readily activated as nucleophilic reagent due to the α -saturated C–H bond next to ketone, which can result in side-reaction.²⁴⁰ Therefore, it is highly desired to develop a strategy to overcome these disadvantages.

Carbohydrates as chiral auxiliaries in stereoselective synthesis^{134,241} and stereochemistry of transition metal complexes controlled by the metallo-anomeric effect²⁴² have drawn much attention recently. Depending on the pK_a values of aqueous solutions, the α/β -anomers of D-glucosamine exist with adjustable ratios.^{36,48} Inspired by these different stereochemical structures of α/β -anomers, we demonstrate herein a novel strategy for the cleavage of C–N bond in D-glucosamine that is enabled by α -anomer through the formation of a key (non-covalent) seven-membered ring transition state for the synthesis of diverse imidazo[1,5-a]pyridines without any metal catalysts. Various inaccessible substrates from existing methods for imidazo[1,5-a]pyridines are synthetically accessible by this protocol.

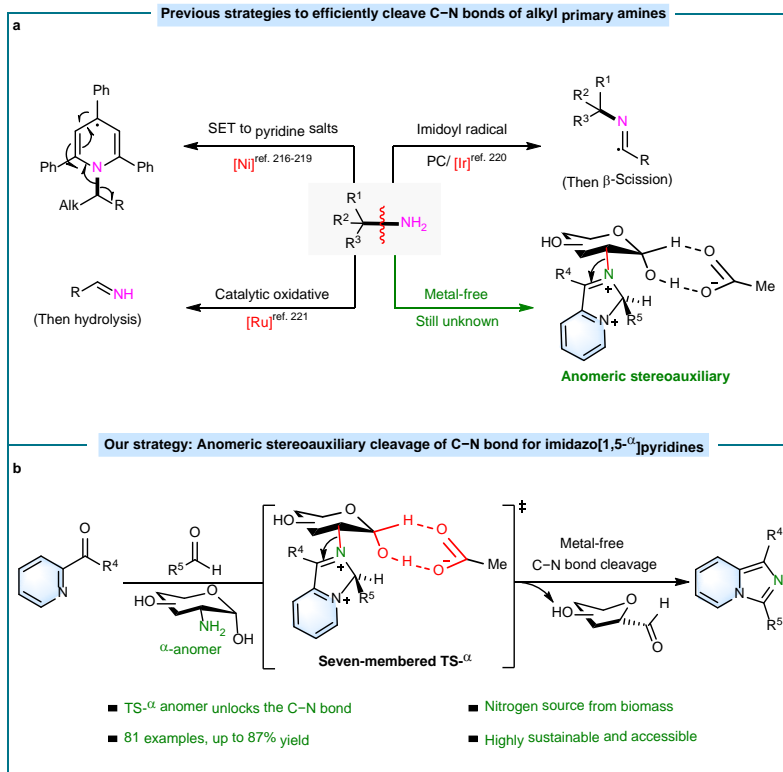


Figure 22 (a) The representative example of previous strategies for the cleavage of the C–N bonds of alkyl primary amines. **(b)** Our strategy: cleavage of C–N bond for imidazo[1,5-*a*]pyridines via anomeric stereoauxiliary.

3.1.2 Optimization studies

We commenced our study by probing various reaction conditions for imidazo[1,5-*a*]pyridines by using 2-acetylpyridine (**1a**), 2-methylbenzaldehyde (**2a**) and diverse nitrogen sources (**Table S1** of **Publication 1**). After extensive experimentation, we got the optimal condition for the efficient synthesis of imidazo[1,5-*a*]pyridines with 74% yield using D-glucosamine as nitrogen source in a solvent mixture ($V_{\text{AcOH}}:V_{\text{H}_2\text{O}}$ of 9:1) at 120 °C under Ar gas atmosphere (**Figure 23a**). In parallel, commercial acetylated amine sugars as stabilized α -anomer (**3b**) and β -anomer (**3c**) were used for the reaction under the optimal conditions (**Figure 23a**). The α -anomer of acetylated D-glucosamine led to 30% yield, while only trace of the product was detected using the β -anomer. Therefore, the α -anomer of D-glucosamine with the hydroxyl group at the neighbor C1 position is preferred for the synthesis of imidazo[1,5-*a*]pyridines. Besides, the

scope with D-mannosamine under the same conditions led to a yield of 41% in the presence of a major β -anomer distribution ($\alpha/\beta = 0.79/1$). This result further verified that the configuration of amine and hydroxyl group should be on the same side to cooperatively cleave C–N bonds for imidazo[1,5-a]pyridines. Various amines (**3e** to **3i**) were also investigated. As a result, only **3e** fulfilling these configuration requirements provided the desired product with the highest yield of 16%. To explore the correlation between the yield of imidazo[1,5-a]pyridine **4** and the anomer of D-glucosamine in solvents with diverse pK_a , solvent mixtures with various pK_a (0.9 mL) and H₂O (0.1 mL) were investigated under optimal conditions (**Figure 23b**). In general, the ratio between α - and β -anomer of D-glucosamine (refers as α/β) highly depends on the pK_a of the respective solvent. Solvents with higher pK_a , such as HFIP (pK_a : 9.30), Et₃N (pK_a : 10.76) and H₂O (pK_a : 15.75), result in lower α/β ratios of glucosamine and significantly lower yields of imidazo[1,5-a]pyridine **4**. The predominant reason for this result should be the presence of the β -anomer of D-glucosamine as the major isomer. In comparison, those suitable acidic reaction media, such as CF₃COOH (pK_a : 0.30), HCOOH (pK_a : 3.75) and AcOH (pK_a : 4.76), result in higher α/β ratios of glucosamine and higher yields of **4**, showing the facilitating effect of the α -anomer of D-glucosamine on the synthesis of **4**. These results further indicate that Brønsted acids with suitable pK_a stabilize the methyl group of alkyl(pyridine-2-yl)methanone, and thus hinder the deprotonation of the methyl group.²⁴³

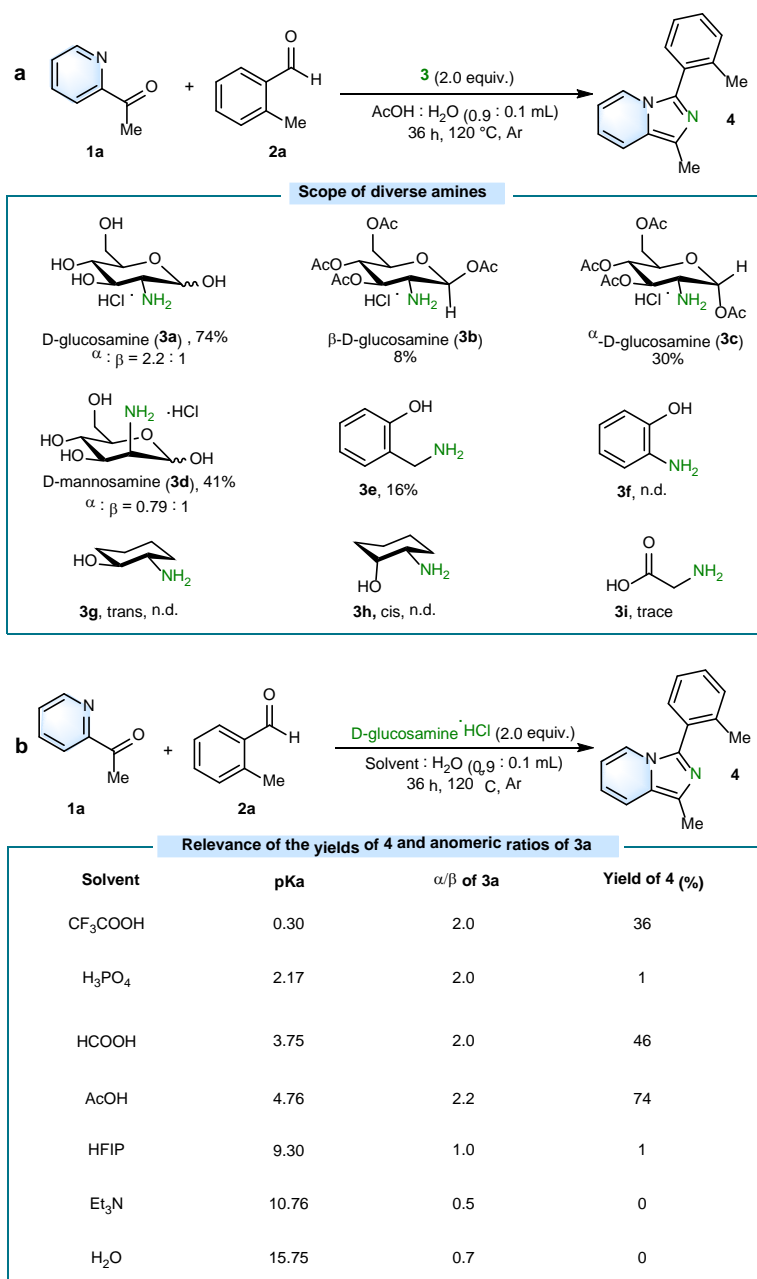


Figure 23 Reaction development. **(a)** Diverse amino compounds **3a-3i** were used in reactions in the solvent mixtures (vAcOH:vH₂O of 9:1) at 120 °C under Ar gas atmosphere. **(b)** Various mixtures of solvents with various pK_a (0.9 mL) and H₂O (0.1 mL) were used for the reactions under the optimal conditions. pK_a of CF₃COOH,²⁴⁴ H₃PO₄,²⁴⁵ HCOOH,²⁴⁶ CH₃COOH,²⁴⁷ 1,1,1,3,3,3-hexafluoro-2-propanol (HFIP),²⁴⁸ Et₃N,²⁴⁹ and H₂O²⁵⁰ are determined at 25 °C. The ratios of α- and β-anomers were measured by ¹H-NMR analysis at room temperature; Yields were determined by ¹H-NMR analysis with CH₂Br₂ as the internal standard.

3.1.3 Scope of substrates

With the optimized reaction conditions in hand, we next probed the scope of various aldehydes using 2-acetypyridine as a representative heteroaryl ketone (**Figure 24**). An array of aromatic aldehydes, including those with electron-donating or -withdrawing groups at different positions (*ortho*, *meta* or *para*), were efficiently transformed to the corresponding products **4-23**. A variety of valuable functional groups at diverse positions, such as methoxy (**11** and **12**), halogens (**14-18**), trifluoromethyl (**19**), nitro (**20**), nitrile (**21**) and ester (**22**), were well compatible with these conditions. Particularly, free *para*-dialdehyde (**23**) and *ortho*-phenolic hydroxyl (**13**) were also tolerated in this protocol. The structure of **20** was further confirmed by X-ray crystallographic analysis, and those of other products in Table 1 were assigned by analogy. Moreover, 2-phenylacetaldehyde (product **24**), cinnamaldehyde (product **25**), 1-naphthaldehyde (product **26**) and heterocyclic aldehydes (product **27-28**) were also well compatible with this approach. Furthermore, a series of aliphatic aldehydes, including cyclic aldehydes (product **29-30**) and aldehydes with aliphatic chains (product **31-34**), could also be transformed into desired products. Hence, this facile and efficient approach has been proved for the successful preparation of saturated 1-alkylimidazo[1,5-*a*]pyridine compounds, with unprecedented use of inexpensive and commercially available aromatic/aliphatic aldehydes.

We further explored heteroaryl ketones (**Figure 25**). Di(pyridin-2-yl)methanone (product **35**) and pyridin-2-yl(pyridin-4-yl)methanone (product **36**) were tolerated in this reaction. Various aromatic pyridine ketones, including those having electron-donating or -withdrawing groups at distinct positions (*ortho*, *meta* or *para*), were well transformed into the corresponding products **37-43**. The functional groups at diverse positions, such as methyl (**38-39**), methoxyl (**40**), trifluoromethyl (**41**), mono-Br- (**42**) and di-Br-substituted arenes (**43**), were fully compatible with our conditions. The cyclic aliphatic pyridine ketone was also tolerated under these conditions (**44**). In addition, our protocol was also capable for the assembly of diverse tridentate (**45-47**), bidentate ligands (**48-60**), and heterocyclic backbones with fluorescent properties. The structure of **51** was further assigned with X-ray crystallographic analysis.

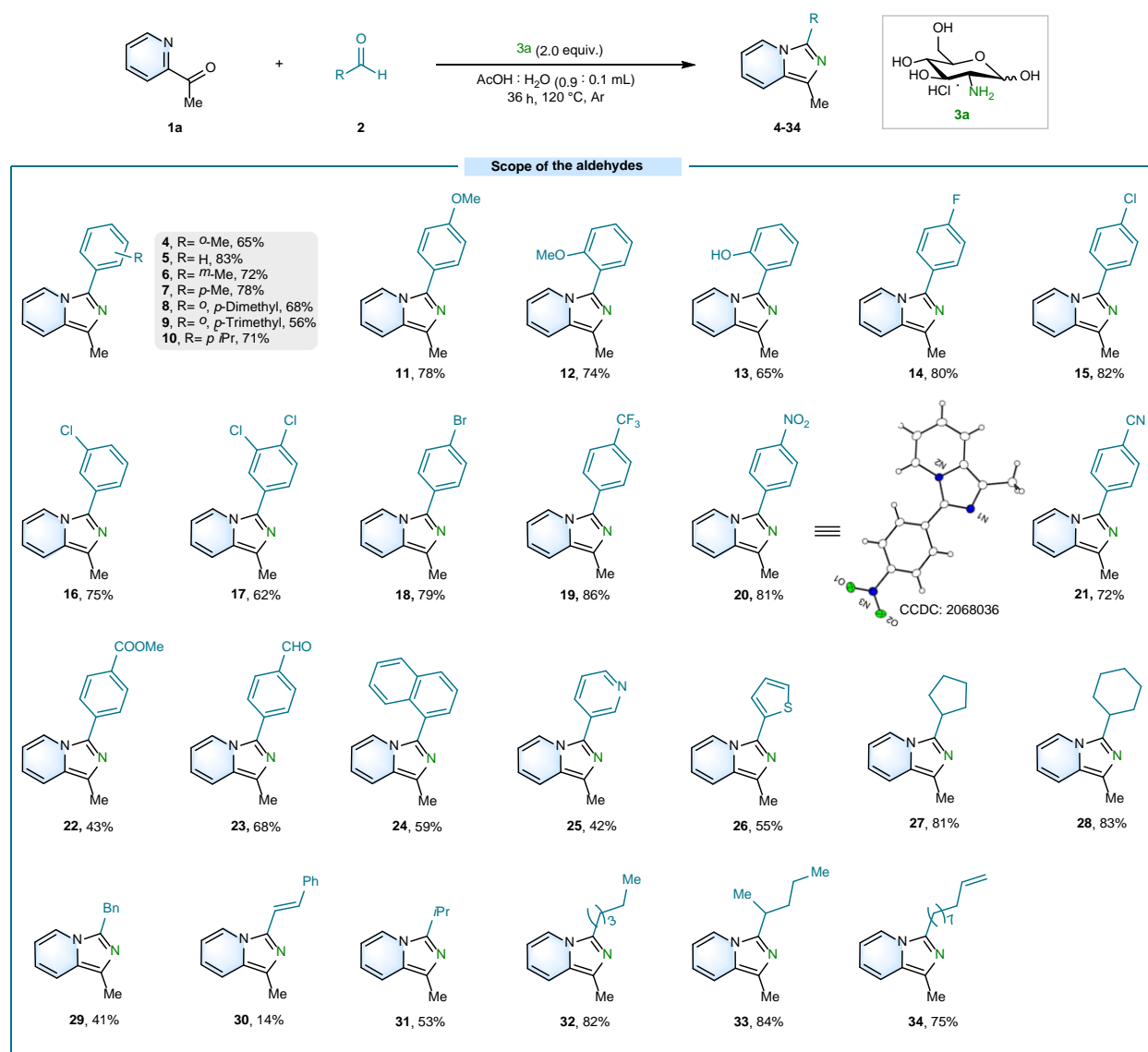


Figure 24 Scope of aldehydes. Reactions were carried out at 120 °C with 2-acetylpyridine, aldehydes, D-glucosamine·HCl (**3a**) in AcOH : H₂O (0.9 mL : 0.1 mL), under Ar gas with stirring for 36 h. Yields are those of the isolated product.

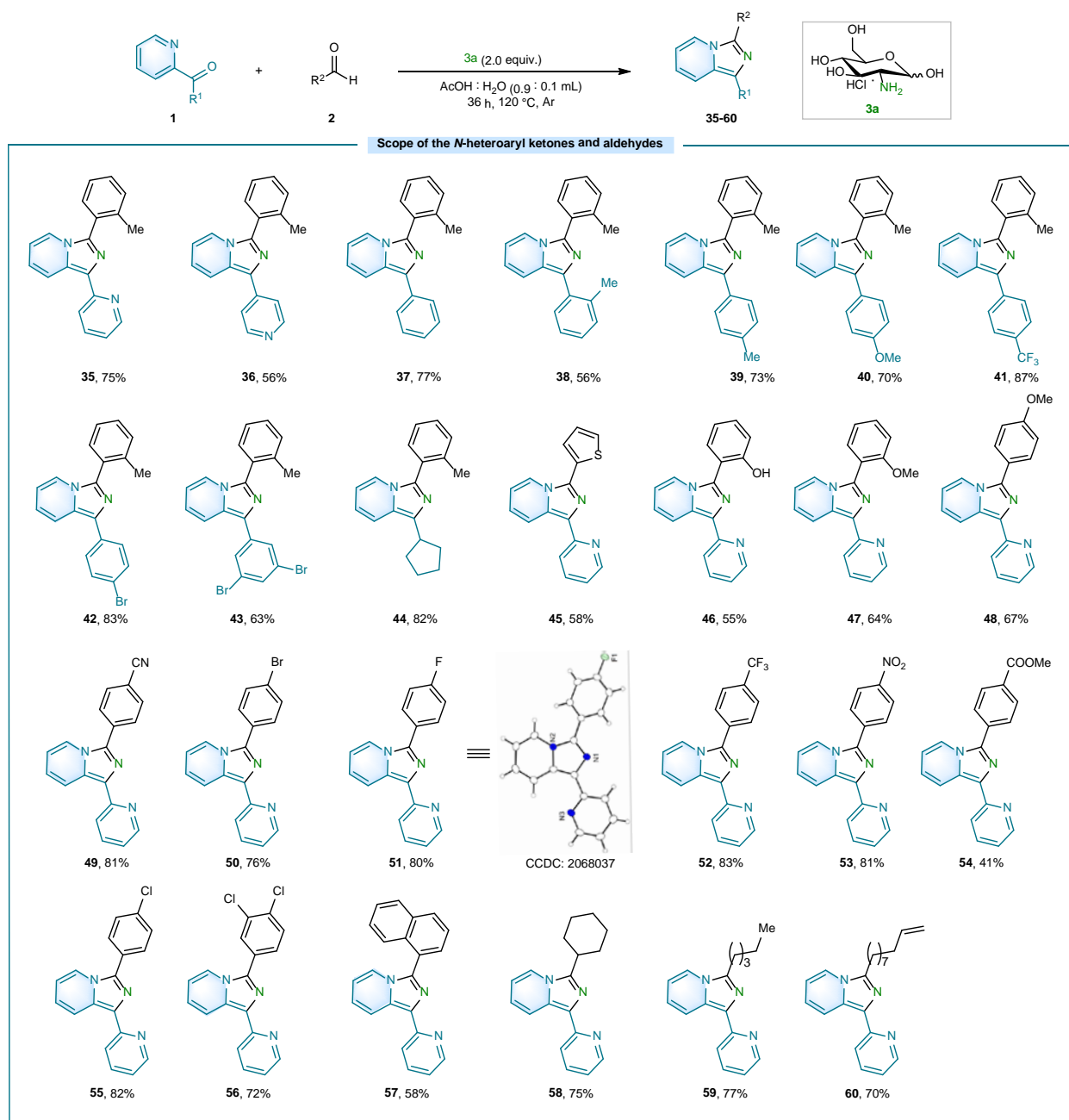


Figure 25 Scope of *N*-heteroaryl ketones and aldehydes. Reactions were carried out at 120 °C with *N*-heteroaryl ketones, aldehydes, D-glucosamine-HCl (3a) in AcOH : H₂O (0.9 mL : 0.1 mL), under Ar gas with stirring for 36 h. Yields are those of the isolated product.

3.1.4 Applications

Certain imidazo[1,5-a]pyridines with multiple substitutions show interesting optical properties and ligand effects due to the conjugation feasibility and the presence of lone pair electrons in nitrogen and oxygen atoms. Because of the difficulty for the regioselective functionalization and the interference of potential side reactions, there is still no efficient method to synthesize such compounds so far. With our method, the challenging bi-functionalization of dialdehyde (product **61**) was achieved smoothly (**Figure 26a**). Motivated by this result, 1,4-phenylenebis(pyridin-2-ylmethanone) (product **62**) was also prepared *via* our synthetic route (**Figure 26b**). In addition, starting from **50**, products **63** and **64** were readily obtained with yields of 68% and 72% after the reaction with diphenylphosphine oxide and phenylboronic acid, respectively (**Figure 26c-26d**). Moreover, imidazo[1,2-a:3,4-a']dipyridin-10-ium (**65**) was accessed concisely after two steps with standard conditions (**Figure 26e**).

Isotope labeling, such as deuterated fine chemicals, has a broad range of applications, for instance for drug absorption, distribution, metabolism and excretion, for the investigation of reaction processes and for imaging.²⁵¹⁻²⁵⁴ The first deuterated drug, deutetrabenazine, was approved by FDA in 2017.²⁵⁵ Because of the versatile functionalities of imidazo[1,5-a]pyridines that are interesting for diverse fields ranging from material science to pharmaceuticals, efficient synthetic methods for deuterated building blocks of imidazo[1,5-a]pyridines derivatives are highly desired.

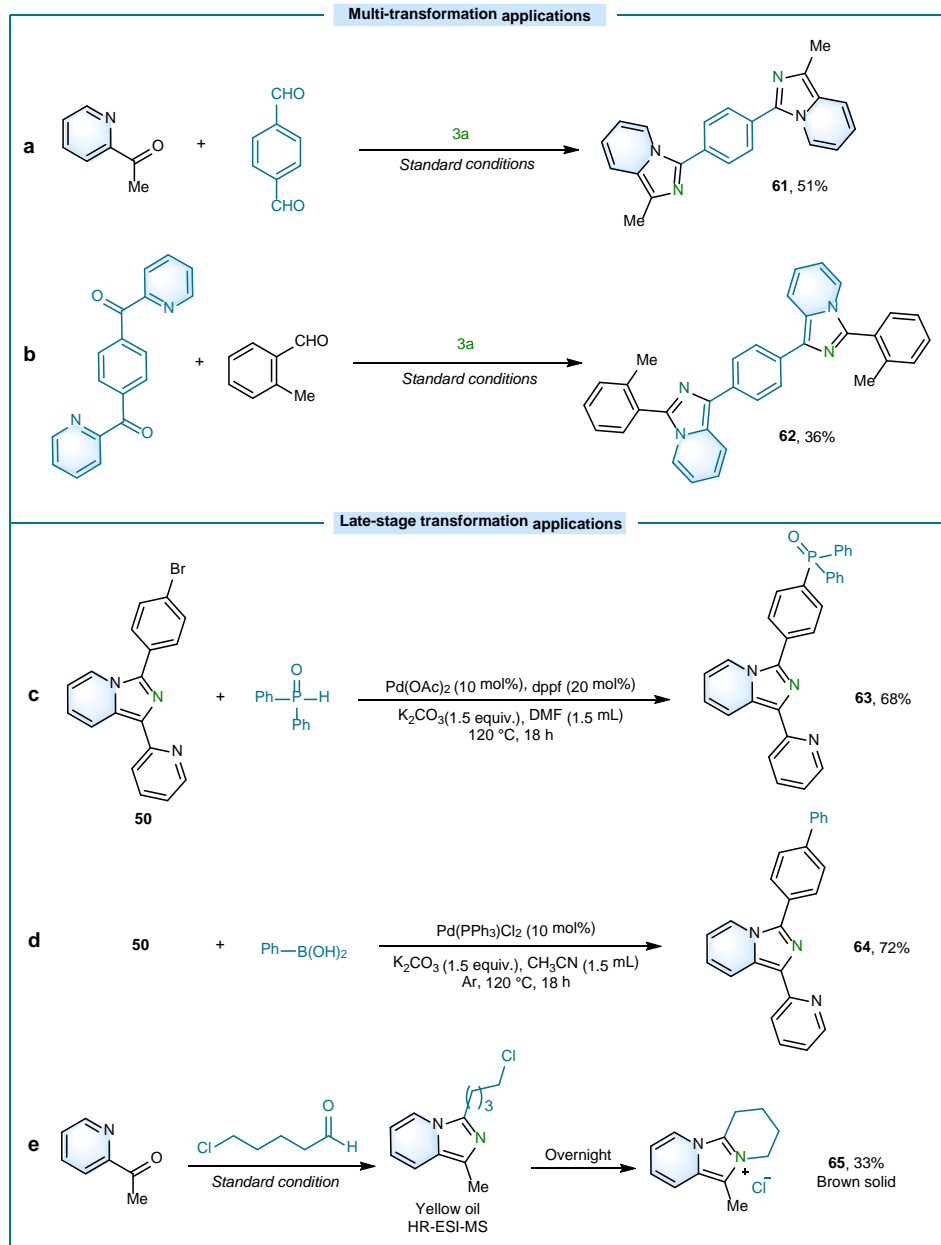


Figure 26 Synthetic applications. **(a), (b)** Multi-transformations of aldehyde and *N*-heteroaryl ketone. **(c)-(e)** Late-stage transformation of imidazo[1,5-*a*]pyridine.

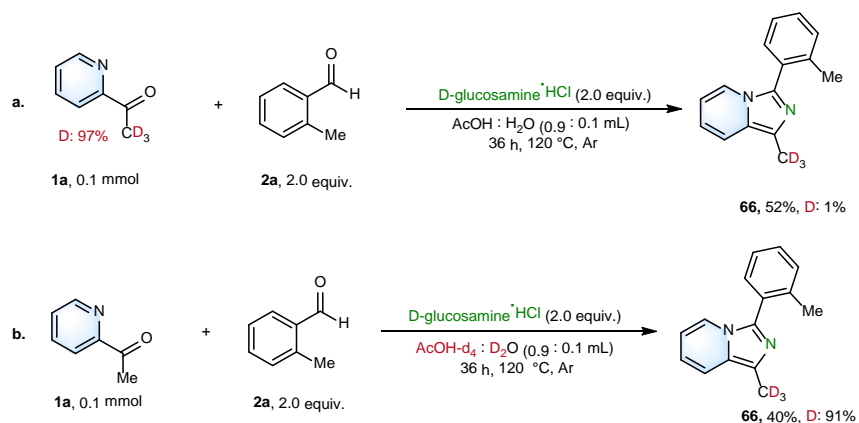


Figure 27 H/D exchange experiments.

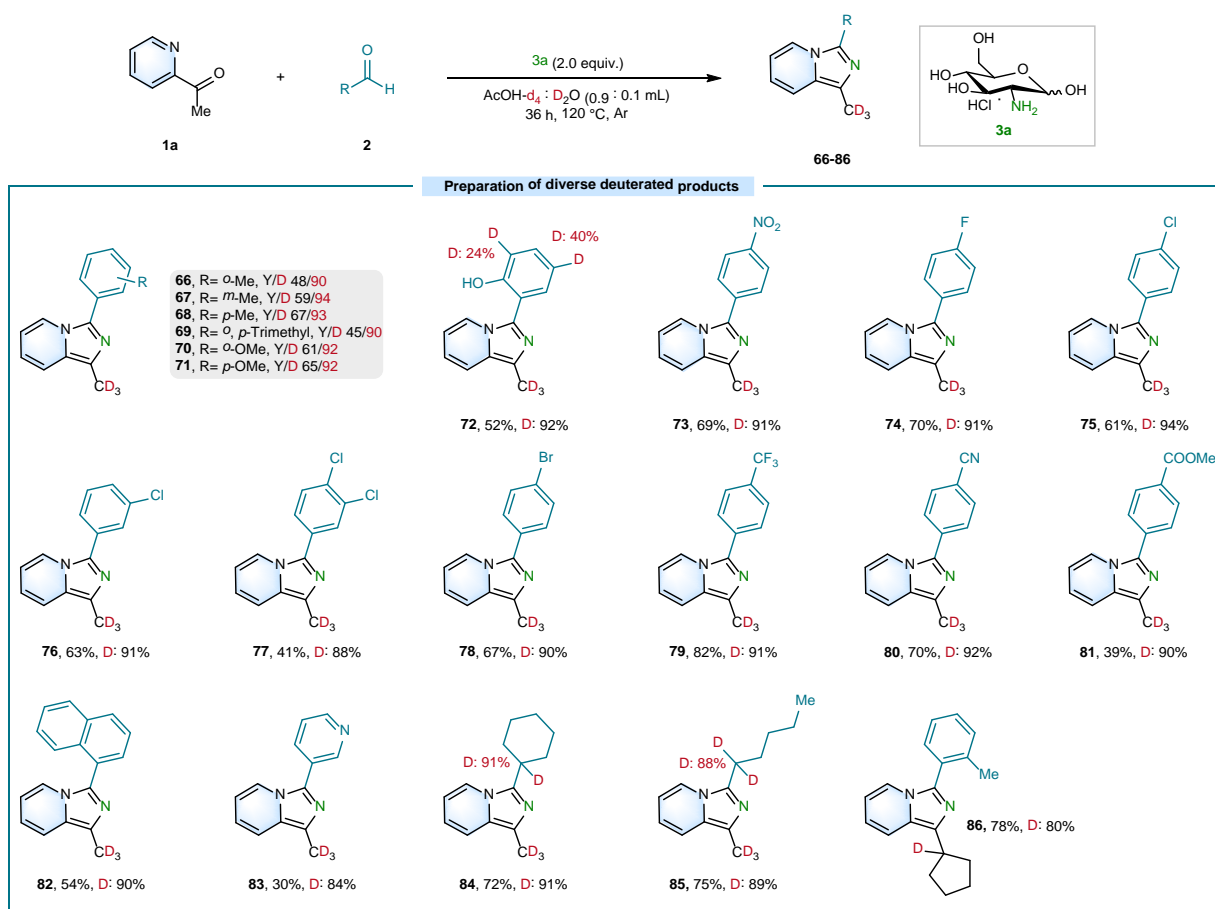


Figure 28 One-pot synthetic applications for diverse deuterated imidazo[1,5-a]pyridines. All yields are isolated products and the D incorporation was measured by ^1H -NMR analysis.

The protons at the α -position of pyridine ketone and aliphatic aldehydes could reversibly exchange with acidic aqueous surroundings (**Figure 27**). With our protocol, deuterated imidazo[1,5-a]pyridines were readily synthesized *via* a one-pot process with the simultaneous cleavage of C–N bond of D-glucosamine. The aromatic aldehydes with electro-withdrawing and electro-donating groups at diverse positions were transformed into deuterated products with high yields (**66-81**) (**Figure 28**). Moreover, 1-naphthaldehyde, pyridine aldehyde and cyclopentyl(pyridin-2-yl)methanone were also compatible with the reaction condition (products **82**, **83** and **86**). In addition, the products **84** and **85** even achieved the efficient deuteration at multiple positions.

Process mass intensity (PMI) is a key mass-based metric to evaluate the green credentials of reactions during process and chemical development.²⁵⁶ The calculations of the PMI for our current work as well as for representative approaches are shown comparatively^{234, 237-239} (**Figure 29**). The PMI_{RRC} (expressed as the amount of reagents, reactants and catalyst) of our strategy is slightly higher than the previous presentative approaches, while the PMI_{Solv} (solvent relative to the amount of isolated product) of the approach from Gevorgyan's group²³⁷ is higher than ours and the other two approaches.^{234, 238,239} It should be noted that PMI_{Solv} does not take into account of any solvent consumed during purification processes since the reference values for the comparative works are not available.

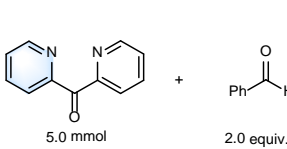
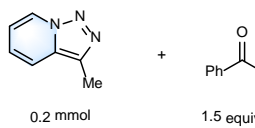
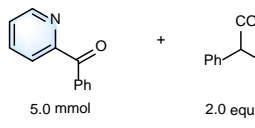
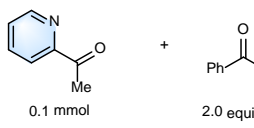
Ref.	Representative approaches		Yield	PMI _{RRC} g·g ⁻¹	PMI _{Solv} g·g ⁻¹	PMI g·g ⁻¹
236			81%	3.6	47.8	51.4
239			73%	3.4	87.0	90.4
240			93%	3.8	0.7	4.5
This work			83%	4.4	60.5	64.9

Figure 29. The calculations of process mass intensity.

3.1.5 Mechanistic studies

To gain insight into the mechanism, a set of control experiments were conducted (**Figure 30-31**). D-glucosamine is able to form imines and azomethines with aldehydes and ketones⁸⁵ and the hydroxyl groups of D-glucosamine can be modified *via* substitution reactions.⁵¹ In the first group of control experiment, intermediates **3j** and **3k** were used to verify the reaction order (**Figure 30a-30b**).^{85,257} As a result, product **13** was detected *via* ¹H-NMR spectroscopy and further confirmed *via* HR-ESI-MS (**Figure 30a**), while product **4** was not detectable (**Figure 30b**). We therefore suggest that D-glucosamine reacted with aldehyde at first to form the imine intermediate.

In the second group of control experiment, *N*-acetyl-D-glucosamine was examined under standard conditions (**Figure 30c**). The results rule out the cleavage pathway *via* *N*-acetylation of D-glucosamine. In the absence of aldehyde and 2-acetylpyridine, only traces of ammonium acetate were detected by ¹H-NMR (**Figure 30d**). The ammonium acetate was further verified by two-dimensional ¹H-¹⁵N-heteronuclear single quantum coherence (HSQC)-NMR measurement (**Figure 31**), which excluded the pathway of thermo-cleavage of the C–N bond in D-glucosamine. Moreover, the intermediates **3o**, **3p**, **3q** and **5** were detected by HR-ESI-MS and ESI-MS, which reveals the late-stage pathway with the formation of derivatives of imidazo[1,5-a]pyridines and furanoses as the intermediates after the cleavage of the C–N bond of D-glucosamine (**Figure 30e**). The isolation of furanoses is rather difficult due to the unstable properties under the acidic conditions at high temperature (**Figure S25** of **Publication 1**). A group of molecules can be detected (99 g/mol, 131 g/mol and 159 g/mol *et al.*). This indicates that the furanose is not stable and it might degrade into smaller molecules. Even though furanose has been obtained as by-product from D-glucosamine in this approach, furanose with a pending aldehyde group and hydroxyl groups represents an interesting starting material for platform chemicals including synthesis of furanose sugars,²⁵⁸ nucleophilic substitution at the anomeric position of furanose,²⁵⁹ equilibrium and non-equilibrium furanose selection in the ribose isomerization network²⁶⁰ and synthesis of furanose-based carbohydrates.²⁶¹

Furthermore, picolinaldehyde (**1j**) and formaldehyde (**2c**) were used for the same protocol under standard conditions, which excluded the pathway of post dealkylation of imidazo[1,5-a]pyridinium salts (**Figure 30f**).²⁶² 1-(pyridin-2-yl)propan-2-one (**1k**) reacts with benzaldehyde (**2b**) under the same conditions. In contrast, the *N*-transfer process from glucosamine to the *N*-heterocyclic chemical (**3s**) failed (**Figure 30g**).

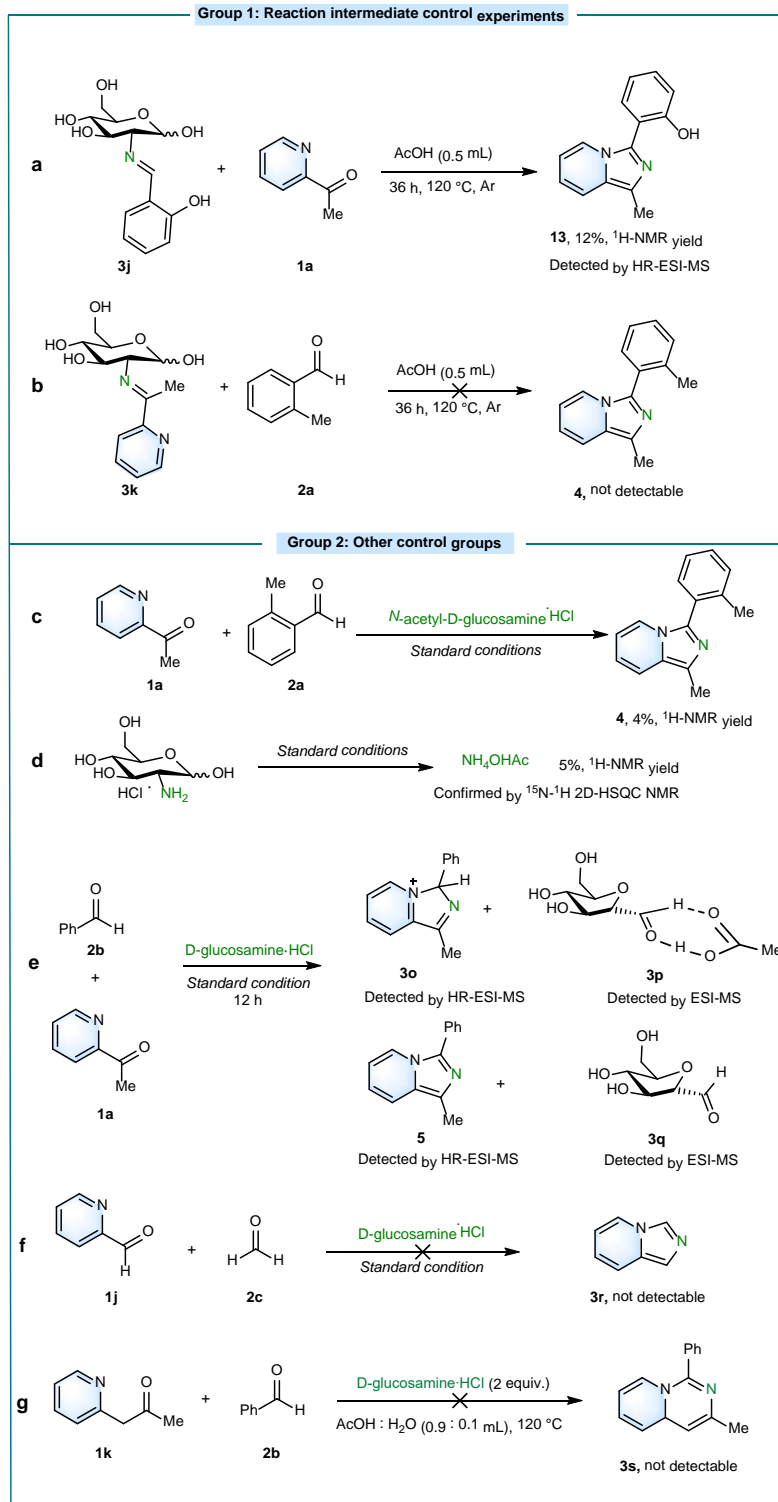


Figure 30 Reaction pathway control experiments.

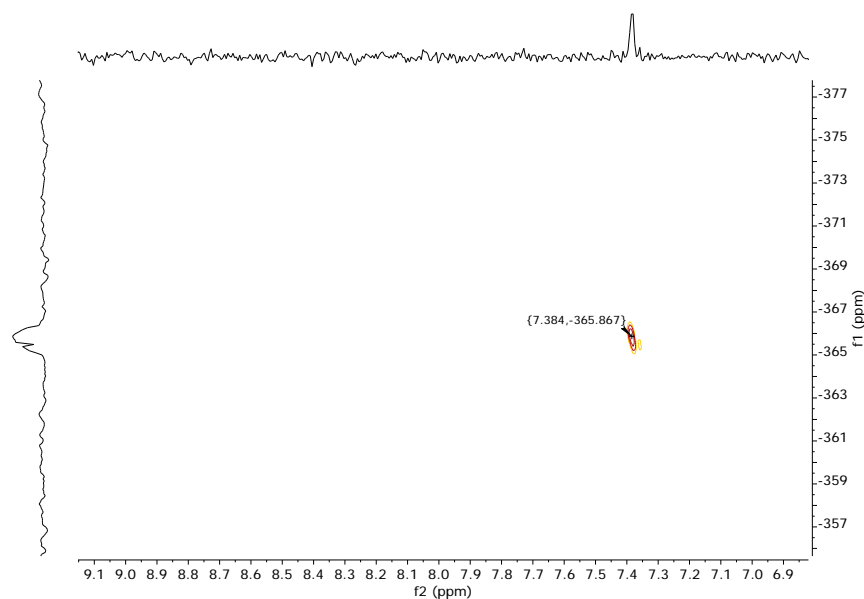


Figure 31 2D- $^{15}\text{N}^1\text{H}$ -HSQC-NMR analysis of ammonium acetate in $\text{CH}_3\text{OH-d}_4$.

In further control groups, 1,3,4,6-tetra-*O*-acetyl-2-amino-2-deoxy- β -D-glucopyranose hydrochloride (**3b**) and 1,3,4,6-tetra-*O*-acetyl-2-amino-2-deoxy- α -D-glucopyranose hydrochloride (**3c**) were tested under the same conditions (Group 3 in **Figure 32a-32b**). **3b** and **3c** show stable and pure β -anomer and α -anomer structures, respectively, while the anomer of D-glucoseamine is unstable. Therefore, **3b** and **3c** can be used to verify whether the α/β -anomer can affect the yield of product **4**. As a result, product **4** with 30% yield was obtained using **3c** (α -anomer), while **3b** (β -anomer) could only deliver 8% yield. Based on the results shown in **Figure 30**, plausible reaction pathways for TS_α and TS_β are proposed in **Figure 32** to explain the distinct reaction activities between **3c** (α -anomer) and **3b** (β -anomer). First, in comparison to **3b** (β -anomer), **3c** (α -anomer) should favor the formation of the *E* isomer of imine due to the steric hinderance (**Figure 32a**). Moreover, the α -anomer promotes the formation of a seven-membered ring transition state with the acetate anion in solutions via hydrogen bonds. This ring of the α -anomer transition state (TS_α) not only helps to stabilize the intermediate during the cleavage of the C–N bond, but also shows a favorable alignment with the aromatic ring. **3b** (β -anomer) forms a seven-membered ring transition state with β -anomer *via* hydrogen bond between the acetyl group of the β -anomer and the acetate anion (**Figure 32b**). The *E* isomer of imine forms more easily,⁸⁵ and the stronger steric shielding from seven-membered ring transition state also contributes to the formation. The ring of the β -anomer transition state (TS_β) shows a disfavorable alignment with the aromatic ring. The energy states of both TS_α and TS_β *via* D-glucosamine were calculated by electronic structure calculations (**Figure 33b**). Hence, based on the results shown in **Figure 23, 30** and **32**, a seven-membered ring of α -anomer transition state

(TS _{α}) formed *via* hydrogen bonds, which favors the following cleavage of C–N bond.²⁶³⁻²⁶⁵ Combining all results, a plausible mechanism is proposed (**Figure 33a**). First, D-glucosamine reacts with aldehyde to form imine **A**. Then, **A** attacks the ketone of pyridine ketone via electrophilic addition to generate the intermediate **B**.²⁶⁶ Under acidic conditions with acetic acid, a seven-membered ring of α -anomer transition state (TS _{α}) forms. The nitrogen of pyridine attacks the imine *via* nucleophilic addition to form the intermediate **C**. Under acidic conditions, the intermediate **D** forms via dehydration. The seven-membered ring of α -anomer transition state (TS _{α}) helps to stabilize the transition state when the C–N bond of the intermediate **D** is cleaved. The cleavage of the C–N bond in **D** results in intermediates **E** and **F**. Intermediate **F** shows a favorable alignment with the seven-membered ring. Due to the unstable transition state of **F**, **H** forms rapidly *via* the ring opening of **F** and further leads to **I**. In parallel, the deprotonation of **E** results in the product **G**.

Based on the proposed mechanism and control experiments, theoretical calculations were performed for the reaction step of the C–N bond cleavage (**D**→**E**+**H**) with the consideration of the stereoselectivity to further support the proposed mechanism. The calculated final Gibbs free energy of the transition state of the α -anomer (TS _{α} in **Figure 33b**) was 0.9 kcal/mol, which is lower than that of the β -anomer (TS _{β}). Since the Gibbs free energies of reactant connected to TS _{α} (**D** _{α}) was 0.7 kcal/mol higher than that connected to TS _{β} (**D** _{β}), the reaction barrier for the α -anomer is thus 1.6 kcal/mol and lower than that of the β -anomer (22.2 vs. 23.8 kcal/mol). Given that the two anomers do not stand in kinetic competition (they are utilized in separate reactions), the latter value should be taken as the actual barrier difference. The acetate molecule stabilizes the transition state via the hydrogen bond as depicted in **Figure 32a**, which is ultimately transferred. The ring system, as schematically shown in **Figure 32b**, aligns with the carboxylic group *via* dispersion forces that could reduce the barrier. This stands as a further example for the importance of London forces in stereoselectivity.²⁶⁷

The α/β -ratio for the mixture of D-glucosamine and HCl was determined using the same theoretical method. Three conformers (**Figure 33c**) were taken into consideration for each anomer, where the chloride might interact with each of the hydrogen atom of the protonated amine group. The Gibbs free energy of the α -v1-conformer was taken as reference for all the energy terms listed in **Figure 33c**. For each anomer the Gibbs free energy was obtained by averaging the Gibbs free energies of the three conformers with their Boltzmann-factors and applying conformational entropy corrections. The resulting final Gibbs free energy was -0.1 kcal/mol for the α -anomer and 0.8 kcal/mol for the β -anomer, respectively. The energy difference of 0.9 kcal/mol corresponds to an α/β -ratio of 3.1 at the reaction temperature of 393.15 K. This difference would be reduced to 0.55 kcal/mol if one excludes the chloride

anion. Such energy difference corresponding to an α/β -ratio of 2.0 gives us a range, which comfortably accommodates the experimental observations.

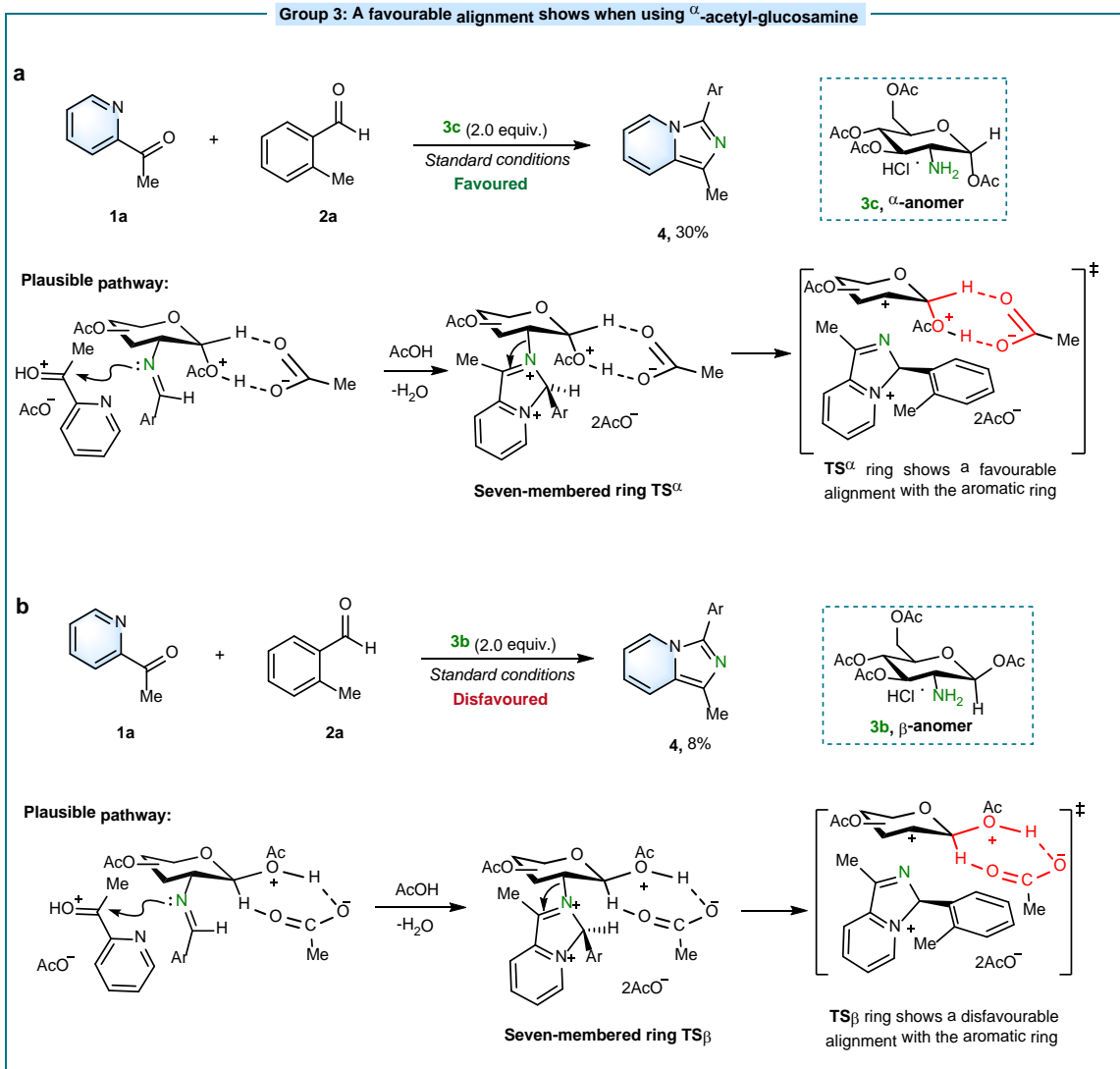


Figure 32 Anomeric stereoauxiliary control experiments.

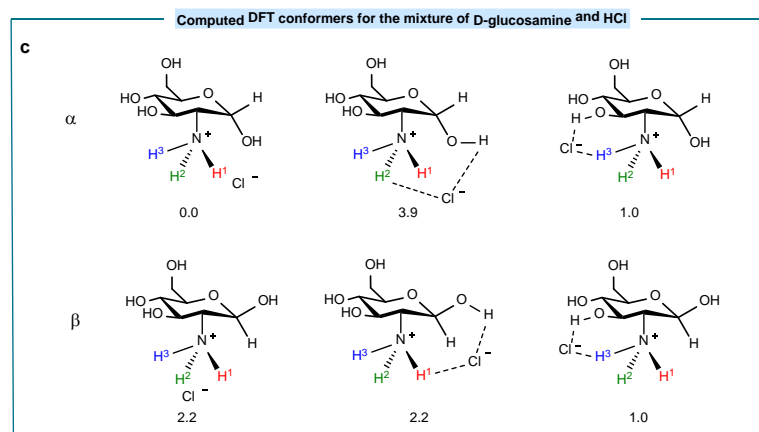
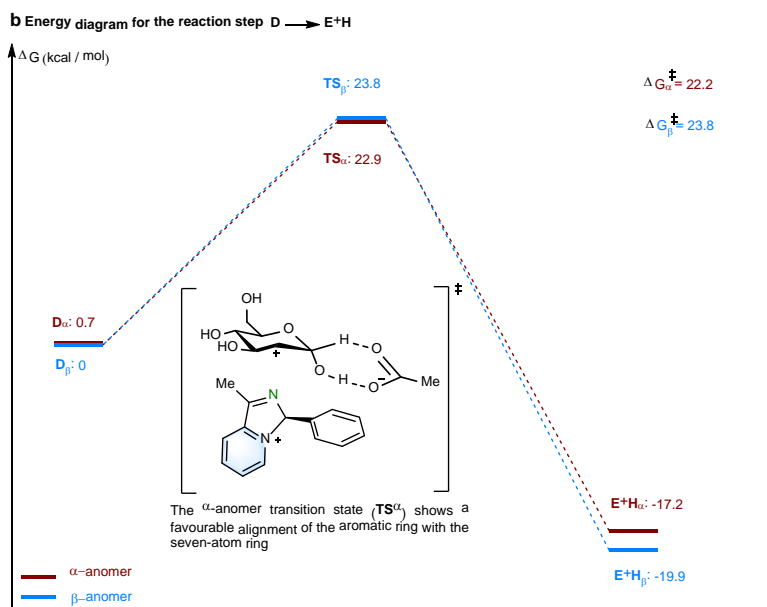
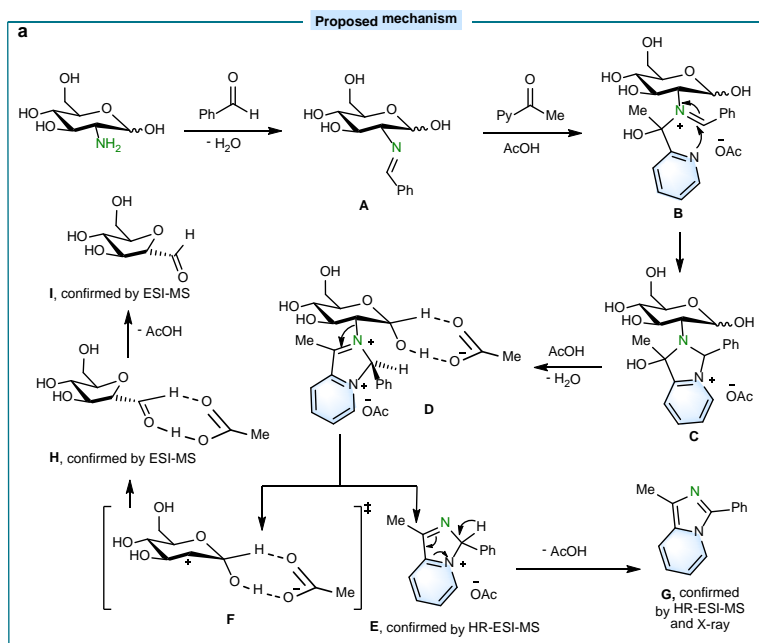


Figure 33 (a) Proposed mechanism. (b) Density functional theory calculations. (c) Simplified scheme of the computed DFT conformers for the mixture of D-glucosamine and HCl, discriminated according to the respective anomers. The relative energies (in kcal/mol) in respect to the most stable conformer are provided. Conformers in c are v1, v2 and v3 from left to right.

3.2 Publication 2: Direct Nitrogen Interception from Chitin/chitosan for the Preparation of Nitrogen-containing Chemicals

3.2.1 Background

Chitin and chitosan are abundant native linear polymers composed of randomly distributed units, namely, *N*-acetyl-D-glucosamine and D-glucosamine linked by β -1,4-linkages.²⁶⁸ The quantities of *N*-acetyl-D-glucosamine units are generally more than 50% in chitin backbone, while chitosan contains less than 50% *N*-acetyl-D-glucosamine.^{52,269,270} Human processing of sustainable chitin-containing raw materials, such as crustaceans shells, exhibits only a very minor fraction of all the chitin produced annually in nature,²⁷¹⁻²⁷⁶ while the major fraction maintains intact.³⁶ On the other hand, low molecular weight nitrogen-containing chemicals (NCCs) play a pivotal role in modern life, from pharmaceutical, agriculture and food fields to material fields.^{12,277-280} The nitrogen source of prevailing industrial processes for NCCs is mainly from NH_3 , NO_3^- and NO_2^- etc., which are obtained from N_2 fixation.²⁸¹⁻²⁸⁹ The quantities of nitrogen element fixed annually in chitin *via* the biochemical process are much more than the N_2 fixed in the Haber-Bosch process.¹³ Therefore, the transformation of renewable chitin containing fixed nitrogen into value-added NCCs has drawn much attention.¹²

Four types of strategies are currently known for the activation of amines in chitin/chitosan (**Figure 34a**). The first strategy (i) referred to the direct modifications of amines on the chitosan/chitin backbones without C–N bond cleavage.²⁷¹ The second strategy (ii) involves the cleavage of C–N bonds *via* strong oxidants or acidic conditions with the simultaneous release of N_2 or limited types of organic and primarily inorganic low-value NCCs, such as acetamide and ammonium salts.^{116,290,291} These two strategies cannot generate value-added NCCs, in particular more complicated organic compounds, which are highly desired in the modern life. Recently, the third strategy (iii) emerges with the biorefinery by converting chitin/chitosan into a preliminary C6 backbone *via* a depolymerization (e.g. monomeric and oligomeric molecules) and by further conversion of the C6 backbone into diversified products *via* breakage and

rearrangement (e.g. 3-acetamido-5-acetylfuran et al.).^{12,13} Although various protocols have been established through enzymatic, catalytic and/or hydrothermal treatments pathway, only more than 10 NCCs examples (including sugars derivatives, amino alcohols, furanic amides) have been obtained with complicated conditions and a low efficiency. The fourth strategy (iv) involves the cleavage of C–N bond

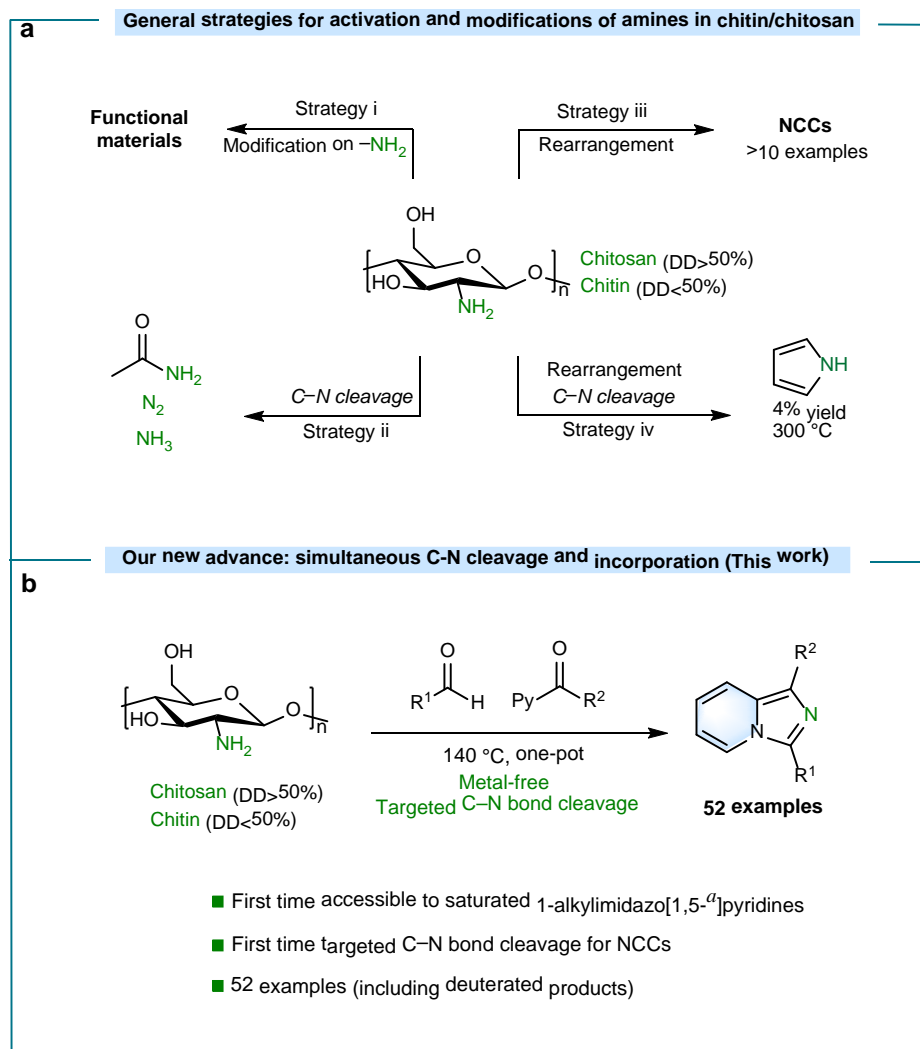


Figure 34 **a)** General strategies for activation of amines in chitin/chitosan. **b)** This work: direct incorporation of nitrogen from chitosan/chitin for imidazo[1,5-*a*]pyridines.

of chitin for the assembly of pyrrole with a low yield of 4%, which was realized in alkali aqueous solution at 300 °C in 2016.¹¹⁵ In particular, it should be stressed that the synthesis of *N*-heterocycles from chitin/chitosan biomass is challenging and introducing an external nitrogen source is the main pathway

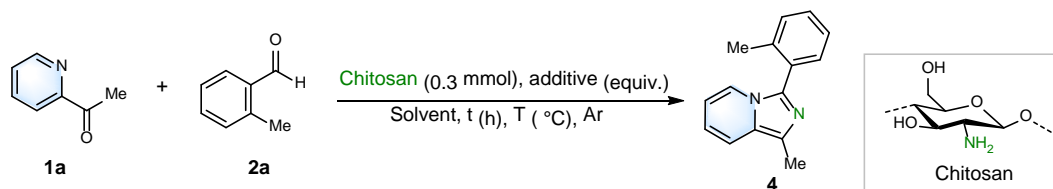
for the construction of *N*-heterocycles from biomass.²⁹²⁻²⁹⁴ The C–N bond of chitin/chitosan that offers a potential reactive site for various versatile chemical diversifications generally remains intact. Therefore, a one-pot protocol enabling the targeted efficient incorporation of nitrogen from chitin/chitosan into diverse valuable NCCs like *N*-heterocycles is highly attractive, which will advance the existing methodologies while expanding the library of NCCs derived from renewable sources.

Herein, we developed an efficient catalyst-free one-pot protocol for the direct integration of nitrogen from the renewable feedstock chitosan/chitin that yields various valuable imidazo[1,5-*a*]pyridines (52 examples with yields up to 92%) under mild conditions (**Figure 34b**), which provides ready access to 1-alkylimidazo[1,5-*a*]pyridines (39 examples) and 1-arylimidazo[1,5-*a*]pyridines (13 examples).

3.2.2 Optimization studies

We initiated our studies by using 2-acetylpyridine (**1a**) and 2-methylbenzaldehyde (**2a**) as substrates to evaluate the envisioned nitrogen interception from chitosan for desired imidazo[1,5-*a*]pyridine **3** employing metal-free condition. At the outset, product **4** was observed by ¹H NMR analysis on a condition with solvent mixture of AcOH/H₂O (0.9/0.1 mL) (**Table 4**, entries 1-2). After that, several silver salt additives were investigated. Interestingly, the yield of **4** was increased obviously from 3% to 13% when the silver trifluoroacetate (AgTFA) was exploited, while silver acetate (AgOAc) was not helpful for the efficacy (**Table 4**, entries 3-4). These results indicated that the anion ion ⁻OCOCF₃ in AgTFA might facilitate the transformation. Therefore, a solvent mixture of CF₃COOH/H₂O (0.9/0.1 mL) was employed as a substitute of the previous solvent AcOH/H₂O (0.9/0.1 mL). To our delight, the yield was improved to 29% when CF₃COOH/H₂O was used as reaction media (**Table 4**, entry 5). Then, a higher yield (61%) was achieved with CF₃COOH (1.0 mL) as the solvent (**Table 4**, entry 6). Besides, diverse reaction temperatures, ranging from 90 °C to 140 °C, were examined (**Table 4**, entries 7-8). Thus, the optimal temperature was determined to be 140 °C and a 78% yield was obtained. Moreover, the reaction was conducted at 12 h, 24 h and 36 h with the yield of 31%, 59% and 78%, respectively. The lower equivalent of chitosan leads to inferior yield. Combining all factors regarding the equivalent of chitosan, reaction temperature and time, the optimal reaction condition was identified as shown in entry 8: 140 °C for 36 h (**Table 4**).

Table 4. Optimization of the reaction conditions



Entry	Additives	Solvent (mL)	t(h) / T(°C)	Yield (%) ^a
1	-	AcOH	36 / 120	n.d.
2	-	AcOH : H ₂ O (0.9 : 0.1)	36 / 120	3
3	AgTFA	AcOH : H ₂ O (0.9 : 0.1)	36 / 120	13
4	AgOAc	AcOH : H ₂ O (0.9 : 0.1)	36 / 120	<1
5	-	CF ₃ COOH : H ₂ O (0.9 : 0.1)	36 / 120	29
6 ^b	-	CF ₃ COOH (1.0)	36 / 120	61
7 ^b	-	CF ₃ COOH (1.0)	36 / 90	4
8^b	-	CF₃COOH (1.0)	36 / 140	78
9 ^b	-	CF ₃ COOH (1.0)	12 / 140	31
10 ^b	-	CF ₃ COOH (1.0)	24 / 140	59
11 ^{b, c}	-	CF ₃ COOH (1.0)	36 / 140	66

^a Reactions were carried out with chitosan (0.2 mmol based on AGU 161 g/mol), 2-acetylpyridine (0.1 mmol), 2-methylbenzaldehyde (0.2 mmol) and solvent (1.0 mL). Yields were determined by ¹H-NMR analysis with CH₂Br₂ as the internal standard. ^b Chitosan was dried at 100 °C oven overnight. ^c Chitosan (0.2 mmol per AGU).

3.2.3 Scope of substrates

With the optimized reaction conditions in hand, we next explored the versatility of metal-free nitrogen interception from chitosan with various aldehydes for NCCs (**Figure 35**). Firstly, a series of aromatic aldehydes, including those having electron-donating or electro-withdrawing groups at different positions (*ortho*, *meta* or *para*) were subjected to the optimized conditions. As a result, these aldehydes could be efficiently transformed into the corresponding products **4-23** and **61**. Aldehydes with electro-donating groups, such as methyl, isopropyl and methoxy group (**4-12**), are well compatible with the transformations, while a lower yield was obtained for substrate with hydroxyl group (**13**). In addition,

electro-withdrawing substituents, including halogen and trifluoromethyl groups, were well accommodated. Moderate yields were achieved for aldehydes with nitro and nitrile groups. It is noteworthy that the challenging *para*-dialdehyde also delivered the corresponding products with diverse conditions (**23** and **61**). Furthermore, (*E*)-3-(4-(dimethylamino)phenyl)acrylaldehyde (**87**), 1-naphthaldehyde (**24**), nicotinaldehyde (**25**) and thiophene-2-carbaldehyde (**26**) were also amenable in our protocol. In addition to aromatic aldehydes, those aliphatic aldehydes (**28**, **30**, **31** and **33**) were well compatible with the conditions.

We further investigated the viable scope of differently substituted pyridine ketone **1** as the general coupling partners for this transformation (**Figure 36**). The di(pyridin-2-yl)methanone and phenyl(pyridin-2-yl)methanone were well compatible with the protocol (**35** and **37**). Pyridine ketones with electron-donating groups (**38-40**) or electron-withdrawing groups (**41-43**) displayed good reactivity and delivered the corresponding products efficiently. These results indicated that the current protocol was not sensitive to the electronic or stereoscopic properties of pyridine ketones. Moreover, aliphatic cyclopentyl(pyridin-2-yl)methanone (**44**) was also transformed efficiently.

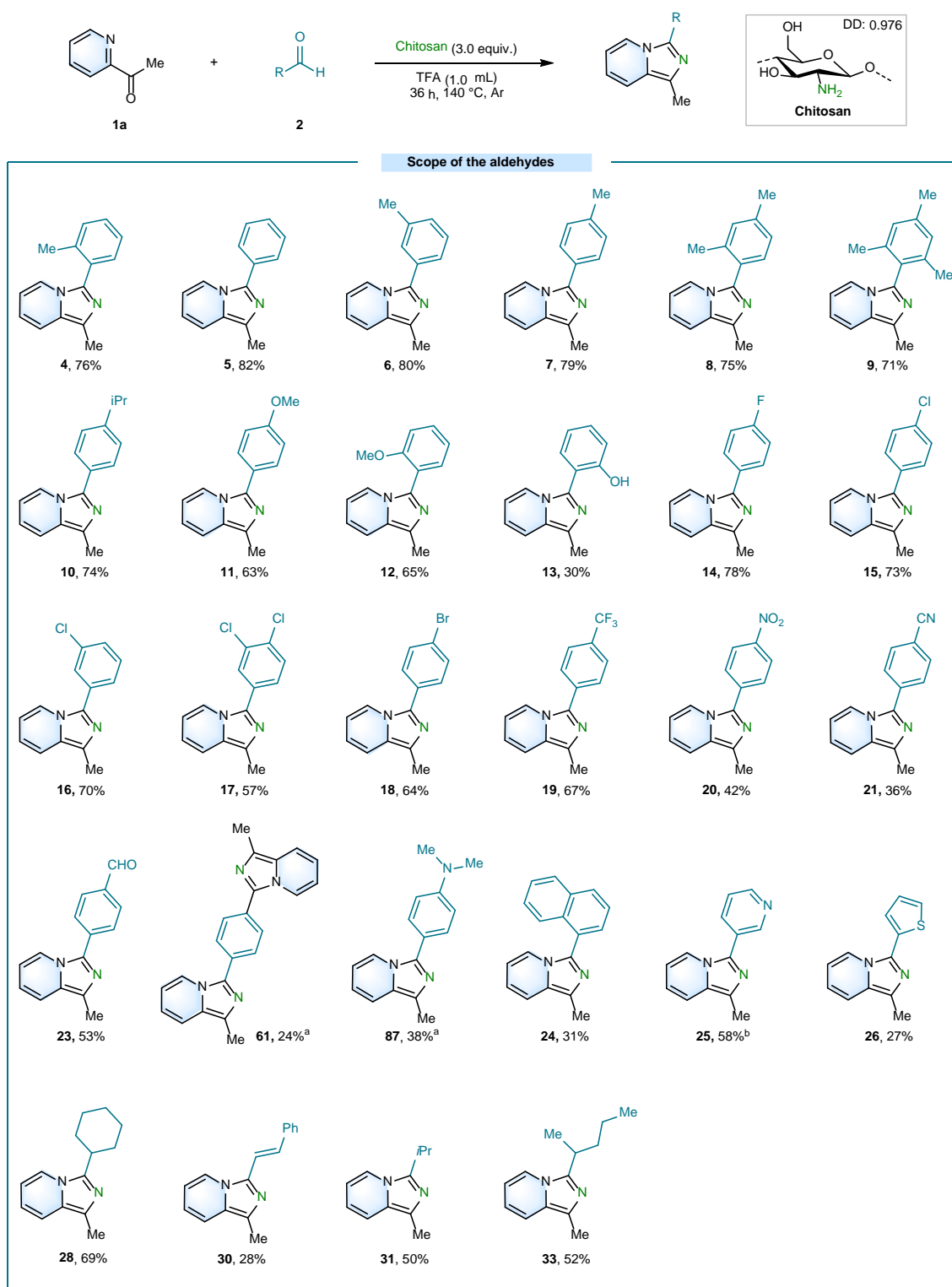


Figure 35 Scope of aldehydes for synthesis of imidazo[1,5-a]pyridines. A mixture of 2-acetylpyridine (0.1 mmol), aldehydes (2.0 equiv.) and chitosan (3.0 equiv.) in CF₃COOH (1.0 mL) was stirred at 140 °C under Ar atmosphere for 36 h. Chitosan was dried at 100 °C overnight. Yields are those of the isolated

products. ^a2-acetylpyridine (0.1 mmol), aldehyde (2.0 equiv.), chitosan (3.0 equiv) and CF₃COOH (0.7 mL), 4 days. ^b2-acetylpyridine (0.1 mmol), aldehyde (4.0 equiv.), chitosan (2.5 equiv) and CF₃COOH (0.7 mL), 36 h.

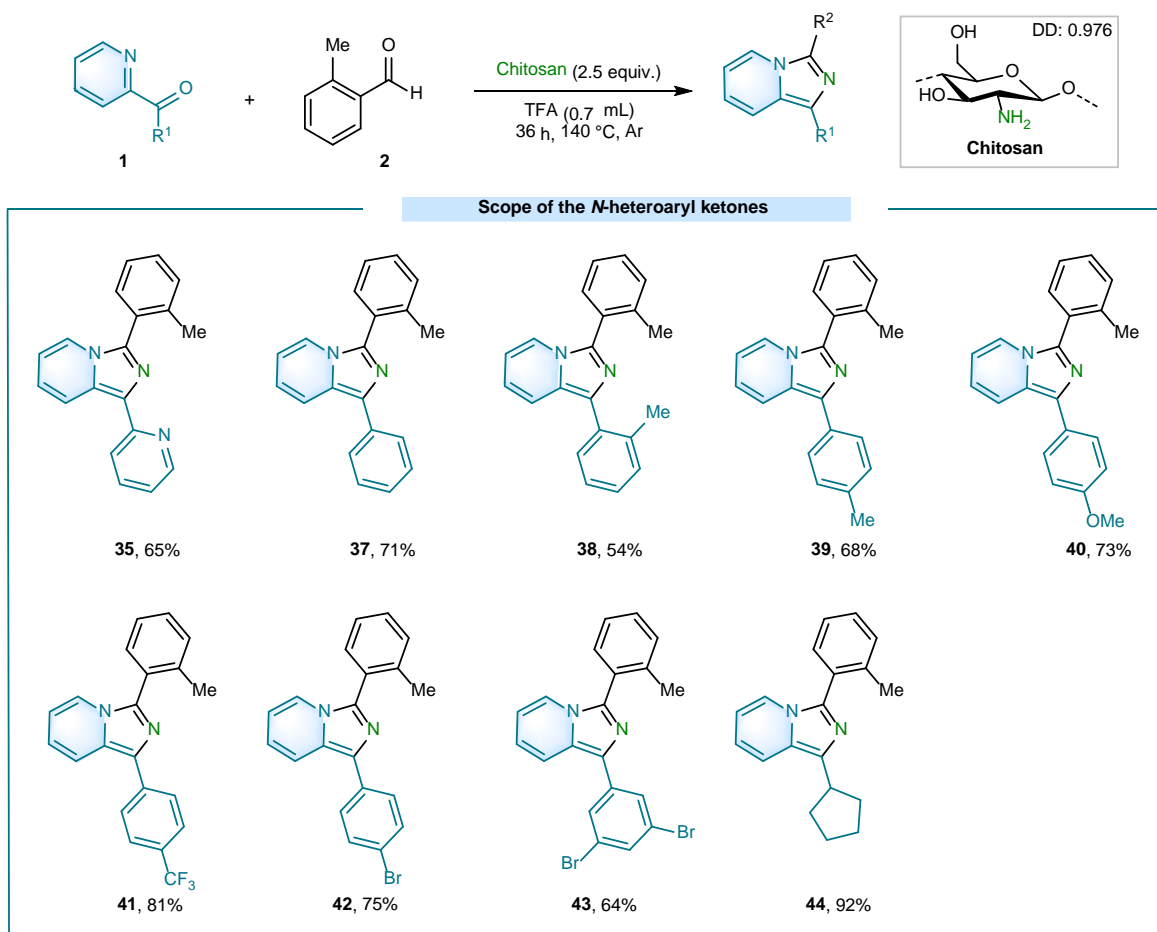


Figure 36 Scope of the pyridine ketones for synthesis of imidazo[1,5-a]pyridines. A mixture of pyridine ketone (0.1 mmol), 2-methylbenzaldehyde (4.0 equiv.) and chitosan (2.5 equiv.) in CF₃COOH (0.7 mL) was stirred at 140 °C under Ar atmosphere for 36 h. Chitosan was dried at 100 °C overnight. Yields are those of the isolated products.

3.2.4 Applications

N-heterocyclic compounds play an important role in materials and organic synthesis fields, especially as ligands in chemical transformations. Thus, the utilization of a sustainable protocol for the assembling of *N*-heterocyclic tridentate ligands is highly desirable. Herein, these tridentate ligands, 2-(1-(pyridin-2-

yl)imidazo[1,5-a]pyridin-3-yl)phenol (**46**) and 3-(2-methoxyphenyl)-1-(pyridin-2-yl)imidazo [1,5-a]pyridine (**47**), were accessed concisely through the interception of nitrogen from chitosan which further demonstrates the practical synthetic use of our method (**Figure 37**).

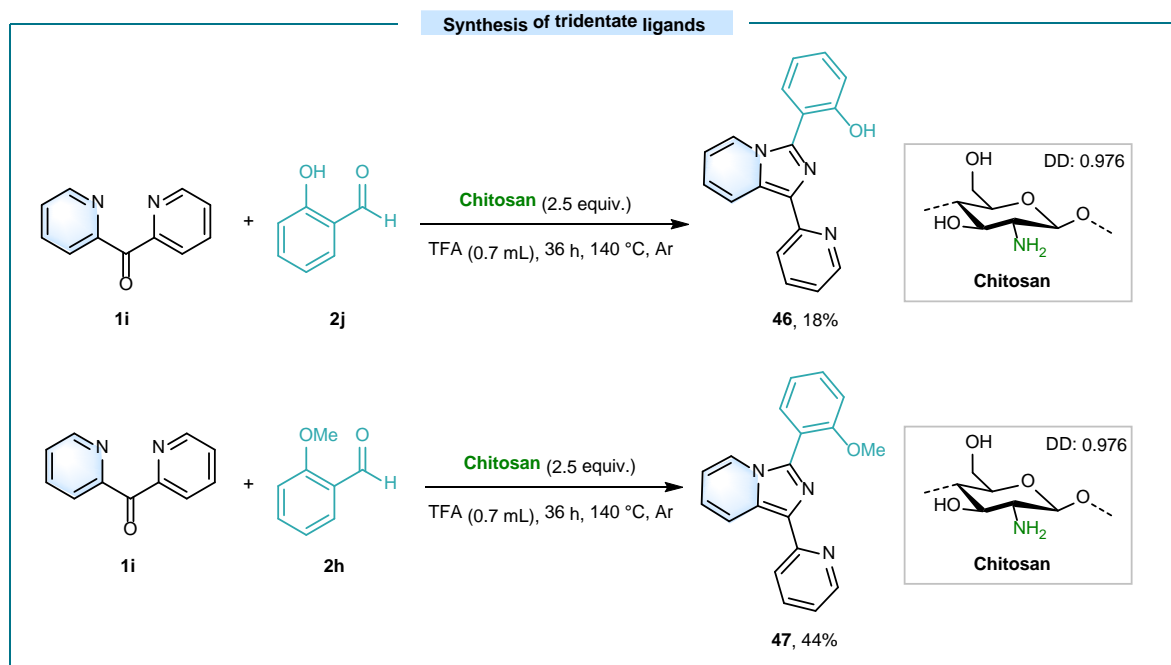


Figure 37 Synthesis of value-added tridentate ligands *via* incorporation of nitrogen from renewable chitosan. A mixture of di(pyridin-2-yl)methanone (0.1 mmol), aldehydes (4.0 equiv.) and chitosan (2.5 equiv.) in CF₃COOH (0.7 mL) was stirred at 140 °C under Ar atmosphere for 36 h. Chitosan was dried at 100 °C overnight. Yields are those of the isolated products.

Isotope labeling showed important applications on drug absorption, distribution, metabolism and excretion.^{251,252,255} Inspired by the H–D exchange between α -C–H of pyridine ketone and CF₃COOD solvent, a variety of C(sp³)–H deuterated imidazo[1,5-a]pyridines derivatives was synthesized in one-pot procedure (**Figure 38**). Those aromatic aldehydes, pyridine aldehyde and cyclic aliphatic aldehyde were synthesized with deuterated incorporation of products (**45–54**). In addition, it is worth noting that the proton of the aromatic aldehyde with a hydroxy group and methoxy group could also be deuterated under this condition (**46–48**).

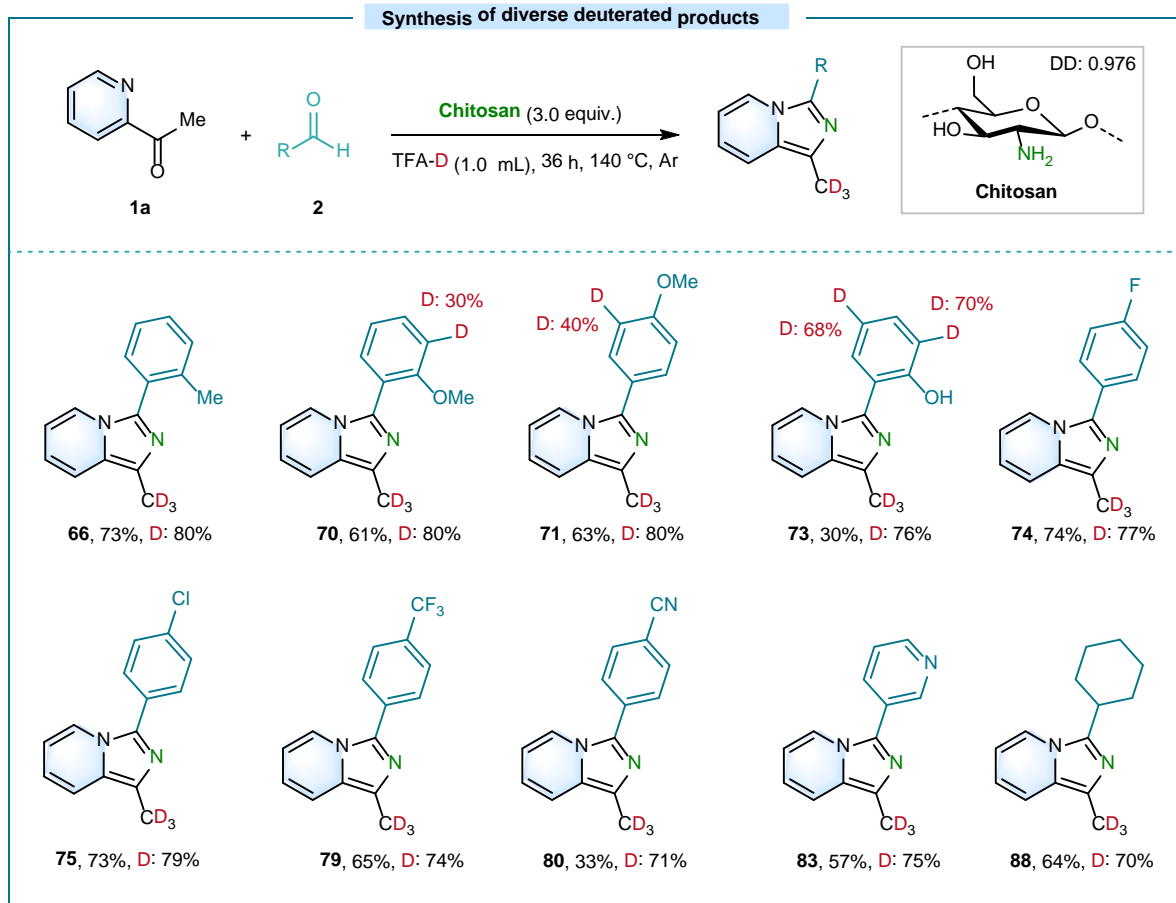


Figure 38 One-pot synthesis of deuterated C(sp³)-H bonds of imidazo[1,5-a]pyridine derivatives via incorporation of nitrogen from renewable chitosan. A mixture of 2-acetylpyridine (0.1 mmol), aldehydes (2.0 equiv.) and chitosan (3.0 equiv.) in CF_3COOD (1.0 mL) was stirred at 140 °C under Ar atmosphere for 36 h. Chitosan was dried at 100 °C overnight. Yields are those of the isolated products. D incorporation was tested by ¹H-NMR analysis.

3.2.5 Synthesis of NCCs with chitin

Furthermore, sustainable biomass chitin is proved to be a viable nitrogen source as well in this transformation, which could further bypass the deacylation process from chitin to chitosan and make the reaction more useful (**Figure 39**). As a result, a group of imidazo[1,5-a]pyridines were obtained under the standard conditions, including **3** (71%), **13** (71%), **15** (53%), **23** (29%), **26** (17%), **31** (26%), **39** (89%) and **41** (16%). Using chitosan as nitrogen sources in Fig. 2-5, these products can be obtained with higher yields, including **3** (76%), **13** (78%), **15** (70%), **23** (31%), **26** (28%), **31** (65%), **39** (92%) and **41** (44%). Despite the more advantages for these NCCs by chitin than chitosan, compared the yields of these

products by chitin with these by chitosan, chitosan shows better reaction efficiency to obtain higher yield of these products.

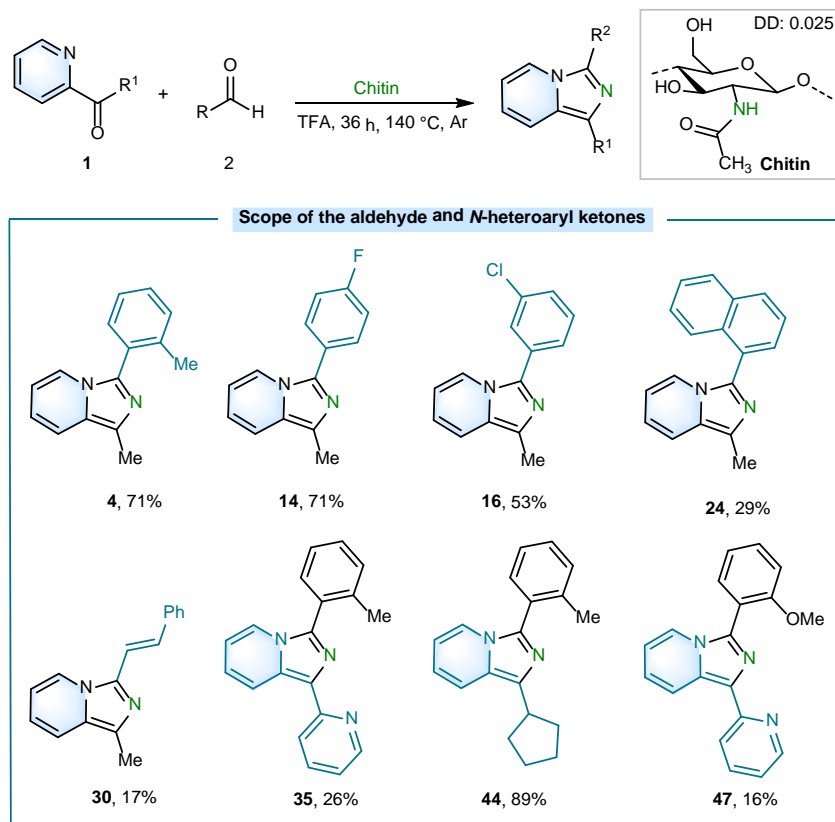


Figure 39 Direct utilization of chitin for the synthesis of imidazo[1,5-*a*]pyridines. Method for products **3**, **13**, **15**, **23** and **26**, a mixture of 2-acetylpyridine (0.1 mmol), aldehydes (2.0 equiv.) and chitosan (3.0 equiv.) in CF₃COOH (1.0 mL) was stirred at 140 °C under Ar atmosphere for 36 h. Method for products **31**, **39** and **41**, a mixture of pyridine ketones (0.1 mmol), aldehydes (4.0 equiv.) and chitosan (2.5 equiv.) in CF₃COOH (0.7 mL) was stirred at 140 °C under Ar atmosphere for 36 h. Chitosan was dried at 100 °C overnight. Yields are those of the isolated products.

3.2.6 Mechanistic studies

11% of ammonium was trapped *via* ¹H-NMR analysis, which could provide a support for the mainly release of nitrogen after the condensation to an intermediate, such as imidazo[1,5-*a*]pyridine-2,4-dium or imidazo[1,5-*a*]pyridin-4-ium (**Figure 40a**). Besides, an imidazo[1,5-*a*]pyridin-4-ium **4'** was confirmed by ESI-HRMS analysis (calc. for C₁₅H₁₅N₂⁺ [M]: 223.1230, Found: 223.1232) (**Figure 40b**), which reveals a

pathway for imidazo[1,5-*a*]pyridine **4** *via* deprotonation of **4'**. Based on the control experiment (Figure 40a-40b), a plausible reaction pathway was proposed. First, deacetylation takes place which enables the transformation of chitin to chitosan.²⁹⁵ Then, an imine intermediate is generated *via* the dehydration between an aldehyde substrate and an amine group of chitosan. The imine intermediate attacks a pyridine ketone substrate to form an iminium intermediate. Next, an imidazo[1,5-*a*]pyridine-2,4-dium intermediate forms *via* the intramolecular nucleophilic addition between pyridine and iminium in the backbone of iminium intermediate.²⁹⁶ With the help of TFA, the imidazo[1,5-*a*]pyridine-2,4-dium transfers to an imidazo[1,5-*a*]pyridin-4-ium (Figure 40b) *via* plausible pathways of dehydration and C–N bonds cleavage. Finally, an imidazo[1,5-*a*]pyridine generated *via* the deprotonation of the imidazo[1,5-*a*]pyridin-4-ium.

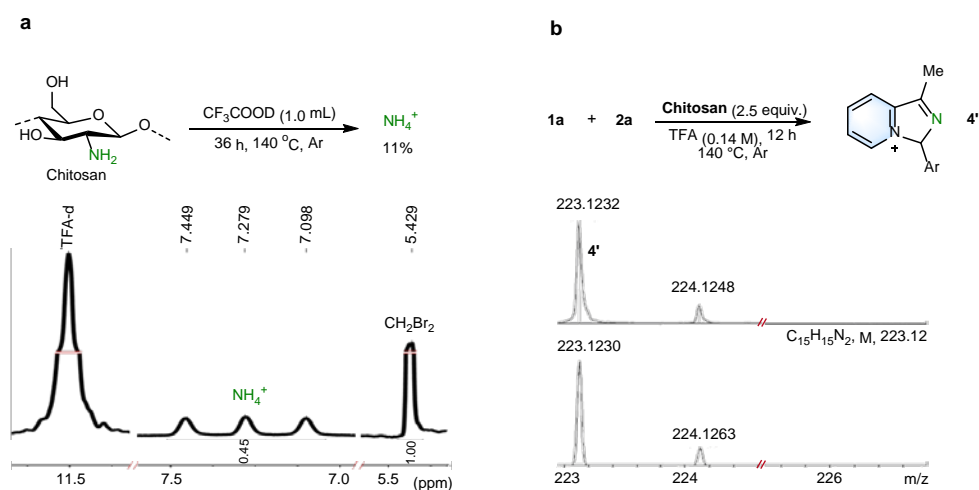


Figure 40 Control experiments. (a) Control experiment for thermal deamination. (b) ESI-HRMS analysis of intermediate.

3.3 Publication 3: Anomeric Stereoauxiliary-organocatalyzed One-pot Site-selective C_2 -aldehylation for Trisubstituted Indolizine-2-carbaldehydes

3.3.1 Background

Indolizines are an important group of *N*-heterocyclic compounds that play pivotal roles in pharmaceuticals (Figure 41a),²⁹⁷⁻³⁰³ materials science,³⁰⁴ and chemical feedstocks.^{305,306} The easily modifiable carbaldehyde group makes indolizine-carbaldehyde a precursor for versatile indolizine products. Although tremendous progress has been made in the preparation of indolizines mainly by Scholtz

reaction,^{307,308} Tschitschibabin reaction,^{309,310} pyridinium *N*-methylides,^{311,312} and cyclization of alkynes with heteroaromatic compounds,^{236,313,314} the chemical construction of indolizine-carbaldehydes with a direct site-selectivity (C₁/C₂/C₃) has been a long-standing challenge for synthetic chemists.

Various strategies for direct site selectivity on C₁/C₃ of indolizine-carbaldehydes have been reported so far. **i)** Representative strategies for C₁ site-selectivity. In 2018, Li group reported the iron-catalyzed aerobic oxidation of pyridine and allenolate for the preparation of trisubstituted indolizine-1-carbaldehydes.³¹⁵ It is highly dependent on specific allenolate substrates with electron-withdrawing groups and the traditional pyridinium *N*-methylides route. **ii)** Representative strategies for C₃ site selectivity. In 1984 and 2017, a four-step synthesis of trisubstituted indolizine-3-carbaldehydes was reported.^{316,317} This method first requires the synthesis of indolizines *via* a traditional pyridinium *N*-methylide strategy, followed by the introduction of a carbaldehyde group at the C₃ position *via* a non-green Vilsmeier reagent. In addition, in 2017, Wang group developed a novel amine-*N*-heterocyclic carbene relay catalysis strategy for the preparation of trisubstituted indolizine-3-carbaldehyde.³¹⁸ Although indolizine-3-carbaldehyde is efficiently synthesized *via* a relay co-organocatalyst, this method is still limited to pyridine substrates with 2-substituted electron-withdrawing groups, which greatly limits the accessible substrates.

In contrast, there are few effective strategies for the site-selective C₂-aldehylation of indolizine-carbaldehyde, especially for trisubstituted indolizines. In 2016, Biagetti *et al.* synthesized 1,2-disubstituted indolizine-2-carbaldehyde derivatives in 6 steps (**Figure 41b**).²⁰² Two years later, a monosubstituted indolizine-2-carbaldehyde was prepared in two steps by using a commercial indolizine-2-carboxylic acid, for just one example.²⁰³ However, these two C₂ site-selective strategies are highly rely on multiple synthetic steps using non-green expensive reagents (first four-steps construction of disubstituted indolizines with C₂-substituted carboxyl group *via* precious metal catalyst, followed by introduction of a carbaldehyde on indolizine through reduction and oxidation processes)²⁰² or expensive and rare commercial feedstocks such as monosubstituted indolizine-2-carboxylic acid (395.57 €/g from Sigma company) *via* amidation and reduction processes.²⁰³ In addition, only a few examples of monosubstituted indolizine-2-carbaldehyde have been constructed, while the high value-added 1,2,3-trisubstituted indolizine-2-carbaldehyde is still not available. In 2021, Zhao *et al.* reported a metal-free catalytic method using acetic acid as a solvent for 2-acylindolizine and indolizine-2-carbaldehydes.³¹⁹ This method is limited to internal enones and 1-aryl-indolizine. In particular, the method was significantly less efficient for 1-alkyl-2-acylindolizine (20%-24%) and cyclic indolizine (8%).

Due to the challenges of sensitive functional groups, regioselectivity, and strong steric shielding, there is currently no efficient one-pot method for the construction of 1,2,3-trisubstituted indolizine-2-carbaldehydes derivatives by dehydration [3+2] cyclization of cinnamaldehyde and 2-acetylpyridine. As

the most severe challenge, 2-acetylpyridine readily attacks the carbonyl group of cinnamaldehyde with a base,²⁴⁰ while 2-acetylpyridine activated by metal-based Lewis acid can attack the β -position of α,β -unsaturated aldehydes in the presence of secondary organoamine as catalysts.³²⁰

Asymmetric organocatalysis is one of the most thriving research domains in contemporary organic synthesis.¹⁶¹⁻¹⁶⁶ Until today, aminocatalysis with iminium and enamine catalysis modes has drawn much attention.^{178-183,185,186} However, the majority of chiral skeleton aminocatalysts are provided by artificially designed adjustable steroauxiliary group, which provide strong steric shielding for regioselectivity and stereoselectivity control. Thus, it is attractive to utilize an inexpensive aminocatalyst with a natural charil backbone. Commercially available D-glucosamine as an amino-containing monosaccharide is ubiquitous in nature derived from chitin.²⁶⁸ Sugar-derived thiourea organocatalysis has only been reported as a few examples.¹⁹⁰⁻¹⁹² Up to now, there are no examples of aminocatalysis using carbohydrates alone. Inspired by the regioselective-controlled aminocatalysis strategy and the different stereochemical structures of α/β -anomers of D-glucosamine, we demonstrate for the first time a novel synthetic strategy using β -anomeric glucosamine as a stereoeauxiliary aminocatalyst for one-pot site-selective construction of 1,2,3-trisubstituted indolizine-2-carbaldehydes (**Figure 41c**). To overcome the above challenges, we used Brønsted acids to stabilize the methyl group of 2-acetylpyridine and hinder the deprotonation of the methyl group.²⁴³ In addition, lithium cations are used to stabilize oxygen anions and activate cinnamaldehyde.³²¹

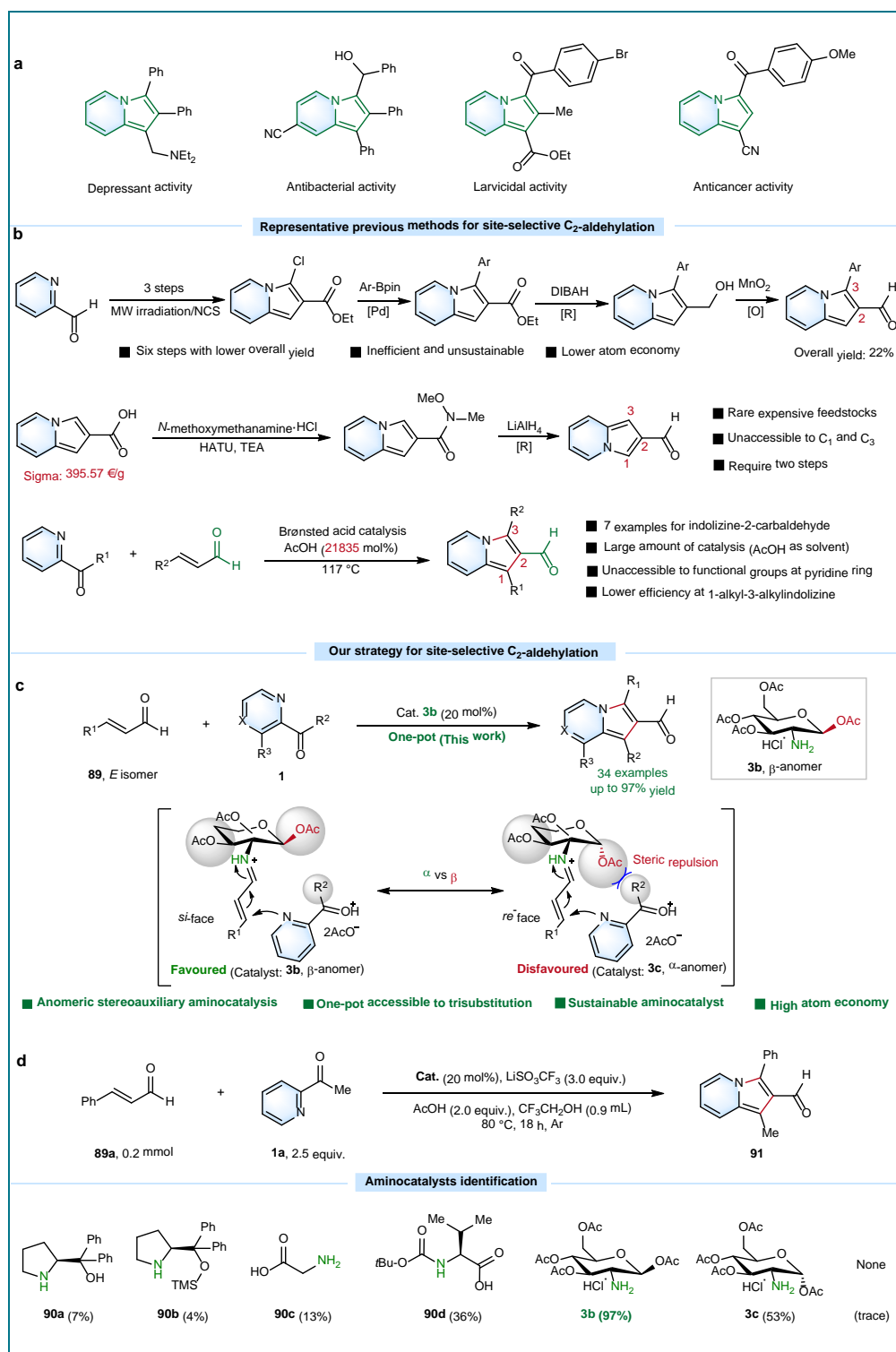


Figure 41 (a) Pharmaceuticals derived from indolizine. (b) The representative example of previous approaches for indolizine-carbaldehydes with site-selective C₂-aldehylation. [O] Means oxidation process. [R] Means reduction process. [A] Means amidation process. (c) Our strategy: An anomeric

stereoauxiliary organocatalysis by glucosamine enables one-pot site-selective for trisubstituted indolizine-2-carbaldehydes. **(d)** Optimized aminocatalysts for 1,2,3-trisubstituted indolizine-2-carbaldehydes *via* one-pot reaction. NCS means *N*-chlorosuccinimide. DIBAL means diisobutylaluminum hydride.

3.3.2 Optimization studies

We initiated our studies using cinnamaldehyde (**89a**) and 2-acetylpyridine (**1a**) as substrates to evaluate the envisioned aminocatalyzed [3+2] cyclization reaction for the synthesis of desired 1-methyl-3-phenylindolizine-2-carbaldehyde (**4**). At the outset, without catalyst, the reaction was tested with a trace yield of product **4** with a mixture of **89a** (0.2 mmol), **1a** (2.5 equiv.), LiSO₃CF₃ (3.0 equiv.) and acetic acid (2.0 equiv.) in CF₃CH₂OH (0.9 mL) for 18 h under Ar gas atmosphere (**Figure 41d**). Then, we examined various aminocatalysts (**Figure 41d**). By using (S)-diphenyl(pyrrolidin-2-yl)methanol (**90a**) and (S)-2-(diphenyl(trimethylsilyl)oxy)methylpyrrolidine (**90b**) as catalysts, a low yield of product **4** was achieved. Next, diverse sustainable aminocatalysts, including glycine (**90c**), (*tert*-butoxycarbonyl)-L-valine (**90d**), 1,3,4,6-tetra-*O*-acetyl-2-amino-2-deoxy- β -D-glucopyranose hydrochlorid (**3b**) and 1,3,4,6-tetra-*O*-acetyl-2-amino-2-deoxy- α -D-glucopyranose hydrochloride (**3c**), were investigated under the same conditions. As a result, a low yield of **4** was obtained with **90c** and **90d**. Surprisingly, 97% yield of **4** was smoothly delivered by using catalyst **3b**, while **3c** only achieved 53% yield of **4**. Based on all these results, **3b** was taken as the optimal aminocatalyst. In addition to amine-containing catalysts showing the central function for the efficient reaction, acetic acid plays an important role. Without acetic acid, the yield of **4** decreased obviously from 97% to 44%.²⁴³ As well, the amount of LiSO₃CF₃ (2 equiv.) and 2-acetylpyridine (1.5 equiv.), reaction time (12 h) and reaction temperature (25 °C and 50 °C) also affected the yields.

3.3.3 Mechanistic studies

To gain insight into the reaction mechanism, a set of control experiments were conducted (**Figure 42a-42d**). First, under standard condition, catalyst **3b** with β -anomer was examined to smoothly achieve 97% yield of **4**, while catalyst **3c** with α -anomer only yielded 53% of **4** (**Figure 42a** and **42c**), which obviously demonstrates the presence of a strong steric shielding from α -anomer that affects the efficient conversion for the desired product **4**. In the further control groups, **90p** with β -anomer and **90q** with α -anomer were tested under the same conditions (**Figure 42b** and **42d**). Interestingly, product **4** with 51% yield was obtained using **90p** (β -anomer), while **90q** (α -anomer) could only deliver 16% yield. Thus, the imine

reaction pathway *via* aminocatalyst preferentially reacting with α,β -unsaturated aldehydes is reasonable. Besides, the lower yield with **90q** (α -anomer) further provides a strong support for the existing bulky steric hinderance from acetyl group at C₁-position in **90q**. Therefore, control experiments in **Figure 42** clearly demonstrated that a stereoauxiliary effect favored by β -anomer as well as a steric shielding effect from α -anomer obviously show a tremendous difference during the [3+2] cyclization for the preparation of 1,2,3-trisubstituted indolizine-2-carbaldehydes.

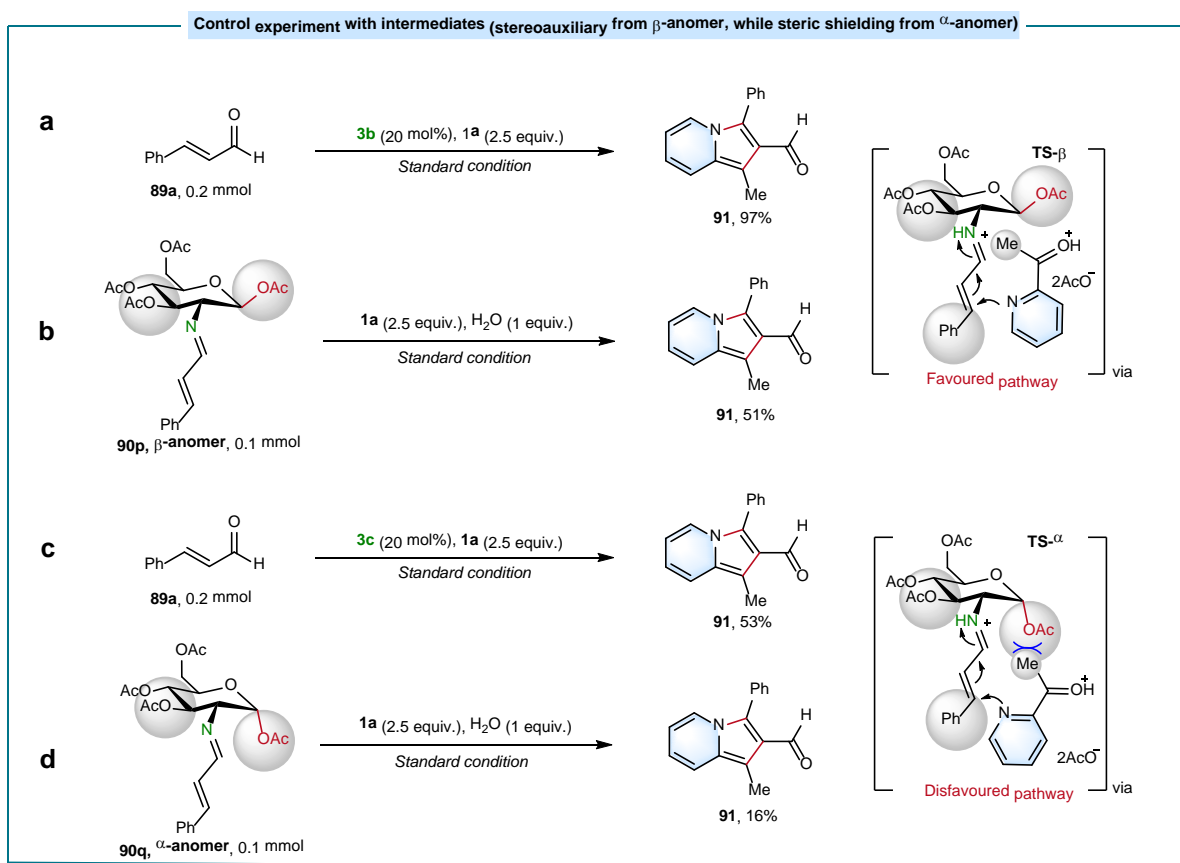


Figure 42 Stereoauxiliary control experiments. **(a)-(d)** Control experiments with intermediates of different anomers: stereoauxiliary from β -anomer, while steric shielding from α -anomer.

Combining all results, the plausible mechanism is proposed (**Figure 43**). First, aminocatalyst **A** reacts with α,β -unsaturated aldehyde **B** to form imine ion **D**.¹⁸⁰ Then, the 2-acetylpyridine attacks the imine ion **D** *via* Michael addition reaction to generate an enamine **E**.^{163,180} Because there is the bulky steric hinder between R¹ and R², which could hinder the cyclization of **E**. Thus, an enamine **F** was formed by

the conversion of **E** *via* a rotation. Next, an intermediate **G** forms *via* the intramolecular cyclization reaction in the enamine **F**. Then, an intermediate **H** generates through a dehydration reaction in the intermediate **G**. After that, an intermediate **I** forms from the deprotonation reaction of **H**. Finally, an indolizine **J** generates *via* the hydrolysis reaction of intermediate **I** and the catalyst **A** regenerates to inter next catalysis cycle.

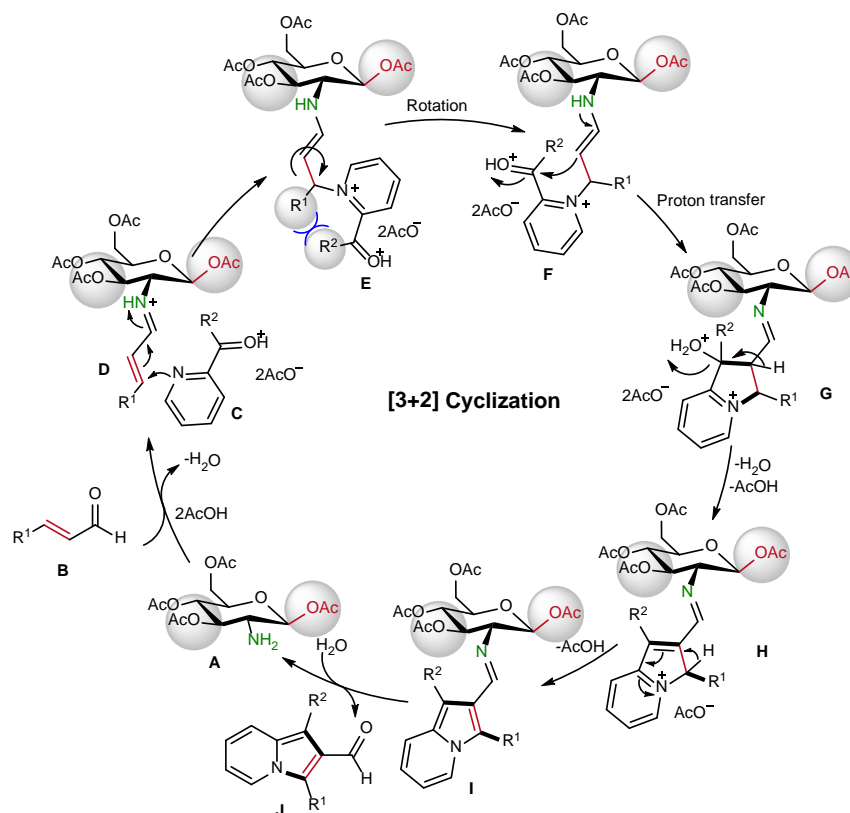


Figure 43 Proposed mechanism.

3.3.4 Scope of substrates

With the optimized reaction conditions in hand, we next probed the scope of various α,β -unsaturated aldehydes using 2-acetylpyridine as a representative heteroaryl ketone (**Figure 44a**). A series of α,β -unsaturated aldehydes, including those with electron-donating or -withdrawing groups at different positions (*ortho*, *meta* or *para*), were unprecedentedly delivered to the corresponding products **91-100**. An array of valuable products **91-95** were efficiently accessed with this aminocatalyzed protocol. Notably,

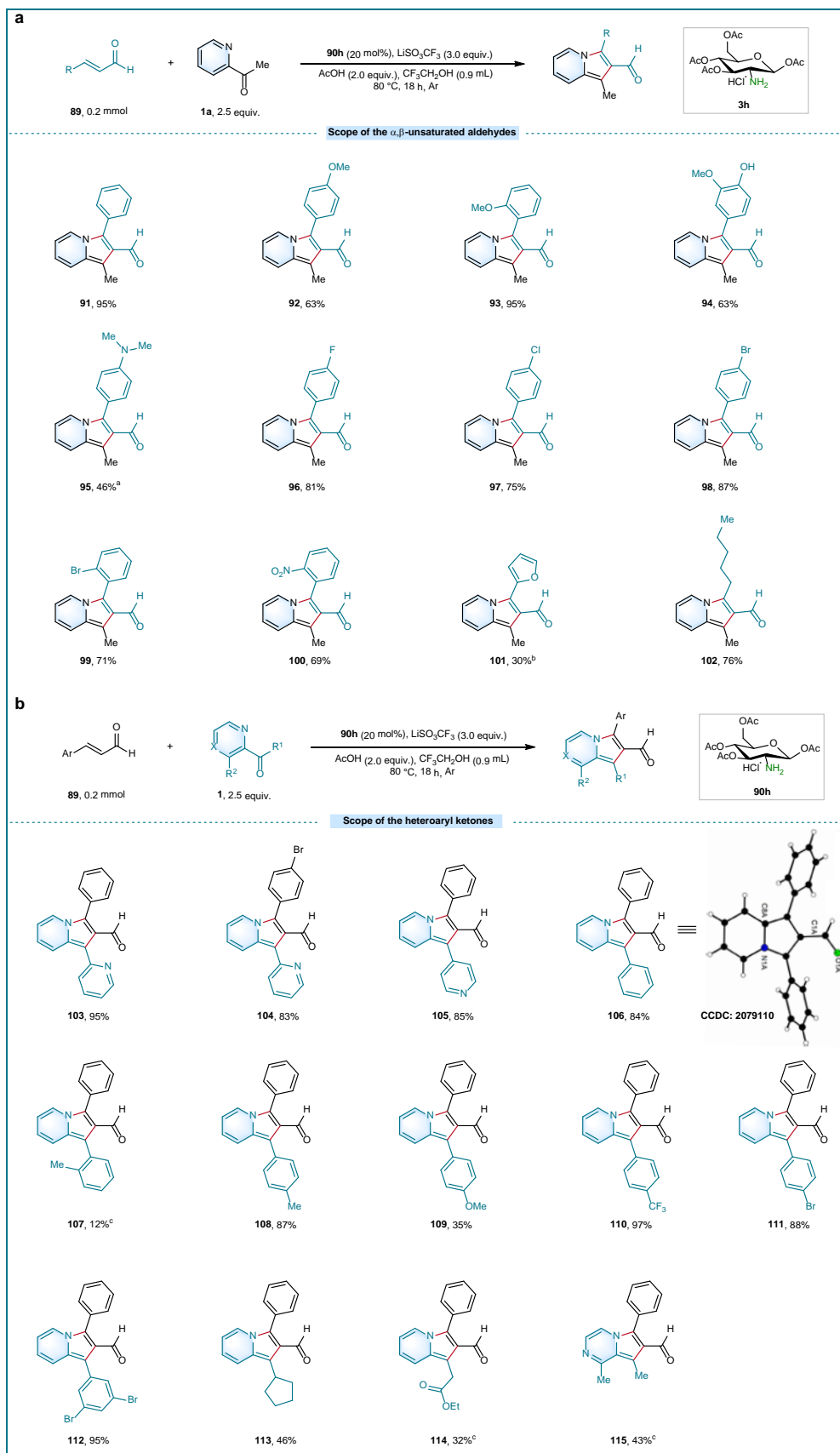


Figure 44 [3+2] annulations for indolizine-2-carbaldehydes. **(a)** Scope of aldehydes. **(b)** Scope of the heteroaryl ketones. ^aReaction for 36 h in AcOH : CF₃CH₂OH (0.4 : 0.5 mL). ^bAcOH (4.0 equiv.) and reaction for 42 h at r.t. ^cReaction time 36 h. Yields are those of isolated products.

in our system, a substrate with an electron-donating methoxy group at *ortho* position (**93**, 95%) could even achieve a higher yield than those at *para* position (**92**, 63%). Surprisingly, a native valuable substrate from *gliricidia sepium* with a hydroxyl group and a methoxy group was smoothly transformed into a value-added indolizine-2-aldehyde with a moderate yield of 63% (**94**). Interestingly, an important substrate for detection of catechins was also tolerant under this method with a 46% yield (**95**). Meanwhile, a variety of valuable functional groups at diverse positions, such as fluoro (**96**), chloro (**97**), bromo (**98**, **99**), and nitro groups (**100**), were well compatible with these conditions. Particularly, the sensitive (*E*)-3-(furan-2-yl)acrylaldehyde was also tolerated in this protocol and was successfully transformed into desired product (**101**). Moreover, aliphatic α,β -unsaturated aldehyde was also well compatible under the optimal conditions (**102**).

We further explored heteroaryl ketones with cinnamaldehyde as a standard substrate (**Figure 44b**). Di(pyridin-2-yl)methanone and pyridin-2-yl(pyridin-4-yl)methanone were well compatible with the conditions and smoothly achieved 95% (**103**), 83% (**104**) and 85% yield (**105**), respectively. Diverse aromatic pyridine ketones, including those having electron-donating or -withdrawing groups at distinct positions (*ortho*, *meta* or *para*) were well transformed into the corresponding products (**106-112**). Various valuable functional groups at distinct positions (*ortho*, *meta* or *para*), including methoxy (**109**), trifluoromethyl (**110**), bromo (**111**) and dibromo (**112**), were well tolerated under this condition. The structure of **106** was further confirmed by X-ray crystallographic analysis, and those of other products in **Figure 44** were assigned by analogy. It is worth to note that cyclopentyl(pyridin-2-yl)methanone (**113**) and ethyl 3-oxo-3-(pyridin-2-yl)propanoate (**114**) were also unprecedentedly transformed into valuable 1-alkyl-3-arylindolizine-2-carbaldehydes with the protocol. Surprisingly, 1-(3-methylpyrazin-2-yl)ethan-1-ol was also compatible with the condition (**115**).

3.3.5 Applications

Indolizines show important biological activities. To obtain diverse value-added indolizine, it is attractive to introduce a modifiable aldehyde group in an indolizine backbone. Thus, these valuable indolizine-2-carbaldehydes were achieved *via* 6 steps with complex conditions²⁰² and 2 steps with rare expensive feedstocks (**Figure 45a**).³²² Compared with their protocols *via* carboxylation and reduction for the desired

products, we efficiently achieved a group of value-added 1,2,3-trisubstituted indolizine-2-carbaldehydes *via* totally 2 steps *via* aminocatalyzed [3+2] cyclization reaction. Besides, a group of important bioactive molecules or drugs was investigated in our protocol (**Figure 45b**). Surprisingly, an important fluvastatin intermediate was unprecedentedly accessed by our protocol for the preparation of value-added indolizine-2-carbaldehyde (**116**). As well, (*E*)-3-(4-hydroxy-3-methoxyphenyl)acrylaldehyde from *gliricidia sepium* was also tolerant under the optimal conditions, which led to 3-(4-hydroxy-3-methoxyphenyl)-1-(pyridin-2-yl)indolizine-2-carbaldehyde (**117**) with 79% yield. Interestingly, (*E*)-3-(4-(dimethylamino)phenyl)acrylaldehyde that is often used to detect catechins³²³ was also smoothly transformed into 3-(4-(dimethylamino)phenyl)-1-(pyridin-2-yl)indolizine-2-carbaldehyde (**118**). Furthermore, the obtained indolizine-2-carbaldehydes could be readily late-stage diversified, thus providing more complex molecules in an efficient manner (**Figure 45c**). Namely, 3-(4-bromophenyl)-1-(pyridin-4-yl)indolizine-2-carbaldehyde (**104**) underwent reduction (**119**), arylation (**120**), acylation (**121**) or dehydration [5+1] annulations (**122**), showcasing the synthetic power of 1,2,3-trisubstituted indolizine-2-carbaldehydes that were assembled *via* our robust stereoauxiliary aminocatalysis. Finally, our anomeric stereoauxiliary aminocatalysis strategy was first time expanded beyond the library of indolizine-2-carbaldehydes to 2-acylindolizines (**123**) and (**124**) (**Figure 45d**).

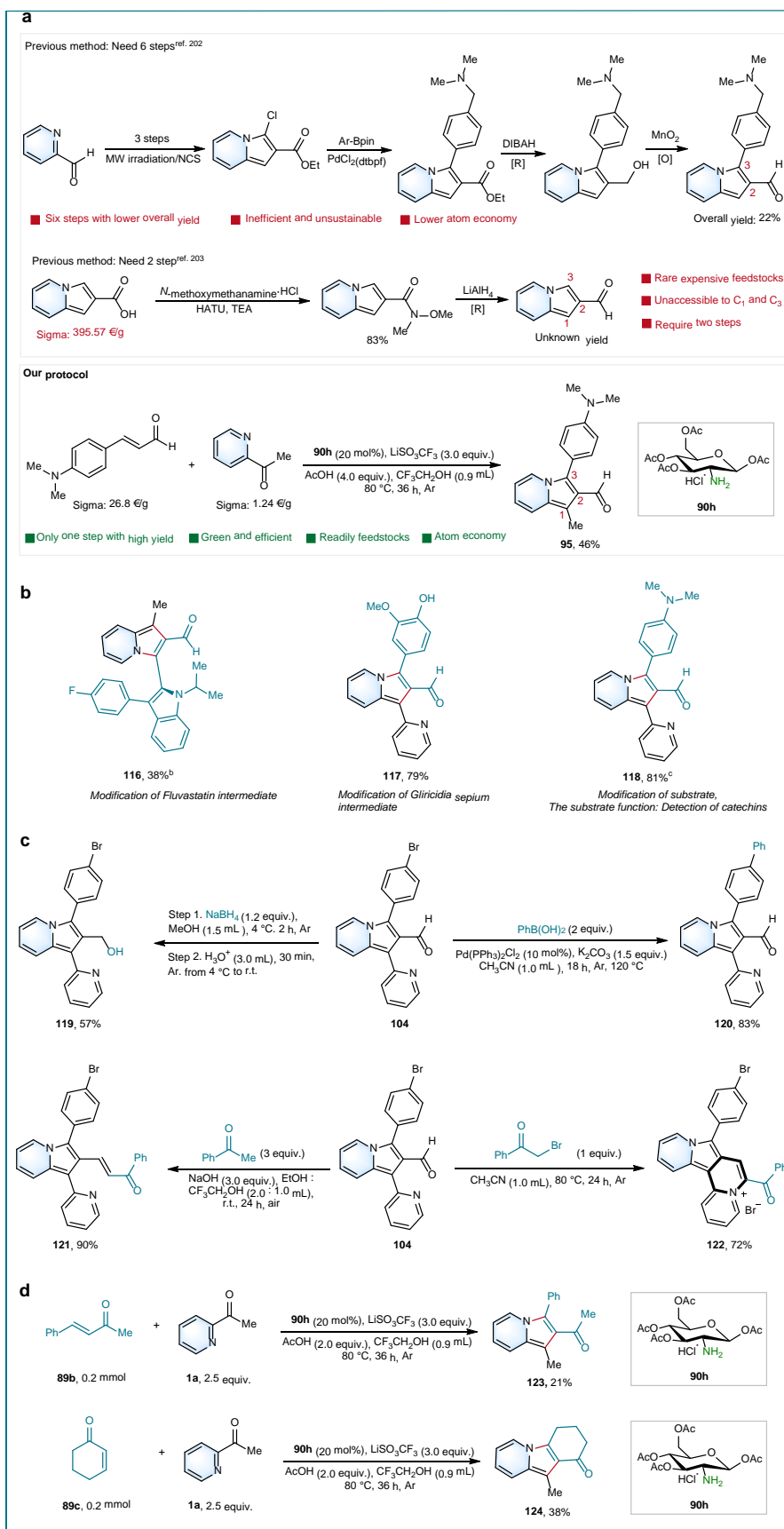


Figure 45 Synthetic applications. **(a)** Comparison previous protocols with our strategy for indolizine-2-carbaldehyde. **(b)** Late-stage selective modifications of bioactive molecules and drugs. **(c)** Late-stage transformation applications. **(d)** Preliminary expansion of the anomeric stereoauxiliary aminocatalysis strategy beyond α,β -unsaturated aldehyde to enone. ^aYields are those of isolated products. ^bReaction for 42 h in AcOH : CF₃CH₂OH (0.45 : 0.45 mL). ^cReaction for 42 h in AcOH : CF₃CH₂OH (0.4 : 0.5 mL).

4. General conclusions and perspectives

Chitin/chitosan/glucosamine has been increasingly used as a sustainable, environmentally-benign and economically-attractive *N*-containing biomass for organic synthesis of value-added *N*-heterocyclic products. The work in this thesis focuses on developing sustainable strategies for the preparation of value-added imidazo[1,5-*a*]pyridines and indolizines *via* cyclization reactions from the *N*-containing biomass.

In the first project, we developed a novel α -anomeric stereoauxiliary strategy for the facile preparation of a broad range of imidazo[1,5-*a*]pyridines, which features the cleavage of C–N bonds of D-glucosamine through a seven-membered ring transition state intermediate and the simultaneous incorporation of amine moieties into valuable imidazo[1,5-*a*]pyridines (**Figure 46**). This method unlocks efficient access to diverse imidazo[1,5-*a*]pyridine derivatives bearing sensitive functional groups that are inaccessible with conventional approaches. Various control experiments and DFT calculations revealed that the hydroxyl group of α -anomer promoted the formation of a seven-membered ring transition state with the acetate anion *via* hydrogen bonds. The ring structure in the α -anomer transition state (TS $_{\alpha}$) not only helped to stabilize the intermediate during the C–N bond cleavage, but also profited from the dispersion interactions brought by the neighboring aromatic ring. Given the importance of imidazo[1,5-*a*]pyridines and C–N bonds cleavage of aliphatic amine, we believe that this approach with combined anomer-assisting C–N bonds cleavage by using native stereochemistry of D-glucosamine and the synthesis of imidazo[1,5-*a*]pyridines will be of significant and general interest for many fields, and opens a new window for chemical synthesis.

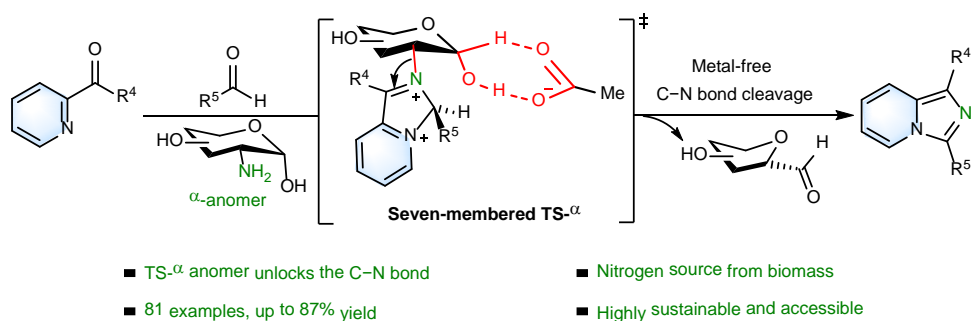


Figure 46 Anomeric stereoauxiliary strategy enables efficient synthesis of wide-ranging imidazo[1,5-*a*]pyridines

In the second project, we have developed a new strategy for the rapid assembly of NCCs with an ample scope through the direct use of native nitrogen sources from biomass (**Figure 47**). This strategy is achieved through a one-pot conversion approach of chitin/chitosan by cleaving the C–N bonds and simultaneously integrating the nitrogen in the synthesis of a broad range of imidazo[1,5-*a*]pyridines (52 examples) that show diverse potential applications. A broad group of previously inaccessible products including saturated 1-alkylimidazo[1,5-*a*]pyridines is unprecedentedly synthesized by this protocol. The amine groups of chitin/chitosan backbone are intercepted *via* aldehyde and pyridine ketone assisted with CF₃COOH under metal-free conditions. We believe this approach will initiate research endeavors for the targeted efficient incorporation of nitrogen from biomass for high-value NCCs.

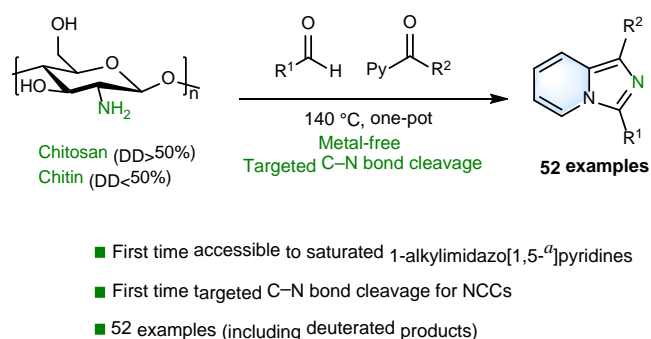


Figure 47 Direct nitrogen interception from chitin/chitosan for imidazo[1,5-*a*]pyridines.

In the third project, we developed a novel β -anomeric stereoauxiliary glucosamine-catalyzed strategy that first time allows one-pot site-selective C₂-aldehylation for efficient preparation of 1,2,3-trisubstituted indolizine-2-carbaldehydes (**Figure 48**). Utilization of this aminocatalysis strategy with native chiral skeletons for the highly regioselective control, this one-pot approach expands the scope of readily accessible trisubstituted indolizine-2-carbaldehydes relative to existing state-of-the-art methods by multi-steps to introduce carbaldehyde only for mono/di-substituted indolizine-2-carbaldehyde. This method not only enable the efficient C₂-aldehylation of a range of commercial α,β -unsaturated aldehydes and bioactive molecules or drugs, it has been expanded beyond α,β -unsaturated aldehyde to enone. Overall, our anomeric stereoauxiliary catalytic system provides a promising solution towards addressing the challenge associated with indolizine formation with site-selective C₂ aldehylation, on which ongoing work is targeted to apply this strategy towards developing a wider range of catalytic applications.

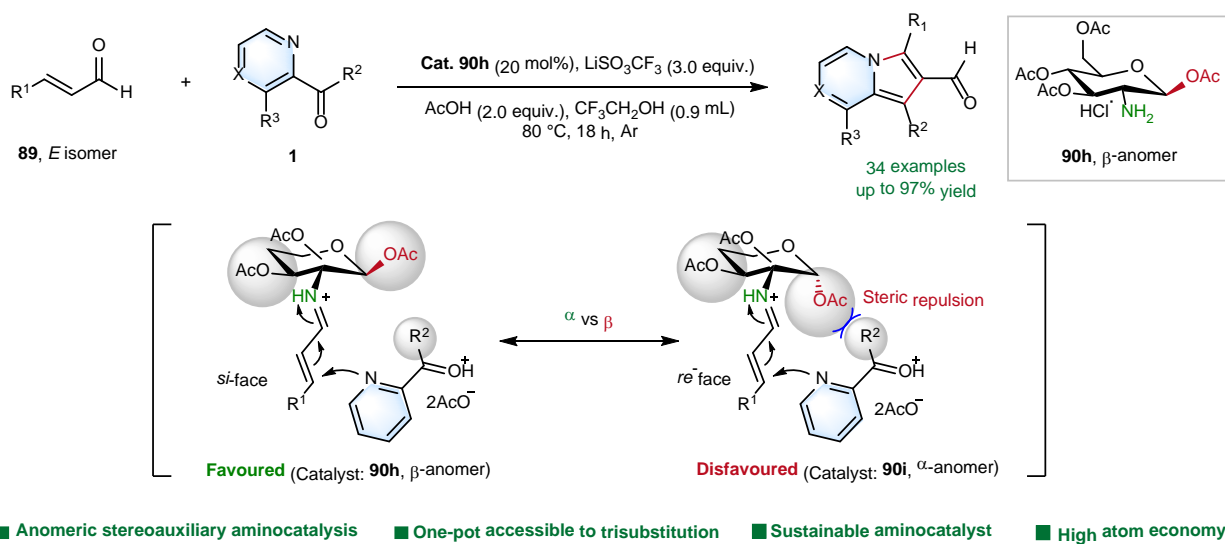


Figure 48 Anomeric stereoauxiliary-organocatalyzed one-pot site-selective C₂-aldehylation for trisubstituted indolizine-2-carbaldehydes

5. Experimental section

5.1 General Remarks

NMR spectra were recorded on Inova 500 or Bruker Avance III 300, Avance III 400 and Avance III HD 500 in the solvent indicated. Chemical shifts are provided in ppm and spectra refer to non-deuterated solvent signal. Yields refer to isolated compounds estimated to be >95% pure as determined by ¹H-NMR. Chromatography was carried out on silica gel (40–63 μm). ESI mass spectra were recorded on Bruker Daltonic spectrometers maXis (ESI-QTOF-MS) and micrOTOF (ESI-TOF-MS). GC-MS spectra was performed on an Agilent Technologies chromatograph 7820A GC System and Agilent Technologies 5977E system. GC calibrations were carried out with authentic samples and n-dodecane as an internal standard. FT-IR spectra were recorded on Alpha FT-IR Spectrometer (Bruker, Germany) at room temperature. All samples were measured between 4000 and 500 cm⁻¹ with a resolution of 4 cm⁻¹ using Platinum ATR and accumulated 24 scans. Melt point were recorded on melting point apparatus, Electrothermal IA 9200. The following starting chemicals were synthesized according to previously described methods: **1d-1k**.³²⁴ Unless otherwise specified, the chemicals were obtained commercially and used without further purification. All reactions were performed under an atmosphere of Ar unless specified otherwise.

5.2 General Procedures

General procedure A: Preparation of pyridines ketones substrates

The following starting chemicals were synthesized according to previously described methods: **1d-1k**.³²⁴ A solution of dry THF (15 mL) with bromobenzene (10.0 mmol, 1.00 equiv.) was dropwised into magnesium (12 mmol, 1.2 equiv.) and the mixture solution was stirred in Ar gas in room temperature. After the formation of the Grignard reagent (the color changed to gray), then stopped it. At the same time, picolinonitrile (8 mmol, 0.8 equiv.) was dissolved in dry THF (10 mL), which was dropwised into the former mixture solution of Grignard reagent at 0 °C. After the reaction completely, it was quenched by saturated NH₄Cl aqueous solution. The organic layer was separated and extracted twice by CH₂Cl₂. After evaporation, the organic layer was re-dissolved in Et₂O (30 mL) and 6 M HCl (6 mL) was added into the solution. After 30 min, the organic layer was separated. The aqueous layer was basified by saturated NaHCO₃ aqueous solution and then extracted three times by CH₂Cl₂. The combined organic layers were dried over Na₂SO₄ and evaporated in rotary evaporator. The residue was purified by column chromatography with *n*-hexane and ethyl acetate to afford 1b. Other pyridine ketones 1c–1i were prepared with the similar procedures, and characterized by NMR analysis.

General procedure B: Anomeric stereoauxiliary cleavage of the C–N bond of D-glucosamine for the preparation of imidazo[1,5-a]pyridines

General procedure B1

A mixture of pyridine ketone (0.1 mmol), aldehydes (2 equiv.) and D-glucosamine·HCl (2.0 equiv.) in AcOH : H₂O (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h.

General procedure B2

A mixture of pyridine ketone (0.1 mmol), aldehydes (4 equiv.) and D-glucosamine·HCl (2.0 equiv.) in AcOH : H₂O (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h.

Workup: The reaction was conducted in a sealed Schlenk flask and heated by an IKA magnetic heating agitator with oil bath. The reaction temperature was directly read from temperature detector of IKA apparatus and was calibrated by thermometer. After cooling to room temperature, the reaction mixture was concentrated with rotary evaporator, the crude product was purified with flash chromatography on silica gel (ethyl acetate : *n*-hexane : Et₃N) to give products **4-62**. The deuterated products **66-86** were synthesized through an analogous procedure to method B1 using AcOH-d₄ : D₂O (0.9 mL : 0.1 mL) as solvent. More experimental details and characterization are available in the Supporting Information.

General procedure C: Direct nitrogen interception from chitin/chitosan for imidazo[1,5-a]pyridines

General procedure C1

A mixture of 2-acetylpyridines (0.1 mmol), aldehydes (2.0 equiv.) and chitosan (3.0 equiv.) in CF₃COOH (1.0 mL) was stirred at 140 °C under Ar atmosphere for 36 h. Chitosan was dried at 100 °C overnight.

General procedure C2

A mixture of pyridine ketone (0.1 mmol), 2-methylbenzaldehyde (4.0 equiv.) and chitosan (2.5 equiv.) in CF₃COOH (0.7 mL) was stirred at 140 °C under Ar atmosphere for 36 h. Chitosan was dried at 100 °C overnight.

General procedure C3

A mixture of pyridine ketone (0.1 mmol), aldehydes (2.0 equiv.) and chitin (3.0 equiv.) in CF₃COOH (1.0 mL) was stirred at 140 °C under Ar atmosphere for 36 h. Chitin was dried at 100 °C overnight.

General procedure C4

A mixture of pyridine ketone (0.1 mmol), aldehydes (4.0 equiv.) and chitin (2.5 equiv.) in CF₃COOH (0.7 mL) was stirred at 140 °C under Ar atmosphere for 36 h. Chitin was dried at 100 °C overnight.

General procedure C5

A mixture of 2-acetylpyridines (0.1 mmol), aldehydes (2.0 equiv.) and chitosan (3.0 equiv.) in CF₃COOD (1.0 mL) was stirred at 140 °C under Ar atmosphere for 36 h. Chitosan was dried under 100 °C oven overnight.

Workup: The reaction temperature was directly read from temperature detector of IKA apparatus and was calibrated by a thermometer. After cooling to room temperature, the reaction mixture was basified up to pH = 7 *via* std. Na₂CO₃ aqueous solution and then extracted by Et₂O (3×3.0 mL) and dried over anhydrous Na₂SO₄. After filtration and concentration in a rotary evaporator, the crude product was purified with flash chromatography on silica gel (EtOAc: *n*-hexane: Et₃N) to give the desired products.

General procedure D: Anomeric stereoauxiliary-organocatalyzed one-pot site-selective C₂-aldehylation for trisubstituted indolizine-2-carbaldehydes

General procedure D1

A mixture of α,β -unsaturated aldehydes or ketones (0.2 mmol), heteroaryl ketones (2.5 equiv.), catalyst **3b** (0.04 mmol), AcOH (2.0 equiv.) and LiSO₃CF₃ (3.0 equiv.) in the CF₃CH₂OH (0.9 mL) were stirred at 80 °C under Ar atmosphere for 18 h.

General procedure D2

A mixture of α,β -unsaturated aldehydes (0.2 mmol), heteroaryl ketones (2.5 equiv.), catalyst **3b** (0.04 mmol) and LiSO₃CF₃ (3.0 equiv.) in the CF₃CH₂OH : AcOH (0.5 : 0.4 mL) were stirred at 80 °C under Ar atmosphere for 36 h.

General procedure D3

A mixture of α,β -unsaturated aldehydes (0.2 mmol), heteroaryl ketones (2.5 equiv.), catalyst **3b** (0.04 mmol), AcOH (4.0 equiv.) and LiSO₃CF₃ (3.0 equiv.) in the CF₃CH₂OH (0.9 mL) were stirred at room temperature under Ar atmosphere for 42 h.

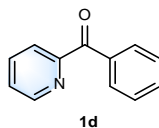
General procedure **D4**

A mixture of α,β -unsaturated aldehydes (0.2 mmol), heteroaryl ketones (2.5 equiv.), catalyst **3b** (0.04 mmol), AcOH (2.0 equiv.) and LiSO_3CF_3 (3.0 equiv.) in the $\text{CF}_3\text{CH}_2\text{OH}$ (0.9 mL) were stirred at 80 °C under Ar atmosphere for 36 h.

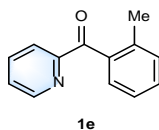
Workup: The reactions were conducted in a sealed Schlenk tube and heated by an IKA magnetic heating agitator with heating block. The reaction temperature was directly read from temperature detector of IKA apparatus and was calibrated by thermometer. After cooling to room temperature, the reaction mixture was basified up to pH 7 *via* Na_2CO_3 aqueous solution, then extracted by diether (3×3 mL) and dried over anhydrous Na_2SO_4 . After filtration and concentrated in rotary evaporator, the crude product was purified with flash chromatography on silica gel (ethyl acetate : *n*-hexane) to give products.

5.3 Characterization Data

5.3.1 Preparation of pyridines ketones substrates

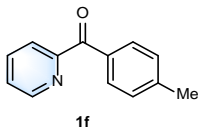


phenyl(pyridin-2-yl)methanone (**1d**): Following the **General procedure A**, work-up gave product **1d** (1189.5 mg, 6.5 mmol, isolated yield 65%) as a yellow oil. ^1H NMR (400 MHz, CDCl_3): δ 8.61 (ddd, J = 4.8, 1.6, 0.8 Hz, 1H), 7.98 – 7.95 (m, 2H), 7.92 (d, J = 8.0 Hz, 1H), 7.77 (td, J = 8.0, 1.6 Hz, 1H), 7.50 – 7.46 (m, 1H), 7.39 – 7.34 (m, 3H). ^{13}C NMR (100 MHz, CDCl_3): δ 193.64, 154.9, 148.4, 136.9, 136.1, 132.7, 130.8, 128.0, 126.0, 124.4. The compound is known, and the NMR data is in accordance with the previous literature.³²⁵

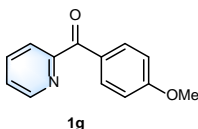


pyridin-2-yl(o-tolyl)methanone (**1e**): Following the **General procedure A**, work-up gave product **1e** (1004.7 mg, 5.1 mmol, isolated yield 51%) as a yellow oil. ^1H NMR (400 MHz, CDCl_3): δ 8.61 (d, J = 5.2 Hz, 1H), 8.00 (d, J = 8.0 Hz, 1H), 7.80 (t, J = 7.6 Hz, 1H), 7.39 – 7.30 (m, 3H), 7.21 – 7.15 (m, 2H), 2.30 (s, 3H). ^{13}C NMR (100 MHz, CDCl_3): δ 197.4, 155.0, 149.1, 137.8, 137.3, 136.9, 131.1, 130.9,

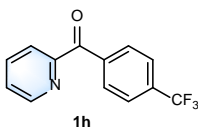
129.9, 126.4, 125.0, 124.1, 20.4. The compound is known, and the NMR data is in accordance with the previous literature.³²⁶



pyridin-2-yl(p-tolyl)methanone (1f): Following the **General procedure A**, work-up gave product **1f** (1241.1 mg, 6.3 mmol, isolated yield 63%) as a brown oil. **¹H NMR** (400 MHz, CDCl₃): δ 8.61 (d, *J* = 4.4 Hz, 1H), 7.91 – 7.87 (m, 3H), 7.79 – 7.75 (m, 1H), 7.38 – 7.34 (m, 1H), 7.18 (d, *J* = 8.0 Hz, 2H), 2.32 (s, 3H). **¹³C NMR** (100 MHz, CDCl₃): δ 193.4, 155.3, 148.4, 143.7, 136.9, 133.5, 131.0, 128.8, 125.9, 124.4, 21.6. The compound is known, and the NMR data is in accordance with the previous literature.³²⁵

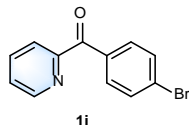


(4-methoxyphenyl)(pyridin-2-yl)methanone (1g): Following the **General procedure A**, work-up gave product **1g** (1597.5 mg, 7.5 mmol, isolated yield 75%) as a white solid. **¹H NMR** (400 MHz, CDCl₃): δ 8.64 (d, *J* = 4.4 Hz, 1H), 8.05 (d, *J* = 7.2 Hz, 2H), 7.92 (d, *J* = 8.0 Hz, 1H), 7.81 (t, *J* = 7.6 Hz, 1H), 7.41 – 7.38 (m, 1H), 6.90 (d, *J* = 8.8 Hz, 2H), 3.81 (s, 3H). **¹³C NMR** (100 MHz, CDCl₃): δ 192.3, 163.5, 155.8, 148.3, 137.0, 133.5, 129.0, 125.8, 124.5, 113.5, 55.5. The compound is known, and the NMR data is in accordance with the previous literature.³²⁵

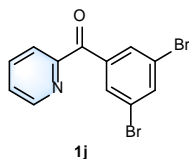


pyridin-2-yl(4-(trifluoromethyl)phenyl)methanone (1h): Following the **General procedure A**, work-up gave product **1h** (1129.5 mg, 4.5 mmol, isolated yield 45%) as a yellow oil. **¹H NMR** (400 MHz, CDCl₃): δ 8.64 – 8.62 (m, 1H), 8.09 (d, *J* = 8.0 Hz, 2H), 8.03 (d, *J* = 8.0 Hz, 1H), 7.84 (td, *J* = 8.0, 1.6 Hz, 1H), 7.65 (d, *J* = 8.0 Hz, 2H), 7.43 (ddd, *J* = 7.6 Hz, 4.8, 1.2 Hz, 1H). **¹³C NMR** (100 MHz, CDCl₃): δ 192.7, 154.1, 148.6, 139.3 (q, ⁴*J*_{C-F} = 1.2 Hz), 137.25, 133.85 (q, ²*J*_{C-F} = 32.6 Hz), 131.2, 126.7, 125.0 (q,

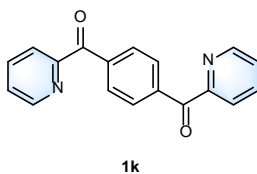
$^3J_{\text{C-F}} = 3.8$ Hz), 124.7, 123.7 (q, $^1J_{\text{C-F}} = 270.9$ Hz), 115.4. ^{19}F NMR (375 MHz, CDCl_3): δ -63.13. The compound is known, and the NMR data is in accordance with the previous literature.³²⁷



pyridin-2-yl(4-(trifluoromethyl)phenyl)methanone (1i): Following the **General procedure A**, work-up gave product **1i** (1357.2 mg, 5.2 mmol, isolated yield 52%) as a pale yellow solid. ^1H NMR (400 MHz, CDCl_3): δ 8.61 (d, $J = 5.5$ Hz, 1H), 7.98 – 7.88 (m, 3H), 7.83 – 7.79 (m, 1H), 7.53 (d, $J = 8.6$ Hz, 2H), 7.41 – 7.38 (m, 1H). ^{13}C NMR (100 MHz, CDCl_3): δ 192.5, 154.5, 148.4, 137.1, 134.9, 132.5, 131.3, 128.1, 126.4, 124.6. **ESI-HRMS**: m/z calcd. for $\text{C}_{12}\text{H}_8\text{BrNO}$ $[\text{M}+\text{H}]^+$: 261.9868, found 263.9846. The compound is known, and the NMR data is in accordance with the previous literature.³²⁸



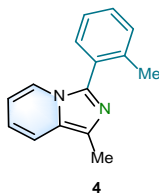
(3,5-dibromophenyl)(pyridin-2-yl)methanone (1j): Following the **General procedure A**, work-up gave product **1j** (1186.2 mg, 3.5 mmol, isolated yield 35%) as a white green solid. ^1H NMR (400 MHz, CDCl_3): δ 8.67 – 8.66 (m, 1H), 8.11 (s, 2H), 8.02 (d, $J = 8.0$ Hz, 1H), 7.87 – 7.83 (m, 1H), 7.80 (s, 1H), 7.48 – 7.44 (m, 1H). ^{13}C NMR (100 MHz, CDCl_3): δ 190.6, 153.7, 148.7, 139.2, 137.8, 137.3, 132.6, 126.9, 124.8, 122.7. **ESI-HRMS**: m/z calcd. for $\text{C}_{12}\text{H}_7\text{Br}_2\text{NO}$ $[\text{M}+\text{H}]^+$: 339.8973, found 339.8969.



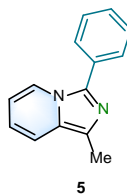
1,4-phenylenebis(pyridin-2-ylmethanone) (1k): Following the **General procedure A**, work-up gave product **1k** (316.8 mg, 1.1 mmol, isolated yield 11%) as a yellow solid. ^1H NMR (400 MHz, CDCl_3): δ 8.67 – 8.65 (m, 2H), 8.09 (d, $J = 0.8$ Hz, 4H), 8.02 (dd, $J = 7.6, 1.2$ Hz, 2H), 7.88 – 7.83 (m, 2H), 7.46 – 7.43 (m, 2H). ^{13}C NMR (100 MHz, CDCl_3): δ 193.3, 154.4, 148.6, 139.4, 137.1, 130.5, 126.5, 124.6.

ESI-HRMS: m/z calcd. for $C_{18}H_{12}N_2O_2$ $[M+H]^+$: 289.0977, found 289.0972. The compound is known, and the NMR data is in accordance with the previous literature.³²⁸

5.3.2 Anomeric stereoauxiliary strategy enables efficient synthesis of wide-ranging imidazo[1,5-a]pyridines

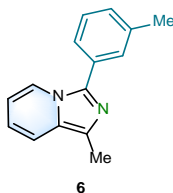


Methyl-3-(o-tolyl)imidazo[1,5-a]pyridine: Following the **General procedure B1**, a mixture of 2-acetylpyridine (0.1 mmol), 2-methylbenzaldehyde (2 equiv.) and D-glucosamine·HCl (2.0 equiv.) in AcOH : H₂O (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up gave product **4** (14.4 mg, 0.065 mmol, isolated yield 65%) as a yellow liquid. **FT-IR:** ν (cm⁻¹): 3066, 2924, 2852, 1671, 1630, 1605, 1554, 1515, 1451, 1412, 1362, 1319, 1247, 1140, 1107, 1072, 1037, 1006, 940, 865, 808, 773, 734, 694, 657, 596, 575, 546, 492, 447, 418. **¹H NMR** (300 MHz, CDCl₃): δ 7.41 (d, J = 7.2 Hz, 1H), 7.35 – 7.17 (m, 5H), 6.53 – 6.48 (m, 1H), 6.35 – 6.30 (m, 1H), 2.49 (s, 3H), 2.12 (s, 3H). **¹³C NMR** (100 MHz, CDCl₃): δ 138.1, 136.2, 130.6, 130.4, 129.5, 129.1, 127.9, 126.7, 125.9, 121.2, 118.0, 116.6, 112.2, 19.6, 12.6. **ESI-HRMS:** m/z calcd. for $C_{15}H_{14}N_2$ $[M+H]^+$: 223.1235, found 223.1230.

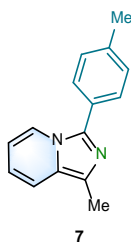


1-methyl-3-phenylimidazo[1,5-a]pyridine: Following the **General procedure B1**, a mixture of 2-acetylpyridine (0.1 mmol), benzaldehyde (2 equiv.) and D-glucosamine·HCl (2.0 equiv.) in AcOH : H₂O (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up gave product **5** (17.3 mg, 0.083 mmol, isolated yield 83%) as a yellow liquid. **FT-IR:** ν (cm⁻¹): 3056, 2918, 2850, 1659, 1603, 1572, 1517, 1451, 1409, 1362, 1257, 1076, 1024, 1004, 942, 915, 775, 738, 688, 616, 480, 420. **¹H NMR** (400 MHz, CDCl₃): δ 8.09 (d, J = 7.2 Hz, 1H), 7.68 (d, J = 7.6 Hz, 2H), 7.41 (t, J = 7.6 Hz, 2H), 7.33 – 7.30 (m, 2H), 6.56 – 6.52 (m, 1H), 6.41 (t, J = 6.8 Hz, 1H), 2.48 (s, 3H). **¹³C NMR** (100 MHz, CDCl₃): δ

136.5, 130.3, 128.9, 128.4, 127.9, 127.7, 121.1, 118.3, 117.0, 112.9, 12.5. The compound is known, and the NMR data is in accordance with the previous literature.³²⁹

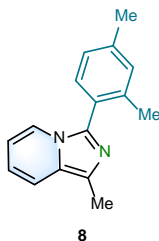


1-methyl-3-(m-tolyl)imidazo[1,5-a]pyridine: Following the **General procedure B1**, a mixture of 2-acetylpyridine (0.1 mmol), 3-methylbenzaldehyde (2 equiv.) and D-glucosamine·HCl (2.0 equiv.) in AcOH : H₂O (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up gave product **6** (16.0 mg, 0.072 mmol, isolated yield 72%) as a yellow solid. **Mp**: 105–107 °C. **FT-IR**: ν (cm⁻¹): 2963, 2920, 2852, 1659, 1607, 1583, 1519, 1463, 1432, 1366, 1261, 1201, 1094, 1026, 913, 882, 793, 738, 723, 696, 618, 548, 521, 439, 420. **¹H NMR** (400 MHz, CDCl₃): δ 8.10 (d, J = 7.2 Hz, 1H), 7.52 (s, 1H), 7.47 (d, J = 7.6 Hz, 1H), 7.31 (d, J = 8.4 Hz, 2H), 7.14 (d, J = 7.6 Hz, 1H), 6.56 – 6.52 (m, 1H), 6.41 (t, J = 6.8 Hz, 1H), 2.48 (s, 3H), 2.34 (s, 3H). **¹³C NMR** (100 MHz, CDCl₃): δ 138.8, 136.7, 132.7, 130.2, 129.2, 128.7, 128.6, 127.9, 124.5, 121.2, 118.3, 117.0, 112.8, 21.4, 12.5. **ESI-HRMS**: m/z calcd. for C₁₅H₁₄N₂ [M+H]⁺: 223.1235, found 223.1231.

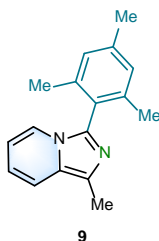


1-methyl-3-(p-tolyl)imidazo[1,5-a]pyridine: Following the **General procedure B1**, a mixture of 2-acetylpyridine (0.1 mmol), 4-methylbenzaldehyde (2 equiv.) and D-glucosamine·HCl (2.0 equiv.) in AcOH : H₂O (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up gave product **7** (17.3 mg, 0.078 mmol, isolated yield 78%) as a yellow solid. **Mp**: 62–64 °C. **FT-IR**: ν (cm⁻¹): 2914, 2852, 1661, 1611, 1570, 1529, 1492, 1405, 1360, 1259, 1183, 1111, 1076, 1034, 1020, 946, 820, 732, 690, 579, 492, 427. **¹H NMR** (400 MHz, CDCl₃): δ 8.06 (d, J = 7.3 Hz, 1H), 7.58 (d, J = 8.1 Hz, 2H), 7.30 (d, J = 9.1 Hz, 1H), 7.23 (d, J = 7.9 Hz, 2H), 6.52 (dd, J = 9.0, 6.3 Hz, 1H), 6.40 – 6.37 (m, 1H), 2.48 (s, 3H),

2.34 (s, 3H). ^{13}C NMR (100 MHz, CDCl_3): δ 138.3, 136.7, 129.6, 128.6, 127.7, 127.6, 127.5, 121.2, 118.3, 116.8, 112.6, 21.4, 12.5. **ESI-HRMS**: m/z calcd. for $\text{C}_{15}\text{H}_{14}\text{N}_2$ $[\text{M}+\text{H}]^+$: 223.1235, found 223.1234.

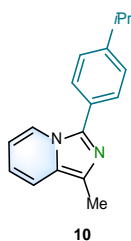


3-(2,4-dimethylphenyl)-1-methylimidazo[1,5-a]pyridine: Following the **General procedure B1**, a mixture of 2-acetylpyridine (0.1 mmol), 2,3-dimethylbenzaldehyde (2 equiv.) and D-glucosamine·HCl (2.0 equiv.) in $\text{AcOH} : \text{H}_2\text{O}$ (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up gave product **8** (16.0 mg, 0.068 mmol, isolated yield 68%) as a brown solid. **Mp**: 81–83 °C. **FT-IR**: ν (cm^{-1}): 2920, 2856, 1663, 1616, 1447, 1414, 1362, 1234, 1152, 1109, 1072, 1032, 1001, 950, 878, 818, 740, 696, 591, 546, 424. ^1H NMR (400 MHz, CDCl_3): δ 7.41 (d, J = 6.8 Hz, 1H), 7.31 (d, J = 9.2 Hz, 1H), 7.22 – 7.19 (m, 1H), 7.08 (s, 1H), 7.02 (d, J = 7.6 Hz, 1H), 6.55 – 6.51 (m, 1H), 6.36 – 6.33 (m, 1H), 2.48 (s, 3H), 2.31 (s, 3H), 2.08 (s, 3H). ^{13}C NMR (100 MHz, CDCl_3): δ 139.1, 137.9, 136.3, 132.0, 131.4, 130.4, 127.6, 126.7, 126.4, 121.3, 118.1, 116.7, 112.2, 21.3, 19.6, 12.5. **ESI-HRMS**: m/z calcd. for $\text{C}_{16}\text{H}_{16}\text{N}_2$ $[\text{M}+\text{H}]^+$: 237.1392, found 237.1386.

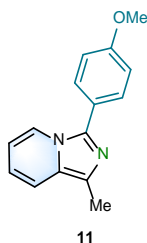


3-mesityl-1-methylimidazo[1,5-a]pyridine: Following the **General procedure B1**, a mixture of 2-acetylpyridine (0.1 mmol), 2,4,6-trimethylbenzaldehyde (2 equiv.) and D-glucosamine·HCl (2.0 equiv.) in $\text{AcOH} : \text{H}_2\text{O}$ (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up gave product **9** (14.0 mg, 0.056 mmol, isolated yield 56%) as a yellow liquid. **FT-IR**: ν (cm^{-1}): 2924, 2852, 1669, 1630, 1611, 1574, 1550, 1519, 1459, 1442, 1412, 1377, 1360, 1319, 1189, 1119, 1072, 1032, 1001, 925, 849, 773, 738, 699, 596, 560, 544, 422. ^1H NMR (400 MHz, CDCl_3): δ 7.31 (d, J = 9.2 Hz, 1H),

7.14 – 7.12 (m, 1H), 6.88 (s, 2H), 6.52 – 6.48 (m, 1H), 6.32 – 6.28 (m, 1H), 2.50 (s, 3H), 2.27 (s, 3H), 1.88 (s, 6H). ¹³C NMR (100 MHz, CDCl₃): δ 139.1, 139.1, 135.5, 128.3, 127.5, 126.2, 126.2, 120.9, 118.0, 116.3, 112.0, 21.2, 19.5, 12.7. **ESI-HRMS**: m/z calcd. for C₁₇H₁₈N₂ [M+H]⁺: 251.1548, found 251.1542.

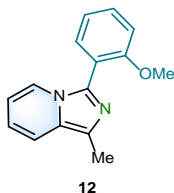


3-(4-isopropylphenyl)-1-methylimidazo[1,5-a]pyridine: Following the **General procedure B1**, a mixture of 2-acetylpyridine (0.1 mmol), 4-isopropylbenzaldehyde (2 equiv.) and D-glucosamine·HCl (2.0 equiv.) in AcOH : H₂O (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up gave product **10** (17.8 mg, 0.071 mmol, isolated yield 71%) as a brown solid. **Mp**: 61–63 °C. **FT-IR**: ν (cm⁻¹): 2961, 2922, 2864, 1671, 1607, 1527, 1457, 1414, 1383, 1362, 1306, 1255, 1199, 1175, 1127, 1076, 1053, 1018, 948, 835, 797, 777, 736, 717, 690, 637, 593, 532, 414. ¹H NMR (400 MHz, CDCl₃): δ 8.09 (d, *J* = 7.6 Hz, 1H), 7.63 – 7.60 (m, 2H), 7.32 – 7.27 (m, 3H), 6.55 – 6.51 (m, 1H), 6.41 – 6.37 (m, 1H), 2.93 – 2.86 (m, 1H), 2.48 (s, 3H), 1.22 (dd, *J* = 6.8, 1.2 Hz, 6H). ¹³C NMR (75 MHz, CDCl₃): δ 149.4, 136.5, 128.3, 127.8, 127.7, 127.5, 127.0, 121.2, 118.3, 117.0, 112.8, 34.0, 23.9, 12.3. **ESI-HRMS**: m/z calcd. for C₁₇H₁₈N₂ [M+H]⁺: 251.1548, found 251.1542.

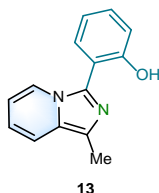


3-(4-methoxyphenyl)-1-methylimidazo[1,5-a]pyridine: Following the **General procedure B1**, a mixture of 2-acetylpyridine (0.1 mmol), 4-methoxybenzaldehyde (2 equiv.) and D-glucosamine·HCl (2.0 equiv.) in AcOH : H₂O (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up gave product **11** (18.6 mg, 0.078 mmol, isolated yield 78%) as a yellow solid. **Mp**: 61–63 °C. **FT-IR**: ν (cm⁻¹):

2961, 2922, 2838, 1661, 1605, 1498, 1459, 1362, 1300, 1251, 1173, 1107, 1022, 835, 797, 734, 690, 581, 515. **¹H NMR** (400 MHz, CDCl₃): δ 8.00 (d, *J* = 7.2 Hz, 1H), 7.63 – 7.59 (m, 2H), 7.29 – 7.26 (m, 1H), 6.96 – 6.93 (m, 2H), 6.51 – 6.47 (m, 1H), 6.38 – 6.34 (m, 1H), 3.78 (s, 3H), 2.48 (s, 3H). **¹³C NMR** (100 MHz, CDCl₃): δ 159.7, 136.5, 129.2, 128.3, 127.5, 122.9, 121.0, 118.3, 116.6, 114.3, 112.6, 55.3, 12.4. **ESI-HRMS**: *m/z* calcd. for C₁₅H₁₄N₂O [M+H]⁺: 239.1184, found 239.1181.

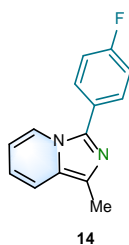


3-(2-methoxyphenyl)-1-methylimidazo[1,5-a]pyridine: Following the **General procedure B1**, a mixture of 2-acetylpyridine (0.1 mmol), 2-methoxybenzaldehyde (2 equiv.) and D-glucosamine·HCl (2.0 equiv.) in AcOH : H₂O (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up gave product **12** (17.6 mg, 0.074 mmol, isolated yield 74%) as a yellow solid. **Mp**: 55–57 °C. **FT-IR**: *ν* (cm⁻¹): 2916, 2848, 1735, 1663, 1601, 1581, 1512, 1467, 1436, 1418, 1362, 1238, 1179, 1160, 1109, 1045, 1018, 940, 797, 732, 690, 651, 583, 565, 534, 490, 449, 424. **¹H NMR** (400 MHz, CDCl₃): δ 7.51 (dd, *J* = 7.6, 1.8 Hz, 1H), 7.42 – 7.40 (m, 1H), 7.36 – 7.27 (m, 2H), 7.01 – 6.97 (m, 1H), 6.92 (d, *J* = 8.00 Hz, 1H), 6.54 – 6.50 (m, 1H), 6.35 – 6.31 (m, 1H), 3.70 (s, 3H), 2.49 (s, 3H). **¹³C NMR** (100 MHz, CDCl₃): δ 157.1, 134.3, 132.3, 130.3, 128.1, 127.5, 122.8, 121.0, 119.3, 117.6, 116.6, 111.4, 111.0, 55.4, 12.5. **ESI-HRMS**: *m/z* calcd. for C₁₅H₁₄N₂O [M+H]⁺: 239.0402, found 239.0397.

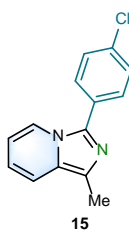


2-(1-methylimidazo[1,5-a]pyridin-3-yl)phenol: Following the **General procedure B1**, a mixture of 2-acetylpyridine (0.1 mmol), 2-hydroxybenzaldehyde (2 equiv.) and D-glucosamine·HCl (2.0 equiv.) in AcOH : H₂O (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up gave product **13** (14.6 mg, 0.065 mmol, isolated yield 65%) as a white solid. **Mp**: 137–138 °C. **FT-IR**: *ν* (cm⁻¹): 2922, 2856, 1735, 1653, 1609, 1583, 1519, 1469, 1436, 1381, 1364, 1290, 1253, 1236, 1181, 1156, 1102, 1076, 1037, 1012, 933, 812, 752, 734, 690, 655, 596, 577, 548, 534, 455, 420. **¹H NMR** (400 MHz, CDCl₃): δ

8.35 (d, $J = 7.2$ Hz, 1H), 7.67 – 7.64 (m, 1H), 7.37 – 7.34 (m, 1H), 7.20 – 7.17 (m, 1H), 7.08 – 7.05 (m, 1H), 6.92 – 6.88 (m, 1H), 6.64 – 6.60 (m, 1H), 6.54 – 6.50 (m, 1H), 2.48 (s, 3H). ^{13}C NMR (100 MHz, CDCl_3): δ 156.3, 134.3, 129.4, 127.2, 127.0, 123.8, 121.9, 118.8, 118.6, 117.7, 117.7, 114.4, 113.7, 12.3. **ESI-HRMS**: m/z calcd. for $\text{C}_{14}\text{H}_{12}\text{N}_2\text{O}_1$ $[\text{M}+\text{H}]^+$: 225.1028, found 225.1028.

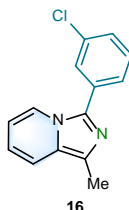


3-(4-fluorophenyl)-1-methylimidazo[1,5-a]pyridine: Following the **General procedure B1**, a mixture of 2-acetylpyridine (0.1 mmol), 4-fluorobenzaldehyde (2 equiv.) and D-glucosamine·HCl (2.0 equiv.) in $\text{AcOH} : \text{H}_2\text{O}$ (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up gave product **14** (18.1 mg, 0.080 mmol, isolated yield 80%) as a yellow solid. **Mp**: 80–82 °C. **FT-IR**: ν (cm^{-1}): 2963, 2920, 2852, 1665, 1603, 1496, 1412, 1368, 1257, 1222, 1156, 1092, 1012, 837, 791, 738, 688, 577, 507. ^1H NMR (400 MHz, CDCl_3): δ 8.04 (d, $J = 7.6$ Hz, 1H), 7.75 – 7.70 (m, 2H), 7.38 – 7.34 (m, 1H), 7.17 – 7.11 (m, 2H), 6.69 – 6.60 (m, 1H), 6.57 – 6.52 (m, 1H), 2.55 (s, 3H). ^{13}C NMR (100 MHz, CDCl_3) δ 163.0 (d, $^1J_{\text{C-F}} = 248.6$ Hz), 134.4, 129.9 (d, $^3J_{\text{C-F}} = 8.5$ Hz), 127.6, 127.14, 124.4 (d, $^4J_{\text{C-F}} = 3.4$ Hz), 120.8, 118.4, 118.3, 116.3 (d, $^2J_{\text{C-F}} = 21.9$ Hz), 114.4, 11.6. ^{19}F NMR (375 MHz, CDCl_3): δ -110.68. **ESI-HRMS**: m/z calcd. for $\text{C}_{14}\text{H}_{11}\text{FN}_2$ $[\text{M}+\text{H}]^+$: 227.0985, found 227.0980.

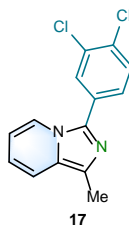


3-(4-chlorophenyl)-1-methylimidazo[1,5-a]pyridine: Following the **General procedure B1**, a mixture of 2-acetylpyridine (0.1 mmol), 4-chlorobenzaldehyde (2 equiv.) and D-glucosamine·HCl (2.0 equiv.) in $\text{AcOH} : \text{H}_2\text{O}$ (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up gave product **15** (19.8 mg, 0.082 mmol, isolated yield 82%) as a yellow solid. **Mp**: 63–65 °C. **FT-IR**: ν (cm^{-1}): 2959, 2926, 2852, 2096, 1896, 1663, 1589, 1504, 1405, 1257, 1086, 1010, 797, 736, 709, 686, 573, 544, 499,

484, 424. **¹H NMR** (500 MHz, CDCl₃): δ 8.04 (d, *J* = 7.5 Hz, 1H), 7.64 (d, *J* = 8.5 Hz, 2H), 7.39 (d, *J* = 8.5 Hz, 2H), 7.33 (d, *J* = 9.0 Hz, 1H), 6.58 – 6.55 (m, 1H), 6.46 – 6.43 (m, 1H), 2.48 (s, 3H). **¹³C NMR** (125 MHz, CDCl₃): δ 135.3, 134.1, 129.2, 129.1, 128.9, 128.8, 128.2, 120.9, 118.4, 117.2, 113.2, 12.5. **ESI-HRMS**: *m/z* calcd. for C₁₄H₁₁ClN₂ [M+H]⁺: 243.0689, found 243.0687.

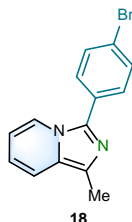


3-(3-chlorophenyl)-1-methylimidazo[1,5-a]pyridine: Following the **General procedure B1**, a mixture of 2-acetylpyridine (0.1 mmol), 3-chlorobenzaldehyde (2 equiv.) and D-glucosamine·HCl (2.0 equiv.) in AcOH : H₂O (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up gave product **16** (18.2 mg, 0.075 mmol, isolated yield 75%) as a brown solid. **Mp**: 70–72 °C. **FT-IR**: *ν* (cm⁻¹): 2965, 2918, 2850, 1653, 1560, 1407, 1257, 1069, 1010, 795, 692. **¹H NMR** (400 MHz, CDCl₃): δ 8.10 (d, *J* = 7.2 Hz, 1H), 7.71 (s, 1H), 7.60 – 7.58 (m, 1H), 7.35 – 7.28 (m, 3H), 6.61 – 6.57 (m, 1H), 6.49 – 6.46 (m, 1H), 2.48 (s, 3H). **¹³C NMR** (100 MHz, CDCl₃): δ 134.9, 132.1, 131.1, 130.2, 129.3, 128.4, 128.3, 127.6, 125.6, 121.0, 118.5, 117.4, 113.4, 12.5. **ESI-HRMS**: *m/z* calcd. for C₁₄H₁₁ClN₂ [M+H]⁺: 243.0689, found 243.0685.

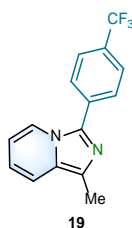


3-(3,4-dichlorophenyl)-1-methylimidazo[1,5-a]pyridine: Following the **General procedure B1**, a mixture of 2-acetylpyridine (0.1 mmol), 3,4-dichlorobenzaldehyde (2 equiv.) and D-glucosamine·HCl (2.0 equiv.) in AcOH : H₂O (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up gave product **17** (17.1 mg, 0.062 mmol, isolated yield 62%) as a yellow solid. **Mp**: 117–118 °C. **FT-IR**: *ν* (cm⁻¹): 2959, 1665, 1628, 1589, 1550, 1498, 1440, 1414, 1395, 1257, 1078, 1010, 872, 797, 732, 696, 666, 521, 436. **¹H NMR** (400 MHz, CDCl₃): δ 8.05 (d, *J* = 7.6 Hz, 1H), 7.83 (s, 1H), 7.57 – 7.54 (m, 1H), 7.49 – 7.46 (m, 1H), 7.35 – 7.32 (m, 1H), 6.62 – 6.57 (m, 1H), 6.51 – 6.47 (m, 1H), 2.48 (s, 3H). **¹³C NMR**

(100 MHz, CDCl₃): δ 134.0, 133.2, 132.0, 130.8, 130.4, 129.6, 129.2, 128.6, 126.4, 120.8, 118.5, 117.5, 113.6, 12.5. **ESI-HRMS**: m/z calcd. for C₁₄H₁₀Cl₂N₂ [M+H]⁺: 277.0299, found 277.0295.

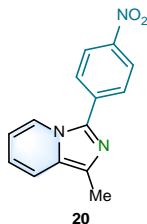


3-(4-bromophenyl)-1-methylimidazo[1,5-a]pyridine: Following the **General procedure B1**, a mixture of 2-acetylpyridine (0.1 mmol), 4-bromobenzaldehyde (2 equiv.) and D-glucosamine·HCl (2.0 equiv.) in AcOH : H₂O (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up gave product **18** (22.6 mg, 0.079 mmol, isolated yield 79%) as a yellow solid. **Mp**: 90–92 °C. **FT-IR**: ν (cm⁻¹): 2920, 2850, 1659, 1589, 1506, 1459, 1401, 1362, 1255, 1230, 1105, 1069, 1008, 942, 826, 734, 684, 492, 420. **¹H NMR** (400 MHz, CDCl₃): δ 8.05 (d, J = 7.6 Hz, 1H), 7.60 – 7.54 (m, 4H), 7.33 (d, J = 8.8 Hz, 1H), 6.58 – 6.55 (m, 1H), 6.46 – 6.43 (m, 1H), 2.49 (s, 3H). **¹³C NMR** (100 MHz, CDCl₃): δ 135.4, 132.1, 129.4, 129.3, 129.1, 128.3, 122.2, 120.9, 118.5, 117.2, 113.2, 12.6. **ESI-HRMS**: m/z calcd. for C₁₄H₁₁BrN₂ [M+H]⁺: 287.0184, found 287.0181.

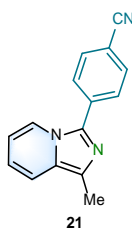


1-methyl-3-(4-(trifluoromethyl)phenyl)imidazo[1,5-a]pyridine: Following the **General procedure B1**, a mixture of 2-acetylpyridine (0.1 mmol), 4-(trifluoromethyl)benzaldehyde (2 equiv.) and D-glucosamine·HCl (2.0 equiv.) in AcOH : H₂O (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up gave product **19** (23.7 mg, 0.086 mmol, isolated yield 86 %) as a yellow solid. **Mp**: 73–74 °C. **FT-IR**: ν (cm⁻¹): 3062, 2918, 2858, 1675, 1618, 1521, 1418, 1321, 1255, 1166, 1111, 1065, 1012, 843, 802, 738, 684, 604, 499, 437. **¹H NMR** (400 MHz, CDCl₃): δ 8.13 (d, J = 7.2 Hz, 1H), 7.85 (d, J = 8.0 Hz, 2H), 7.67 (d, J = 8.0 Hz, 2H), 7.37 – 7.34 (m, 1H), 6.62 – 6.59 (m, 1H), 6.51 – 6.47 (m, 1H), 2.50 (s, 3H). **¹³C NMR** (100 MHz, CDCl₃): δ 134.9, 133.9 (q, ⁴ J_{C-F} = 1.1 Hz), 129.8 (q, ² J_{C-F} = 32.5 Hz), 129.8, 128.7, 127.6, 125.9 (q, ³ J_{C-F} = 3.8 Hz), 124.0 (q, ¹ J_{C-F} = 270.3 Hz), 120.9, 118.5, 117.6, 113.6, 12.6.

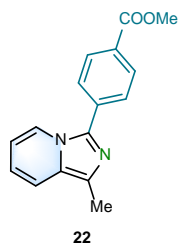
^{19}F NMR (375 MHz, CDCl_3): δ -62.60. **ESI-HRMS**: m/z calcd. for $\text{C}_{15}\text{H}_{11}\text{F}_3\text{N}_2$ $[\text{M}+\text{H}]^+$: 277.0953, found 277.0950.



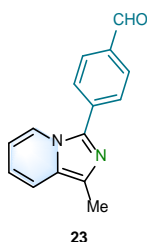
1-methyl-3-(4-nitrophenyl)imidazo[1,5-a]pyridine: Following the **General procedure B1**, a mixture of 2-acetylpyridine (0.1 mmol), 4-nitrobenzaldehyde (2 equiv.) and D-glucosamine·HCl (2.0 equiv.) in $\text{AcOH} : \text{H}_2\text{O}$ (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up gave product **20** (20.5 mg, 0.081 mmol, isolated yield 81%) as a red solid. **Mp**: 142–144 °C. **FT-IR**: ν (cm^{-1}): 2912, 1677, 1628, 1589, 1496, 1447, 1424, 1309, 1257, 1173, 1107, 1069, 1016, 948, 847, 814, 750, 738, 692, 676, 546, 523, 490, 476, 433, 422. **^1H NMR** (400 MHz, CDCl_3): δ 8.29 – 8.26 (m, 2H), 8.22 (d, J = 7.2 Hz, 1H), 7.95 – 7.92 (m, 2H), 7.41 (d, J = 9.1 Hz, 1H), 6.71 – 6.67 (m, 1H), 6.61 – 6.57 (m, 1H), 2.52 (s, 3H). **^{13}C NMR** (100 MHz, CDCl_3): δ 146.7, 136.5, 134.0, 131.0, 129.6, 127.3, 124.4, 121.0, 118.7, 118.4, 114.3, 12.6. **ESI-HRMS**: m/z calcd. for $\text{C}_{14}\text{H}_{11}\text{N}_3\text{O}_2$ $[\text{M}+\text{H}]^+$: 254.0930, found 254.0928.



4-(1-methylimidazo[1,5-a]pyridin-3-yl)benzonitrile: Following the **General procedure B1**, a mixture of 2-acetylpyridine (0.1 mmol), 4-formylbenzonitrile (2 equiv.) and D-glucosamine·HCl (2.0 equiv.) in $\text{AcOH} : \text{H}_2\text{O}$ (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up gave product **21** (16.8 mg, 0.072 mmol, isolated yield 72%) as a yellow solid. **Mp**: 89–91 °C. **FT-IR**: ν (cm^{-1}): 2957, 2910, 2844, 1655, 1605, 1552, 1510, 1395, 1257, 1082, 1010, 857, 789, 734, 686, 519. **^1H NMR** (400 MHz, CDCl_3): δ 8.15 (d, J = 7.6 Hz, 1H), 7.87 – 7.84 (m, 2H), 7.69 – 7.67 (m, 2H), 7.38 (d, J = 9.2 Hz, 1H), 6.65 (dd, J = 8.8, 6.4 Hz, 1H), 6.56 – 6.52 (m, 1H), 2.49 (s, 3H). **^{13}C NMR** (100 MHz, CDCl_3): δ 134.6, 134.3, 132.7, 130.5, 129.2, 127.5, 120.9, 118.7, 118.7, 118.2, 114.1, 111.1, 12.5. **ESI-HRMS**: m/z calcd. for $\text{C}_{15}\text{H}_{11}\text{N}_3$ $[\text{M}+\text{H}]^+$: 234.1031, found 234.1030.

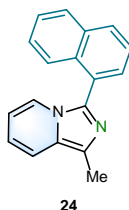


methyl 4-(1-methylimidazo[1,5-a]pyridin-3-yl)benzoate: Following the **General procedure B1**, a mixture of 2-acetylpyridine (0.1 mmol), methyl 4-formylbenzoate (2 equiv.) and D-glucosamine·HCl (2.0 equiv.) in AcOH : H₂O (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up gave product **22** (12.0 mg, 0.043 mmol, isolated yield 43%) as a brown solid. **Mp:** 92–94 °C. **FT-IR:** ν (cm⁻¹): 2953, 2920, 2850, 1717, 1661, 1607, 1434, 1269, 1177, 1102, 1016, 960, 859, 797, 775, 736, 702, 568, 540, 490, 420. **¹H NMR** (500 MHz, CDCl₃): δ 8.17 (d, J = 7.5 Hz, 1H), 8.08 (d, J = 8.5 Hz, 2H), 7.80 (d, J = 8.5 Hz, 2H), 7.36 (d, J = 9.0 Hz, 1H), 6.62 (dd, J = 9.0, 6.0 Hz, 1H), 6.52 – 6.49 (m, 1H), 3.87 (s, 3H), 2.49 (s, 3H). **¹³C NMR** (125 MHz, CDCl₃): δ 166.6, 134.5, 130.2, 129.8, 129.4, 128.8, 127.4, 127.1, 121.1, 118.5, 117.8, 52.2, 12.5. **ESI-HRMS:** m/z calcd. for C₁₆H₁₄N₂O₂ [M+H]⁺: 267.1134, found 267.1132.

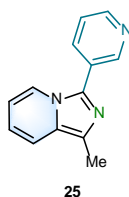


4-(1-methylimidazo[1,5-a]pyridin-3-yl)benzaldehyde: Following the **General procedure B1**, a mixture of 2-acetylpyridine (0.1 mmol), terephthalaldehyde (2 equiv.) and D-glucosamine·HCl (2.0 equiv.) in AcOH : H₂O (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up gave product **23** (16.0 mg, 0.068 mmol, isolated yield 68%) as a yellow solid. **Mp:** 313–315 °C. **FT-IR:** ν (cm⁻¹): 2914, 1698, 1599, 1381, 1012, 795, 670. **¹H NMR** (400 MHz, CDCl₃): δ 9.97 (s, 1H), 8.21 (dd, J = 7.2, 0.8 Hz, 1H), 7.92 – 7.92 (m, 4H), 7.37 (dd, J = 9.2, 1.2 Hz, 1H), 6.66 – 6.62 (m, 1H), 6.55 – 6.51 (m, 1H), 2.51 (s, 3H). **¹³C NMR** (100 MHz, CDCl₃): δ 191.5, 136.0, 135.3, 135.0, 130.4, 130.3, 129.1, 127.4,

121.2, 118.6, 118.0, 113.8, 12.6. **ESI-HRMS:** m/z calcd. for $C_{15}H_{12}N_2O$ $[M+H]^+$: 237.1028, found 237.1026.

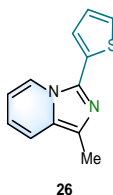


1-methyl-3-(naphthalen-1-yl)imidazo[1,5-a]pyridine: Following the **General procedure B1**, a mixture of 2-acetylpyridine (0.1 mmol), 1-naphthaldehyde (2 equiv.) and D-glucosamine·HCl (2.0 equiv.) in AcOH : H₂O (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up gave product **24** (15.2 mg, 0.059 mmol, isolated yield 59%) as a brown solid. **Mp:** 91–93 °C. **FT-IR:** ν (cm⁻¹): 3054, 2918, 2854, 1659, 1585, 1513, 1461, 1416, 1366, 1249, 1212, 1142, 1092, 1020, 995, 933, 865, 804, 773, 738, 694, 645, 577, 532, 503, 412. **¹H NMR** (400 MHz, CDCl₃): δ 8.00 – 7.95 (m, 2H), 7.76 – 7.73 (m, 2H), 7.63 – 7.58 (m, 2H), 7.57 – 7.53 (m, 1H), 7.50 – 7.46 (m, 2H), 6.69 – 6.65 (m, 1H), 6.44 – 6.40 (m, 1H), 2.67 (s, 3H). **¹³C NMR** (100 MHz, CDCl₃): δ 135.3, 134.0, 131.8, 129.6, 128.6, 128.6, 128.5, 127.6, 127.5, 126.9, 126.2, 125.7, 125.4, 121.6, 118.2, 117.0, 112.3, 12.7. **ESI-HRMS:** m/z calcd. for $C_{18}H_{14}N_2$ $[M+H]^+$: 259.1235, found 259.1233.

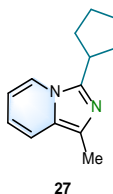


1-methyl-3-(pyridin-3-yl)imidazo[1,5-a]pyridine: Following the **General procedure B1**, a mixture of 2-acetylpyridine (0.1 mmol), nicotinaldehyde (2 equiv.) and D-glucosamine·HCl (2.0 equiv.) in AcOH : H₂O (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up gave product **25** (8.8 mg, 0.042 mmol, isolated yield 42%) as a brown liquid. **FT-IR:** ν (cm⁻¹): 3035, 2922, 2858, 1675, 1630, 1587, 1568, 1510, 1496, 1420, 1362, 1327, 1300, 1255, 1179, 1150, 1123, 1078, 1022, 1006, 944, 808, 736, 709, 688, 616, 546, 422. **¹H NMR** (400 MHz, CDCl₃): δ 8.99 (d, J = 2.0 Hz, 1H), 8.56 (dd, J = 4.8, 1.5 Hz, 1H), 8.09 (d, J = 7.2 Hz, 1H), 8.05 – 8.03 (m, 1H), 7.38 – 7.35 (m, 2H), 6.61 (dd, J = 9.1, 6.3 Hz,

1H), 6.52 – 6.48 (m, 1H), 2.50 (s, 3H). ¹³C NMR (100 MHz, CDCl₃): δ 149.1, 148.2, 135.1, 133.4, 129.8, 128.6, 126.8, 123.8, 120.7, 118.5, 117.6, 113.6, 12.5. **ESI-HRMS**: *m/z* calcd. for C₁₃H₁₁N₃ [M+H]⁺: 210.1031, found 210.1027.

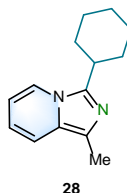


1-methyl-3-(thiophen-2-yl)imidazo[1,5-a]pyridine: Following the **General procedure B1**, a mixture of 2-acetylpyridine (0.1 mmol), thiophene-2-carbaldehyde (2 equiv.) and D-glucosamine·HCl (2.0 equiv.) in AcOH : H₂O (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up gave product **26** (11.8 mg, 0.055 mmol, isolated yield 55%) as a brown solid. **Mp**: 79–81 °C. **FT-IR**: ν (cm⁻¹): 3074, 2959, 2920, 2854, 1735, 1649, 1603, 1513, 1463, 1401, 1364, 1253, 1037, 909, 843, 804, 682, 416. ¹H NMR (400 MHz, CDCl₃): δ 8.18 – 8.16 (m, 1H), 7.42 – 7.40 (m, 1H), 7.35 – 7.30 (m, 2H), 7.10 (ddd, *J* = 4.8, 3.6, 1.2 Hz, 1H), 6.60 – 6.50 (m, 2H), 2.49 (s, 3H). ¹³C NMR (100 MHz, CDCl₃): δ 132.7, 131.3, 129.3, 128.2, 127.6, 125.4, 124.0, 121.5, 118.4, 117.0, 113.5, 12.6. **ESI-HRMS**: *m/z* calcd. for C₁₂H₁₀N₂S [M+H]⁺: 215.0643, found 215.0641.

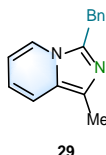


3-cyclopentyl-1-methylimidazo[1,5-a]pyridine: Following the **General procedure B1**, a mixture of 2-acetylpyridine (0.1 mmol), cyclopentanecarbaldehyde (2 equiv.) and D-glucosamine·HCl (2.0 equiv.) in AcOH : H₂O (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up gave product **27** (16.2 mg, 0.081 mmol, isolated yield 81%) as a brown liquid. **FT-IR**: ν (cm⁻¹): 2945, 2864, 1667, 1504, 1449, 1434, 1339, 1314, 1284, 1241, 1199, 1158, 1094, 993, 938, 795, 732, 692, 620, 587, 422, 402. ¹H NMR (400 MHz, CDCl₃): δ 7.57 (d, *J* = 7.2 Hz, 1H), 7.19 (d, *J* = 9.6 Hz, 1H), 6.44 – 6.39 (m, 1H), 6.35 – 6.31 (m, 1H), 3.25 (p, *J* = 8.3 Hz, 1H), 2.40 (s, 3H), 2.08 – 2.01 (m, 2H), 1.96 – 1.89 (m, 2H), 1.80 – 1.75 (m, 2H), 1.64 – 1.61 (m, 2H). ¹³C NMR (100 MHz, CDCl₃): δ 140.5, 126.6, 126.0, 120.4, 118.1,

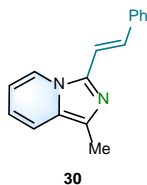
115.6, 111.5, 36.6, 30.8, 25.5, 12.4. **ESI-HRMS**: m/z calcd. for $C_{13}H_{16}N_2$ $[M+H]^+$: 201.1392, found 201.1384.



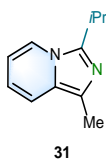
3-cyclohexyl-1-methylimidazo[1,5-a]pyridine: Following the **General procedure B1**, a mixture of 2-acetylpyridine (0.1 mmol), cyclohexanecarbaldehyde (2 equiv.) and D-glucosamine·HCl (2.0 equiv.) in AcOH : H₂O (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up gave product **28** (17.8 mg, 0.083 mmol, isolated yield 83%) as a white solid. **Mp**: 92–94 °C. **FT-IR**: ν (cm⁻¹): 2920, 2850, 1663, 1591, 1480, 1445, 1259, 1170, 1090, 1018, 890, 795, 742, 694, 643, 542, 420. **¹H NMR** (400 MHz, CDCl₃): δ 7.58 (d, J = 7.2 Hz, 1H), 7.21 (d, J = 9.2 Hz, 1H), 6.44 – 6.40 (m, 1H), 6.35 – 6.32 (m, 1H), 2.88 – 2.80 (m, 1H), 2.41 (s, 3H), 1.95 – 1.90 (m, 2H), 1.85 – 1.80 (m, 2H), 1.74 – 1.64 (m, 3H), 1.41 – 1.25 (m, 3H). **¹³C NMR** (100 MHz, CDCl₃): δ 141.4, 126.4, 126.2, 120.2, 118.3, 115.6, 111.5, 35.7, 30.8, 26.4, 26.0, 12.5. **ESI-HRMS**: m/z calcd. for $C_{14}H_{18}N_2$ $[M+H]^+$: 215.1548, found 215.1546.



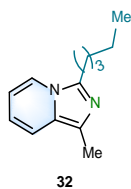
3-benzyl-1-methylimidazo[1,5-a]pyridine: Following the **General procedure B1**, a mixture of 2-acetylpyridine (0.1 mmol), 2-phenylacetaldehyde (2 equiv.) and D-glucosamine·HCl (2.0 equiv.) in AcOH : H₂O (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up gave product **29** (9.1 mg, 0.041 mmol, isolated yield 41%) as a brown solid. **Mp**: 89–91 °C. **FT-IR**: ν (cm⁻¹): 3054, 3029, 2961, 2926, 1663, 1599, 1494, 1453, 1370, 1261, 1222, 1156, 1074, 1030, 800, 746, 696, 565, 460. **¹H NMR** (400 MHz, CDCl₃): δ 7.40 (d, J = 7.2 Hz, 1H), 7.20 – 7.18 (m, 3H), 7.15 – 7.13 (m, 1H), 7.11 – 7.09 (m, 2H), 6.47 – 6.43 (m, 1H), 6.30 – 6.26 (m, 1H), 4.31 (s, 2H), 2.45 (s, 3H). **¹³C NMR** (100 MHz, CDCl₃): δ 136.6, 135.2, 129.8, 128.8, 128.6, 128.2, 126.7, 120.5, 118.2, 116.1, 112.1, 33.5, 12.5. **ESI-HRMS**: m/z calcd. for $C_{15}H_{14}N_2$ $[M+H]^+$: 223.1235, found 223.1231.



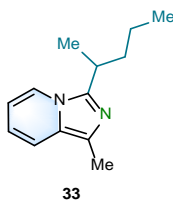
(E)-1-methyl-3-styrylimidazo[1,5-a]pyridine: Following the **General procedure B1**, a mixture of 2-acetylpyridine (0.1 mmol), cinnamaldehyde (2 equiv.) and D-glucosamine·HCl (2.0 equiv.) in AcOH : H₂O (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up gave product **30** (3.3 mg, 0.014 mmol, isolated yield 14%) as a brown solid. **Mp**: 110–112 °C. **FT-IR**: ν (cm⁻¹): 3031, 2918, 2848, 2089, 1659, 1628, 1597, 1548, 1513, 1492, 1449, 1412, 1374, 1333, 1261, 1241, 1197, 1129, 1094, 1072, 1026, 954, 795, 750, 696, 550, 495, 416. **¹H NMR** (400 MHz, CDCl₃): δ 7.89 (d, J = 7.2 Hz, 1H), 7.61 (d, J = 16.0 Hz, 1H), 7.50 (d, J = 8.0 Hz, 2H), 7.32 – 7.28 (m, 3H), 7.21 (d, J = 8.4 Hz, 1H), 7.11 (d, J = 15.6 Hz, 1H), 6.59 – 6.51 (m, 2H), 2.50 (s, 3H). **¹³C NMR** (100 MHz, CDCl₃): δ 137.0, 134.9, 130.0, 128.8, 127.9, 127.9, 126.6, 120.5, 118.5, 117.3, 113.4, 111.7, 12.6. **ESI-HRMS**: m/z calcd. for C₁₆H₁₄N₂ [M+H]⁺: 235.1235, found 235.1229.



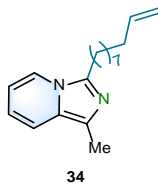
3-isopropyl-1-methylimidazo[1,5-a]pyridine: Following the **General procedure B1**, a mixture of 2-acetylpyridine (0.1 mmol), isobutyraldehyde (2 equiv.) and D-glucosamine·HCl (2.0 equiv.) in AcOH : H₂O (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up gave product **31** (9.2 mg, 0.053 mmol, isolated yield 53%) as a brown solid. **Mp**: 94–95 °C. **FT-IR**: ν (cm⁻¹): 2959, 2928, 2866, 1659, 1587, 1455, 1368, 1199, 1175, 1125, 1096, 800, 740, 717, 694, 620, 587, 422. **¹H NMR** (300 MHz, CDCl₃): δ 7.57 (d, J = 7.2 Hz, 1H), 7.24 – 7.20 (m, 1H), 6.44 (dd, J = 9.0, 6.3 Hz, 1H), 6.38 – 6.33 (m, 1H), 3.20 (hept, J = 6.9 Hz, 1H), 2.42 (s, 3H), 1.35 (d, J = 6.9 Hz, 6H). **¹³C NMR** (100 MHz, CDCl₃): δ 141.7, 126.4, 120.2, 118.3, 115.9, 112.0, 25.9, 20.5, 12.3. **ESI-HRMS**: m/z calcd. for C₁₁H₁₄N₂ [M+H]⁺: 175.1235, found 175.1224.



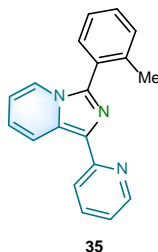
1-methyl-3-pentylimidazo[1,5-a]pyridine: Following the **General procedure B1**, a mixture of 2-acetylpyridine (0.1 mmol), hexanal (2 equiv.) and D-glucosamine·HCl (2.0 equiv.) in AcOH : H₂O (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. work-up gave product **32** (16.6 mg, 0.082 mmol, isolated yield 82%) as a brown liquid. **FT-IR:** ν (cm⁻¹): 3342, 3186, 2957, 2928, 2858, 1665, 1500, 1463, 1377, 1228, 1181, 1092, 997, 859, 795, 732, 694, 620, 418. **¹H NMR** (400 MHz, CDCl₃): δ 7.51 (d, J = 7.2 Hz, 1H), 7.20 (d, J = 10.0 Hz, 1H), 6.43 – 6.40 (m, 1H), 6.35 – 6.32 (m, 1H), 2.83 (t, J = 7.8 Hz, 2H), 2.39 (s, 3H), 1.72 (p, J = 7.2 Hz, 2H), 1.31 – 1.27 (m, 4H), 0.82 (t, J = 6.3 Hz, 3H). **¹³C NMR** (100 MHz, CDCl₃): δ 137.3, 126.3, 126.3, 120.2, 118.1, 115.5, 111.7, 31.6, 26.9, 26.5, 22.3, 13.9, 12.3. **ESI-HRMS:** m/z calcd. for C₁₃H₁₈N₂ [M+H]⁺: 203.1548, found 203.1543.



1-methyl-3-(pentan-2-yl)imidazo[1,5-a]pyridine: Following the **General procedure B1**, a mixture of 2-acetylpyridine (0.1 mmol), 2-methylpentanal (2 equiv.) and D-glucosamine·HCl (2.0 equiv.) in AcOH : H₂O (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up gave product **33** (17.0 mg, 0.084 mmol, isolated yield 84%) as a brown liquid. **FT-IR:** ν (cm⁻¹): 3334, 3180, 2963, 2928, 2875, 1665, 1461, 1459, 1374, 1337, 1282, 1238, 1188, 1154, 1094, 1055, 997, 868, 787, 738, 694, 622, 416. **¹H NMR** (400 MHz, CDCl₃): δ 7.57 (d, J = 7.2 Hz, 1H), 7.19 (d, J = 9.2 Hz, 1H), 6.43 – 6.39 (m, 1H), 6.34 – 6.31 (m, 1H), 3.07 (h, J = 7.0 Hz, 1H), 2.40 (s, 3H), 1.85 – 1.76 (m, 1H), 1.65 – 1.56 (m, 1H), 1.31 (d, J = 6.8 Hz, 3H), 1.27 – 1.17 (m, 2H), 0.80 (t, J = 7.3 Hz, 3H). **¹³C NMR** (100 MHz, CDCl₃): δ 141.3, 126.4, 126.1, 120.1, 118.2, 115.5, 111.7, 37.5, 30.9, 20.6, 18.6, 13.9, 12.3. **ESI-HRMS:** m/z calcd. for C₁₃H₁₈N₂ [M+H]⁺: 203.1548, found 203.1543.

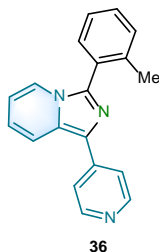


3-(dec-9-en-1-yl)-1-methylimidazo[1,5-a]pyridine: Following the **General procedure B1**, a mixture of 2-acetylpyridine (0.1 mmol), undec-10-enal (2 equiv.) and D-glucosamine·HCl (2.0 equiv.) in AcOH : H₂O (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up gave product **34** (20.3 mg, 0.075 mmol, isolated yield 75%) as a brown liquid. **FT-IR:** ν (cm⁻¹): 3357, 3182, 3079, 2924, 2854, 1659, 1632, 1465, 1424, 1372, 1243, 995, 909, 789, 746, 721, 699, 633, 554, 523, 474. **¹H NMR** (400 MHz, CDCl₃): δ 7.60 (d, J = 7.2 Hz, 1H), 7.31 – 7.26 (m, 1H), 6.53 – 6.48 (m, 1H), 6.45 – 6.41 (m, 1H), 5.86 – 5.75 (m, 1H), 5.02 – 4.91 (m, 2H), 2.94 – 2.90 (m, 2H), 2.48 (s, 3H), 2.06 – 2.00 (m, 2H), 1.84 – 1.77 (m, 2H), 1.42 – 1.26 (m, 10H). **¹³C NMR** (100 MHz, CDCl₃): δ 139.1, 137.3, 126.4, 126.3, 120.2, 118.1, 115.5, 114.1, 111.7, 33.7, 29.4, 29.3, 29.2, 29.0, 28.8, 27.2, 26.6, 12.4. **ESI-HRMS:** m/z calcd. for C₁₈H₂₆N₂ [M+H]⁺: 271.2174, found 271.2171.

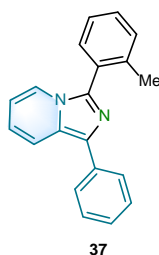


1-(pyridin-2-yl)-3-(o-tolyl)imidazo[1,5-a]pyridine: Following the **General procedure B2**, a mixture of di(pyridine-2-yl)methanone (0.1 mmol), 2-methylbenzaldehyde (4 equiv.) and D-glucosamine·HCl (2.0 equiv.) in AcOH : H₂O (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up gave product **35** (21.4 mg, 0.075 mmol, isolated yield 75%) as a yellow solid. **Mp:** 90–92 °C. **FT-IR:** ν (cm⁻¹): 2924, 2848, 1634, 1605, 1589, 1562, 1533, 1508, 1482, 1465, 1445, 1426, 1399, 1354, 1327, 1315, 1259, 1236, 1185, 1148, 1137, 1107, 1090, 1049, 1006, 942, 890, 868, 791, 777, 758, 744, 723, 711, 674, 624, 604, 542, 495, 443, 427, 404. **¹H NMR** (400 MHz, CDCl₃): δ 8.66 – 8.61 (m, 1H), 8.58 – 8.55 (m, 1H), 8.18 – 8.14 (m, 1H), 7.65 – 7.61 (m, 1H), 7.59 – 7.55 (m, 1H), 7.46 – 7.43 (m, 1H), 7.37 – 7.25 (m, 3H), 7.05 – 6.99 (m, 1H), 6.89 – 6.83 (m, 1H), 6.57 – 6.51 (m, 1H), 2.19 (s, 3H). **¹³C NMR** (100 MHz, CDCl₃):

δ 155.2, 149.0, 138.6, 137.7, 136.2, 130.8, 130.6, 129.9, 129.7, 129.2, 129.2, 126.1, 121.7, 121.6, 120.9, 120.3, 119.8, 113.6, 19.7. **ESI-HRMS**: m/z calcd. for $C_{19}H_{15}N_3$ $[M+H]^+$: 286.1344, found 286.1341.

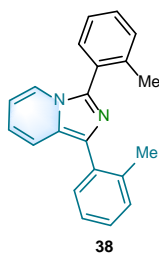


1-(pyridin-4-yl)-3-(o-tolyl)imidazo[1,5-a]pyridine: Following the **General procedure B2**, a mixture of pyridin-2-yl(pyridin-4-yl)methanone (0.1 mmol), 2-methylbenzaldehyde (4 equiv.) and D-glucosamine·HCl (2.0 equiv.) in AcOH : H₂O (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up gave product **36** (16.0 mg, 0.056 mmol, isolated yield 56%) as a yellow liquid. **FT-IR**: ν (cm⁻¹): 2957, 2924, 2850, 1735, 1673, 1622, 1595, 1537, 1498, 1463, 1407, 1379, 1304, 1278, 1197, 1148, 1100, 987, 942, 824, 773, 723, 649, 616, 515, 451, 420. **¹H NMR** (400 MHz, CDCl₃): δ 8.56 – 8.55 (m, 2H), 7.87 (d, J = 9.2 Hz, 1H), 7.85 – 7.83 (m, 2H), 7.63 (d, J = 7.2 Hz, 1H), 7.42 (d, J = 7.6 Hz, 1H), 7.38 – 7.34 (m, 2H), 7.31 – 7.29 (m, 1H), 6.92 (dd, J = 9.3, 6.4 Hz, 1H), 6.60 – 6.56 (m, 1H), 2.19 (s, 3H). **¹³C NMR** (100 MHz, CDCl₃): δ 149.3, 143.1, 139.0, 138.5, 131.0, 130.5, 130.0, 128.6, 127.6, 126.3, 122.7, 121.9, 121.0, 120.3, 118.5, 113.4, 19.7. **ESI-HRMS**: m/z calcd. for $C_{19}H_{15}N_3$ $[M+H]^+$: 286.1344, found 286.1340.

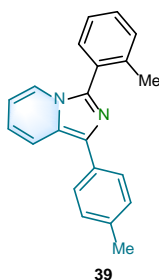


1-phenyl-3-(o-tolyl)imidazo[1,5-a]pyridine: Following the **General procedure B2**, a mixture of phenyl(pyridin-2-yl)methanone (0.1 mmol), 2-methylbenzaldehyde (4 equiv.) and D-glucosamine·HCl (2.0 equiv.) in AcOH : H₂O (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up gave product **37** (21.9 mg, 0.077 mmol, isolated yield 77%) as a yellow liquid. **FT-IR**: ν (cm⁻¹): 3048,

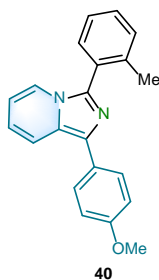
2927, 2851, 1663, 1628, 1599, 1508, 1490, 1447, 1354, 1315, 1242, 1179, 1142, 1102, 1074, 1004, 981, 946, 909, 771, 736, 694, 664, 616, 602, 499, 447, 418. **¹H NMR** (400 MHz, CDCl₃): δ 7.90 – 7.87 (m, 2H), 7.80 – 7.77 (m, 1H), 7.54 – 7.51 (m, 1H), 7.43 – 7.36 (m, 3H), 7.34 – 7.28 (m, 2H), 7.27 – 7.17 (m, 2H), 6.72 – 6.68 (m, 1H), 6.45 – 6.41 (m, 1H), 2.18 (s, 3H). **¹³C NMR** (100 MHz, CDCl₃): δ 138.5, 137.8, 135.1, 131.1, 130.8, 130.6, 129.5, 129.2, 128.7, 126.6, 126.6, 126.3, 126.1, 121.9, 119.5, 118.9, 112.8, 19.8. **ESI-HRMS**: *m/z* calcd. for C₂₀H₁₆N₂ [M+H]⁺: 285.1392, found 285.1390.



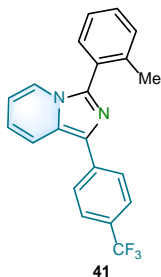
1,3-di-*o*-tolylimidazo[1,5-*a*]pyridine: Following the **General procedure B2**, a mixture of pyridin-2-yl(*o*-tolyl)methanone (0.1 mmol), 2-methylbenzaldehyde (4 equiv.) and D-glucosamine·HCl (2.0 equiv.) in AcOH : H₂O (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up gave product **38** (16.7 mg, 0.056 mmol, isolated yield 56%) as a green solid. **Mp**: 112–114 °C. **FT-IR**: *ν* (cm⁻¹): 3056, 2918, 2854, 1663, 1601, 1517, 1482, 1455, 1354, 1306, 1257, 1133, 1102, 1039, 1010, 950, 793, 769, 758, 723, 703, 659, 616, 542, 453, 422. **¹H NMR** (500 MHz, CDCl₃): δ 7.63 (d, *J* = 7.0 Hz, 1H), 7.46 – 7.44 (m, 2H), 7.42 (d, *J* = 9.0 Hz, 1H), 7.34 – 7.32 (m, 2H), 7.28 – 7.26 (m, 2H), 7.22 – 7.20 (m, 2H), 6.74 – 6.69 (m, 1H), 6.50 – 6.50 (m, 1H), 2.42 (s, 3H), 2.25 (s, 3H). **¹³C NMR** (100 MHz, CDCl₃): δ 138.5, 137.3, 130.9, 130.8, 130.3, 130.2, 129.4, 127.4, 127.4, 126.0, 125.5, 121.5, 118.9, 112.9, 20.7, 19.9. **ESI-HRMS**: *m/z* calcd. for C₂₁H₁₈N₂ [M+H]⁺: 299.1548, found 299.1545.



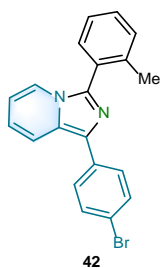
3-(o-tolyl)-1-(p-tolyl)imidazo[1,5-a]pyridine: Following the **General procedure B2**, a mixture of pyridin-2-yl(*p*-tolyl)methanone (0.1 mmol), 2-methylbenzaldehyde (4 equiv.) and D-glucosamine·HCl (2.0 equiv.) in AcOH : H₂O (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up gave product **39** (21.8 mg, 0.073 mmol, isolated yield 73%) as a yellow liquid. **FT-IR:** ν (cm⁻¹): 3031, 2926, 2856, 1909, 1661, 1607, 1539, 1508, 1409, 1358, 1311, 1259, 1245, 1179, 1107, 1039, 1001, 946, 818, 771, 746, 715, 645, 600, 571, 507, 451, 424. **¹H NMR** (400 MHz, CDCl₃): δ 7.80 – 7.75 (m, 3H), 7.54 – 7.51 (m, 1H), 7.42 (d, *J* = 7.2 Hz, 1H), 7.35 – 7.28 (m, 2H), 7.27 – 7.23 (m, 1H), 7.21 – 7.18 (m, 2H), 6.71 – 6.67 (m, 1H), 6.45 – 6.41 (m, 1H), 2.33 (s, 3H), 2.18 (s, 3H). **¹³C NMR** (100 MHz, CDCl₃): δ 138.5, 137.6, 136.0, 132.3, 131.3, 130.8, 130.6, 129.5, 129.4, 129.3, 126.5, 126.3, 126.1, 121.8, 119.2, 119.1, 112.7, 21.2, 19.8. **ESI-HRMS:** *m/z* calcd. for C₂₁H₁₈N₂ [M+H]⁺: 299.1548, found 299.1543.



1-(4-methoxyphenyl)-3-(o-tolyl)imidazo[1,5-a]pyridine: Following the **General procedure B2**, a mixture of (4-methoxyphenyl)(pyridin-2-yl)methanone (0.1 mmol), 2-methylbenzaldehyde (4 equiv.) and D-glucosamine·HCl (2.0 equiv.) in AcOH : H₂O (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up gave product **40** (22.0 mg, 0.070 mmol, isolated yield 70%) as a yellow liquid. **FT-IR:** ν (cm⁻¹): 3066, 2930, 2848, 1657, 1599, 1539, 1500, 1459, 1412, 1356, 1298, 1284, 1245, 1179, 1102, 1028, 999, 942, 835, 771, 750, 725, 703, 647, 600, 577, 521, 449, 422. **¹H NMR** (400 MHz, CDCl₃): δ 7.81 (d, *J* = 8.4 Hz, 2H), 7.73 (d, *J* = 9.3 Hz, 1H), 7.52 (d, *J* = 7.2 Hz, 1H), 7.43 (d, *J* = 7.4 Hz, 1H), 7.35 – 7.29 (m, 2H), 7.27 – 7.24 (m, 1H), 6.94 (d, *J* = 8.5 Hz, 2H), 6.68 (dd, *J* = 9.2, 6.4 Hz, 1H), 6.45 – 6.41 (m, 1H), 3.79 (s, 3H), 2.19 (s, 3H). **¹³C NMR** (100 MHz, CDCl₃): δ 158.4, 138.5, 137.5, 131.1, 130.8, 130.6, 129.5, 129.4, 127.9, 127.8, 126.1, 126.0, 121.8, 119.0, 118.9, 114.2, 112.7, 55.3, 19.8. **ESI-HRMS:** *m/z* calcd. for C₂₁H₁₈N₂O [M+H]⁺: 315.1497, found 315.1495.

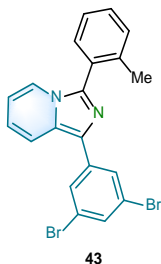


3-(o-tolyl)-1-(4-(trifluoromethyl)phenyl)imidazo[1,5-a]pyridine: Following the **General procedure B2**, a mixture of pyridin-2-yl(4-(trifluoromethyl)phenyl)methanone (0.1 mmol), 2-methylbenzaldehyde (4 equiv.) and D-glucosamine·HCl (2.0 equiv.) in AcOH : H₂O (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up gave product **41** (30.6 mg, 0.087 mmol, isolated yield 87%) as a yellow liquid. **FT-IR:** ν (cm⁻¹): 3066, 2963, 2918, 2848, 1922, 1677, 1616, 1545, 1508, 1469, 1407, 1319, 1253, 1160, 1107, 1061, 1010, 981, 948, 847, 802, 771, 760, 729, 715, 703, 674, 618, 604, 544, 509, 455, 443, 424, 410. **¹H NMR** (400 MHz, CDCl₃): δ 8.01 (d, J = 8.1 Hz, 2H), 7.80 (d, J = 9.3 Hz, 1H), 7.64 – 7.58 (m, 3H), 7.43 (d, J = 7.6 Hz, 1H), 7.37 – 7.26 (m, 3H), 6.84 – 6.80 (m, 1H), 6.52 (t, J = 6.8 Hz, 1H), 2.19 (s, 3H). **¹³C NMR** (100 MHz, CDCl₃): δ 138.7, 138.5, 138.4, 131.2, 130.9, 130.6, 129.8, 129.4, 128.9, 127.8 (q, $^2J_{C-F}$ = 32.1 Hz), 127.5, 126.3, 126.2, 125.6 (q, $^3J_{C-F}$ = 3.8 Hz), 124.5 (q, $^1J_{C-F}$ = 269.8 Hz), 122.3, 120.7, 118.6, 113.1, 19.8. **¹⁹F NMR** (375 MHz, CDCl₃): δ -62.20. **ESI-HRMS:** m/z calcd. for C₂₁H₁₅F₃N₂ [M+H]⁺: 353.1266, found 353.1263.

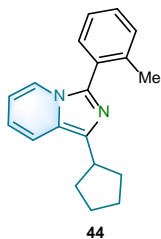


1-(4-bromophenyl)-3-(o-tolyl)imidazo[1,5-a]pyridine: Following the **General procedure B2**, a mixture of (4-bromophenyl)(pyridin-2-yl)methanone (0.1 mmol), 2-methylbenzaldehyde (4 equiv.) and D-glucosamine·HCl (2.0 equiv.) in AcOH : H₂O (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up gave product **42** (30.0 mg, 0.083 mmol, isolated yield 83%) as a yellow solid. **Mp:** 123–125 °C. **FT-IR:** ν (cm⁻¹): 2918, 2854, 2122, 1653, 1583, 1521, 1484, 1457, 1395, 1377, 1346, 1311, 1261, 1241, 1160, 1133, 1098, 1067, 1047, 1006, 981, 946, 828, 800, 771, 756, 723, 711, 699,

678, 633, 610, 598, 583, 505, 455, 437, 422. **¹H NMR** (300 MHz, CDCl₃): δ 7.78 – 7.72 (m, 3H), 7.55 (d, *J* = 7.2 Hz, 1H), 7.50 (d, *J* = 8.4 Hz, 2H), 7.41 (d, *J* = 7.2 Hz, 1H), 7.35 – 7.26 (m, 3H), 6.76 (dd, *J* = 9.2, 6.3 Hz, 1H), 6.50 – 6.46 (m, 1H), 2.18 (s, 3H). **¹³C NMR** (100 MHz, CDCl₃): δ 138.5, 138.0, 134.1, 131.7, 130.9, 130.6, 129.9, 129.7, 129.0, 128.0, 126.8, 126.1, 122.1, 120.1, 120.0, 118.7, 112.9, 19.8. **ESI-HRMS**: *m/z* calcd. for C₂₀H₁₅BrN₂ [M+H]⁺: 363.0497, found 363.0493.

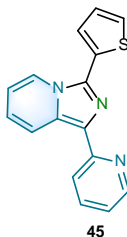


1-(3,5-dibromophenyl)-3-(o-tolyl)imidazo[1,5-a]pyridine: Following the **General procedure B2**, a mixture of (3,5-dibromophenyl)(pyridin-2-yl)methanone (0.1 mmol), 2-methylbenzaldehyde (4 equiv.) and D-glucosamine·HCl (2.0 equiv.) in AcOH : H₂O (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up gave product **43** (27.7 mg, 0.063 mmol, isolated yield 63%) as a yellow solid. **Mp**: 139–140 °C. **FT-IR**: *ν* (cm⁻¹): 3052, 2951, 2916, 2958, 1663, 1583, 1545, 1508, 1459, 1432, 1409, 1383, 1341, 1315, 1276, 1247, 1148, 1100, 1051, 1014, 983, 948, 861, 839, 806, 777, 738, 721, 705, 670, 620, 536, 451, 422. **¹H NMR** (400 MHz, CDCl₃): δ 7.98 (s, 2H), 7.76 (d, *J* = 9.2 Hz, 1H), 7.57 (d, *J* = 7.2 Hz, 1H), 7.47 (s, 1H), 7.41 (d, *J* = 7.6 Hz, 1H), 7.36 – 7.26 (m, 3H), 6.87 – 6.83 (m, 1H), 6.55 – 6.51 (m, 1H), 2.17 (s, 3H). **¹³C NMR** (100 MHz, CDCl₃): δ 138.7, 138.5, 138.4, 131.3, 130.9, 130.6, 129.9, 128.7, 127.9, 127.7, 127.4, 126.2, 123.2, 122.3, 121.1, 118.4, 113.2, 19.7. **ESI-HRMS**: *m/z* calcd. for C₂₀H₁₄Br₂N₂ [M+H]⁺: 440.9602, found 440.9609.

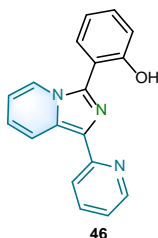


1-cyclopentyl-3-(o-tolyl)imidazo[1,5-a]pyridine: Following the **General procedure B2**, a mixture of cyclopentyl(pyridin-2-yl)methanone (0.1 mmol), 2-methylbenzaldehyde (4 equiv.) and D-

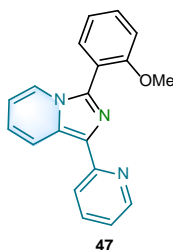
glucosamine·HCl (2.0 equiv.) in AcOH : H₂O (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up gave product **44** (22.6 mg, 0.082 mmol, isolated yield 82%) as a brown solid. **Mp**: 72–74 °C. **FT-IR**: ν (cm⁻¹): 3060, 2951, 2862, 1653, 1628, 1603, 1453, 1416, 1364, 1304, 1238, 1105, 1045, 995, 946, 771, 746, 723, 620, 453, 418. **¹H NMR** (400 MHz, CDCl₃): δ 7.42 (d, J = 6.8 Hz, 1H), 7.36 (d, J = 9.2 Hz, 1H), 7.32 (d, J = 7.6 Hz, 1H), 7.26 – 7.22 (m, 2H), 7.18 (t, J = 6.7 Hz, 1H), 6.48 (dd, J = 9.2, 6.4 Hz, 1H), 6.33 – 6.29 (m, 1H), 3.36 – 3.28 (m, 1H), 2.11 (s, 3H), 2.04 – 1.98 (m, 2H), 1.91 – 1.86 (m, 2H), 1.82 – 1.76 (m, 2H), 1.65 – 1.59 (m, 2H). **¹³C NMR** (100 MHz, CDCl₃): δ 138.4, 136.1, 135.9, 130.7, 130.4, 129.6, 129.1, 125.9, 125.8, 121.2, 118.2, 116.5, 112.2, 38.3, 33.4, 25.7, 19.7. **ESI-
HRMS**: m/z calcd. for C₁₉H₂₀N₂ [M+H]⁺: 277.1705, found 277.1702.



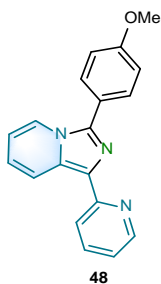
1-(pyridin-2-yl)-3-(thiophen-2-yl)imidazo[1,5-a]pyridine: Following the **General procedure B2**, a mixture of di(pyridin-2-yl)methanone (0.1 mmol), thiophene-2-carbaldehyde (4 equiv.) and D-glucosamine·HCl (2.0 equiv.) in AcOH : H₂O (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up gave product **45** (16.1 mg, 0.058 mmol, isolated yield 58%) as a yellow solid. **¹H NMR** (400 MHz, CDCl₃): δ 8.66 (d, J = 9.2 Hz, 1H), 8.55 (d, J = 4.0 Hz, 1H), 8.30 – 8.28 (m, 1H), 8.18 (d, J = 8.0 Hz, 1H), 7.68 – 7.63 (m, 1H), 7.51 (d, J = 3.6 Hz, 1H), 7.39 (d, J = 5.2, 1H), 7.16 – 7.13 (m, 1H), 7.05 – 7.02 (m, 1H), 6.90 – 6.86 (m, 1H), 6.74 – 6.67 (m, 1H). **¹³C NMR** (100 MHz, CDCl₃): δ 154.8, 149.0, 136.3, 132.6, 132.1, 130.8, 130.4, 127.7, 126.4, 125.5, 122.0, 121.9, 121.1, 120.7, 120.1, 114.5. The compound is known, and the NMR data is in accordance with the previous literature.²³⁴



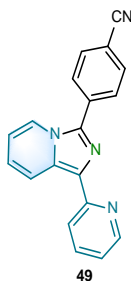
2-(1-(pyridin-2-yl)imidazo[1,5-a]pyridin-3-yl)phenol: Following the **General procedure B2**, a mixture of di(pyridin-2-yl)methanone (0.1 mmol), 2-hydroxybenzaldehyde (4 equiv.) and D-glucosamine·HCl (2.0 equiv.) in AcOH : H₂O (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up gave product **46** (15.8 mg, 0.055 mmol, isolated yield 55%) as a yellow solid. ¹H NMR (400 MHz, CDCl₃): δ 8.72 (d, *J* = 9.2 Hz, 1H), 8.57 – 8.55 (m, 1H), 8.47 (d, *J* = 7.6 Hz, 1H), 8.04 (d, *J* = 8.0 Hz, 1H), 7.71 (dd, *J* = 7.6, 1.6 Hz, 1H), 7.67 (td, *J* = 8.0, 2.0 Hz, 1H), 7.28 – 7.24 (m, 1H), 7.11 (dd, *J* = 8.0, 1.2 Hz, 1H), 7.07 – 7.04 (m, 1H), 6.97 – 6.89 (m, 2H), 6.71 – 6.67 (m, 1H). ¹³C NMR (100 MHz, CDCl₃): δ 156.3, 153.8, 149.1, 136.4, 135.5, 130.1, 129.6, 128.6, 124.5, 122.3, 122.2, 121.7, 120.9, 119.8, 119.1, 117.8, 114.8, 114.0. The compound is known, and the NMR data is in accordance with the previous literature.³³⁰



3-(2-methoxyphenyl)-1-(pyridin-2-yl)imidazo[1,5-a]pyridine: Following the **General procedure B2**, a mixture of di(pyridin-2-yl)methanone (0.1 mmol), 2-methoxybenzaldehyde (4 equiv.) and D-glucosamine·HCl (2.0 equiv.) in AcOH : H₂O (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up gave product **47** (19.3 mg, 0.064 mmol, isolated yield 64%) as a yellow solid. ¹H NMR (400 MHz, CDCl₃): δ 8.63 – 8.61 (m, 1H), 8.56 – 8.54 (m, 1H), 8.18 – 8.15 (m, 1H), 7.64 – 7.58 (m, 2H), 7.55 – 7.53 (m, 1H), 7.44 – 7.39 (m, 1H), 7.09 – 7.04 (m, 1H), 7.01 – 6.97 (m, 2H), 6.88 – 6.84 (m, 1H), 6.55 – 6.51 (m, 1H), 3.73 (s, 3H). ¹³C NMR (100 MHz, CDCl₃): δ 157.6, 155.3, 148.9, 136.2, 136.1, 132.8, 131.0, 130.2, 130.0, 123.3, 121.3, 121.2, 120.9, 120.2, 119.8, 119.1, 112.7, 111.2, 55.6. The compound is known, and the NMR data is in accordance with the previous literature.²²⁶

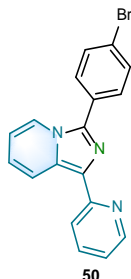


3-(4-methoxyphenyl)-1-(pyridin-2-yl)imidazo[1,5-a]pyridine: Following the **General procedure B2**, a mixture of di(pyridin-2-yl)methanone (0.1 mmol), 4-methoxybenzaldehyde (4 equiv.) and D-glucosamine·HCl (2.0 equiv.) in AcOH : H₂O (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up gave product **48** (21.2 mg, 0.067 mmol, isolated yield 67%) as a yellow solid. **¹H NMR** (400 MHz, CDCl₃): δ 8.60 (d, *J* = 9.2 Hz, 1H), 8.55 (d, *J* = 4.8 Hz, 1H), 8.16 (d, *J* = 8.0 Hz, 1H), 8.10 (d, *J* = 7.2 Hz, 1H), 7.68 (d, *J* = 8.7 Hz, 2H), 7.65 – 7.61 (m, 1H), 7.02 – 6.98 (m, 3H), 6.82 (dd, *J* = 9.2, 6.4 Hz, 1H), 6.56 – 6.53 (m, 1H), 3.81 (s, 3H). **¹³C NMR** (100 MHz, CDCl₃): δ 160.1, 155.1, 148.9, 138.1, 136.2, 130.2, 129.9, 129.9, 122.5, 121.7, 121.6, 120.8, 120.3, 119.9, 114.5, 113.7, 55.4. The compound is known, and the NMR data is in accordance with the previous literature.³³⁰

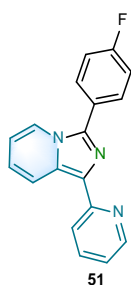


4-(1-(pyridin-2-yl)imidazo[1,5-a]pyridin-3-yl)benzonitrile: Following the **General procedure B2**, a mixture of di(pyridin-2-yl)methanone (0.1 mmol), 4-formylbenzonitrile (4 equiv.) and D-glucosamine·HCl (2.0 equiv.) in AcOH : H₂O (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up gave product **49** (24.0 mg, 0.081 mmol, isolated yield 81 %) as a yellow solid. **Mp**: 140–142 °C. **FT-IR**: ν (cm⁻¹): 3384, 3309, 3200, 3058, 2928, 2850, 2230, 2217, 1667, 1603, 1591, 1564, 1513, 1484, 1457, 1434, 1409, 1352, 1311, 1278, 1255, 1170, 1148, 1105, 1084, 1061, 1008, 993, 946, 845, 816, 779, 748, 734, 721, 694, 661, 628, 616, 585, 558, 544, 525, 486, 433. **¹H NMR** (400 MHz, CDCl₃): δ 8.70 (d, *J* = 9.2 Hz, 1H), 8.57 (d, *J* = 4.8 Hz, 1H), 8.24 (d, *J* = 7.2 Hz, 1H), 8.16 (d, *J* = 8.0 Hz, 1H), 7.96 – 7.94 (m, 2H), 7.75 (d, *J* = 7.6, 2H), 7.69 – 7.65 (m, 1H), 7.08 – 7.05 (m, 1H), 6.96 – 6.91 (m, 1H), 6.71 – 6.68 (m, 1H). **¹³C NMR** (100 MHz, CDCl₃): δ 154.5, 149.0, 136.4, 135.7, 134.5, 132.8, 131.9,

131.8, 131.1, 130.1, 128.2, 122.2, 121.8, 121.3, 120.9, 120.0, 118.6, 115.0, 111.8. **ESI-HRMS:** m/z calcd. for $C_{19}H_{12}N_4$ $[M+H]^+$: 297.1140, found 297.1136.

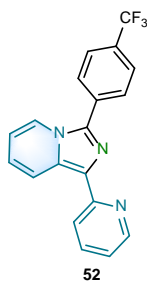


3-(4-bromophenyl)-1-(pyridin-2-yl)imidazo[1,5-a]pyridine: Following the **General procedure B2**, a mixture of di(pyridin-2-yl)methanone (0.1 mmol), 4-bromobenzaldehyde (4 equiv.) and D-glucosamine·HCl (2.0 equiv.) in AcOH : H₂O (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. work-up gave product **50** (26.5 mg, 0.076 mmol, isolated yield 76%) as a brown solid. **FT-IR:** ν (cm⁻¹): 2920, 2852, 2125, 1733, 1667, 1589, 1508, 1477, 1397, 1313, 1247, 1140, 1121, 1067, 1006, 950, 830, 785, 736, 696, 626, 593, 542, 488, 431, 406. **¹H NMR** (400 MHz, CDCl₃): δ 8.65 (d, J = 9.2 Hz, 1H), 8.56 – 8.55 (m, 1H), 8.16 – 8.13 (m, 2H), 7.68 – 7.60 (m, 5H), 7.05 – 7.02 (m, 1H), 6.89 – 6.85 (m, 1H), 6.61 (t, J = 6.8 Hz, 1H). **¹³C NMR** (100 MHz, CDCl₃): δ 154.8, 149.0, 136.8, 136.3, 132.3, 130.9, 130.4, 129.7, 129.1, 122.9, 122.0, 121.3, 121.2, 120.6, 119.9, 114.3. The compound is known, and the NMR data is in accordance with the previous literature.³³¹

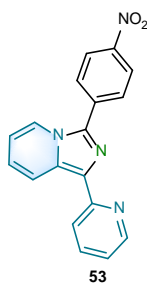


3-(4-fluorophenyl)-1-(pyridin-2-yl)imidazo[1,5-a]pyridine: Following the **General procedure B2**, a mixture of di(pyridin-2-yl)methanone (0.1 mmol), 4-fluorobenzaldehyde (4 equiv.) and D-glucosamine·HCl (2.0 equiv.) in AcOH : H₂O (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up gave product **51** (23.1 mg, 0.080 mmol, isolated yield 80%) as a yellow solid. **Mp:** 116–117 °C. **FT-IR:** ν (cm⁻¹): 2930, 2112, 1630, 1591, 1562, 1527, 1513, 1480, 1405, 1350, 1311, 1280,

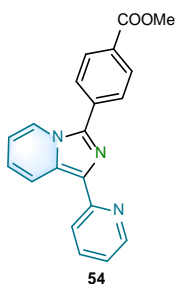
1251, 1216, 1193, 1154, 1088, 1010, 948, 837, 810, 787, 734, 713, 690, 657, 628, 587, 565, 513, 433, 406. **¹H NMR** (400 MHz, CDCl₃): δ 8.62 (d, *J* = 9.2 Hz, 1H), 8.56 – 8.54 (m, 1H), 8.14 (d, *J* = 8.0, 1H), 8.08 (d, *J* = 7.2 Hz, 1H), 7.75 – 7.72 (m, 2H), 7.65 – 7.61 (m, 1H), 7.18 – 7.14 (m, 2H), 7.03 – 7.00 (m, 1H), 6.86 – 6.82 (m, 1H), 6.59 – 6.55 (m, 1H). **¹³C NMR** (100 MHz, CDCl₃): δ 163.0 (d, ¹*J*_{C-F} = 249.4 Hz), 154.9, 149.0, 137.0, 136.2, 130.6, 130.3 (d, ³*J*_{C-F} = 8.4 Hz), 130.1, 126.3 (d, ⁴*J*_{C-F} = 3.4 Hz), 121.9, 121.3, 121.0, 120.5, 119.8, 116.2 (d, ²*J*_{C-F} = 21.8 Hz), 114.1. **¹⁹F NMR** (375 MHz, CDCl₃): δ -111.55. **ESI-
HRMS**: *m/z* calcd. for C₁₈H₁₂FN₃ [M+H]⁺: 290.1094, found 290.1090.



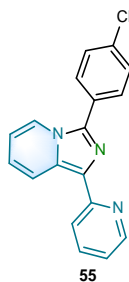
1-(pyridin-2-yl)-3-(4-(trifluoromethyl)phenyl)imidazo[1,5-a]pyridine: Following the **General procedure B2**, a mixture of di(pyridin-2-yl)methanone (0.1 mmol), 4-(trifluoromethyl)benzaldehyde (4 equiv.) and D-glucosamine·HCl (2.0 equiv.) in AcOH : H₂O (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up gave product **52** (28.1 mg, 0.083 mmol, isolated yield 83%) as a yellow solid. **¹H NMR** (400 MHz, CDCl₃): δ 8.69 – 8.66 (m, 1H), 8.57 – 8.55 (m, 1H), 8.12 – 8.15 (m, 2H), 7.92 (d, *J* = 8.0 Hz, 2H), 7.72 (d, *J* = 8.1 Hz, 2H), 7.68 – 7.63 (m, 1H), 7.06 – 7.02 (m, 1H), 6.91 – 6.87 (m, 1H), 6.66 – 6.62 (m, 1H). **¹³C NMR** (100 MHz, CDCl₃): δ 154.7, 149.0, 136.3, 136.3, 133.6 (q, ⁴*J*_{C-F} = 1.5 Hz), 131.3, 130.7, 130.5 (q, ²*J*_{C-F} = 32.7 Hz), 128.3, 126.0 (q, ³*J*_{C-F} = 3.8 Hz), 124.0 (q, ¹*J*_{C-F} = 272.1 Hz), 122.1, 121.4, 121.3, 120.7, 120.0, 114.6. **¹⁹F NMR** (375 MHz, CDCl₃): δ -62.66. The compound is known, and the NMR data is in accordance with the previous literature.³³²



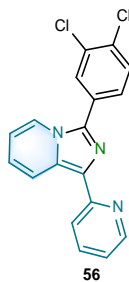
3-(4-nitrophenyl)-1-(pyridin-2-yl)imidazo[1,5-a]pyridine: Following the **General procedure B2**, a mixture of di(pyridin-2-yl)methanone (0.1 mmol), 4-nitrobenzaldehyde (4 equiv.) and D-glucosamine·HCl (2.0 equiv.) in AcOH : H₂O (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up gave product **53** (25.6 mg, 0.081 mmol, isolated yield 81%) as a red solid. **Mp:** 225–226 °C. **FT-IR:** ν (cm⁻¹): 2916, 1683, 1589, 1566, 1519, 1504, 1480, 1449, 1412, 1346, 1321, 1306, 1249, 1175, 1100, 1084, 1014, 950, 851, 785, 729, 694, 628, 591, 525, 478, 424. **¹H NMR** (300 MHz, CDCl₃): δ 8.75 (dt, J = 9.2, 1.2 Hz, 1H), 8.59 (ddd, J = 4.9, 1.7, 0.9 Hz, 1H), 8.36 – 8.23 (m, 3H), 8.21 – 8.18 (m, 1H), 8.06 – 8.01 (m, 2H), 7.73 – 7.67 (m, 1H), 7.09 (ddd, J = 7.4, 4.9, 1.1 Hz, 1H), 6.98 (ddd, J = 9.2, 6.5, 0.9 Hz, 1H), 6.74 (ddd, J = 7.1, 6.4, 1.2 Hz, 1H). **¹³C NMR** (100 MHz, CDCl₃): δ 148.9, 147.2, 136.6, 136.3, 136.3, 135.8, 135.5, 131.4, 128.2, 124.4, 122.3, 122.1, 121.4, 121.1, 120.2, 115.3. **ESI-HRMS:** m/z calcd. for C₁₈H₁₂N₄O₂ [M+H]⁺: 317.1039, found 317.1036.



methyl 4-(1-(pyridin-2-yl)imidazo[1,5-a]pyridin-3-yl)benzoate: Following the **General procedure B2**, a mixture of di(pyridin-2-yl)methanone (0.1 mmol), methyl-4-formylbenzoate (4 equiv.) and D-glucosamine·HCl (2.0 equiv.) in AcOH : H₂O (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up gave product **54** (13.5 mg, 0.041 mmol, isolated yield 41%) as a yellow liquid. **FT-IR:** ν (cm⁻¹): 2918, 1719, 1681, 1581, 1523, 1467, 1430, 1317, 1278, 1189, 1105, 1018, 993, 946, 868, 824, 777, 746, 696, 664, 614, 534, 439. **¹H NMR** (300 MHz, CDCl₃): δ 8.70 (d, J = 9.3 Hz, 1H), 8.58 (ddd, J = 5.1, 1.5, 0.9 Hz, 1H), 8.28 – 8.25 (m, 1H), 8.21 – 8.18 (m, 1H), 8.17 – 8.13 (m, 2H), 7.93 – 7.89 (m, 2H), 7.71 – 7.65 (m, 1H), 7.07 (ddd, J = 7.2, 5.1, 1.2 Hz, 1H), 6.95 – 6.90 (m, 1H), 6.70 – 6.65 (m, 1H), 3.91 (s, 3H). **¹³C NMR** (100 MHz, CDCl₃): δ 166.6, 149.7, 149.2, 137.0, 136.8, 130.3, 129.8, 129.0, 127.9, 126.4, 125.3, 123.5, 122.0, 121.6, 121.3, 120.8, 52.3. **ESI-HRMS:** m/z calcd. for C₁₈H₁₅N₃O₂ [M+H]⁺: 330.1243, found 330.1236.

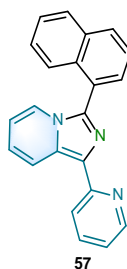


3-(4-chlorophenyl)-1-(pyridin-2-yl)imidazo[1,5-a]pyridine: Following the **General procedure B2**, a mixture of di(pyridin-2-yl)methanone (0.1 mmol), 4-chlorobenzaldehyde (4 equiv.) and D-glucosamine·HCl (2.0 equiv.) in AcOH : H₂O (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up gave product **55** (25.0 mg, 0.082 mmol, isolated yield 82%) as a yellow solid. **Mp:** 137–138 °C. **FT-IR:** ν (cm⁻¹): 2922, 2854, 2100, 1673, 1587, 1560, 1502, 1480, 1403, 1313, 1245, 1197, 1173, 1146, 1123, 1086, 1010, 952, 833, 781, 740, 713, 699, 628, 608, 596, 581, 492, 429, 406. **¹H NMR** (400 MHz, CDCl₃): δ 8.65 (d, J = 10.4 Hz, 1H), 8.56 (s, 1H), 8.17 – 8.14 (m, 2H), 7.74 – 7.71 (m, 2H), 7.67 – 7.64 (m, 1H), 7.47 – 7.44 (m, 2H), 7.05 – 7.03 (m, 1H), 6.89 – 6.86 (m, 1H), 6.63 – 6.60 (m, 1H). **¹³C NMR** (100 MHz, CDCl₃): δ 154.9, 149.0, 136.8, 136.3, 134.8, 130.9, 130.4, 129.5, 129.3, 128.6, 122.0, 121.4, 121.1, 120.6, 119.9, 114.2. **ESI-HRMS:** m/z calcd. for C₁₈H₁₂ClN₃ [M+H]⁺: 306.0798, found 306.0795.

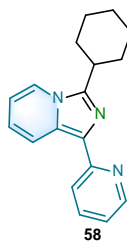


3-(3,4-dichlorophenyl)-1-(pyridin-2-yl)imidazo[1,5-a]pyridine: Following the **General procedure B2**, a mixture of di(pyridin-2-yl)methanone (0.1 mmol), 3,4-dichlorobenzaldehyde (4 equiv.) and D-glucosamine·HCl (2.0 equiv.) in AcOH : H₂O (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up gave product **56** (24.4 mg, 0.072 mmol, isolated yield 72%) as a yellow solid. **Mp:** 177–179 °C. **FT-IR:** ν (cm⁻¹): 2922, 2850, 2112, 1669, 1632, 1591, 1566, 1531, 1506, 1477, 1453, 1401, 1381, 1352, 1313, 1247, 1133, 1088, 1030, 1012, 954, 878, 828, 812, 783, 727, 703, 672, 628, 600, 532, 507, 455, 435, 412. **¹H NMR** (400 MHz, CDCl₃): δ 8.67 (d, J = 9.2 Hz, 1H), 8.56 (d, J = 4.0 Hz, 1H), 8.16 (d, J = 7.2 Hz, 2H), 7.92 (s, 1H), 7.68 – 7.63 (m, 2H), 7.55 (d, J = 8.0 Hz, 1H), 7.07 – 7.04 (m, 1H),

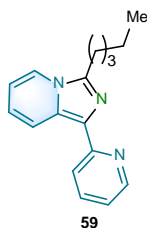
6.92 – 6.88 (m, 1H), 6.68 – 6.65 (m, 1H). ^{13}C NMR (100 MHz, CDCl_3): δ 154.7, 149.0, 136.3, 135.5, 133.4, 132.8, 131.2, 131.0, 130.6, 130.1, 130.0, 127.1, 122.1, 121.4, 121.2, 120.8, 120.0, 114.6. **ESI-HRMS**: m/z calcd. for $\text{C}_{18}\text{H}_{11}\text{Cl}_2\text{N}_3$ $[\text{M}+\text{H}]^+$: 340.0408, found 340.0404.



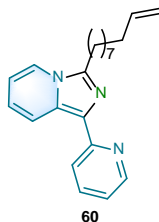
3-(naphthalen-1-yl)-1-(pyridin-2-yl)imidazo[1,5-a]pyridine: Following the **General procedure B2**, a mixture of di(pyridin-2-yl)methanone (0.1 mmol), 1-naphthaldehyde (4 equiv.) and D-glucosamine·HCl (2.0 equiv.) in AcOH : H_2O (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up gave product **57** (18.6 mg, 0.058 mmol, isolated yield 58%) as a yellow solid. **Mp**: 71–73 °C. **FT-IR**: ν (cm^{-1}): 3050, 2924, 2850, 1663, 1585, 1529, 1508, 1473, 1426, 1401, 1352, 1304, 1245, 1214, 1142, 1076, 999, 969, 938, 802, 775, 740, 729, 705, 670, 616, 596, 530, 511, 418. ^1H NMR (400 MHz, CDCl_3): δ 8.70 (d, J = 8.8 Hz, 1H), 8.60 (d, J = 4.0 Hz, 1H), 8.21 (d, J = 7.6 Hz, 1H), 7.96 (d, J = 8.0 Hz, 1H), 7.90 (d, J = 8.0 Hz, 1H), 7.73 (d, J = 7.2 Hz, 1H), 7.68 – 7.55 (m, 4H), 7.50 – 7.38 (m, 2H), 7.06 – 7.03 (m, 1H), 6.91 – 6.87 (m, 1H), 6.51 – 6.48 (m, 1H). ^{13}C NMR (100 MHz, CDCl_3): δ 155.2, 149.0, 136.8, 136.3, 134.0, 132.0, 130.4, 130.1, 129.8, 129.0, 128.6, 127.1, 126.4, 125.5, 125.4, 122.0, 121.7, 121.2, 120.4, 120.0, 113.6. **ESI-HRMS**: m/z calcd. for $\text{C}_{22}\text{H}_{15}\text{N}_3$ $[\text{M}+\text{H}]^+$: 322.1344, found 322.1342.



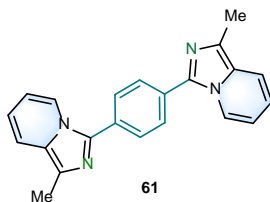
3-cyclohexyl-1-(pyridin-2-yl)imidazo[1,5-a]pyridine: Following the **General procedure B2**, a mixture of di(pyridin-2-yl)methanone (0.1 mmol), cyclohexanecarbaldehyde (4 equiv.) and D-glucosamine·HCl (2.0 equiv.) in AcOH : H₂O (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up gave product **58** (20.8 mg, 0.075 mmol, isolated yield 75%) as a yellow solid. **Mp:** 107–109 °C. **FT-IR:** ν (cm⁻¹): 2926, 2852, 1632, 1587, 1562, 1533, 1519, 1471, 1445, 1428, 1401, 1346, 1284, 1224, 1146, 1086, 1059, 1001, 958, 890, 822, 791, 736, 703, 633, 616, 598, 509, 424, 408. **¹H NMR** (400 MHz, CDCl₃): δ 8.62 – 8.58 (m, 2H), 8.18 (d, J = 8.0 Hz, 1H), 7.82 (d, J = 7.2 Hz, 1H), 7.72 – 7.68 (m, 1H), 7.07 – 7.04 (m, 1H), 6.84 (dd, J = 8.8, 6.4 Hz, 1H), 6.62 – 6.58 (m, 1H), 3.01 (tt, J = 11.3, 3.0 Hz, 1H), 2.10 (d, J = 13.2 Hz, 2H), 1.98 – 1.83 (m, 5H), 1.53 – 1.40 (m, 3H). **¹³C NMR** (100 MHz, CDCl₃): δ 155.3, 148.9, 142.8, 136.0, 128.9, 128.7, 121.6, 120.7, 119.9, 119.9, 119.7, 112.7, 35.8, 30.6, 26.2, 25.9. **ESI-HRMS:** m/z calcd. for C₁₈H₁₉N₃ [M+H]⁺: 278.1657, found 278.1654.



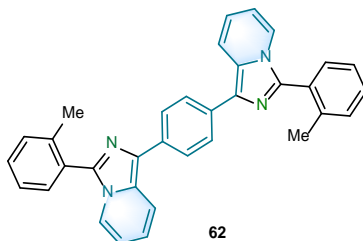
3-pentyl-1-(pyridin-2-yl)imidazo[1,5-a]pyridine: Following the **General procedure B2**, a mixture of di(pyridin-2-yl)methanone (0.1 mmol), hexanal (4 equiv.) and D-glucosamine·HCl (2.0 equiv.) in AcOH : H₂O (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up gave product **59** (20.4 mg, 0.077 mmol, isolated yield 77%) as a brown liquid. **FT-IR:** ν (cm⁻¹): 3037, 2936, 2858, 1686, 1681, 1630, 1587, 1533, 1519, 1463, 1428, 1405, 1374, 1341, 1323, 1286, 1241, 1220, 1179, 1148, 1115, 1088, 1065, 1028, 1004, 954, 886, 830, 787, 762, 734, 723, 699, 674, 631, 614, 579, 418, 404. **¹H NMR** (400 MHz, CDCl₃): δ 8.52 – 8.48 (m, 2H), 8.04 (d, J = 8.4 Hz, 1H), 7.66 (d, J = 7.2 Hz, 1H), 7.60 (td, J = 7.6, 1.6 Hz, 1H), 6.96 (ddd, J = 7.2, 4.8, 1.2 Hz, 1H), 6.76 (ddd, J = 9.2, 6.4, 0.8 Hz, 1H), 6.55 – 6.51 (m, 1H), 2.93 (t, J = 8.0 Hz, 2H), 1.78 (p, J = 8.0 Hz, 2H), 1.38 – 1.28 (m, 4H), 0.84 (t, J = 6.8 Hz, 3H). **¹³C NMR** (100 MHz, CDCl₃): δ 155.1, 148.9, 139.1, 136.1, 129.0, 128.7, 121.6, 120.8, 120.0, 120.0, 119.6, 113.1, 31.6, 26.8, 26.7, 22.4, 14.0. **ESI-HRMS:** m/z calcd. for C₁₇H₁₉N₃ [M+H]⁺: 266.1657, found 266.1654.



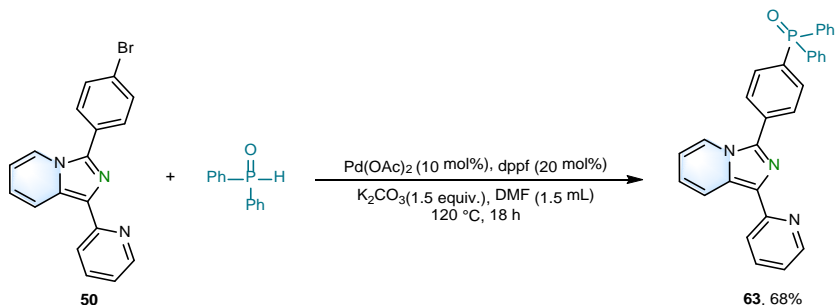
3-(dec-9-en-1-yl)-1-methylimidazo[1,5-a]pyridine: Following the **General procedure B2**, a mixture of di(pyridin-2-yl)methanone (0.1 mmol), undec-10-enal (4 equiv.) and D-glucosamine·HCl (2.0 equiv.) in AcOH : H₂O (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up gave product **60** (23.3 mg, 0.070 mmol, isolated yield 70%) as a dark liquid. **FT-IR:** ν (cm⁻¹): 3338, 3079, 2926, 2856, 1679, 1640, 1587, 1537, 1521, 1473, 1428, 1403, 1381, 1321, 1276, 1238, 1144, 1086, 1049, 995, 960, 909, 787, 738, 725, 678, 631, 612, 420, 402. **¹H NMR** (400 MHz, CDCl₃): δ 8.61 – 8.57 (m, 2H), 8.13 (d, J = 8.0 Hz, 1H), 7.74 (d, J = 7.2 Hz, 1H), 7.71 – 7.66 (m, 1H), 7.07 – 7.03 (m, 1H), 6.86 – 6.82 (m, 1H), 6.63 – 6.59 (m, 1H), 5.87 – 5.77 (m, 1H), 5.03 – 4.92 (m, 2H), 3.04 – 2.99 (m, 2H), 2.05 (q, J = 7.0 Hz, 2H), 1.87 (p, J = 7.7 Hz, 2H), 1.49 – 1.44 (m, 2H), 1.40 – 1.35 (m, 4H), 1.34 – 1.29 (m, 4H). **¹³C NMR** (100 MHz, CDCl₃): δ 155.2, 148.9, 139.1, 139.0, 136.0, 129.0, 128.7, 121.5, 120.7, 119.9, 119.9, 119.5, 114.1, 113.0, 33.7, 29.4, 29.3, 29.2, 29.0, 28.8, 27.1, 26.7. **ESI-HRMS:** m/z calcd. for C₂₂H₂₇N₃ [M+H]⁺: 334.2283, found 334.2280.



1,4-bis(1-methylimidazo[1,5-a]pyridin-3-yl)benzene: a mixture of 2-acetylpyridine (0.3 mmol), terephthalaldehyde (0.1 mmol) and D-glucosamine·HCl (0.4 mmol) in AcOH : H₂O (1.8 : 0.2 mL) were stirred at 120 °C under Ar atmosphere for 3 days. Work-up gave product **61** (17.2 mg, 0.051 mmol, isolated yield 51%) as a brown red solid. **Mp:** 212–214 °C. **FT-IR:** ν (cm⁻¹): 3070, 3035, 2920, 2858, 2110, 1927, 1663, 1630, 1537, 1510, 1442, 1424, 1407, 1356, 1327, 1306, 1253, 1179, 1115, 1078, 1008, 942, 837, 736, 678, 581, 548, 503, 422, 410. **¹H NMR** (400 MHz, CDCl₃): δ 8.18 – 8.16 (m, 2H), 7.87 – 7.86 (m, 4H), 7.37 – 7.33 (m, 2H), 6.60 – 6.55 (m, 2H), 6.49 – 6.45 (m, 2H), 2.52 (s, 6H). **¹³C NMR** (100 MHz, ν): δ 136.0, 130.1, 129.3, 128.3, 128.0, 121.2, 118.4, 117.2, 113.2, 12.6. **ESI-HRMS:** m/z calcd. for C₂₂H₁₈N₄ [M+H]⁺: 338.1531, found 338.1528.

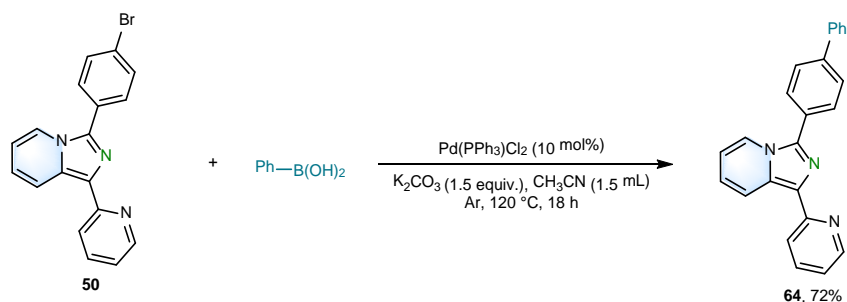


1,4-bis(3-(o-tolyl)imidazo[1,5-a]pyridin-1-yl)benzene: A mixture of 1,4-phenylenebis(pyridin-2-ylmethanone) (0.1 mmol), 2-methylbenzaldehyde (0.6 mmol) and D-glucosamine·HCl (0.4 mmol) in AcOH : H₂O (2.7 : 0.3 mL) were stirred at 120 °C under Ar atmosphere for 3 days. Work-up gave product **62** (17.6 mg, 0.036 mmol, isolated yield 36%) as a brown solid. **Mp**: 98–100 °C. **FT-IR**: ν (cm⁻¹): 3058, 2920, 2852, 2110, 1869, 1745, 1661, 1599, 1510, 1455, 1428, 1377, 1356, 1306, 1276, 1241, 1158, 1105, 1043, 1004, 942, 843, 804, 773, 746, 727, 709, 616, 579, 519, 449, 424. **¹H NMR** (400 MHz, CDCl₃): δ 8.01 (s, 4H), 7.87 – 7.84 (m, 2H), 7.55 (d, J = 7.6 Hz, 2H), 7.46 (d, J = 7.5 Hz, 2H), 7.36 – 7.25 (m, 6H), 6.76 – 6.72 (m, 2H), 6.48 – 6.45 (m, 2H), 2.21 (s, 6H). **¹³C NMR** (100 MHz, CDCl₃): δ 138.6, 137.8, 133.1, 131.2, 130.8, 130.7, 129.5, 129.3, 126.8, 126.7, 126.1, 121.9, 119.4, 119.2, 112.8, 19.8. **ESI-MS**: m/z calcd. for C₃₄H₂₆N₄ [M+H]⁺: 491.2236, found 491.2240.

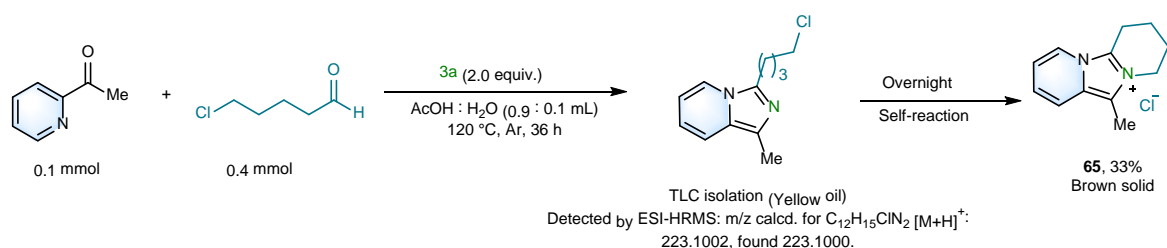


A mixture of 3-(4-bromophenyl)-1-(pyridin-2-yl)imidazo[1,5-a]pyridine **50** (0.1 mmol), diphenylphosphine oxide (1.5 equiv.), Pd(OAc)₂ (10 mol%), dppf (20 mol%) and K₂CO₃ (1.5 equiv.) in the DMF (1.5 mL) were stirred at 120 °C under Ar atmosphere for 18 h. The reactions were conducted in a sealed Schlenk tube and heated by an IKA magnetic heating agitator with heating block. The reaction temperature was directly read from temperature detector of IKA apparatus and was calibrated by thermometer. After the reaction completely, the reaction solution was filtrated and concentrated in rotary evaporator, the crude product was purified with flash chromatography on silica gel (ethyl acetate : *n*-hexane : Et₃N) to give products **63** (32.0 mg, isolated yield 68%) as a yellow solid. **Mp**: 102–104 °C; **FT-IR**: ν (cm⁻¹): 2918, 2623, 2498, 1679, 1589, 1510, 1477, 1434, 1395, 1311, 1251, 1175, 1109, 1010, 995, 948, 841, 791, 744, 719, 692, 643, 552, 538, 422. **¹H NMR** (400 MHz, CDCl₃): δ 8.65 (d, J = 9.2 Hz, 1H),

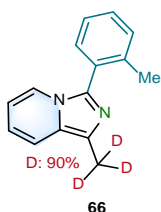
8.53 (d, $J = 2.4$ Hz, 1H), 8.21 (d, $J = 7.2$ Hz, 1H), 8.14 (d, $J = 8.0$ Hz, 1H), 7.89 (d, $J = 8.0$ Hz, 2H), 7.78 – 7.73 (m, 2H), 7.65 – 7.60 (m, 5H), 7.51 – 7.47 (m, 2H), 7.42 – 7.39 (m, 4H), 7.03 – 7.00 (m, 1H), 6.90 – 6.85 (m, 1H), 6.63 – 6.60 (m, 1H). ^{13}C NMR (100 MHz, CDCl_3): δ 154.5, 149.1, 148.8, 136.6 (d, $J_{\text{C-P}} = 15.6$ Hz), 136.3, 133.5 (d, $J_{\text{C-P}} = 2.9$ Hz), 132.7 (d, $J_{\text{C-P}} = 10.1$ Hz), 132.1 (d, $J_{\text{C-P}} = 2.7$ Hz), 132.0 (d, $J_{\text{C-P}} = 10.0$ Hz), 131.1, 130.7, 128.5 (d, $^2J_{\text{C-P}} = 12.2$ Hz), 127.8 (d, $^2J_{\text{C-P}} = 12.2$ Hz), 125.8 (d, $^1J_{\text{C-P}} = 116.6$ Hz), 121.9, 121.5, 121.4, 120.6, 119.9, 114.5. ^{31}P NMR (162 MHz, CDCl_3): δ 29.0. **EI-HRMS (FTMS + p EI Full ms)**: m/z calcd. for $\text{C}_{30}\text{H}_{22}\text{N}_3\text{OP}$ [M]: 471.1500, found 471.1486.



A mixture of 3-(4-bromophenyl)-1-(pyridin-2-yl)imidazo[1,5-a]pyridine **50** (0.1 mmol), phenylboronic acid (1.5 equiv.), $\text{Pd}(\text{PPh}_3)\text{Cl}_2$ (10 mol%) and K_2CO_3 (1.5 equiv.) in the CH_3CN (1.5 mL) were stirred at 120 °C under Ar atmosphere for 18 h. The reactions were conducted in a sealed Schlenk tube and heated by an IKA magnetic heating agitator with heating block. The reaction temperature was directly read from temperature detector of IKA apparatus and was calibrated by thermometer. After reaction completely, the reaction mixture was filtrated and concentrated in rotary evaporator, the crude product was purified with flash chromatography on silica gel (ethyl acetate : *n*-hexane : Et_3N) to give products **64** (25.0 mg, isolated yield 72%) as a green solid. **Mp**: 160–162 °C; **FT-IR**: ν (cm^{-1}): 2965, 2920, 2850, 2112, 1673, 1585, 1560, 1529, 1513, 1496, 1480, 1447, 1430, 1403, 1335, 1313, 1302, 125, 1193, 1140, 1086, 1008, 948, 853, 789, 760, 738, 723, 696, 664, 631, 596, 550, 517, 482, 435. ^1H NMR (300 MHz, CDCl_3): δ 8.68 (d, $J = 9.0$ Hz, 1H), 8.58 (d, $J = 4.2$ Hz, 1H), 8.24 (dd, $J = 14.5, 7.6$ Hz, 2H), 7.87 (d, $J = 8.1$ Hz, 2H), 7.71 (d, $J = 7.8$ Hz, 2H), 7.66 (d, $J = 8.1$ Hz, 1H), 7.61 (d, $J = 8.4$ Hz, 2H), 7.42 (t, $J = 7.5$ Hz, 2H), 7.34 (d, $J = 6.9$ Hz, 1H), 7.07 – 7.03 (m, 1H), 6.89 (dd, $J = 9.0, 6.6$ Hz, 1H), 6.63 (t, $J = 6.8$ Hz, 1H). ^{13}C NMR (100 MHz, CDCl_3): δ 154.8, 148.8, 141.7, 140.4, 137.9, 136.5, 132.6, 130.4, 129.1, 128.9, 128.7, 127.7, 127.2, 127.1, 121.9, 121.7, 121.2, 120.5, 120.1, 114.1. **ESI-HRMS**: m/z calcd. for $\text{C}_{24}\text{H}_{17}\text{N}_3$ $[\text{M}+\text{H}]^+$: 348.1501, found 348.1498.

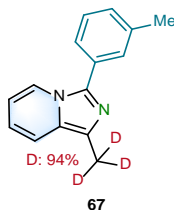


A mixture of 2-acetylpyridine (0.1 mmol), 5-chloropentanal (4.0 equiv.), D-glucosamine·HCl (2.0 equiv.) in the AcOH : H₂O (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. The reactions were conducted in a sealed Schlenk tube and heated by an IKA magnetic heating agitator with heating block. The reaction temperature was directly read from temperature detector of IKA apparatus and was calibrated by thermometer. After cooling to room temperature, the reaction mixture was filtrated and concentrated in rotary evaporator, the crude product was purified with flash chromatography on silica gel (ethyl acetate : *n*-hexane : Et₃N) to give products 3-(4-chlorobutyl)-1-methylimidazo[1,5-a]pyridine as a yellow oil, which then transform to final product **65** in glass bottle through self-reaction under air atmosphere in room temperature overnight. Finally the crude product **65** was washed by *n*-hexane (2 mL) for 3 times and ethyl acetate (2 mL) for 3 times to obtain pure product **65** (25.0 mg, overall isolated yield 33%). **Mp**: 115–117 °C. **FT-IR**: ν (cm⁻¹): 3436, 3379, 3031, 2953, 2926, 2858, 1723, 1653, 1611, 1533, 1434, 1358, 1255, 1041, 806, 744, 536. **¹H NMR** (300 MHz, CD₃OD): δ 8.20 (d, J = 7.2 Hz, 1H), 7.77 (d, J = 9.3 Hz, 1H), 7.18 – 7.06 (m, 2H), 4.39 (t, J = 5.7 Hz, 2H), 3.29 (t, J = 6.2 Hz, 2H), 2.65 (s, 3H), 2.31 – 2.17 (m, 4H). **¹³C NMR** (100 MHz, CD₃OD): δ 134.7, 127.1, 123.5, 122.4, 121.6, 118.7, 118.0, 45.7, 22.6, 21.5, 19.0, 7.8. **ESI-HRMS**: m/z calcd. for $C_{12}H_{15}N_2$ $[M]$: 187.1230, found 187.1240.

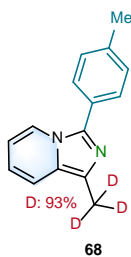


1-(methyl-d₃)-3-(o-tolyl)imidazo[1,5-a]pyridine: Following the **General procedure B1**, a mixture of 2-acetylpyridine (0.1 mmol), 2-methylbenzaldehyde (2 equiv.) and D-glucosamine·HCl (2.0 equiv.) in AcOH-d₄ : D₂O (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up gave product **66** (10.8 mg, 0.048 mmol, isolated yield 48%) as a yellow solid. **Mp**: 60–61 °C. **FT-IR**: ν (cm⁻¹): 3050, 2918, 2852, 1669, 1517, 1453, 1412, 1362, 1247, 1100, 1044, 1016, 993, 958, 773, 727, 678, 649, 451, 422. D incorporation was tested by ¹H NMR: 90%. **¹H NMR** (300 MHz, CDCl₃): δ 7.43 (d, J = 7.2

Hz, 1H), 7.36 – 7.22 (m, 5H), 6.53 (dd, $J = 9.0, 6.3$ Hz, 1H), 6.37 – 6.33 (m, 1H), 2.47 (0.3H), 2.13 (s, 3H). **ESI-HRMS**: m/z calcd. for $C_{15}H_{11}D_3N_2$ $[M+H]^+$: 226.1424, found 226.1420.

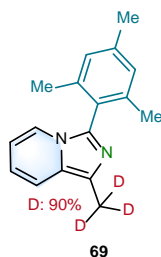


1-(methyl-d3)-3-(m-tolyl)imidazo[1,5-a]pyridine: Following the **General procedure B1**, a mixture of 2-acetylpyridine (0.1 mmol), 3-methylbenzaldehyde (2 equiv.) and D-glucosamine·HCl (2.0 equiv.) in $AcOH-d_4 : D_2O$ (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up gave product **67** (13.3 mg, 0.059 mmol, isolated yield 59%) as a yellow liquid. **FT-IR**: ν (cm^{-1}): 3050, 2918, 2854, 1671, 1605, 1583, 1517, 1465, 1362, 1255, 1203, 1090, 1028, 997, 971, 884, 791, 734, 696, 657, 618, 528, 439, 420. D incorporation was tested by 1H NMR: 94%. **1H NMR** (300 MHz, $CDCl_3$): δ 8.11 (d, $J = 7.2$ Hz, 1H), 7.54 (s, 1H), 7.49 (d, $J = 7.8$ Hz, 1H), 7.33 – 7.28 (m, 2H), 7.15 (d, $J = 7.8$ Hz, 1H), 6.54 (dd, $J = 9.1, 6.3$ Hz, 1H), 6.41 (t, $J = 6.9$ Hz, 1H), 2.47 (0.19H), 2.36 (s, 3H). **ESI-HRMS**: m/z calcd. for $C_{15}H_{11}D_3N_2$ $[M+H]^+$: 226.1424, found 226.1420.

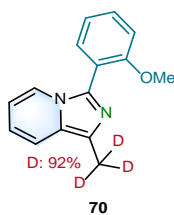


1-(methyl-d3)-3-(p-tolyl)imidazo[1,5-a]pyridine: Following the **General procedure B1**, a mixture of 2-acetylpyridine (0.1 mmol), 4-methylbenzaldehyde (2 equiv.) and D-glucosamine·HCl (2.0 equiv.) in $AcOH-d_4 : D_2O$ (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up gave product **68** (15.1 mg, 0.067 mmol, isolated yield 67%) as a yellow solid. **Mp**: 49-50 °C. **FT-IR**: ν (cm^{-1}): 3019, 2916, 2854, 1661, 1609, 1529, 1457, 1409, 1364, 1251, 1177, 1111, 1041, 1016, 958, 820, 725, 676, 637, 573, 490, 422. D incorporation was tested by 1H NMR: 93%. **1H NMR** (300 MHz, $CDCl_3$): δ 8.07 (d, $J = 7.2$ Hz, 1H), 7.59 (d, $J = 8.1$ Hz, 2H), 7.30 (d, $J = 9.0$ Hz, 1H), 7.23 (d, $J = 7.8$ Hz, 2H), 6.53 (dd, $J =$

9.1, 6.3 Hz, 1H), 6.42 – 6.37 (m, 1H), 2.46 (0.22H), 2.34 (s, 3H). **ESI-HRMS**: m/z calcd. for $C_{15}H_{11}D_3N_2$ $[M+H]^+$: 226.1424, found 226.1419.

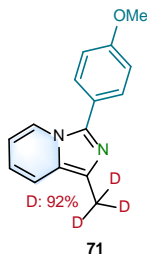


3-mesityl-1-(methyl-d₃)imidazo[1,5-a]pyridine: Following the **General procedure B1**, a mixture of 2-acetylpyridine (0.1 mmol), 2,4,6-trimethylbenzaldehyde (2 equiv.) and D-glucosamine·HCl (2.0 equiv.) in AcOH-d₄ : D₂O (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up gave product **69** (11.4 mg, 0.045 mmol, isolated yield 45%) as a yellow liquid. **FT-IR**: ν (cm⁻¹): 2918, 2850, 1611, 1521, 1461, 1412, 1362, 1321, 1243, 1187, 1127, 1034, 1016, 993, 966, 940, 851, 773, 734, 682, 585, 420. D incorporation was tested by ¹H NMR: 90%. **¹H NMR** (300 MHz, CDCl₃): δ 7.31 (d, J = 9.3 Hz, 1H), 7.14 (d, J = 7.2 Hz, 1H), 6.89 (s, 2H), 6.51 (dd, J = 9.1, 6.3 Hz, 1H), 6.33 – 6.29 (m, 1H), 2.48 (0.30H), 2.27 (s, 3H), 1.88 (s, 6H). **ESI-HRMS**: m/z calcd. for $C_{17}H_{15}D_3N_2$ $[M+H]^+$: 254.1737, found 254.1733.

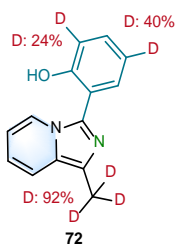


3-(2-methoxyphenyl)-1-(methyl-d₃)imidazo[1,5-a]pyridine: Following the **General procedure B1**, a mixture of 2-acetylpyridine (0.1 mmol), 2-methoxybenzaldehyde (2 equiv.) and D-glucosamine·HCl (2.0 equiv.) in AcOH-d₄ : D₂O (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up gave product **70** (14.7 mg, 0.061 mmol, isolated yield 61%) as a yellow solid. **Mp**: 66–68 °C. **FT-IR**: ν (cm⁻¹): 2926, 2838, 1657, 1632, 1601, 1578, 1519, 1463, 1434, 1414, 1370, 1241, 1160, 1105, 1045, 1016,

960, 793, 750, 732, 678, 649, 573, 521, 492, 422. D incorporation was tested by ^1H NMR: 92%. ^1H NMR (300 MHz, CDCl_3): δ 7.52 (dd, $J = 7.5, 1.5$ Hz, 1H), 7.44 (d, $J = 7.2$ Hz, 1H), 7.40 – 7.30 (m, 2H), 7.04 – 7.00 (m, 1H), 6.96 (d, $J = 8.4$ Hz, 1H), 6.55 (dd, $J = 9.0, 6.3$ Hz, 1H), 6.39 – 6.34 (m, 1H), 3.73 (s, 3H), 2.48 (0.25H). **ESI-HRMS**: m/z calcd. for $\text{C}_{15}\text{H}_{11}\text{D}_3\text{N}_2\text{O}$ $[\text{M}+\text{H}]^+$: 242.1373, found 242.1368.

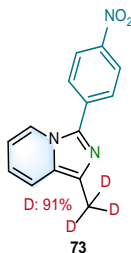


3-(4-methoxyphenyl)-1-(methyl- d_3)imidazo[1,5-a]pyridine: Following the **General procedure B1**, a mixture of 2-acetylpyridine (0.1 mmol), 4-methoxybenzaldehyde (2 equiv.) and D-glucosamine·HCl (2.0 equiv.) in AcOH-d_4 : D_2O (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up gave product **71** (15.7 mg, 0.065 mmol, isolated yield 65%) as a yellow solid. **Mp**: 50–52 °C. **FT-IR**: ν (cm^{-1}): 2998, 2932, 2835, 1665, 1607, 1570, 1527, 1461, 1438, 1407, 1366, 1300, 1288, 1247, 1170, 1111, 1020, 958, 833, 727, 676, 604, 575, 513, 416. D incorporation was tested by ^1H NMR: 92%. ^1H NMR (300 MHz, CDCl_3): δ 8.02 (d, $J = 7.3$ Hz, 1H), 7.62 (d, $J = 8.7$ Hz, 2H), 7.30 (d, $J = 9.0$ Hz, 1H), 6.96 (d, $J = 8.7$ Hz, 2H), 6.52 (dd, $J = 9.0, 6.3$ Hz, 1H), 6.41 – 6.37 (m, 1H), 3.80 (s, 3H), 2.46 (0.23H). **ESI-HRMS**: m/z calcd. for $\text{C}_{15}\text{H}_{11}\text{D}_3\text{N}_2\text{O}$ $[\text{M}+\text{H}]^+$: 242.1373, found 242.1368.

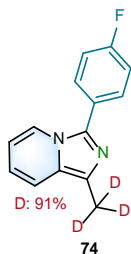


2-(1-(methyl- d_3)imidazo[1,5-a]pyridin-3-yl)phen-4,6- d_2 -ol: Following the **General procedure B1**, a mixture of 2-acetylpyridine (0.1 mmol), 2-hydroxybenzaldehyde (2 equiv.) and D-glucosamine·HCl (2.0 equiv.) in AcOH-d_4 : D_2O (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up

gave product **72** (11.9 mg, 0.052 mmol, isolated yield 52%) as a white solid. **Mp**: 129–131 °C. **FT-IR**: ν (cm⁻¹): 2924, 2850, 2120, 1741, 1634, 1607, 1578, 1519, 1436, 1416, 1372, 1292, 1259, 1234, 1179, 1096, 1041, 1018, 964, 913, 839, 808, 750, 727, 680, 641, 583, 544, 523, 420. D incorporation was tested by ¹H NMR: 92%, 40% and 24%. **¹H NMR** (300 MHz, CDCl₃): δ 8.37 (d, J = 7.3 Hz, 1H), 7.69 – 7.66 (m, 1H), 7.37 (d, J = 9.0 Hz, 1H), 7.19 (s, 1H), 7.07 (0.60H), 6.91 (0.76H), 6.63 (dd, J = 9.0, 6.3 Hz, 1H), 6.57 – 6.52 (m, 1H), 2.45 (0.26H). **ESI-HRMS**: m/z calcd. for C₁₄H₇D₅N₂O [M+H]⁺: 230.1342, found 230.1330.

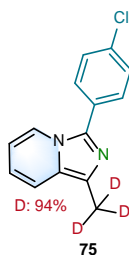


1-(methyl-d₃)-3-(4-nitrophenyl)imidazo[1,5-a]pyridine: Following the **General procedure B1**, a mixture of 2-acetylpyridine (0.1 mmol), 4-nitrobenzaldehyde (2 equiv.) and D-glucosamine·HCl (2.0 equiv.) in AcOH-d₄ : D₂O (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up gave product **73** (17.7 mg, 0.069 mmol, isolated yield 69%) as a red solid. **Mp**: 49–51 °C. **FT-IR**: ν (cm⁻¹): 2951, 2918, 2850, 1681, 1591, 1510, 1422, 1344, 1309, 1255, 1175, 1098, 1078, 1016, 956, 913, 847, 750, 736, 719, 690, 666, 521, 490, 474, 420. D incorporation was tested by ¹H NMR: 91%. **¹H NMR** (400 MHz, CDCl₃): δ 8.30 – 8.27 (m, 2H), 8.23 (d, J = 7.2 Hz, 1H), 7.95 – 7.93 (m, 2H), 7.41 (d, J = 8.8 Hz, 1H), 6.71 – 6.68 (m, 1H), 6.62 – 6.58 (m, 1H), 2.49 (0.27H). **ESI-HRMS**: m/z calcd. for C₁₄H₈D₃N₃O₂ [M+H]⁺: 257.1118, found 257.1117.

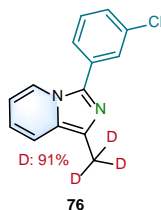


3-(4-fluorophenyl)-1-(methyl-d₃)imidazo[1,5-a]pyridine: Following the **General procedure B1**, a mixture of 2-acetylpyridine (0.1 mmol), 4-fluorobenzaldehyde (2 equiv.) and D-glucosamine·HCl (2.0 equiv.) in AcOH-d₄ : D₂O (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up

gave product **74** (16.0 mg, 0.070 mmol, isolated yield 70%) as a yellow solid. **Mp**: 99-100 °C. **FT-IR**: ν (cm⁻¹): 2955, 2921, 2852, 1659, 1603, 1525, 1506, 1461, 1414, 1377, 1222, 1156, 1096, 1014, 962, 839, 812, 765, 736, 673, 577, 517. D incorporation was tested by ¹H NMR: 91%. **¹H NMR** (300 MHz, CDCl₃): δ 8.02 (d, J = 6.9 Hz, 1H), 7.68 (dd, J = 8.4, 5.4 Hz, 2H), 7.32 (d, J = 9.3 Hz, 1H), 7.12 (t, J = 8.7 Hz, 2H), 6.58 – 6.53 (m, 1H), 6.45 – 6.41 (m, 1H), 2.45 (0.26H). **ESI-HRMS**: m/z calcd. for C₁₄H₈D₃N₂F [M+H]⁺: 230.1173, found 230.1168.

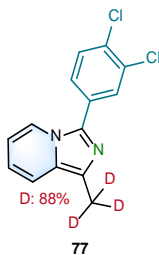


3-(4-chlorophenyl)-1-(methyl-d₃)imidazo[1,5-a]pyridine: Following the **General procedure B1**, a mixture of 2-acetylpyridine (0.1 mmol), 4-chlorobenzaldehyde (2 equiv.) and D-glucosamine·HCl (2.0 equiv.) in AcOH-d₄ : D₂O (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up gave product **75** (14.9 mg, 0.061 mmol, isolated yield 61%) as a yellow solid. **Mp**: 83–85 °C. **FT-IR**: ν (cm⁻¹): 2951, 2914, 2844, 1659, 1591, 1506, 1480, 1403, 1374, 1253, 1173, 1088, 1012, 960, 833, 725, 670, 523, 484, 424. D incorporation was tested by ¹H NMR: 94%. **¹H NMR** (300 MHz, CDCl₃): δ 8.05 (d, J = 7.2 Hz, 1H), 7.65 (d, J = 8.1 Hz, 2H), 7.41 – 7.31 (m, 3H), 6.57 (dd, J = 9.0, 6.3 Hz, 1H), 6.47 – 6.43 (m, 1H), 2.46 (0.18H). **ESI-HRMS**: m/z calcd. for C₁₄H₈D₃N₂Cl [M+H]⁺: 246.0877, found 246.0872.

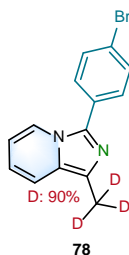


3-(3-chlorophenyl)-1-(methyl-d₃)imidazo[1,5-a]pyridine: Following the **General procedure B1**, a mixture of 2-acetylpyridine (0.1 mmol), 3-chlorobenzaldehyde (2 equiv.) and D-glucosamine·HCl (2.0

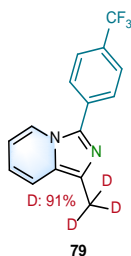
equiv.) in AcOH-d₄ : D₂O (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up gave product **76** (15.4 mg, 0.063 mmol, isolated yield 63%) as a yellow solid. **Mp**: 89–90 °C. **FT-IR**: ν (cm⁻¹): 2917, 1665, 1592, 1564, 1455, 1428, 1377, 1251, 1080, 995, 884, 787, 734, 688, 614, 414. D incorporation was tested by ¹H NMR: 91%. **¹H NMR** (300 MHz, CDCl₃): δ 8.10 (d, *J* = 7.2 Hz, 1H), 7.72 (s, 1H), 7.60 (d, *J* = 7.5 Hz, 1H), 7.36 – 7.28 (m, 3H), 6.61 – 6.56 (m, 1H), 6.49 – 6.45 (m, 1H), 2.46 (0.27H). **ESI-HRMS**: *m/z* calcd. for C₁₄H₈D₃ClN₂ [M+H]⁺: 246.0877, found 246.0873.



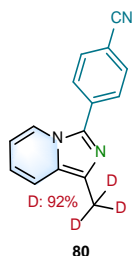
3-(3,4-dichlorophenyl)-1-(methyl-d₃)imidazo[1,5-a]pyridine: Following the **General procedure B1**, a mixture of 2-acetylpyridine (0.1 mmol), 3,4-dichlorobenzaldehyde (2 equiv.) and D-glucosamine·HCl (2.0 equiv.) in AcOH-d₄ : D₂O (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up gave product **77** (11.4 mg, 0.041 mmol, isolated yield 41%) as a yellow solid. **Mp**: 97–99 °C. **FT-IR**: ν (cm⁻¹): 2916, 2852, 1671, 1589, 1554, 1496, 1451, 1416, 1397, 1249, 1131, 1028, 977, 874, 826, 729, 666, 573, 536, 519, 435, 420. D incorporation was tested by ¹H NMR: 88%. **¹H NMR** (300 MHz, CDCl₃): δ 8.05 (d, *J* = 7.5 Hz, 1H), 7.82 (s, 1H), 7.55 (dd, *J* = 8.4, 1.5 Hz, 1H), 7.47 (d, *J* = 8.4 Hz, 1H), 7.34 (d, *J* = 9.0 Hz, 1H), 6.60 (dd, *J* = 9.0, 6.3 Hz, 1H), 6.52 – 6.47 (m, 1H), 2.45 (0.37H). **ESI-HRMS**: *m/z* calcd. for C₁₄H₇D₃Cl₂N₂ [M+H]⁺: 280.0488, found 280.0476.



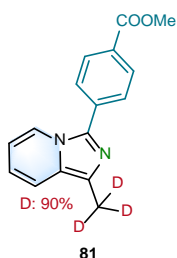
3-(4-bromophenyl)-1-(methyl-d₃)imidazo[1,5-a]pyridine: Following the **General procedure B1**, a mixture of 2-acetylpyridine (0.1 mmol), 4-bromobenzaldehyde (2 equiv.) and D-glucosamine·HCl (2.0 equiv.) in AcOH-d₄ : D₂O (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up gave product **78** (19.4 mg, 0.067 mmol, isolated yield 67%) as a yellow solid. **Mp**: 86–88 °C. **FT-IR**: ν (cm⁻¹): 2928, 2114, 1660, 1591, 1508, 1418, 1399, 1362, 1253, 1175, 1102, 1067, 1008, 958, 826, 723, 672, 490, 420. D incorporation was tested by ¹H NMR: 90%. **¹H NMR** (300 MHz, CDCl₃): δ 8.05 (d, J = 6.6 Hz, 1H), 7.60 – 7.56 (m, 4H), 7.33 (d, J = 9.0 Hz, 1H), 6.59 – 6.54 (m, 1H), 6.47 – 6.42 (m, 1H), 2.45 (0.3H). **ESI-HRMS**: m/z calcd. for C₁₄H₈D₃N₂Br [M+H]⁺: 290.0372, found 290.0364.



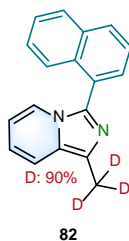
1-(methyl-d₃)-3-(4-(trifluoromethyl)phenyl)imidazo[1,5-a]pyridine: Following the **General procedure B1**, a mixture of 2-acetylpyridine (0.1 mmol), 4-(trifluoromethyl)benzaldehyde (2 equiv.) and D-glucosamine·HCl (2.0 equiv.) in AcOH-d₄ : D₂O (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up gave product **79** (22.9 mg, 0.082 mmol, isolated yield 82%) as a yellow solid. **Mp**: 73–74 °C. **FT-IR**: ν (cm⁻¹): 2930, 1684, 1616, 1519, 1422, 1317, 1253, 1162, 1107, 1063, 1014, 958, 839, 732, 670, 602, 523, 497, 427. D incorporation was tested by ¹H NMR: 91%. **¹H NMR** (300 MHz, CDCl₃): δ 8.13 (d, J = 7.2 Hz, 1H), 7.85 (d, J = 8.4 Hz, 2H), 7.67 (d, J = 8.1 Hz, 2H), 7.36 (d, J = 9.0 Hz, 1H), 6.62 (dd, J = 9.1, 6.4 Hz, 1H), 6.52 – 6.48 (m, 1H), 2.47 (0.28H). **ESI-HRMS**: m/z calcd. for C₁₅H₈D₃F₃N₂ [M+H]⁺: 280.1141, found 280.1139.



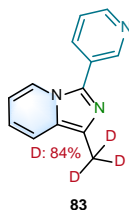
4-(1-(methyl-d₃)imidazo[1,5-a]pyridin-3-yl)benzonitrile: Following the **General procedure B1**, a mixture of 2-acetylpyridine (0.1 mmol), 4-formylbenzonitrile (2 equiv.) and D-glucosamine·HCl (2.0 equiv.) in AcOH-d₄ : D₂O (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up gave product **80** (16.5 mg, 0.070 mmol, isolated yield 70%) as a yellow solid. **Mp**: 117–118 °C. **FT-IR**: ν (cm⁻¹): 3079, 2953, 2920, 2856, 2221, 2125, 1737, 1671, 1605, 1513, 1414, 1333, 1255, 1168, 1105, 1018, 956, 843, 727, 672, 593, 550, 533, 420. D incorporation was tested by ¹H NMR: 92%. **¹H NMR** (300 MHz, CDCl₃): δ 8.17 (d, J = 7.2 Hz, 1H), 7.87 (d, J = 8.4 Hz, 2H), 7.70 (d, J = 8.4 Hz, 2H), 7.39 (d, J = 9.0 Hz, 1H), 6.66 (dd, J = 9.0, 6.6 Hz, 1H), 6.55 (t, J = 6.6 Hz, 1H), 2.47 (0.23H). **ESI-HRMS**: m/z calcd. for C₁₅H₈D₃N₃ [M+H]⁺: 237.1220, found 237.1219.



methyl 4-(1-(methyl-d₃)imidazo[1,5-a]pyridin-3-yl)benzoate: Following the **General procedure B1**, a mixture of 2-acetylpyridine (0.1 mmol), methyl-4-formylbenzoate (2 equiv.) and D-glucosamine·HCl (2.0 equiv.) in AcOH-d₄ : D₂O (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. work-up gave product **81** (10.5 mg, 0.039 mmol, isolated yield 39%) as a brown solid. **Mp**: 79–81 °C. **FT-IR**: ν (cm⁻¹): 2916, 1714, 1609, 1436, 1278, 1175, 1102, 1014, 956, 859, 773, 701, 528, 486, 420. D incorporation was tested by ¹H NMR: 90%. **¹H NMR** (300 MHz, CDCl₃): δ 8.17 (d, J = 7.2 Hz, 1H), 8.08 (d, J = 8.4 Hz, 2H), 7.81 (d, J = 8.4 Hz, 2H), 7.36 (d, J = 9.0 Hz, 1H), 6.62 (dd, J = 9.0, 6.3 Hz, 1H), 6.53 – 6.48 (m, 1H), 3.88 (s, 3H), 2.47 (0.29H). **ESI-HRMS**: m/z calcd. for C₁₆H₁₁D₃O₂N₂ [M+H]⁺: 270.1322, found 270.1316.

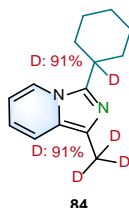


1-(methyl-d₃)-3-(naphthalen-1-yl)imidazo[1,5-a]pyridine: Following the **General procedure B1**, a mixture of 2-acetylpyridine (0.1 mmol), 1-naphthaldehyde (2 equiv.) and D-glucosamine·HCl (2.0 equiv.) in AcOH-d₄ : D₂O (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up gave product **82** (14.1 mg, 0.054 mmol, isolated yield 54%) as a yellow solid. **Mp:** 59–61 °C. **FT-IR:** ν (cm⁻¹): 3056, 2918, 2850, 1646, 1587, 1508, 1442, 1409, 1364, 1327, 1247, 1210, 1144, 1004, 942, 864, 802, 773, 734, 680, 639, 573, 530, 495, 412. D incorporation was tested by ¹H NMR: 90%. **¹H NMR** (400 MHz, CDCl₃): δ 7.90 – 7.85 (m, 2H), 7.65 – 7.63 (m, 2H), 7.53 – 7.48 (m, 2H), 7.43 (d, *J* = 6.8 Hz, 1H), 7.39 – 7.36 (m, 2H), 6.59 – 6.55 (m, 1H), 6.34 – 6.30 (m, 1H), 2.53 (0.30H). **ESI-HRMS:** *m/z* calcd. for C₁₈H₁₁D₃N₂ [M+H]⁺: 262.1424, found 262.1419.

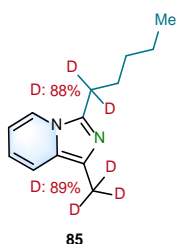


1-(methyl-d₃)-3-(pyridin-3-yl)imidazo[1,5-a]pyridine: Following the **General procedure B1**, a mixture of 2-acetylpyridine (0.1 mmol), nicotinaldehyde (2 equiv.) and D-glucosamine·HCl (2.0 equiv.) in AcOH-d₄ : D₂O (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up gave product **83** (6.4 mg, 0.030 mmol, isolated yield 30%) as a yellow solid. **Mp:** 47–48 °C. **FT-IR:** ν (cm⁻¹): 3037, 2926, 2854, 1677, 1630, 1568, 1517, 1422, 1366, 1327, 1255, 1181, 1154, 1022, 958, 810, 736, 705, 676, 616, 521, 420. D incorporation was tested by ¹H NMR: 84%. **¹H NMR** (400 MHz, CDCl₃): δ 9.00 (d, *J* = 1.2 Hz, 1H), 8.60 (dd, *J* = 4.8, 1.6 Hz, 1H), 8.15 – 8.12 (m, 2H), 7.44 – 7.39 (m, 2H), 6.69 (dd, *J* = 9.1,

6.4 Hz, 1H), 6.60 – 6.56 (t, $J = 6.7$ Hz, 1H), 2.52 (0.47H). **ESI-HRMS**: m/z calcd. for $C_{13}H_8D_3N_3$ $[M+H]^+$: 213.1220, found 213.1214.

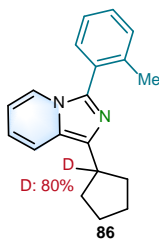


3-(cyclohexyl-1-d)-1-(methyl-d3)imidazo[1,5-a]pyridine: Following the **General procedure B1**, a mixture of 2-acetylpyridine (0.1 mmol), cyclohexanecarbaldehyde (2 equiv.) and D-glucosamine·HCl (2.0 equiv.) in $AcOH-d_4 : D_2O$ (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up gave product **84** (15.7 mg, 0.072 mmol, isolated yield 72 %) as a yellow solid. **Mp**: 60–61 °C. **FT-IR**: ν (cm^{-1}): 3342, 3173, 2928, 2854, 1626, 1445, 1405, 1245, 1142, 1090, 1022, 995, 960, 837, 816, 727, 696, 661, 587, 517, 499, 418. D incorporation was tested by 1H NMR: 91% and 91%. **1H NMR** (300 MHz, $CDCl_3$): δ 7.58 (d, $J = 7.2$ Hz, 1H), 7.20 (d, $J = 9.0$ Hz, 1H), 6.44 – 6.31 (m, 2H), δ 2.89 – 2.80 (0.09H), 2.38 (0.26H), 1.94 – 1.45 (m, 7H), 1.41 – 1.33 (m, 3H). **ESI-HRMS**: m/z calcd. for $C_{14}H_{14}D_4N_2$ $[M+H]^+$: 219.1799, found 219.1796.



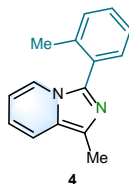
1-(methyl-d3)-3-(pentyl-1,1-d2)imidazo[1,5-a]pyridine: Following the **General procedure B1**, a mixture of 2-acetylpyridine (0.1 mmol), hexanal (2 equiv.) and D-glucosamine·HCl (2.0 equiv.) in $AcOH-d_4 : D_2O$ (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up gave product **85** (15.5 mg, 0.075 mmol, isolated yield 75%) as a brown solid. **Mp**: 106–107 °C. **FT-IR**: ν (cm^{-1}): 3184, 2959, 2930, 2858, 1653, 1461, 1377, 1247, 1043, 995, 783, 734, 694, 618, 560, 418. D incorporation was tested by 1H NMR: 89% and 88%. **1H NMR** (300 MHz, $CDCl_3$): δ 7.53 (d, $J = 7.2$ Hz,

1H), 7.22 (d, $J = 9.3$ Hz, 1H), 6.44 (dd, $J = 9.0, 6.3$ Hz, 1H), 6.38 – 6.34 (m, 1H), 2.83 (0.24H), 2.37 (0.33H), 1.72 (t, $J = 6.3$ Hz, 2H), 1.32 – 1.29 (m, 4H), 0.83 (t, $J = 6.9$ Hz, 3H). **ESI-HRMS**: m/z calcd. for $C_{13}H_{13}D_5N_2$ $[M+H]^+$: 208.1862, found 208.1858.



1-(cyclopentyl-1-d)-3-(o-tolyl)imidazo[1,5-a]pyridine: Following the **General procedure B1**, a mixture of cyclopentyl(pyridin-2-yl)methanone (0.1 mmol), 2-methylbenzaldehyde (2 equiv.) and D-glucosamine·HCl (2.0 equiv.) in $AcOH-d_4 : D_2O$ (0.9 : 0.1 mL) were stirred at 120 °C under Ar atmosphere for 36 h. Work-up gave product **86** (21.6 mg, 0.078 mmol, isolated yield 78%) as a brown liquid. **FT-IR**: ν (cm^{-1}): 3064, 2951, 2866, 1688, 1632, 1517, 1453, 1412, 1362, 1306, 1243, 1222, 1137, 1105, 1043, 999, 944, 773, 746, 721, 616, 544, 453, 418. D incorporation was tested by 1H NMR: 80%. **1H NMR** (400 MHz, $CDCl_3$): δ 7.42 – 7.40 (m, 1H), 7.36 – 7.31 (m, 2H), 7.24 – 7.21 (m, 2H), 7.19 – 7.15 (m, 1H), 6.48 – 6.44 (m, 1H), 6.30 – 6.26 (m, 1H), 3.36 – 3.28 (0.2H), 2.11 (s, 3H), 2.03 – 1.98 (m, 2H), 1.92 – 1.85 (m, 2H), 1.81 – 1.77 (m, 2H), 1.64 – 1.59 (m, 2H). **ESI-HRMS**: m/z calcd. for $C_{19}H_{19}DN_2$ $[M+H]^+$: 278.1768, found 278.1759.

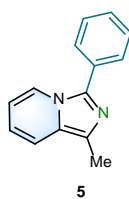
5.3.3 Direct nitrogen interception from chitin/chitosan for imidazo[1,5-a]pyridine



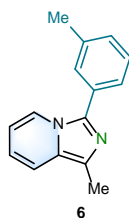
Methyl-3-(o-tolyl)imidazo[1,5-a]pyridine: Following the **General procedure C1**, a mixture of 2-acetylpyridine (0.1 mmol), 2-methylbenzaldehyde (2.0 equiv.) and chitosan (3.0 equiv.) in CF_3COOH (1.0 mL) was stirred at 140 °C under Ar atmosphere for 36 h. Chitosan was dried at 100 °C overnight.

Work-up gave product **3** (16.9 mg, 0.076 mmol, isolated yield 76%) as a yellow liquid. Following the **General procedure C3**, a mixture of 2-acetylpyridine (0.1 mmol), 2-methylbenzaldehyde (2.0 equiv.) and chitin (3.0 equiv.) in the CF₃COOH (1.0 mL) was stirred at 140 °C under Ar atmosphere for 36 h. Chitin was dried at 100 °C overnight. Work-up gave product **3** (15.8 mg, 0.071 mmol, isolated yield 71%) as a yellow liquid.

¹H NMR (300 MHz, CDCl₃) δ 7.41 (d, *J* = 7.2 Hz, 1H), 7.35–7.17 (m, 5H), 6.53 – 6.48 (m, 1H), 6.35–6.30 (m, 1H), 2.49 (s, 3H), 2.12 (s, 3H). ¹³C NMR (100 MHz, CDCl₃) δ 138.1, 136.2, 130.6, 130.4, 129.5, 129.1, 127.9, 126.7, 125.9, 121.2, 118.0, 116.6, 112.2, 19.6, 12.6.

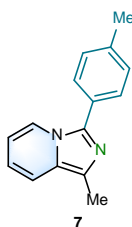


1-Methyl-3-phenylimidazo[1,5-a]pyridine: Following the **General procedure C1**, a mixture of 2-acetylpyridine (0.1 mmol), benzaldehyde (2.0 equiv.) and chitosan (3.0 equiv.) in CF₃COOH (1.0 mL) was stirred at 140 °C under Ar atmosphere for 36 h. Chitosan was dried at 100 °C overnight. Work-up gave product **5** (17.1 mg, 0.082 mmol, isolated yield 82%) as a yellow liquid. ¹H NMR (400 MHz, CDCl₃) δ 8.09 (d, *J* = 7.2 Hz, 1H), 7.69 (d, *J* = 7.6 Hz, 2H), 7.41 (t, *J* = 7.6 Hz, 2H), 7.34 – 7.30 (m, 2H), 6.56 – 6.52 (m, 1H), 6.43 – 6.39 (m, 1H), 2.48 (s, 3H). ¹³C NMR (100 MHz, CDCl₃) 136.5, 130.3, 128.9, 128.4, 127.9, 127.7, 121.1, 118.3, 117.1, 112.9, 12.4. **ESI-HRMS:** *m/z* calcd. for C₁₅H₁₄N₂ [M+H]⁺: 223.1235, found 223.1230.

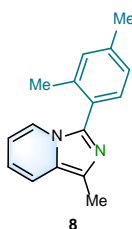


1-Methyl-3-(m-tolyl)imidazo[1,5-a]pyridine: Following the **General procedure C1**, a mixture of 2-acetylpyridine (0.1 mmol), 3-methylbenzaldehyde (2.0 equiv.) and chitosan (3.0 equiv.) in CF₃COOH (1.0 mL) was stirred at 140 °C under Ar atmosphere for 36 h. Chitosan was dried at 100 °C overnight.

Work-up gave product **6** (17.8 mg, 0.080 mmol, isolated yield 80%) as a yellow solid. **m.p.**: 104-106 °C. **¹H NMR** (400 MHz, CDCl₃) δ 8.10 (d, *J* = 7.2 Hz, 1H), 7.53 (s, 1H), 7.47 (d, *J* = 7.6 Hz, 1H), 7.31 (d, *J* = 8.0 Hz, 2H), 7.14 (d, *J* = 7.6 Hz, 1H), 6.54 (dd, *J* = 9.2, 6.4 Hz, 1H), 6.42 – 6.39 (m, 1H), 2.48 (s, 3H), 2.34 (s, 3H). **¹³C NMR** (100 MHz, CDCl₃) δ 138.8, 136.7, 132.7, 130.2, 129.2, 128.7, 128.6, 127.9, 124.5, 121.2, 118.3, 117.0, 112.8, 21.4, 12.5. **ESI-HRMS**: *m/z* calcd. for C₁₅H₁₄N₂ [M+H]⁺: 223.1235, found 223.1233.

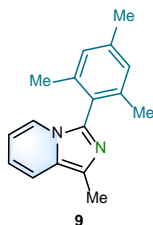


1-methyl-3-(p-tolyl)imidazo[1,5-a]pyridine: Following the **General procedure C1**, a mixture of 2-acetylpyridine (0.1 mmol), 4-methylbenzaldehyde (2.0 equiv.) and chitosan (3.0 equiv.) in CF₃COOH (1.0 mL) was stirred at 140 °C under Ar atmosphere for 36 h. Chitosan was dried at 100 °C overnight. Work-up gave product **7** (17.6 mg, 0.079 mmol, isolated yield 79%) as a yellow solid. **m.p.**: 62-63 °C. **¹H NMR** (400 MHz, CDCl₃) δ 8.06 (d, *J* = 7.2 Hz, 1H), 7.58 (d, *J* = 8.0 Hz, 2H), 7.30 (d, *J* = 9.2 Hz, 1H), 7.23 (d, *J* = 8.0 Hz, 2H), 6.52 (dd, *J* = 9.2, 6.4 Hz, 1H), 6.41 – 6.37 (m, 1H), 2.49 (s, 3H), 2.34 (s, 3H). **¹³C NMR** (100 MHz, CDCl₃) δ 138.4, 136.7, 129.6, 128.5, 127.7, 127.7, 127.4, 121.2, 118.3, 116.8, 112.7, 21.4, 12.5. **ESI-HRMS**: *m/z* calcd. for C₁₅H₁₄N₂ [M+H]⁺: 223.1235, found 223.1234.

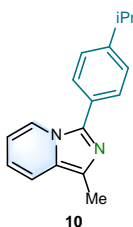


3-(2,4-dimethylphenyl)-1-methylimidazo[1,5-a]pyridine: Following the **General procedure C1**, a mixture of 2-acetylpyridine (0.1 mmol), 2,4-dimethylbenzaldehyde (2.0 equiv.) and chitosan (3.0 equiv.) in CF₃COOH (1.0 mL) was stirred at 140 °C under Ar atmosphere for 36 h. Chitosan was dried at 100 °C overnight. Work-up gave product **8** (16.7 mg, 0.075 mmol, isolated yield 75%) as a brown solid. **m.p.**: 82-83 °C. **¹H NMR** (400 MHz, CDCl₃) δ 7.41 (d, *J* = 7.2 Hz, 1H), 7.30 (d, *J* = 9.2 Hz, 1H), 7.21 (d, *J* = 8.0 Hz, 1H), 7.07 (s, 1H), 7.02 (d, *J* = 8.0 Hz, 1H), 6.54 – 6.50 (m, 1H), 6.36 – 6.32 (m, 1H), 2.48 (s, 3H),

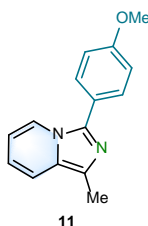
2.31 (s, 3H), 2.07 (s, 3H). ^{13}C NMR (100 MHz, CDCl_3) δ 139.1, 137.9, 136.3, 131.4, 130.4, 127.6, 126.7, 126.4, 121.3, 118.0, 116.7, 112.2, 21.3, 19.5, 12.5. **ESI-HRMS**: m/z calcd. for $\text{C}_{16}\text{H}_{16}\text{N}_2$ $[\text{M}+\text{H}]^+$: 237.1392, found 237.1389.



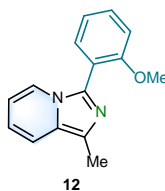
3-mesityl-1-methylimidazo[1,5-a]pyridine: Following the **General procedure C1**, a mixture of 2-acetylpyridine (0.1 mmol), 2,4,6-trimethylbenzaldehyde (2.0 equiv.) and chitosan (3.0 equiv.) in CF_3COOH (1.0 mL) was stirred at 140 °C under Ar atmosphere for 36 h. Chitosan was dried at 100 °C overnight. Work-up gave product **9** (17.8 mg, 0.071 mmol, isolated yield 71%) as a yellow liquid. ^1H NMR (400 MHz, CDCl_3) δ 7.31 (d, J = 8.8 Hz, 1H), 7.13 (d, J = 7.2 Hz, 1H), 6.88 (s, 2H), 6.52 – 6.48 (m, 1H), 6.32 – 6.28 (m, 1H), 2.50 (s, 3H), 2.27 (s, 3H), 1.88 (s, 6H). ^{13}C NMR (100 MHz, CDCl_3) δ 139.1, 135.5, 128.3, 127.5, 126.2, 126.2, 120.9, 118.0, 116.3, 112.1, 21.2, 19.5, 12.7. **ESI-HRMS**: m/z calcd. for $\text{C}_{17}\text{H}_{18}\text{N}_2$ $[\text{M}+\text{H}]^+$: 251.1548, found 251.1545.



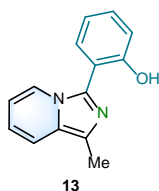
3-(4-isopropylphenyl)-1-methylimidazo[1,5-a]pyridine: Following the **General procedure C1**, a mixture of 2-acetylpyridine (0.1 mmol), 4-isopropylbenzaldehyde (2.0 equiv.) and chitosan (3.0 equiv.) in CF_3COOH (1.0 mL) was stirred at 140 °C under Ar atmosphere for 36 h. Chitosan was dried at 100 °C overnight. Work-up gave product **10** (18.5 mg, 0.074 mmol, isolated yield 74%) as a brown solid. **m.p.**: 61–63 °C. ^1H NMR (300 MHz, CDCl_3) δ 8.08 (d, J = 7.2 Hz, 1H), 7.62 (d, J = 8.1 Hz, 2H), 7.31 – 7.26 (m, 3H), 6.53 (dd, J = 9.0, 6.4 Hz, 1H), 6.42 – 6.37 (m, 1H), 2.89 (hept, J = 6.9 Hz, 1H), 2.49 (s, 3H), 1.21 (d, J = 6.9 Hz, 6H). ^{13}C NMR (100 MHz, CDCl_3) δ 149.4, 136.5, 128.3, 127.8, 127.7, 127.4, 127.0, 121.2, 118.3, 117.0, 112.9, 34.0, 23.8, 12.3. **ESI-HRMS**: m/z calcd. for $\text{C}_{17}\text{H}_{18}\text{N}_2$ $[\text{M}+\text{H}]^+$: 251.1548, found 251.1544.



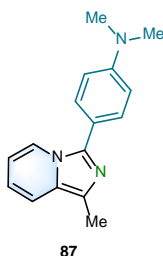
3-(4-methoxyphenyl)-1-methylimidazo[1,5-a]pyridine: Following the **General procedure C1**, a mixture of 2-acetylpyridine (0.1 mmol), 4-methoxybenzaldehyde (2.0 equiv.) and chitosan (3.0 equiv.) in CF₃COOH (1.0 mL) was stirred at 140 °C under Ar atmosphere for 36 h. Chitosan was dried at 100 °C overnight. Work-up gave product **11** (15.5 mg, 0.065 mmol, isolated yield 65%) as a yellow solid. **m.p.:** 55–57 °C. **¹H NMR** (400 MHz, CDCl₃) δ 8.01 (d, *J* = 7.2 Hz, 1H), 7.61 (d, *J* = 8.4 Hz, 2H), 7.29 (d, *J* = 9.2 Hz, 1H), 6.95 (d, *J* = 8.4 Hz, 2H), 6.51 (dd, *J* = 9.2, 6.0 Hz, 1H), 6.40 – 6.37 (m, 1H), 3.79 (s, 3H), 2.47 (s, 3H). **¹³C NMR** (100 MHz, CDCl₃) δ 159.7, 136.5, 129.2, 128.3, 127.5, 122.8, 121.1, 118.3, 116.7, 114.4, 112.7, 55.3, 12.4. **ESI-HRMS:** *m/z* calcd. for C₁₅H₄N₂O [M+H]⁺: 239.0402, found 239.0397.



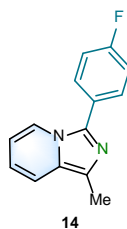
3-(2-methoxyphenyl)-1-methylimidazo[1,5-a]pyridine: Following the **General procedure C1**, a mixture of 2-acetylpyridine (0.1 mmol), 2-methoxybenzaldehyde (2.0 equiv.) and chitosan (3.0 equiv.) in CF₃COOH (1.0 mL) was stirred at 140 °C under Ar atmosphere for 36 h. Chitosan was dried at 100 °C overnight. Work-up gave product **12** (14.99 mg, 0.063 mmol, isolated yield 63%) as a yellow solid. **m.p.:** 62–63 °C. **¹H NMR** (400 MHz, CDCl₃) δ 7.50 (dd, *J* = 7.2, 1.2 Hz, 1H), 7.42 – 7.40 (m, 1H), 7.36 – 7.28 (m, 2H), 7.01 – 6.97 (m, 1H), 6.93 (d, *J* = 8.4 Hz, 1H), 6.54 – 6.50 (m, 1H), 6.36 – 6.32 (m, 1H), 3.70 (s, 3H), 2.49 (s, 3H). **¹³C NMR** (100 MHz, CDCl₃) δ 157.1, 134.3, 132.3, 130.3, 128.1, 127.5, 122.8, 120.9, 119.2, 117.6, 116.7, 111.5, 111.0, 55.4, 12.5. **ESI-HRMS:** *m/z* calcd. for C₁₅H₁₄N₂O [M+H]⁺: 239.1184, found 239.1181.



2-(1-methylimidazo[1,5-a]pyridin-3-yl)phenol: Following the **General procedure C1**, a mixture of 2-acetylpyridine (0.1 mmol), 2-hydroxybenzaldehyde (2.0 equiv.) and chitosan (3.0 equiv.) in CF_3COOH (1.0 mL) was stirred at 140 °C under Ar atmosphere for 36 h. Chitosan was dried at 100 °C overnight. Work-up gave product **13** (6.7 mg, 0.030 mmol, isolated yield 30%) as a white solid. **m.p.:** 136–138 °C. **^1H NMR** (400 MHz, CDCl_3) δ 8.34 (d, J = 7.2 Hz, 1H), 7.66 (d, J = 8.0 Hz, 1H), 7.36 (d, J = 9.2 Hz, 1H), 7.19 – 7.18 (m, 1H), 7.07 (d, J = 8.4 Hz, 1H), 6.92 – 6.88 (m, 1H), 6.64 – 6.60 (m, 1H), 6.55 – 6.51 (m, 1H), 2.48 (s, 3H). **^{13}C NMR** (100 MHz, CDCl_3) δ 156.3, 134.2, 129.4, 127.2, 127.0, 123.9, 121.9, 118.9, 118.6, 117.7, 117.7, 114.3, 113.7, 12.2. **ESI-HRMS:** m/z calcd. for $\text{C}_{14}\text{H}_{12}\text{N}_2\text{O}_1$ $[\text{M}+\text{H}]^+$: 225.1028, found 225.1026.

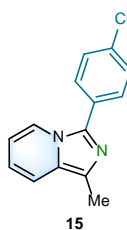


***N,N*-dimethyl-4-(1-methylimidazo[1,5-a]pyridin-3-yl)aniline:** Following the **General procedure C1**, a mixture of 2-acetylpyridine (0.1 mmol), (*E*)-3-(4-(dimethylamino)phenyl)acrylaldehyde (2.0 equiv.) and chitosan (3.0 equiv.) in CF_3COOH (0.7 mL) was stirred at 140 °C under Ar atmosphere for 4 days. Chitosan was dried at 100 °C overnight. Work-up gave product **87** (10.5 mg, 0.038 mmol, isolated yield 38%) as a yellow solid; **m.p.** 57–59 °C. **^1H NMR** (400 MHz, CDCl_3) δ 7.82 (d, J = 6.8 Hz, 1H), 7.52 (d, J = 16.0 Hz, 1H), 7.39 (d, J = 8.4 Hz, 2H), 7.25 (d, J = 9.2 Hz, 1H), 6.88 (d, J = 15.6 Hz, 1H), 6.64 (d, J = 8.8 Hz, 2H), 6.51 – 6.42 (m, 2H), 2.92 (s, 6H), 2.47 (s, 3H). **^{13}C NMR** (100 MHz, CDCl_3) δ 150.3, 135.9, 130.4, 129.7, 129.2, 127.7, 125.3, 120.5, 118.4, 116.5, 112.7, 112.3, 107.6, 40.4, 12.6. **ESI-HRMS:** m/z calcd. for $\text{C}_{18}\text{H}_{19}\text{N}_3$ $[\text{M}+\text{H}]^+$: 278.1652, found 278.1654.

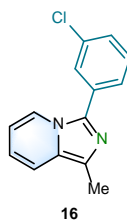


3-(4-fluorophenyl)-1-methylimidazo[1,5-a]pyridine: Following the **General procedure C1**, a mixture of 2-acetylpyridine (0.1 mmol), 4-fluorobenzaldehyde (2.0 equiv.) and chitosan (3.0 equiv.) in CF₃COOH (1.0 mL) was stirred at 140 °C under Ar atmosphere for 36 h. Chitosan was dried at 100 °C overnight. Work-up gave product **14** (17.6 mg, 0.078 mmol, isolated yield 78%) as a yellow solid. **m.p.:** 82–83 °C. Following the **General procedure C3**, a mixture of 2-acetylpyridine (0.1 mmol), 4-fluorobenzaldehyde (2.0 equiv.) and chitin (3.0 equiv.) in CF₃COOH (1.0 mL) was stirred at 140 °C under Ar atmosphere for 36 h. Chitin was dried at 100 °C overnight. Work-up gave product **14** (16.0 mg, 0.071 mmol, isolated yield 71%) as a yellow solid.

¹H NMR (400 MHz, CDCl₃) δ 8.02 (d, *J* = 7.2 Hz, 1H), 7.67 (dd, *J* = 8.8, 5.2 Hz, 2H), 7.33 (d, *J* = 9.2 Hz, 1H), 7.14 – 7.10 (m, 2H), 6.56 (dd, *J* = 9.2, 6.0 Hz, 1H), 6.45 – 6.42 (m, 1H). **¹³C NMR** (100 MHz, CDCl₃) δ 162.7 (d, ¹*J*_{C-F} = 247.1 Hz), 135.5, 129.7 (d, ³*J*_{C-F} = 8.2 Hz), 128.7, 127.9, 126.5 (d, ⁴*J*_{C-F} = 3.3 Hz), 120.8, 118.4, 117.1, 116.1 (d, ²*J*_{C-F} = 21.6 Hz), 113.1, 12.4. **ESI-HRMS:** *m/z* calcd. for C₁₄H₁₁FN₂ [M+H]⁺: 227.0985, found 227.0980.

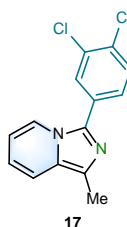


3-(4-chlorophenyl)-1-methylimidazo[1,5-a]pyridine: Following the **General procedure C1**, a mixture of 2-acetylpyridine (0.1 mmol), 4-chlorobenzaldehyde (2.0 equiv.) and chitosan (3.0 equiv.) in CF₃COOH (1.0 mL) was stirred at 140 °C under Ar atmosphere for 36 h. Chitosan was dried at 100 °C overnight. Work-up gave product **15** (17.7 mg, 0.073 mmol, isolated yield 73%) as a yellow solid. **m.p:** 63–64 °C. **¹H NMR** (400 MHz, CDCl₃) δ 8.04 (d, *J* = 7.2 Hz, 1H), 7.63 (d, *J* = 8.4 Hz, 2H), 7.39 (d, *J* = 8.8 Hz, 2H), 7.33 (d, *J* = 10.0 Hz, 1H), 6.57 (dd, *J* = 8.8, 6.4 Hz, 1H), 6.47 – 6.43 (m, 1H), 2.47 (s, 3H). **¹³C NMR** (100 MHz, CDCl₃) δ 135.3, 134.2, 129.2, 129.1, 128.9, 128.8, 128.2, 121.6, 120.9, 118.4, 117.3, 113.3, 12.4. **ESI-HRMS:** *m/z* calcd. for C₁₄H₁₁ClN₂ [M+H]⁺: 243.0689, found 243.0687.

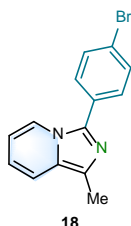


3-(3-chlorophenyl)-1-methylimidazo[1,5-a]pyridine: Following the **General procedure C1**, a mixture of 2-acetylpyridine (0.1 mmol), 3-chlorobenzaldehyde (2.0 equiv.) and chitosan (3.0 equiv.) in CF₃COOH (1.0 mL) was stirred at 140 °C under Ar atmosphere for 36 h. Chitosan was dried at 100 °C overnight. Work-up gave product **16** (17.0 mg, 0.070 mmol, isolated yield 70%) as a brown solid. **m.p.**: 70–71 °C. Following the **General procedure C3**, a mixture of 2-acetylpyridine (0.1 mmol), 3-chlorobenzaldehyde (2.0 equiv.) and chitin (3.0 equiv.) in the CF₃COOH (1.0 mL) was stirred at 140 °C under Ar atmosphere for 36 h. Chitin was dried at 100 °C oven overnight. Work-up gave product **15** (12.8 mg, 0.053 mmol, isolated yield 53%) as a brown solid.

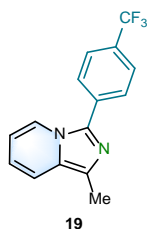
¹H NMR (400 MHz, CDCl₃) δ 8.10 (d, *J* = 7.6 Hz, 1H), 7.71 (s, 1H), 7.59 (d, *J* = 7.6 Hz, 1H), 7.35 – 7.30 (m, 3H), 6.59 (dd, *J* = 9.2, 6.0 Hz, 1H), 6.50 – 6.46 (m, 1H), 2.49 (s, 3H). **¹³C NMR** (100 MHz, CDCl₃) δ 134.9, 132.1, 130.2, 130.0, 129.3, 128.4, 128.3, 127.6, 125.6, 121.0, 118.5, 117.5, 113.4, 12.5. **ESI-HRMS**: *m/z* calcd. for C₁₄H₁₁ClN₂ [M+H]⁺: 243.0689, found 243.0687.



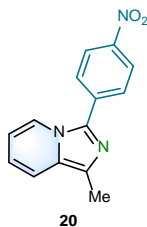
3-(3,4-dichlorophenyl)-1-methylimidazo[1,5-a]pyridine: Following the **General procedure C1**, a mixture of 2-acetylpyridine (0.1 mmol), 3,4-dichlorobenzaldehyde (2.0 equiv.) and chitosan (3.0 equiv.) in CF₃COOH (1.0 mL) was stirred at 140 °C under Ar atmosphere for 36 h. Chitosan was dried at 100 °C overnight. Work-up gave product **17** (15.7 mg, 0.057 mmol, isolated yield 57%) as a yellow solid. **m.p.**: 116–117 °C. **¹H NMR** (400 MHz, CDCl₃) δ 8.05 (d, *J* = 7.2 Hz, 1H), 7.82 (s, 1H), 7.57 – 7.45 (m, 2H), 7.34 (d, *J* = 9.2 Hz, 1H), 6.61 – 6.58 (m, 1H), 6.50 – 6.47 (m, 1H), 2.48 (s, 3H). **¹³C NMR** (100 MHz, CDCl₃) δ 134.0, 133.2, 132.0, 130.8, 130.4, 129.6, 129.2, 128.6, 126.4, 120.8, 118.5, 117.6, 113.6, 12.5. **ESI-HRMS**: *m/z* calcd. for C₁₄H₁₀Cl₂N₂ [M+H]⁺: 277.0299, found 277.0295.



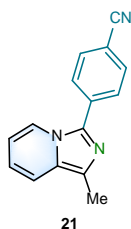
3-(4-bromophenyl)-1-methylimidazo[1,5-a]pyridine: Following the **General procedure C1**, a mixture of 2-acetylpyridine (0.1 mmol), 4-bromobenzaldehyde (2.0 equiv.) and chitosan (3.0 equiv.) in CF₃COOH (1.0 mL) was stirred at 140 °C under Ar atmosphere for 36 h. Chitosan was dried at 100 °C overnight. Work-up gave product **18** (18.31 mg, 0.064 mmol, isolated yield 64%) as a yellow solid. **m.p.:** 90–91 °C. **¹H NMR** (400 MHz, CDCl₃) δ 8.05 – 8.03 (m, 1H), 7.58 – 7.53 (m, 4H), 7.34 – 7.31 (m, 1H), 6.57 (dd, *J* = 9.1, 6.4 Hz, 1H), 6.47 – 6.43 (m, 1H), 2.47 (s, 3H). **¹³C NMR** (100 MHz, CDCl₃) δ 135.3, 132.1, 129.2, 129.1, 129.1, 128.2, 122.3, 120.9, 118.4, 117.3, 113.3, 12.4. **ESI-HRMS:** *m/z* calcd. for C₁₄H₁₁BrN₂ [M+H]⁺: 287.0184, found 287.0181.



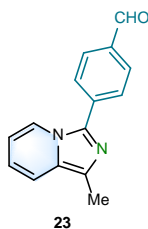
1-methyl-3-(4-(trifluoromethyl)phenyl)imidazo[1,5-a]pyridine: Following the **General procedure C1**, a mixture of 2-acetylpyridine (0.1 mmol), 4-trifluoromethylbenzaldehyde (2.0 equiv.) and chitosan (3.0 equiv.) in CF₃COOH (1.0 mL) was stirred at 140 °C under Ar atmosphere for 36 h. Chitosan was dried at 100 °C overnight. Work-up gave product **19** (18.49 mg, 0.067 mmol, isolated yield 67%) as a yellow solid. **m.p.:** 72–73 °C. **¹H NMR** (400 MHz, CDCl₃) δ 8.15 – 8.12 (m, 1H), 7.85 (d, *J* = 8.0 Hz, 2H), 7.69 – 7.66 (m, 2H), 7.38 – 7.35 (m, 1H), 6.64 – 6.59 (m, 1H), 6.52 – 6.47 (m, 1H), 2.50 (s, 3H). **¹³C NMR** (100 MHz, CDCl₃) δ 134.9, 133.9 (q, ⁴*J*_{C-F} = 1.2 Hz), 129.8 (q, ²*J*_{C-F} = 32.5 Hz), 129.8, 128.7, 127.6, 125.9 (q, ³*J*_{C-F} = 3.8 Hz), 124.0 (q, ¹*J*_{C-F} = 270.4 Hz), 120.9, 118.5, 117.7, 113.6, 12.5. **¹⁹F NMR** (377 MHz, CDCl₃) δ -62.6. **ESI-HRMS:** *m/z* calcd. for C₁₅H₁₁F₃N₂ [M+H]⁺: 277.0953, found 277.0950.



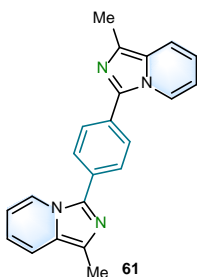
1-methyl-3-(4-nitrophenyl)imidazo[1,5-a]pyridine: Following the **General procedure C1**, a mixture of 2-acetylpyridine (0.1 mmol), 4-nitrobenzaldehyde (2.0 equiv.) and chitosan (3.0 equiv.) in CF₃COOH (1.0 mL) was stirred at 140 °C under Ar atmosphere for 36 h. Chitosan was dried at 100 °C overnight. Work-up gave product **20** (10.6 mg, 0.042 mmol, isolated yield 42%) as a red solid. **m.p.:** 142–143 °C. **¹H NMR** (400 MHz, CDCl₃) δ 8.27 (d, *J* = 8.8 Hz, 2H), 8.22 (d, *J* = 7.2 Hz, 1H), 7.93 (d, *J* = 8.8 Hz, 2H), 7.41 (d, *J* = 9.2 Hz, 1H), 6.71 – 6.67 (m, 1H), 6.61 – 6.57 (m, 1H), 2.52 (s, 3H). **¹³C NMR** (100 MHz, CDCl₃) δ 146.6, 136.5, 134.0, 131.0, 129.6, 127.3, 124.4, 121.0, 118.7, 118.4, 114.3, 12.6. **ESI-HRMS:** *m/z* calcd. for C₁₄H₁₁N₃O₂ [M+H]⁺: 254.0930, found 254.0926.



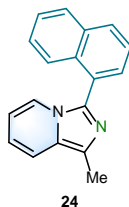
4-(1-methylimidazo[1,5-a]pyridin-3-yl)benzonitrile: Following the **General procedure C1**, a mixture of 2-acetylpyridine (0.1 mmol), 4-formylbenzonitrile (2.0 equiv.) and chitosan (3.0 equiv.) in CF₃COOH (1.0 mL) was stirred at 140 °C under Ar atmosphere for 36 h. Chitosan was dried at 100 °C overnight. Work-up gave product **21** (8.4 mg, 0.036 mmol, isolated yield 36%) as a yellow solid. **m.p.:** 89–90 °C. **¹H NMR** (400 MHz, CDCl₃) δ 8.16 (d, *J* = 7.2 Hz, 1H), 7.87 (d, *J* = 8.0 Hz, 2H), 7.70 (d, *J* = 8.4 Hz, 2H), 7.39 (d, *J* = 9.2 Hz, 1H), 6.66 (dd, *J* = 8.9, 6.5 Hz, 1H), 6.57 – 6.54 (m, 1H), 2.50 (s, 3H). **¹³C NMR** (100 MHz, CDCl₃) δ 134.6, 134.3, 132.7, 130.5, 129.2, 127.4, 120.9, 118.7, 118.6, 118.1, 114.0, 111.0, 12.5. **ESI-HRMS:** *m/z* calcd. for C₁₅H₁₁N₃ [M+H]⁺: 234.1031, found 234.1030.



4-(1-methylimidazo[1,5-a]pyridin-3-yl)benzaldehyde: Following the **General procedure C1**, a mixture of 2-acetylpyridine (0.1 mmol), terephthalaldehyde (2.0 equiv.) and chitosan (3.0 equiv.) in CF_3COOH (1.0 mL) was stirred at 140 °C under Ar atmosphere for 36 h. Chitosan was dried at 100 °C overnight. Work-up gave product **23** (12.5 mg, 0.053 mmol, isolated yield 53%) as a yellow solid. **m.p.:** 310–312 °C. **^1H NMR** (400 MHz, CDCl_3) δ 9.97 (s, 1H), 8.21 (d, J = 7.2 Hz, 1H), 7.92 (s, 4H), 7.38 (d, J = 8.8 Hz, 1H), 6.65 (dd, J = 9.2, 6.4 Hz, 1H), 6.54 (t, J = 6.8 Hz, 1H), 2.51 (s, 3H). **^{13}C NMR** (100 MHz, CDCl_3) δ 191.5, 136.0, 135.3, 135.0, 130.4, 130.3, 129.1, 127.4, 121.2, 118.6, 118.0, 113.9, 12.5. **ESI-HRMS:** m/z calcd. for $\text{C}_{15}\text{H}_{12}\text{N}_2\text{O}$ $[\text{M}+\text{H}]^+$: 237.1028, found 237.1024.

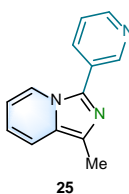


1,4-bis(1-methylimidazo[1,5-a]pyridin-3-yl)benzene: Following the **General procedure C1**, a mixture of 2-acetylpyridine (0.1 mmol), terephthalaldehyde (2.0 equiv.) and chitosan (3.0 equiv.) in CF_3COOH (0.7 mL) was stirred at 140 °C under Ar atmosphere for 4 days. Chitosan was dried at 100 °C overnight. Work-up gave product **61** (8.1 mg, 0.024 mmol, isolated yield 24%) as a brown red solid. **m.p.:** 212–214 °C. **^1H NMR** (300 MHz, CDCl_3) δ 8.17 (d, J = 7.2 Hz, 2H), 7.86 (s, 4H), 7.35 (d, J = 9.3 Hz, 2H), 6.58 (dd, J = 9.0, 6.3 Hz, 2H), 6.49 – 6.44 (m, 2H), 2.52 (s, 6H). **^{13}C NMR** (75 MHz, CDCl_3) δ 136.0, 130.1, 129.3, 128.3, 128.0, 121.2, 118.4, 117.2, 113.2, 12.6. **ESI-HRMS:** m/z calcd. for $\text{C}_{22}\text{H}_{18}\text{N}_4$ $[\text{M}+\text{H}]^+$: 338.1531, found 338.1526.

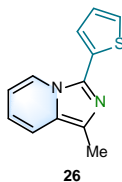


1-methyl-3-(naphthalen-1-yl)imidazo[1,5-a]pyridine: Following the **General procedure C1**, a mixture of 2-acetylpyridine (0.1 mmol), 1-naphthaldehyde (2.0 equiv.) and chitosan (3.0 equiv.) in CF₃COOH (1.0 mL) was stirred at 140 °C under Ar atmosphere for 36 h. Chitosan was dried under 100 °C overnight. Work-up gave product **24** (8.0 mg, 0.031 mmol, isolated yield 31%) as a brown solid. **m.p.:** 90–93 °C. Following the **General procedure C3**, a mixture of 2-acetylpyridines (0.1 mmol), 1-naphthaldehyde (2.0 equiv.) and chitin (3.0 equiv.) in the CF₃COOH (1.0 mL) was stirred at 140 °C under Ar atmosphere for 36 h. Chitin was dried at 100 °C oven overnight. Work-up gave product **24** (7.5 mg, 0.029 mmol, isolated yield 29%) as a brown solid.

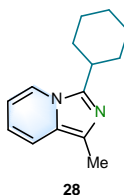
¹H NMR (300 MHz, CDCl₃) δ 7.91 – 7.85 (m, 2H), 7.66 – 7.62 (m, 2H), 7.53 – 7.48 (m, 2H), 7.44 (dd, *J* = 7.8, 1.2 Hz, 1H), 7.41 – 7.35 (m, 2H), 6.57 (dd, *J* = 9.0, 6.0 Hz, 1H), 6.35 – 6.30 (m, 1H), 2.57 (s, 3H). **¹³C NMR** (75 MHz, CDCl₃) δ 135.3, 133.9, 131.8, 129.6, 128.8, 128.5, 128.5, 127.5, 127.4, 126.9, 126.2, 125.7, 125.3, 121.6, 118.1, 117.0, 112.3, 12.7. **ESI-HRMS:** *m/z* calcd. for C₁₈H₁₄N₂ [M+H]⁺: 259.1235, found 259.1233.



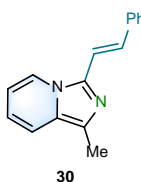
1-methyl-3-(pyridin-3-yl)imidazo[1,5-a]pyridine: Following the **General procedure C1**, a mixture of 2-acetylpyridine (0.1 mmol), nicotinaldehyde (4.0 equiv.) and chitosan (2.5 equiv.) in CF₃COOH (0.7 mL) was stirred at 140 °C under Ar atmosphere for 36 h. Chitosan was dried at 100 °C overnight. Work-up gave product **25** (12.1 mg, 0.058 mmol, isolated yield 58%) as a brown liquid. **¹H NMR** (400 MHz, CDCl₃) δ 8.99 (s, 1H), 8.56 (d, *J* = 4.4 Hz, 1H), 8.10 (d, *J* = 7.2 Hz, 1H), 8.05 (d, *J* = 8.0 Hz, 1H), 7.37 (d, *J* = 8.4 Hz, 2H), 6.64 – 6.60 (m, 1H), 6.50 (t, *J* = 6.8 Hz, 1H), 2.51 (s, 3H). **¹³C NMR** (100 MHz, CDCl₃) δ 149.0, 148.2, 135.2, 133.3, 129.8, 128.6, 126.8, 123.8, 120.7, 118.5, 117.7, 113.7, 12.5. **ESI-HRMS:** *m/z* calcd. for C₁₃H₁₁N₃ [M+H]⁺: 210.1031, found 210.1027.



1-methyl-3-(thiophen-2-yl)imidazo[1,5-a]pyridine: Following the **General procedure C1**, a mixture of 2-acetylpyridine (0.1 mmol), thiophene-2-carbaldehyde (2.0 equiv.) and chitosan (3.0 equiv.) in CF₃COOH (1.0 mL) was stirred at 140 °C under Ar atmosphere for 36 h. Chitosan was dried at 100 °C overnight. Work-up gave product **26** (5.8 mg, 0.027 mmol, isolated yield 27%) as a brown solid. **m.p.:** 79–81 °C. **¹H NMR** (400 MHz, CDCl₃) δ 8.17 (d, *J* = 6.8 Hz, 1H), 7.42 – 7.40 (m, 1H), 7.35 – 7.30 (m, 2H), 7.09 (dd, *J* = 5.2, 4.0 Hz, 1H), 6.60 – 6.56 (m, 1H), 6.54 – 6.50 (m, 1H), 2.48 (s, 3H). **¹³C NMR** (100 MHz, CDCl₃) δ 132.7, 131.3, 129.3, 128.2, 127.6, 125.5, 124.0, 121.5, 118.4, 117.0, 113.5, 12.6. **ESI-HRMS:** *m/z* calcd. for C₁₂H₁₀N₂S [M+H]⁺: 215.0643, found 215.0641.

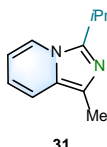


3-cyclohexyl-1-methylimidazo[1,5-a]pyridine: Following the **General procedure C1**, a mixture of 2-acetylpyridine (0.1 mmol), cyclohexanecarbaldehyde (2.0 equiv.) and chitosan (3.0 equiv.) in CF₃COOH (1.0 mL) was stirred at 140 °C under Ar atmosphere for 36 h. Chitosan was dried at 100 °C overnight. Work-up gave product **28** (14.8 mg, 0.069 mmol, isolated yield 69%) as a colorless solid. **m.p.:** 92–93 °C. **¹H NMR** (400 MHz, CDCl₃) δ 7.60 (d, *J* = 7.2 Hz, 1H), 7.22 (dd, *J* = 9.2 Hz, 0.8 Hz, 1H), 6.45 (dd, *J* = 9.2, 6.4 Hz, 1H), 6.39 – 6.35 (m, 1H), 2.85 (tt, *J* = 11.9, 3.4 Hz, 1H), 2.42 (s, 3H), 1.92 (d, *J* = 13.2 Hz, 2H), 1.83 (d, *J* = 12.8 Hz, 2H), 1.71 – 1.68 (m, 3H), 1.37 – 1.29 (m, 3H). **¹³C NMR** (100 MHz, CDCl₃) δ 141.2, 126.2, 126.1, 120.2, 118.3, 115.9, 111.9, 35.7, 30.8, 26.4, 25.9, 12.2. **ESI-HRMS:** *m/z* calcd. for C₁₄H₁₈N₂ [M+H]⁺: 215.1548, found 215.1544.

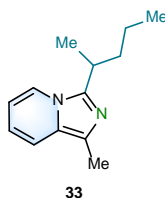


(E)-1-methyl-3-styrylimidazo[1,5-a]pyridine: Following the **General procedure C1**, a mixture of 2-acetylpyridine (0.1 mmol), cinnamaldehyde (2.0 equiv.) and chitosan (3.0 equiv.) in CF₃COOH (1.0 mL) was stirred at 140 °C under Ar atmosphere for 36 h. Chitosan was dried at 100 °C overnight. Work-up gave product **30** (6.6 mg, 0.028 mmol, isolated yield 28%) as a brown solid. **m.p.**: 110–111 °C. Following the **General procedure C3**, a mixture of 2-acetylpyridine (0.1 mmol), cinnamaldehyde (2.0 equiv.) and chitin (3.0 equiv.) in the CF₃COOH (0.1 M) was stirred at 140 °C under Ar atmosphere for 36 h. Chitin was dried at 100 °C overnight. Work-up gave product **30** (4.0 mg, 0.017 mmol, isolated yield 17%) as a brown solid.

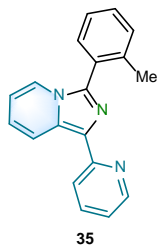
¹H NMR (400 MHz, CDCl₃) δ 7.88 (d, *J* = 7.2 Hz, 1H), 7.56 (d, *J* = 16.0 Hz, 1H), 7.49 (d, *J* = 7.6 Hz, 2H), 7.31 – 7.28 (m, 3H), 7.21 – 7.18 (m, 2H), 6.58 – 6.49 (m, 2H), 2.49 (s, 3H). **¹³C NMR** (100 MHz, CDCl₃) δ 137.1, 129.6, 128.7, 127.8, 126.5, 120.5, 118.5, 117.1, 113.2, 112.0, 12.7. **ESI-HRMS:** *m/z* calcd. for C₁₆H₁₄N₂ [M+H]⁺: 235.1235, found 235.1229.



3-isopropyl-1-methylimidazo[1,5-a]pyridine: Following the **General procedure C1**, a mixture of 2-acetylpyridine (0.1 mmol), isobutyraldehyde (2.0 equiv.) and chitosan (3.0 equiv.) in CF₃COOH (1.0 mL) was stirred at 140 °C under Ar atmosphere for 36 h. Chitosan was dried at 100 °C overnight. Work-up gave product **31** (8.7 mg, 0.050 mmol, isolated yield 50%) as a brown solid. **m.p.**: 93–95 °C. **¹H NMR** (300 MHz, CDCl₃) δ 7.58 (d, *J* = 7.2 Hz, 1H), 7.23 (d, *J* = 9.3 Hz, 1H), 6.46 (dd, *J* = 9.0, 6.3 Hz, 1H), 6.40 – 6.35 (m, 1H), 3.21 (hept, *J* = 6.9 Hz, 1H), 2.43 (s, 3H), 1.36 (d, *J* = 6.9 Hz, 5H). **¹³C NMR** (100 MHz, CDCl₃) δ 141.8, 126.4, 120.2, 118.3, 115.6, 111.7, 25.9, 20.5, 12.5. **ESI-HRMS:** *m/z* calcd. for C₁₁H₁₄N₂ [M+H]⁺: 175.1235, found 175.1227.

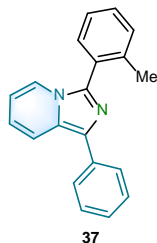


1-methyl-3-(pentan-2-yl)imidazo[1,5-a]pyridine: Following the **General procedure C1**, a mixture of 2-acetylpyridine (0.1 mmol), 2-methylpentanal (2.0 equiv.) and chitosan (3.0 equiv.) in CF₃COOH (1.0 mL) was stirred at 140 °C under Ar atmosphere for 36 h. Chitosan was dried at 100 °C overnight. Work-up gave product **33** (10.5 mg, 0.052 mmol, isolated yield 52%) as a brown liquid. ¹H NMR (300 MHz, CDCl₃) δ 7.65 (d, *J* = 7.2 Hz, 1H), 7.28 (d, *J* = 9.0 Hz, 1H), 6.50 (dd, *J* = 9.0, 6.3 Hz, 1H), 6.44 – 6.40 (m, 1H), 3.15 (h, *J* = 7.0 Hz, 1H), 2.48 (s, 3H), 1.93 – 1.85 (m, 1H), 1.73 – 1.65 (m, 1H), 1.39 (d, *J* = 6.9 Hz, 3H), 1.23 – 1.13 (m, 2H), 0.89 (t, *J* = 7.2 Hz, 3H). ¹³C NMR (75 MHz, CDCl₃) δ 141.4, 126.4, 126.1, 120.1, 118.2, 115.6, 111.7, 37.6, 30.9, 20.6, 18.6, 14.0, 12.3. **ESI-HRMS:** *m/z* calcd. for C₁₃H₁₈N₂ [M+H]⁺: 203.1548, found 203.1543.

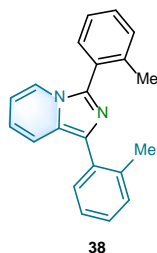


1-(pyridin-2-yl)-3-(o-tolyl)imidazo[1,5-a]pyridine: Following the **General procedure C2**, a mixture of di(pyridin-2-yl)methanone (0.1 mmol), 2-methylbenzaldehyde (4.0 equiv.) and chitosan (2.5 equiv.) in CF₃COOH (0.7 mL) was stirred at 140 °C under Ar atmosphere for 36 h. Chitosan was dried at 100 °C overnight. Work-up gave product **35** (18.5 mg, 0.065 mmol, isolated yield 65%) as a yellow solid. **m.p.:** 90–91 °C. Following the **General procedure C4**, a mixture of di(pyridin-2-yl)methanone (0.1 mmol), 2-methylbenzaldehyde (4.0 equiv.) and chitin (2.5 equiv.) in CF₃COOH (0.7 mL) was stirred at 140 °C under Ar atmosphere for 36 h. Chitin was dried under 100 °C overnight. Work-up gave product **35** (7.4 mg, 0.026 mmol, isolated yield 26%) as a yellow solid.

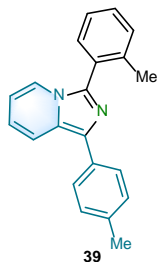
¹H NMR (400 MHz, CDCl₃) δ 8.73 (d, *J* = 9.2 Hz, 1H), 8.65 (d, *J* = 4.0 Hz, 1H), 8.26 (d, *J* = 8.0 Hz, 1H), 7.71 (t, *J* = 7.6 Hz, 1H), 7.64 (d, *J* = 7.2 Hz, 1H), 7.52 (d, *J* = 7.6 Hz, 1H), 7.44 – 7.33 (m, 3H), 7.11 – 7.08 (m, 1H), 6.94 – 6.90 (m, 1H), 6.59 (t, *J* = 6.8 Hz, 1H), 2.28 (s, 3H). ¹³C NMR (100 MHz, CDCl₃) δ 155.1, 148.9, 138.4, 137.6, 136.1, 130.7, 130.5, 129.8, 129.5, 129.1, 129.0, 126.0, 121.5, 121.5, 120.8, 120.2, 119.7, 113.5, 19.6. **ESI-HRMS:** *m/z* calcd. for C₁₉H₁₅N₃ [M+H]⁺: 286.1344, found 286.1342.



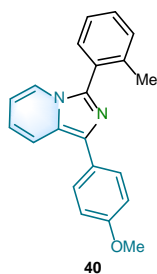
1-phenyl-3-(o-tolyl)imidazo[1,5-a]pyridine: Following the **General procedure C2**, a mixture of phenyl(pyridin-2-yl)methanone (0.1 mmol), 2-methylbenzaldehyde (4.0 equiv.) and chitosan (2.5 equiv.) in CF₃COOH (0.7 mL) was stirred at 140 °C under Ar atmosphere for 36 h. Chitosan was dried at 100 °C overnight. Work-up gave product **37** (20.2 mg, 0.071 mmol, isolated yield 71%) as a yellow liquid. **¹H NMR** (400 MHz, CDCl₃) δ 7.88 (d, *J* = 7.6 Hz, 2H), 7.78 (d, *J* = 9.2 Hz, 1H), 7.52 (d, *J* = 6.8 Hz, 1H), 7.42 – 7.36 (m, 3H), 7.32 – 7.28 (m, 2H), 7.26 – 7.16 (m, 2H), 6.72 – 6.68 (m, 1H), 6.45 – 6.41 (m, 1H), 2.17 (s, 3H). **¹³C NMR** (100 MHz, CDCl₃) δ 138.5, 137.8, 135.1, 131.1, 130.8, 130.6, 129.5, 129.2, 128.6, 126.6, 126.3, 126.1, 121.9, 119.5, 118.9, 112.8, 19.7. **ESI-HRMS:** *m/z* calcd. for C₂₀H₁₆N₂ [M+H]⁺: 285.1392, found 285.1390.



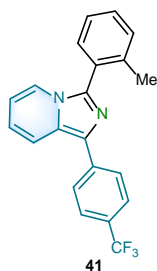
1,3-di-o-tolylimidazo[1,5-a]pyridine: Following the **General procedure C2**, a mixture of pyridin-2-yl(o-tolyl)methanone (0.1 mmol), 2-methylbenzaldehyde (4.0 equiv.) and chitosan (2.5 equiv.) in CF₃COOH (0.7 mL) was stirred at 140 °C under Ar atmosphere for 36 h. Chitosan was dried at 100 °C overnight. Work-up gave product **38** (16.1 mg, 0.054 mmol, isolated yield 54%) as a green solid. **m.p.:** 112–113 °C. **¹H NMR** (400 MHz, CDCl₃) δ 7.62 – 7.60 (m, 1H), 7.45 – 7.39 (m, 3H), 7.31 (d, *J* = 2.8 Hz, 2H), 7.27 – 7.23 (m, 2H), 7.21 – 7.18 (m, 2H), 6.67 – 6.64 (m, 1H), 6.49 – 6.44 (m, 1H), 2.41 (s, 3H), 2.24 (s, 3H). **¹³C NMR** (100 MHz, CDCl₃) δ 138.5, 137.3, 130.9, 130.8, 130.3, 130.2, 129.5, 127.4, 126.0, 125.5, 121.5, 118.9, 112.9, 20.7, 19.9. **ESI-HRMS:** *m/z* calcd. for C₂₁H₁₈N₂ [M+H]⁺: 299.1548, found 299.1546.



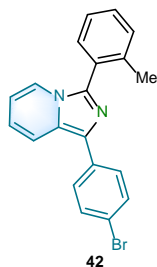
3-(o-tolyl)-1-(p-tolyl)imidazo[1,5-a]pyridine: Following the **General procedure C2**, a mixture of pyridin-2-yl(p-tolyl)methanone (0.1 mmol), 2-methylbenzaldehyde (4.0 equiv.) and chitosan (2.5 equiv.) in CF₃COOH (0.7 mL) was stirred at 140 °C under Ar atmosphere for 36 h. Chitosan was dried at 100 °C overnight. Work-up gave product **39** (20.2 mg, 0.083 mmol, isolated yield 68%) as a yellow liquid. ¹H NMR (400 MHz, CDCl₃) δ 7.79 – 7.75 (m, 3H), 7.52 (d, *J* = 7.2 Hz, 1H), 7.43 (d, *J* = 7.6 Hz, 1H), 7.35 – 7.18 (m, 5H), 6.69 (dd, *J* = 9.3, 6.3 Hz, 1H), 6.45 – 6.42 (m, 1H), 2.33 (s, 3H), 2.18 (s, 3H). ¹³C NMR (100 MHz, CDCl₃) δ 138.5, 137.6, 136.0, 132.2, 131.2, 130.8, 130.6, 129.5, 129.4, 129.3, 126.5, 126.3, 126.1, 121.8, 119.2, 119.1, 112.7, 21.2, 19.8. **ESI-HRMS:** *m/z* calcd. for C₂₁H₁₈N₂ [M+H]⁺: 299.1548, found 299.1544.



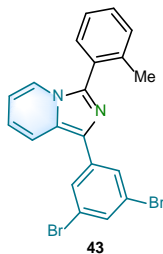
1-(4-methoxyphenyl)-3-(o-tolyl)imidazo[1,5-a]pyridine: Following the **General procedure C2**, a mixture of (4-methoxyphenyl)(pyridin-2-yl)methanone (0.1 mmol), 2-methylbenzaldehyde (4.0 equiv.) and chitosan (2.5 equiv.) in CF₃COOH (0.7 mL) was stirred at 140 °C under Ar atmosphere for 36 h. Chitosan was dried at 100 °C overnight. Work-up gave product **40** (22.9 mg, 0.073 mmol, isolated yield 73%) as a yellow liquid. ¹H NMR (300 MHz, CDCl₃) δ 7.81 (d, *J* = 9.0 Hz, 2H), 7.73 (d, *J* = 9.3 Hz, 1H), 7.52 (d, *J* = 7.2 Hz, 1H), 7.43 (d, *J* = 7.2 Hz, 1H), 7.33 – 7.23 (m, 3H), 6.94 (d, *J* = 8.7 Hz, 2H), 6.68 (ddd, *J* = 9.3, 6.3, 0.9 Hz, 1H), 6.45 – 6.41 (m, 1H), 3.79 (s, 3H), 2.19 (s, 3H). ¹³C NMR (100 MHz, CDCl₃) δ 138.5, 137.3, 130.9, 130.8, 130.3, 130.2, 129.5, 127.4, 126.0, 125.5, 121.5, 118.9, 112.9, 20.7, 19.9. **ESI-HRMS:** *m/z* calcd. for C₂₁H₁₈N₂O [M+H]⁺: 315.1497, found 315.1495.



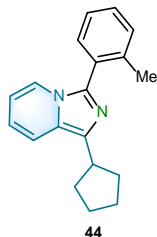
3-(o-tolyl)-1-(4-(trifluoromethyl)phenyl)imidazo[1,5-a]pyridine: Following the **General procedure C2**, a mixture of pyridin-2-yl(4-(trifluoromethyl)phenyl)methanone (0.1 mmol), 2-methylbenzaldehyde (4.0 equiv.) and chitosan (2.5 equiv.) in CF₃COOH (0.7 mL) was stirred at 140 °C under Ar atmosphere for 36 h. Chitosan was dried at 100 °C overnight. Work-up gave product **41** (28.5 mg, 0.081 mmol, isolated yield 81%) as a yellow liquid. ¹H NMR (400 MHz, CDCl₃) δ 8.01 (d, *J* = 8.0 Hz, 2H), 7.81 (d, *J* = 9.2 Hz, 1H), 7.64 – 7.58 (m, 3H), 7.43 (d, *J* = 7.6 Hz, 1H), 7.36 – 7.26 (m, 3H), 6.82 (dd, *J* = 9.2, 6.4 Hz, 1H), 6.54 – 6.50 (m, 1H), 2.19 (s, 3H). ¹³C NMR (100 MHz, CDCl₃) δ 138.7, 138.5, 138.4, 131.2, 130.9, 130.6, 129.8, 129.5, 128.9, 127.9 (q, ²*J*_{C-F} = 33.4 Hz), 127.4, 126.3, 126.2, 125.6 (q, ³*J*_{C-F} = 3.7 Hz), 124.5 (q, ¹*J*_{C-F} = 270.1 Hz), 122.3, 120.8, 118.6, 113.1, 19.8. ¹⁹F NMR (377 MHz, CDCl₃) δ -62.2. **ESI-HRMS:** *m/z* calcd. for C₂₁H₁₅F₃N₂ [M+H]⁺: 353.1266, found 353.1261.



1-(4-bromophenyl)-3-(o-tolyl)imidazo[1,5-a]pyridine: Following the **General procedure C2**, a mixture of (4-bromophenyl)(pyridin-2-yl)methanone (0.1 mmol), 2-methylbenzaldehyde (4.0 equiv.) and chitosan (2.5 equiv.) in CF₃COOH (0.7 mL) was stirred at 140 °C under Ar atmosphere for 36 h. Chitosan was dried at 100 °C overnight. Work-up gave product **42** (27.1 mg, 0.075 mmol, isolated yield 75%) as a yellow solid. **m.p.:** 123–124 °C. ¹H NMR (400 MHz, CDCl₃) δ 7.78 – 7.73 (m, 3H), 7.55 (d, *J* = 6.8 Hz, 1H), 7.50 (d, *J* = 8.4 Hz, 1H), 7.42 (d, *J* = 7.2 Hz, 1H), 7.34 – 7.26 (m, 3H), 6.78 – 6.74 (m, 1H), 6.50 – 6.47 (m, 1H), 2.18 (s, 3H). ¹³C NMR (100 MHz, CDCl₃) δ 138.5, 138.0, 134.1, 131.7, 130.9, 130.6, 129.8, 129.7, 128.0, 126.7, 126.2, 122.1, 120.1, 120.0, 118.7, 113.0, 19.8. **ESI-HRMS:** *m/z* calcd. for C₂₀H₁₅BrN₂ [M+H]⁺: 363.0497, found 363.0494.



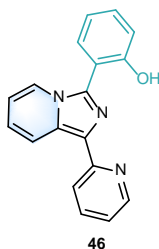
1-(3,5-dibromophenyl)-3-(o-tolyl)imidazo[1,5-a]pyridine: Following the **General procedure C2**, a mixture of (3,5-dibromophenyl)(pyridin-2-yl)methanone (0.1 mmol), 2-methylbenzaldehyde (4.0 equiv.) and chitosan (2.5 equiv.) in CF_3COOH (0.7 mL) was stirred at 140 °C under Ar atmosphere for 36 h. Chitosan was dried at 100 °C overnight. Work-up gave product **43** (28.2 mg, 0.064 mmol, isolated yield 64%) as a yellow solid. **m.p.:** 138–140 °C. $^1\text{H NMR}$ (400 MHz, CDCl_3) δ 7.99 – 7.98 (m, 2H), 7.76 (d, J = 9.2 Hz, 1H), 7.57 (d, J = 7.2 Hz, 1H), 7.47 (dt, J = 3.2, 1.7 Hz, 1H), 7.41 (d, J = 7.2 Hz, 1H), 7.36 – 7.26 (m, 3H), 6.87 – 6.82 (m, 1H), 6.55 – 6.51 (m, 1H), 2.17 (s, 3H). $^{13}\text{C NMR}$ (100 MHz, CDCl_3) δ 138.7, 138.5, 138.4, 131.3, 130.9, 130.6, 129.9, 128.7, 127.9, 127.7, 127.4, 126.2, 123.2, 122.3, 121.1, 118.4, 113.2, 19.7. **ESI-HRMS:** m/z calcd. for $\text{C}_{20}\text{H}_{14}\text{Br}_2\text{N}_2$ $[\text{M}+\text{H}]^+$: 440.9602, found 440.9609.



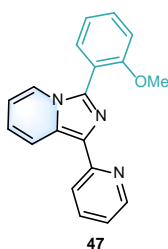
1-cyclopentyl-3-(o-tolyl)imidazo[1,5-a]pyridine: Following the **General procedure C2**, a mixture of cyclopentyl(pyridin-2-yl)methanone (0.1 mmol), 2-methylbenzaldehyde (4.0 equiv.) and chitosan (2.5 equiv.) in CF_3COOH (0.7 mL) was stirred at 140 °C under Ar atmosphere for 36 h. Chitosan was dried at 100 °C overnight. Work-up gave product **44** (25.36 mg, 0.092 mmol, isolated yield 92%) as a brown solid. **m.p.:** 72–73 °C. Following the **General procedure C4**, a mixture of cyclopentyl(pyridin-2-yl)methanone (0.1 mmol), 2-methylbenzaldehyde (4.0 equiv.) and chitin (2.5 equiv.) in CF_3COOH (0.7 mL) was stirred at 140 °C under Ar atmosphere for 36 h. Chitin was dried at 100 °C overnight. Work-up gave product **44** (24.6 mg, 0.089 mmol, isolated yield 89%) as a brown solid.

$^1\text{H NMR}$ (400 MHz, CDCl_3) δ 7.42 (d, J = 7.2 Hz, 1H), 7.37 (d, J = 9.2 Hz, 1H), 7.32 (d, J = 7.2 Hz, 1H), 7.26 – 7.23 (m, 2H), 7.20 – 7.16 (m, 1H), 6.49 (ddd, J = 9.2, 6.3, 0.8 Hz, 1H), 6.34 – 6.30 (m, 1H), 3.37 –

3.28 (m, 1H), 2.10 (s, 3H), 2.04 – 1.99 (m, 2H), 1.91 – 1.87 (m, 2H), 1.81 – 1.78 (m, 2H), 1.64 – 1.60 (m, 2H). ^{13}C NMR (100 MHz, CDCl_3) δ 138.4, 136.1, 135.9, 130.7, 130.4, 129.5, 129.2, 125.9, 125.8, 121.2, 118.2, 116.5, 112.3, 38.3, 33.4, 25.7, 19.6. **ESI-HRMS**: m/z calcd. for $\text{C}_{19}\text{H}_{20}\text{N}_2$ $[\text{M}+\text{H}]^+$: 277.1705, found 277.1703.



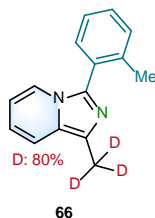
2-(1-(pyridin-2-yl)imidazo[1,5-a]pyridin-3-yl)phenol: Following the **General procedure C2**, a mixture of di(pyridin-2-yl)methanone (0.1 mmol), 2-hydroxybenzaldehyde (4.0 equiv.) and chitosan (2.5 equiv.) in CF_3COOH (0.7 mL) was stirred at 140 °C under Ar atmosphere for 36 h. Chitosan was dried at 100 °C overnight. Work-up gave product **46** (5.2 mg, 0.085 mmol, isolated yield 18%) as a yellow solid. ^1H NMR (400 MHz, CDCl_3) δ 8.73 (d, $J = 9.2$ Hz, 1H), 8.57 (d, $J = 4.0$ Hz, 1H), 8.48 (d, $J = 7.2$ Hz, 1H), 8.05 (d, $J = 8.0$ Hz, 1H), 7.72 (dd, $J = 7.8, 1.4$ Hz, 1H), 7.67 (td, $J = 7.8, 2.0$ Hz, 1H), 7.28 – 7.24 (m, 1H), 7.12 (dd, $J = 8.2, 1.0$ Hz, 1H), 7.06 (ddd, $J = 7.2, 4.8, 1.2$ Hz, 1H), 6.97 – 6.90 (m, 2H), 6.72 – 6.68 (m, 1H). ^{13}C NMR (100 MHz, CDCl_3) δ 156.3, 153.9, 149.1, 136.4, 135.5, 130.1, 129.6, 128.6, 124.5, 122.3, 122.3, 121.7, 120.9, 119.8, 119.1, 117.8, 114.8, 114.0. The compound is known, and the NMR data is in accordance with the previous literature.³³⁰



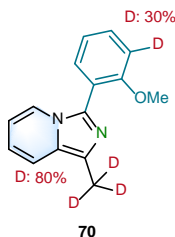
3-(2-methoxyphenyl)-1-(pyridin-2-yl)imidazo[1,5-a]pyridine: Following the **General procedure C2**, a mixture of di(pyridin-2-yl)methanone (0.1 mmol), 2-methoxybenzaldehyde (4.0 equiv.) and chitosan (2.5 equiv.) in CF_3COOH (0.7 mL) was stirred at 140 °C under Ar atmosphere for 36 h. Chitosan was dried at 100 °C overnight. Work-up gave product **47** (13.3 mg, 0.044 mmol, isolated yield 44%) as a yellow solid. Following the **General procedure C4**, a mixture of di(pyridin-2-yl)methanone (0.1 mmol), 2-

methoxybenzaldehyde (4.0 equiv.) and chitin (2.5 equiv.) in CF₃COOH (0.7 mL) was stirred at 140 °C under Ar atmosphere for 36 h. Chitin was dried at 100 °C overnight. Work-up gave product **47** (4.8 mg, 0.016 mmol, isolated yield 16%) as a yellow solid.

¹H NMR (400 MHz, CDCl₃) δ 8.62 (d, *J* = 9.2 Hz, 1H), 8.56 – 8.54 (m, 1H), 8.16 (d, *J* = 8.4 Hz, 1H), 7.64 – 7.58 (m, 2H), 7.54 (d, *J* = 7.2 Hz, 1H), 7.44 – 7.40 (m, 1H), 7.08 – 7.05 (m, 1H), 7.01 – 6.97 (m, 2H), 6.86 (dd, *J* = 9.2, 6.4 Hz, 1H), 6.55 – 6.52 (m, 1H), 3.73 (s, 3H). **¹³C NMR** (100 MHz, CDCl₃) δ 157.6, 155.3, 148.9, 136.1, 132.8, 131.0, 130.1, 130.0, 123.3, 121.2, 121.2, 120.9, 120.2, 119.9, 119.1, 112.7, 111.2, 55.6. The compound is known, and the NMR data is in accordance with the previous literature.²²⁶

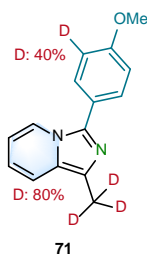


1-(methyl-d₃)-3-(o-tolyl)imidazo[1,5-a]pyridine: Following the **General procedure C5**, a mixture of 2-acetylpyridine (0.1 mmol), 2-methylbenzaldehyde (2.0 equiv.) and chitosan (3.0 equiv.) in CF₃COOD (1.0 mL) was stirred at 140 °C under Ar atmosphere for 36 h. Chitosan was dried at 100 °C overnight. Work-up gave product **66** (16.4 mg, 0.073 mmol, isolated yield 73%) as a yellow solid. **m.p.** 62–64 °C. D incorporation by ¹H NMR: 80%. **¹H NMR** (300 MHz, CDCl₃) δ 7.43 (d, *J* = 7.2 Hz, 1H), 7.37 – 7.19 (m, 5H), 6.53 (dd, *J* = 9.1, 6.3 Hz, 1H), 6.37 – 6.32 (m, 1H), 2.48 (s, 0.59H), 2.13 (s, 3H). **ESI-HRMS:** *m/z* calcd. for C₁₅H₁₁D₃N₂ [M+H]⁺: 226.1424, found 226.1420.

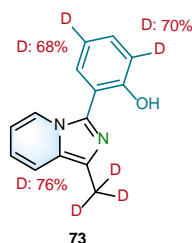


3-(2-3-(2-methoxyphenyl-3-d)-1-(methyl-d₃)imidazo[1,5-a]pyridine: Following the **General procedure C5**, a mixture of 2-acetylpyridine (0.1 mmol), 2-methoxybenzaldehyde (2.0 equiv.) and chitosan (3.0 equiv.) in CF₃COOD (1.0 mL) was stirred at 140 °C under Ar atmosphere for 36 h. Chitosan

was dried at 100 °C overnight. Work-up gave product **70** (14.7 mg, 0.061 mmol, isolated yield 61%) as a yellow solid; **m.p.** 44–46 °C. D incorporation by ^1H NMR: 80% and 30%. ^1H NMR (400 MHz, CDCl_3) δ 7.53 (dd, J = 4.0, 1.6 Hz, 1H), 7.45 – 7.43 (m, 1H), 7.37 – 7.31 (m, 2H), 7.04 – 7.00 (m, 0.70H), 6.96 (dd, J = 8.4, 4.4 Hz, 1H), 6.57 – 6.53 (m, 1H), 6.40 – 6.35 (m, 1H), 3.73 (s, 3H), 2.49 (s, 0.59H). **ESI-HRMS**: m/z calcd. for $\text{C}_{15}\text{H}_{10}\text{D}_4\text{N}_2\text{O}$ $[\text{M}+\text{H}]^+$: 243.1435, found 243.1422.

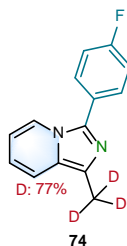


3-(4-methoxyphenyl-3-d)-1-(methyl-d₃)imidazo[1,5-a]pyridine: Following the **General procedure C5**, a mixture of 2-acetylpyridine (0.1 mmol), 4-methoxybenzaldehyde (2.0 equiv.) and chitosan (3.0 equiv.) in CF_3COOD (1.0 mL) was stirred at 140 °C under Ar atmosphere for 36 h. Chitosan was dried at 100 °C overnight. Work-up gave product **71** (15.2 mg, 0.063 mmol, isolated yield 63%) as a yellow solid. D incorporation by ^1H NMR: 80% and 40%. **m.p.** 43–45 °C. ^1H NMR (400 MHz, CDCl_3) δ 8.02 (d, J = 7.6 Hz, 1H), 7.62 (d, J = 8.8 Hz, 2H), 7.29 (d, J = 9.2 Hz, 1H), 6.96 (d, J = 9.2 Hz, 1.59H), 6.51 (ddd, J = 9.2, 6.4, 0.8 Hz, 1H), 6.38 (ddd, J = 7.6, 6.4, 1.2 Hz, 1H), 3.80 (s, 3H), 2.46 (s, 0.60H). **ESI-HRMS**: m/z calcd. for $\text{C}_{15}\text{H}_{10}\text{D}_4\text{N}_2\text{O}$ $[\text{M}+\text{H}]^+$: 243.1435, found 243.1422.

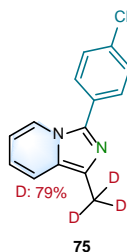


2-(1-(methyl-d₃)imidazo[1,5-a]pyridin-3-yl)phen-4,6-d₂-ol: Following the **General procedure C5**, a mixture of 2-acetylpyridine (0.1 mmol), 2-hydroxybenzaldehyde (2.0 equiv.) and chitosan (3.0 equiv.) in CF_3COOD (1.0 mL) was stirred at 140 °C under Ar atmosphere for 36 h. Chitosan was dried at 100 °C overnight. Work-up gave product **73** (6.9 mg, 0.030 mmol, isolated yield 30%) as a white solid. **m.p.** 130–132 °C. D incorporation by ^1H NMR: 76%, 70% and 68%. ^1H NMR (400 MHz, CDCl_3) δ 8.35 (dd, J = 7.0, 3.6 Hz, 1H), 7.67 – 7.65 (m, 1H), 7.38 – 7.34 (m, 1H), 7.19 (d, J = 4.4 Hz, 1H), 7.07 (dd, J = 8.0,

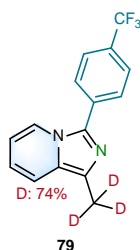
4.0 Hz, 0.32H), 6.93 – 6.88 (m, 0.30H), 6.65 – 6.60 (m, 1H), 6.56 – 6.51 (m, 1H), 2.46 (s, 0.73H). **ESI-HRMS**: m/z calcd. for $C_{14}H_7D_5N_2O$ $[M+H]^+$: 230.1342, found 230.1330.



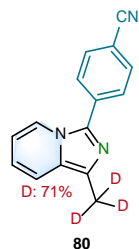
3-(4-fluorophenyl)-1-(methyl- d_3)imidazo[1,5-a]pyridine: Following the **General procedure C5**, a mixture of 2-acetylpyridine (0.1 mmol), 4-fluorobenzaldehyde (2.0 equiv.) and chitosan (3.0 equiv.) in CF_3COOD (1.0 mL) was stirred at 140 °C under Ar atmosphere for 36 h. Chitosan was dried at 100 °C overnight. Work-up gave product **74** (16.9 mg, 0.074 mmol, isolated yield 74%) as a yellow solid. D incorporation by 1H NMR: 77%. **m.p.** 93–95 °C. 1H NMR (300 MHz, $CDCl_3$) δ 8.01 (d, J = 7.2 Hz, 1H), 7.67 (dd, J = 8.4, 5.4 Hz, 2H), 7.31 (d, J = 9.0 Hz, 1H), 7.14 – 7.09 (m, 2H), 6.54 (dd, J = 9.0, 6.3 Hz, 1H), 6.44 – 6.40 (m, 1H), 2.46 (s, 0.68H). **ESI-HRMS**: m/z calcd. for $C_{14}H_8D_3N_2F$ $[M+H]^+$: 230.1173, found 230.1168.



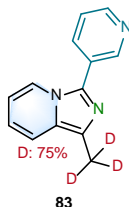
3-(4-chlorophenyl)-1-(methyl- d_3)imidazo[1,5-a]pyridine: Following the **General procedure C5**, a mixture of 2-acetylpyridine (0.1 mmol), 4-chlorobenzaldehyde (2.0 equiv.) and chitosan (3.0 equiv.) in CF_3COOD (1.0 mL) was stirred at 140 °C under Ar atmosphere for 36 h. Chitosan was dried at 100 °C overnight. Work-up gave product **75** (18.6 mg, 0.076 mmol, isolated yield 76%) as a yellow solid. D incorporation by 1H NMR: 79%. **m.p.** 84–86 °C. 1H NMR (400 MHz, $CDCl_3$) δ 8.04 (d, J = 7.2 Hz, 1H), 7.65 (d, J = 8.4 Hz, 2H), 7.40 (d, J = 8.4 Hz, 2H), 7.34 (d, J = 9.2 Hz, 1H), 6.60 (dd, J = 9.2, 6.4 Hz, 1H), 6.50 – 6.46 (m, 1H), 2.47 (s, 0.62H). **ESI-HRMS**: m/z calcd. for $C_{14}H_8D_3N_2Cl$ $[M+H]^+$: 246.0877, found 246.0873.



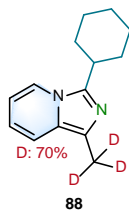
1-(methyl-d₃)-3-(4-(trifluoromethyl)phenyl)imidazo[1,5-a]pyridine: Following the **General procedure C5**, a mixture of 2-acetylpyridine (0.1 mmol), 4-trifluoromethylbenzaldehyde (2.0 equiv.) and chitosan (3.0 equiv.) in CF₃COOD (1.0 mL) was stirred at 140 °C under Ar atmosphere for 36 h. Chitosan was dried at 100 °C overnight. Work-up gave product **79** (18.1 mg, 0.065 mmol, isolated yield 65%) as a yellow solid. D incorporation by ¹H NMR: 74%. **m.p.** 69–71 °C. **¹H NMR** (300 MHz, CDCl₃) δ 8.12 (d, *J* = 7.2 Hz, 1H), 7.84 (d, *J* = 8.1 Hz, 2H), 7.66 (d, *J* = 8.1 Hz, 2H), 7.35 (d, *J* = 9.3 Hz, 1H), 6.63 – 6.58 (m, 1H), 6.51 – 6.46 (m, 1H), 2.47 (s, 0.77H). **ESI-HRMS:** *m/z* calcd. for C₁₅H₈D₃F₃N₂ [M+H]⁺: 280.1141, found 280.1137.



4-(1-methylimidazo[1,5-a]pyridin-3-yl)benzonitrile: Following the **General procedure C5**, a mixture of 2-acetylpyridine (0.1 mmol), 4-formylbenzonitrile (2.0 equiv.) and chitosan (3.0 equiv.) in CF₃COOD (1.0 mL) was stirred at 140 °C under Ar atmosphere for 36 h. Chitosan was dried at 100 °C overnight. Work-up gave product **80** (7.8 mg, 0.033 mmol, isolated yield 33%) as a yellow solid. D incorporation by ¹H NMR: 71%. **m.p.** 120–122 °C. **¹H NMR** (300 MHz, CDCl₃) δ 8.16 (d, *J* = 7.5 Hz, 1H), 7.85 (q, *J* = 8.4 Hz, 4H), 7.36 (d, *J* = 9.9 Hz, 1H), 6.64 – 6.59 (m, 1H), 6.52 – 6.48 (m, 1H), 2.48 (s, 0.86H). **ESI-HRMS:** *m/z* calcd. for C₁₅H₈D₃N₃ [M+H]⁺: 237.1220, found 237.1219.

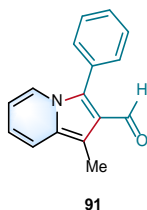


1-(methyl-d₃)-3-(pyridin-3-yl)imidazo[1,5-a]pyridine: Following the **General procedure C5**, a mixture of 2-acetylpyridine (0.1 mmol), nicotinaldehyde (2.0 equiv.) and chitosan (3.0 equiv.) in CF₃COOD (1.0 mL) was stirred at 140 °C under Ar atmosphere for 36 h. Chitosan was dried at 100 °C overnight. Work-up gave product **83** (12.72 mg, 0.060 mmol, isolated yield 60%) as a yellow solid. D incorporation by ¹H NMR: 75%. **m.p.** 45–47 °C. **¹H NMR** (400 MHz, CDCl₃) δ 8.97 (s, 1H), 8.56 – 8.54 (m, 1H), 8.08 (d, *J* = 7.2 Hz, 1H), 8.03 (d, *J* = 8.0 Hz, 1H), 7.35 (dd, *J* = 7.6, 1.2 Hz, 2H), 6.63 – 6.59 (m, 1H), 6.51 – 6.47 (m, 1H), 2.47 (s, 0.75H). **ESI-HRMS:** *m/z* calcd. for C₁₃H₈D₃N₃ [M+H]⁺: 213.1220, found 213.1213.

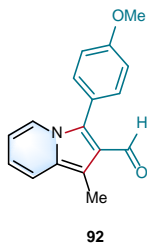


3-cyclohexyl-1-(methyl-d₃)imidazo[1,5-a]pyridine: Following the **General procedure C5**, a mixture of 2-acetylpyridine (0.1 mmol), cyclohexanecarbaldehyde (2.0 equiv.) and chitosan (3.0 equiv.) in CF₃COOD (1.0 mL) was stirred at 140 °C under Ar atmosphere for 36 h. Chitosan was dried at 100 °C overnight. Work-up gave product **88** (13.9 mg, 0.064 mmol, isolated yield 64%) as a white solid. D incorporation by ¹H NMR: 70%. **m.p.** 74–76 °C. **¹H NMR** (300 MHz, CDCl₃) δ 7.58 (d, *J* = 7.2 Hz, 1H), 7.21 (d, *J* = 8.4 Hz, 1H), 6.45 – 6.32 (m, 2H), 2.39 (s, 0.90H), 1.94 – 1.64 (m, 8H), 1.37 – 1.31 (m, 3H). **ESI-HRMS:** *m/z* calcd. for C₁₄H₁₅D₃N₂ [M+H]⁺: 218.1737, found 218.1736.

5.3.4 Anomeric stereoauxiliary-organocatalyzed one-pot site-selective C₂-aldehylation for trisubstituted indolizine-2-carbaldehydes

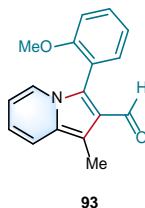


Following the **General procedure D1**, a mixture of α,β -unsaturated aldehydes (0.2 mmol), heteroaryl ketones (2.5 equiv.), catalyst **3b** (0.04 mmol), AcOH (2.0 equiv.) and LiSO₃CF₃ (3.0 equiv.) in the CF₃CH₂OH (0.9 mL) were stirred at 80 °C under Ar atmosphere for 18 h. Work-up gave product 1-methyl-3-phenylindolizine-2-carbaldehyde (**91**, 44.7 mg, 0.19 mmol, isolated yield 95%) as a yellow liquid. **FT-IR**: ν (cm⁻¹): 3052, 2922, 2749, 1661, 1517, 1477, 1430, 1383, 1358, 1319, 1247, 1218, 1142, 1115, 1076, 997, 940, 872, 830, 756, 736, 715, 699, 682, 666, 620, 534, 482, 431. **¹H NMR** (400 MHz, CDCl₃) δ 9.91 (s, 1H), 7.76 (d, J = 7.2 Hz, 1H), 7.47 – 7.45 (m, 2H), 7.42 – 7.40 (m, 3H), 7.32 (d, J = 9.2 Hz, 1H), 6.61 – 6.56 (m, 1H), 6.40 – 6.36 (m, 1H), 2.53 (s, 3H). **¹³C NMR** (100 MHz, CDCl₃) δ 189.7, 131.3, 131.0, 130.6, 129.0, 128.9, 128.8, 123.4, 122.3, 119.1, 117.6, 112.7, 109.9, 9.7. **ESI-HRMS**: m/z calcd. for C₁₆H₁₃NO [M+H]⁺: 236.1075, found 236.1071.

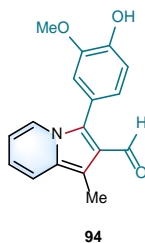


Following the **General procedure D1**, a mixture of α,β -unsaturated aldehyde (0.2 mmol), heteroaryl ketone (2.5 equiv.), catalyst **3b** (0.04 mmol), AcOH (2.0 equiv.) and LiSO₃CF₃ (3.0 equiv.) in the CF₃CH₂OH (0.9 mL) were stirred at 80 °C under Ar atmosphere for 18 h. Work-up gave product 3-(4-methoxyphenyl)-1-methylindolizine-2-carbaldehyde (**92**, 33.4 mg, 0.13 mmol, isolated yield 63%) as a yellow liquid. **FT-IR**: ν (cm⁻¹): 2922, 2840, 2741, 1661, 1607, 1574, 1527, 1482, 1432, 1381, 1356, 1319, 1286, 1245, 1220, 1175, 1150, 1111, 1026, 878, 824, 785, 736, 684, 641, 626, 583, 515, 433, 404. **¹H NMR** (400 MHz, CDCl₃) δ 9.89 (s, 1H), 7.71 (d, J = 7.2 Hz, 1H), 7.34 – 7.28 (m, 3H), 6.98 (d, J = 8.8 Hz, 2H), 6.57 – 6.53 (m, 1H), 6.35 (t, J = 6.6 Hz, 1H), 3.81 (s, 3H), 2.52 (s, 3H). **¹³C NMR** (100 MHz, CDCl₃)

δ 189.7, 160.1, 132.2, 131.4, 130.3, 123.3, 122.3, 120.8, 119.0, 117.4, 114.5, 112.5, 109.6, 55.4, 9.6. **ESI-HRMS**: m/z calcd. for $C_{17}H_{15}NO_2$ $[M+H]^+$: 266.1181, found 266.1177.

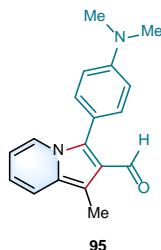


Following the **General procedure D1**, a mixture of α,β -unsaturated aldehyde (0.2 mmol), heteroaryl ketone (2.5 equiv.), catalyst **3b** (0.04 mmol), AcOH (2.0 equiv.) and $LiSO_3CF_3$ (3.0 equiv.) in the CF_3CH_2OH (0.9 mL) were stirred at 80 °C under Ar atmosphere for 18 h. Work-up gave product 3-(2-methoxyphenyl)-1-methylindolizine-2-carbaldehyde (**93**, 50.4 mg, 0.19 mmol, isolated yield 95%) as a yellow liquid. **FT-IR**: ν (cm^{-1}): 2920, 2833, 2745, 1663, 1601, 1576, 1515, 1463, 1432, 1383, 1358, 1321, 1288, 1278, 1245, 1216, 1181, 1152, 1146, 1127, 1100, 1047, 1022, 940, 880, 851, 835, 783, 736, 713, 684, 659, 573, 552, 528, 462, 429. **1H NMR** (400 MHz, $CDCl_3$) δ 9.85 (s, 1H), 7.43 – 7.38 (m, 1H), 7.36 (d, J = 7.2 Hz, 1H), 7.31 (d, J = 9.2 Hz, 1H), 7.25 (dd, J = 7.2, 1.6 Hz, 1H), 7.03 – 6.98 (m, 2H), 6.57 (dd, J = 9.1, 6.3 Hz, 1H), 6.36 (t, J = 6.8 Hz, 1H), 3.70 (s, 3H), 2.54 (s, 3H). **^{13}C NMR** (100 MHz, $CDCl_3$) δ 189.7, 157.9, 133.7, 130.8, 128.2, 123.6, 123.4, 120.7, 118.8, 117.4, 117.2, 112.0, 111.2, 109.5, 55.5, 9.7. **ESI-HRMS**: m/z calcd. for $C_{17}H_{15}NO_2$ $[M+H]^+$: 266.1181, found 266.1176.

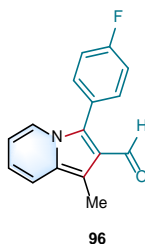


Following the **General procedure D1**, a mixture of α,β -unsaturated aldehyde (0.2 mmol), heteroaryl ketone (2.5 equiv.), catalyst **3b** (0.04 mmol), AcOH (2.0 equiv.) and $LiSO_3CF_3$ (3.0 equiv.) in the CF_3CH_2OH (0.9 mL) were stirred at 80 °C under Ar atmosphere for 18 h. Work-up gave product 3-(4-hydroxy-3-methoxyphenyl)-1-methylindolizine-2-carbaldehyde (**94**, 36.5 mg, 0.13 mmol, isolated yield 63%) as a yellow solid. **mp**: 168–169 °C. **FT-IR**: ν (cm^{-1}): 3171, 2924, 2852, 1638, 1583, 1523, 1488, 1471, 1432, 1416, 1377, 1344, 1319, 1271, 1236, 1208, 1177, 1142, 1125, 1057, 1030, 964, 913, 888, 876, 812, 769, 736, 725, 684, 659, 647, 571, 558, 528, 517, 433, 410. **1H NMR** (400 MHz, $CDCl_3$) δ 9.92 (s,

1H), 7.74 (d, $J = 6.8$ Hz, 1H), 7.30 (d, $J = 8.8$ Hz, 1H), 7.00 (d, $J = 8.0$ Hz, 1H), 6.93 (d, $J = 8.0$ Hz, 1H), 6.86 (s, 1H), 6.58 – 6.54 (m, 1H), 6.37 (t, $J = 6.8$ Hz, 1H), 5.83 (s, 1H), 3.84 (s, 3H), 2.52 (s, 3H). **^{13}C NMR** (100 MHz, CDCl_3) δ 189.8, 146.9, 146.5, 131.6, 130.3, 124.4, 123.3, 122.5, 120.5, 119.1, 117.4, 115.0, 113.4, 112.6, 109.6, 56.1, 9.6. **ESI-HRMS**: m/z calcd. for $\text{C}_{17}\text{H}_{15}\text{NO}_3$ $[\text{M}+\text{H}]^+$: 282.1130, found 282.1124.

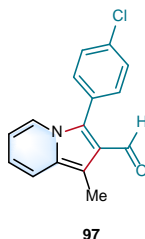


Following the **General procedure D2**, a mixture of α,β -unsaturated aldehyde (0.2 mmol), heteroaryl ketone (2.5 equiv.), catalyst **3b** (0.04 mmol), and LiSO_3CF_3 (3.0 equiv.) in the $\text{AcOH} : \text{CF}_3\text{CH}_2\text{OH}$ (0.4 : 0.5 mL) were stirred at 80 °C under Ar atmosphere for 36 h. Work-up gave product 3-(4-(dimethylamino)phenyl)-1-methylindolizine-2-carbaldehyde (**95**, 25.0 mg, 0.09 mmol, isolated yield 46%) as a yellow liquid. **FT-IR**: ν (cm^{-1}): 2924, 2854, 2800, 1661, 1605, 1531, 1488, 1432, 1352, 1222, 1195, 1164, 1113, 1057, 944, 876, 814, 736, 641, 552, 515, 435. **^1H NMR** (400 MHz, CDCl_3) δ 9.91 (s, 1H), 7.78 (d, $J = 7.6$ Hz, 1H), 7.30 – 7.25 (m, 4H), 6.77 (d, $J = 8.6$ Hz, 2H), 6.53 (dd, $J = 9.0, 6.4$ Hz, 1H), 6.33 (t, $J = 6.6$ Hz, 1H), 2.98 (s, 6H), 2.52 (s, 3H). **^{13}C NMR** (100 MHz, CDCl_3) δ 190.2, 150.6, 132.8, 131.8, 130.1, 123.1, 122.6, 119.0, 117.2, 115.7, 112.2, 112.2, 109.3, 40.3, 9.7. **ESI-HRMS**: m/z calcd. for $\text{C}_{18}\text{H}_{18}\text{N}_2\text{O}$ $[\text{M}+\text{H}]^+$: 279.1497, found 279.1502.

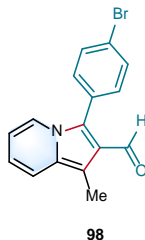


Following the **General procedure D1**, a mixture of α,β -unsaturated aldehyde (0.2 mmol), heteroaryl ketone (2.5 equiv.), catalyst **3b** (0.04 mmol), AcOH (2.0 equiv.) and LiSO_3CF_3 (3.0 equiv.) in the $\text{CF}_3\text{CH}_2\text{OH}$ (0.9 mL) were stirred at 80 °C under Ar atmosphere for 18 h. Work-up gave product 3-(4-fluorophenyl)-1-methylindolizine-2-carbaldehyde (**96**, 40.5 mg, 0.16 mmol, isolated yield 81%) as a

yellow liquid. **FT-IR:** ν (cm⁻¹): 3072, 2924, 2739, 1665, 1601, 1525, 1480, 1434, 1383, 1356, 1319, 1220, 1158, 1113, 1094, 1053, 938, 878, 828, 802, 736, 717, 684, 641, 571, 550, 509, 441, 420. **¹H NMR** (400 MHz, CDCl₃) δ 9.88 (s, 1H), 7.67 (d, J = 7.2 Hz, 1H), 7.39 – 7.36 (m, 2H), 7.31 (d, J = 9.2 Hz, 1H), 7.15 (t, J = 8.6 Hz, 2H), 6.57 (dd, J = 9.2, 6.4 Hz, 1H), 6.40 – 6.37 (m, 1H), 2.51 (s, 3H). **¹³C NMR** (100 MHz, CDCl₃) δ 189.2, 163.0 (d, $^1J_{C-F}$ = 248.2 Hz), 132.8 (d, $^3J_{C-F}$ = 8.3 Hz), 130.6, 129.8, 124.9 (d, $^4J_{C-F}$ = 3.5 Hz), 123.4, 122.1, 119.1, 117.6, 116.2 (d, $^2J_{C-F}$ = 21.6 Hz), 112.9, 110.0, 9.5. **¹⁹F NMR** (375 MHz, CDCl₃) δ -111.5. **ESI-HRMS:** m/z calcd. for C₁₆H₁₂NOF [M+H]⁺: 254.0981, found 254.0975.

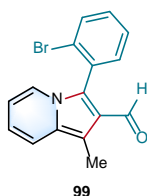


Following the **General procedure D1**, a mixture of α,β -unsaturated aldehyde (0.2 mmol), heteroaryl ketone (2.5 equiv.), catalyst **3b** (0.04 mmol), AcOH (2.0 equiv.) and LiSO₃CF₃ (3.0 equiv.) in the CF₃CH₂OH (0.9 mL) were stirred at 80 °C under Ar atmosphere for 18 h. Work-up gave product 3-(4-chlorophenyl)-1-methylindolizine-2-carbaldehyde (**97**, 40.4 mg, 0.15 mmol, isolated yield 75%) as a yellow solid. **mp:** 105–106 °C. **FT-IR:** ν (cm⁻¹): 3056, 2916, 2846, 1655, 1510, 1475, 1430, 1405, 1385, 1352, 1319, 1247, 1220, 1150, 1111, 1090, 1014, 938, 878, 820, 740, 729, 711, 680, 637, 622, 548, 536, 501, 482, 435, 404. **¹H NMR** (400 MHz, CDCl₃) δ 9.90 (s, 1H), 7.71 (d, J = 7.2 Hz, 1H), 7.44 (d, J = 8.4 Hz, 2H), 7.35 – 7.31 (m, 3H), 6.61 – 6.57 (m, 1H), 6.40 (t, J = 6.8 Hz, 1H), 2.52 (s, 3H). **¹³C NMR** (100 MHz, CDCl₃) δ 189.1, 135.1, 132.1, 130.8, 129.4, 127.3, 123.4, 122.0, 119.2, 117.7, 113.1, 110.3, 9.5. **ESI-HRMS:** m/z calcd. for C₁₆H₁₃ClNO [M+H]⁺: 270.0686, found 270.0685.

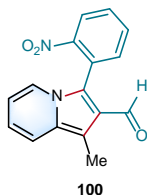


Following the **General procedure D1**, a mixture of α,β -unsaturated aldehyde (0.2 mmol), heteroaryl ketone (2.5 equiv.), catalyst **3b** (0.04 mmol), AcOH (2.0 equiv.) and LiSO₃CF₃ (3.0 equiv.) in the

CF₃CH₂OH (0.9 mL) were stirred at 80 °C under Ar atmosphere for 18 h. Work-up gave product 3-(4-bromophenyl)-1-methylindolizine-2-carbaldehyde (**98**, 53.2 mg, 0.17 mmol, isolated yield 87%) as a yellow liquid. **FT-IR**: ν (cm⁻¹): 2916, 2747, 1663, 1587, 1508, 1471, 1432, 1399, 1381, 1354, 1321, 1249, 1220, 1148, 1115, 1069, 1010, 938, 878, 814, 736, 680, 666, 622, 560, 534, 492, 433, 402. **¹H NMR** (400 MHz, CDCl₃) δ 9.89 (s, 1H), 7.71 (d, *J* = 7.6 Hz, 1H), 7.58 (d, *J* = 8.4 Hz, 2H), 7.31 (d, *J* = 9.2 Hz, 1H), 7.26 (d, *J* = 8.4 Hz, 2H), 6.58 (dd, *J* = 9.2, 6.6 Hz, 1H), 6.41 – 6.37 (m, 1H), 2.50 (s, 3H). **¹³C NMR** (100 MHz, CDCl₃) δ 189.0, 132.3, 132.3, 130.8, 129.3, 127.8, 123.3, 123.2, 122.0, 119.1, 117.7, 113.1, 110.3, 9.5. **ESI-HRMS**: *m/z* calcd. for C₁₆H₁₂NOBr [M+H]⁺: 314.0181, found 314.0164.

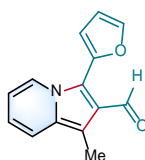


Following the **General procedure D1**, a mixture of α,β -unsaturated aldehyde (0.2 mmol), heteroaryl ketone (2.5 equiv.), catalyst **3b** (0.04 mmol), AcOH (2.0 equiv.) and LiSO₃CF₃ (3.0 equiv.) in the CF₃CH₂OH (0.9 mL) were stirred at 80 °C under Ar atmosphere for 18 h. Work-up gave product 3-(2-bromophenyl)-1-methylindolizine-2-carbaldehyde (**99**, 43.8 mg, 0.14 mmol, isolated yield 71%) as a yellow liquid. **FT-IR**: ν (cm⁻¹): 2918, 2850, 2807, 2739, 1667, 1510, 1432, 1385, 1358, 1321, 1249, 1218, 1154, 1144, 1125, 1115, 1045, 1024, 954, 882, 861, 833, 754, 736, 711, 688, 645, 569, 550, 536, 497, 447, 420. **¹H NMR** (400 MHz, CDCl₃) δ 9.81 (s, 1H), 7.69 – 7.67 (m, 1H), 7.38 – 7.27 (m, 5H), 6.61 (dd, *J* = 8.8, 6.4 Hz, 1H), 6.44 – 6.40 (m, 1H), 2.54 (s, 3H). **¹³C NMR** (100 MHz, CDCl₃) δ 188.8, 134.2, 133.3, 131.0, 130.7, 130.3, 129.2, 127.6, 125.9, 123.6, 122.8, 119.0, 117.6, 112.7, 109.6, 9.6. **ESI-HRMS**: *m/z* calcd. for C₁₆H₁₂NOBr [M+H]⁺: 314.0181, found 314.0164.



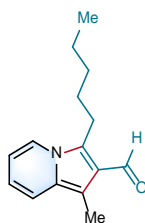
Following the **General procedure D1**, a mixture of α,β -unsaturated aldehyde (0.2 mmol), heteroaryl ketone (2.5 equiv.), catalyst **3b** (0.04 mmol), AcOH (2.0 equiv.) and LiSO₃CF₃ (3.0 equiv.) in the CF₃CH₂OH (0.9 mL) were stirred at 80 °C under Ar atmosphere for 18 h. Work-up gave product 1-

methyl-3-(2-nitrophenyl)indolizine-2-carbaldehyde (**13**, 39.2 mg, 0.14 mmol, isolated yield 69%) as a red liquid. **FT-IR**: ν (cm⁻¹): 2916, 2854, 1667, 1609, 1583, 1523, 1467, 1432, 1385, 1339, 1300, 1249, 1220, 1144, 995, 950, 936, 853, 830, 787, 738, 717, 701, 666, 647, 567, 536, 482, 422. **¹H NMR** (400 MHz, CDCl₃) δ 9.87 (s, 1H), 8.15 (dd, J = 8.0, 1.2 Hz, 1H), 7.68 (td, J = 7.2, 1.6 Hz, 1H), 7.63 – 7.59 (m, 1H), 7.44 (dd, J = 7.6, 1.2 Hz, 1H), 7.36 (d, J = 9.2 Hz, 1H), 7.28 (d, J = 7.2 Hz, 1H), 6.63 (dd, J = 9.2, 6.4 Hz, 1H), 6.44 – 6.41 (m, 1H), 2.53 (s, 3H). **¹³C NMR** (100 MHz, CDCl₃) δ 187.7, 149.7, 134.5, 133.4, 131.3, 130.5, 125.2, 124.5, 123.5, 123.4, 122.2, 119.2, 117.7, 113.6, 111.0, 9.1. **ESI-HRMS**: m/z calcd. for C₁₆H₁₂N₂O₃ [M+H]⁺: 281.0926, found 281.0921.



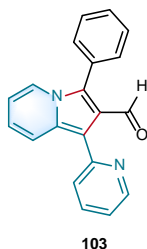
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Following the **General procedure D3**, a mixture of α,β -unsaturated aldehyde (0.2 mmol), heteroaryl ketone (2.5 equiv.), catalyst **3b** (0.04 mmol), AcOH (4.0 equiv.) and LiSO₃CF₃ (3.0 equiv.) in the CF₃CH₂OH (0.9 mL) were stirred at room temperature under Ar atmosphere for 42 h. Work-up gave product 3-(furan-2-yl)-1-methylindolizine-2-carbaldehyde (**101**, 13.5 mg, 0.06 mmol, isolated yield 30%) as a yellow liquid. **FT-IR**: ν (cm⁻¹): 3118, 2920, 2850, 1663, 1510, 1463, 1434, 1381, 1352, 1319, 1249, 1212, 1162, 1144, 1117, 1076, 1016, 958, 888, 874, 808, 7734, 682, 659, 620, 593, 528, 427. **¹H NMR** (400 MHz, CDCl₃) δ 10.17 (s, 1H), 8.13 (d, J = 7.2 Hz, 1H), 7.58 (s, 1H), 7.35 (d, J = 9.2 Hz, 1H), 6.68 – 6.64 (m, 1H), 6.61 (d, J = 3.2 Hz, 1H), 6.55 – 6.50 (m, 2H), 2.52 (s, 3H). **¹³C NMR** (100 MHz, CDCl₃) δ 189.3, 143.6, 143.4, 131.5, 124.0, 124.0, 120.2, 119.0, 118.2, 113.3, 112.2, 111.6, 110.9, 9.6. **ESI-HRMS**: m/z calcd. for C₁₄H₁₁NO₂ [M+H]⁺: 226.0868, found 226.0862.

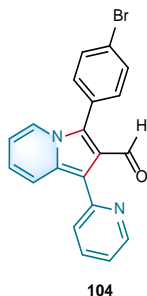


102

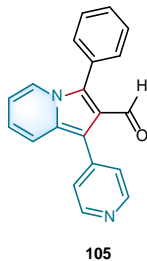
Following the **General procedure D1**, a mixture of α,β -unsaturated aldehyde (0.2 mmol), heteroaryl ketone (2.5 equiv.), catalyst **3b** (0.04 mmol), AcOH (2.0 equiv.) and LiSO_3CF_3 (3.0 equiv.) in the $\text{CF}_3\text{CH}_2\text{OH}$ (0.9 mL) were stirred at 80 °C under Ar atmosphere for 18 h. Work-up gave product 1-methyl-3-pentylindolizine-2-carbaldehyde (**102**, 34.4 mg, 0.15 mmol, isolated yield 76%) as a yellow liquid. **FT-IR**: ν (cm^{-1}): 2955, 2924, 2856, 2734, 1663, 1504, 1447, 1434, 1393, 1321, 1249, 1199, 1142, 1111, 1057, 905, 853, 732, 643, 523, 427. **^1H NMR** (400 MHz, CDCl_3) δ 10.22 (s, 1H), 7.57 (d, $J = 7.2$ Hz, 1H), 7.23 (d, $J = 9.2$ Hz, 1H), 6.48 (dd, $J = 8.8, 6.4$ Hz, 1H), 6.44 – 6.40 (m, 1H), 3.08 – 3.05 (m, 2H), 2.44 (s, 3H), 1.55 (p, $J = 7.4$ Hz, 2H), 1.29 – 1.25 (m, 4H), 0.80 (s, 3H). **^{13}C NMR** (100 MHz, CDCl_3) δ 188.0, 130.3, 129.7, 122.0, 121.6, 119.1, 115.9, 112.6, 109.8, 31.5, 27.8, 23.6, 22.4, 13.9, 8.9. **ESI-HRMS**: m/z calcd. for $\text{C}_{15}\text{H}_{19}\text{NO}$ $[\text{M}+\text{H}]^+$: 230.1545, found 230.1542.



Following the **General procedure D1**, a mixture of α,β -unsaturated aldehyde (0.2 mmol), heteroaryl ketone (2.5 equiv.), catalyst **3b** (0.04 mmol), AcOH (2.0 equiv.) and LiSO_3CF_3 (3.0 equiv.) in the $\text{CF}_3\text{CH}_2\text{OH}$ (0.9 mL) were stirred at 80 °C under Ar atmosphere for 18 h. Work-up gave product 3-phenyl-1-(pyridin-2-yl)indolizine-2-carbaldehyde (**103**, 56.6 mg, 0.19 mmol, isolated yield 95%) as a yellow liquid. **FT-IR**: ν (cm^{-1}): 3060, 2846, 2761, 1673, 1585, 1517, 1473, 1445, 1420, 1381, 1354, 1323, 1278, 1267, 1236, 1191, 1146, 1123, 1096, 1076, 1055, 1039, 1024, 987, 948, 923, 905, 824, 779, 740, 719, 694, 664, 645, 624, 610, 585, 565, 507, 486, 433, 406. **^1H NMR** (400 MHz, CDCl_3) δ 10.01 (s, 1H), 8.61 (d, $J = 4.8$ Hz, 1H), 7.86 (d, $J = 9.2$ Hz, 1H), 7.77 (d, $J = 7.2$ Hz, 1H), 7.65 (d, $J = 3.2$ Hz, 2H), 7.47 – 7.40 (m, 5H), 7.09 (q, $J = 4.4$ Hz, 1H), 6.74 (dd, $J = 9.2, 6.5$ Hz, 1H), 6.47 (t, $J = 6.8$ Hz, 1H). **^{13}C NMR** (100 MHz, CDCl_3) δ 188.6, 153.2, 149.1, 135.7, 132.1, 131.6, 131.1, 129.2, 128.9, 128.7, 125.5, 122.7, 122.2, 120.9, 120.4, 114.0, 113.6. **ESI-HRMS**: m/z calcd. for $\text{C}_{20}\text{H}_{14}\text{N}_2\text{O}$ $[\text{M}+\text{H}]^+$: 299.1184, found 299.1180.

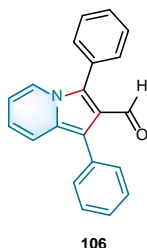


Following the **General procedure D1**, a mixture of α,β -unsaturated aldehyde (0.2 mmol), heteroaryl ketone (2.5 equiv.), catalyst **3b** (0.04 mmol), AcOH (2.0 equiv.) and LiSO_3CF_3 (3.0 equiv.) in the $\text{CF}_3\text{CH}_2\text{OH}$ (0.9 mL) were stirred at 80 °C under Ar atmosphere for 18 h. Work-up gave product 3-(4-bromophenyl)-1-(pyridin-2-yl)indolizine-2-carbaldehyde (**104**, 63.9 mg, 0.17 mmol, isolated yield 83%) as a yellow solid. **mp**: 167–168 °C. **FT-IR**: ν (cm^{-1}): 3052, 2854, 2770, 1671, 1630, 1585, 1562, 1519, 1506, 1469, 1434, 1397, 1333, 1263, 1232, 1193, 1148, 1107, 1094, 1074, 1051, 1037, 1008, 989, 946, 911, 841, 814, 775, 740, 725, 713, 686, 618, 563, 503, 495, 441, 404. **^1H -NMR** (400 MHz, CDCl_3) δ 10.07 (s, 1H), 8.66 (d, J = 3.6 Hz, 1H), 7.86 (d, J = 9.2 Hz, 1H), 7.80 – 7.72 (m, 2H), 7.65 – 7.63 (m, 3H), 7.36 (d, J = 8.1 Hz, 2H), 7.19 (s, 1H), 6.83 (t, J = 7.6 Hz, 1H), 6.57 (t, J = 6.5 Hz, 1H). **^{13}C NMR** (100 MHz, CDCl_3) δ 188.5, 152.8, 149.3, 136.1, 132.7, 132.3, 132.2, 129.1, 127.9, 125.4, 123.7, 122.6, 122.5, 121.1, 121.1, 120.4, 114.1. **ESI-HRMS**: m/z calcd. for $\text{C}_{20}\text{H}_{13}\text{BrN}_2\text{O}$ $[\text{M}+\text{H}]^+$: 377.0290, found 377.0280.

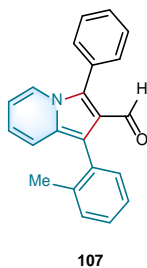


Following the **General procedure D1**, a mixture of α,β -unsaturated aldehyde (0.2 mmol), heteroaryl ketone (2.5 equiv.), catalyst **3b** (0.04 mmol), AcOH (2.0 equiv.) and LiSO_3CF_3 (3.0 equiv.) in the $\text{CF}_3\text{CH}_2\text{OH}$ (0.9 mL) were stirred at 80 °C under Ar atmosphere for 18 h. Work-up gave product 3-phenyl-1-(pyridin-4-yl)indolizine-2-carbaldehyde (**105**, 50.7 mg, 0.17 mmol, isolated yield 85%) as a yellow solid. **mp**: 182–183 °C. **FT-IR**: ν (cm^{-1}): 3050, 2106, 1657, 1628, 1593, 1554, 1533, 1490, 1473, 1436, 1352, 1331, 1300, 1245, 1142, 1119, 1076, 1057, 1006, 950, 863, 824, 769, 756, 723, 711, 696, 686, 668, 635, 622, 612, 546, 499, 443, 414. **^1H NMR** (400 MHz, CDCl_3) δ 9.91 (s, 1H), 8.58 (d, J = 6.0 Hz, 2H), 7.84 (d, J = 7.2 Hz, 1H), 7.52 – 7.45 (m, 6H), 7.40 (d, J = 6.0 Hz, 2H), 6.76 (dd, J = 9.2, 6.4 Hz, 1H),

6.53 (t, $J = 6.8$ Hz, 1H). ^{13}C NMR (100 MHz, CDCl_3) δ 188.0, 149.5, 141.6, 132.7, 131.3, 131.1, 129.5, 129.1, 128.1, 125.2, 123.0, 121.9, 121.1, 118.9, 113.7, 111.9. **ESI-HRMS**: m/z calcd. for $\text{C}_{20}\text{H}_{14}\text{N}_2\text{O}$ $[\text{M}+\text{H}]^+$: 299.1184, found 299.1179.

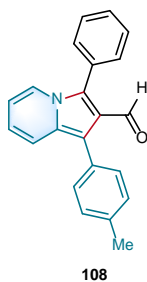


Following the **General procedure D1**, a mixture of α,β -unsaturated aldehyde (0.2 mmol), heteroaryl ketone (2.5 equiv.), catalyst **3b** (0.04 mmol), AcOH (2.0 equiv.) and LiSO_3CF_3 (3.0 equiv.) in the $\text{CF}_3\text{CH}_2\text{OH}$ (0.9 mL) were stirred at 80 °C under Ar atmosphere for 18 h. Work-up gave product 1,3-diphenylindolizine-2-carbaldehyde (**106**, 50.5 mg, 0.17 mmol, isolated yield 84%) as a yellow solid. **mp**: 164–165 °C. **FT-IR**: ν (cm^{-1}): 3029, 1675, 1591, 1525, 1475, 1442, 1428, 1414, 1383, 1354, 1344, 1331, 1315, 1261, 1236, 1218, 1187, 1146, 1076, 1039, 1022, 989, 948, 925, 905, 843, 822, 804, 750, 740, 729, 711, 692, 674, 633, 604, 585, 569, 511, 488, 435, 414. ^1H NMR (400 MHz, CDCl_3) δ 9.95 (s, 1H), 7.79 (d, $J = 7.2$ Hz, 1H), 7.46 – 7.43 (m, 7H), 7.40 – 7.35 (m, 3H), 7.26 (t, $J = 7.2$ Hz, 1H), 6.62 (dd, $J = 9.2$, 6.4 Hz, 1H), 6.43 (t, $J = 6.8$ Hz, 1H). ^{13}C NMR (100 MHz, CDCl_3) δ 188.4, 133.1, 131.0, 130.7, 130.6, 130.2, 129.1, 129.0, 128.9, 128.2, 126.8, 122.6, 122.0, 119.6, 119.4, 116.2, 113.5. **ESI-HRMS**: m/z calcd. for $\text{C}_{21}\text{H}_{15}\text{NO}$ $[\text{M}+\text{H}]^+$: 298.1232, found 298.1227.

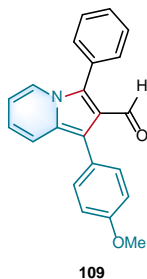


Following the **General procedure D4**, a mixture of α,β -unsaturated aldehyde (0.2 mmol), heteroaryl ketone (2.5 equiv.), catalyst **3b** (0.04 mmol), AcOH (2.0 equiv.) and LiSO_3CF_3 (3.0 equiv.) in the $\text{CF}_3\text{CH}_2\text{OH}$ (0.9 mL) were stirred at 80 °C under Ar atmosphere for 36 h. Work-up gave product 3-phenyl-1-(o-tolyl)indolizine-2-carbaldehyde (**107**, 7.5 mg, 0.02 mmol, isolated yield 12%) as a yellow

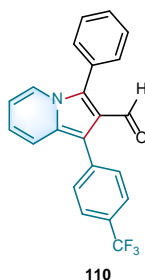
solid. **mp**: 68–69 °C. **FT-IR**: ν (cm⁻¹): 3054, 2918, 2850, 1673, 1624, 1599, 1576, 1523, 1447, 1428, 1379, 1356, 1321, 1257, 1166, 1121, 1072, 1018, 969, 921, 903, 797, 750, 732, 692, 618, 571, 540, 503, 484, 445, 404. **¹H NMR** (400 MHz, CDCl₃) δ 9.95 (s, 1H), 7.79 (d, J = 7.2 Hz, 1H), 7.46 – 7.43 (m, 7H), 7.40 – 7.35 (m, 3H), 7.26 (t, J = 7.2 Hz, 1H), 6.62 (dd, J = 9.2, 6.4 Hz, 1H), 6.43 (t, J = 6.8 Hz, 1H). **¹³C NMR** (100 MHz, CDCl₃) δ 188.4, 138.1, 131.4, 131.3, 131.0, 130.9, 130.7, 130.0, 129.1, 129.1, 129.0, 128.6, 128.5, 127.6, 125.4, 122.5, 119.9, 118.9, 113.3, 20.2. **ESI-HRMS**: m/z calcd. for C₂₂H₁₇NO [M+H]⁺: 312.1388, found 312.1384.



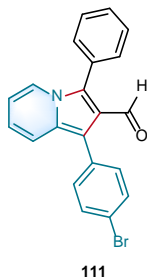
Following the **General procedure D1**, a mixture of α,β -unsaturated aldehyde (0.2 mmol), heteroaryl ketone (2.5 equiv.), catalyst **3b** (0.04 mmol), AcOH (2.0 equiv.) and LiSO₃CF₃ (3.0 equiv.) in the CF₃CH₂OH (0.9 mL) were stirred at 80 °C under Ar atmosphere for 18 h. Work-up gave product 3-phenyl-1-(p-tolyl)indolizine-2-carbaldehyde (**108**, 52.9 mg, 0.17 mmol, isolated yield 87%) as a yellow liquid. **FT-IR**: ν (cm⁻¹): 3019, 2918, 2848, 2747, 1673, 1626, 1601, 1523, 1504, 1475, 1447, 1428, 1381, 1356, 1323, 1259, 1232, 1183, 1121, 1074, 1041, 1016, 971, 950, 923, 903, 812, 746, 729, 696, 565, 507, 441. **¹H NMR** (400 MHz, CDCl₃) δ 9.93 (s, 1H), 7.76 (d, J = 7.2 Hz, 2H), 7.43–7.42 (m, 4H), 7.38 (d, J = 2.4 Hz, 1H), 7.34 – 7.30 (m, 3H), 7.17 (d, J = 8.0 Hz, 2H), 6.58 (ddd, J = 9.2, 6.4, 1.2 Hz, 1H), 6.42–6.38 (m, 1H). **¹³C NMR** (100 MHz, CDCl₃) δ 188.5, 136.4, 131.2, 130.9, 130.4, 130.0, 129.0, 129.0, 129.0, 128.9, 128.4, 122.5, 122.0, 119.7, 119.2, 116.2, 113.4, 21.2. **ESI-HRMS**: m/z calcd. for C₂₂H₁₇NO [M+H]⁺: 312.1388, found 312.1384.



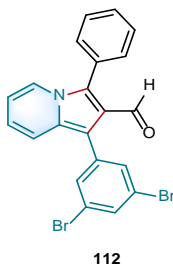
Following the **General procedure D1**, a mixture of α,β -unsaturated aldehyde (0.2 mmol), heteroaryl ketone (2.5 equiv.), catalyst **3b** (0.04 mmol), AcOH (2.0 equiv.) and LiSO_3CF_3 (3.0 equiv.) in the $\text{CF}_3\text{CH}_2\text{OH}$ (0.9 mL) were stirred at 80 °C under Ar atmosphere for 18 h. Work-up gave product 1-(4-methoxyphenyl)-3-phenylindolizine-2-carbaldehyde (**109**, 22.9 mg, 0.07 mmol, isolated yield 35%) as a yellow liquid. **FT-IR**: ν (cm^{-1}): 3015, 2930, 2833, 2749, 1673, 1605, 1537, 1523, 1504, 1463, 1445, 1428, 1381, 1358, 1286, 1243, 1175, 1109, 1076, 1030, 1020, 948, 923, 903, 833, 785, 748, 729, 696, 666, 565, 528, 488, 439, 410. **^1H NMR** (400 MHz, CDCl_3) δ 9.94 (s, 1H), 7.79 (d, $J = 7.2$ Hz, 1H), 7.47 – 7.44 (m, 4H), 7.42 – 7.37 (m, 4H), 6.93 (d, $J = 8.4$ Hz, 2H), 6.63 – 6.59 (m, 1H), 6.45 – 6.41 (m, 1H), 3.79 (s, 3H). **^{13}C NMR** (100 MHz, CDCl_3) δ 188.6, 158.6, 131.7, 131.0, 130.6, 130.2, 129.0, 129.0, 128.9, 125.3, 122.5, 122.0, 119.7, 119.1, 115.9, 113.7, 113.4, 55.3. **ESI-HRMS**: m/z calcd. for $\text{C}_{22}\text{H}_{17}\text{NO}_2$ $[\text{M}+\text{H}]^+$: 328.1338, found 328.1331.



Following the **General procedure D1**, a mixture of α,β -unsaturated aldehyde (0.2 mmol), heteroaryl ketone (2.5 equiv.), catalyst **3b** (0.04 mmol), AcOH (2.0 equiv.) and LiSO_3CF_3 (3.0 equiv.) in the $\text{CF}_3\text{CH}_2\text{OH}$ (0.9 mL) were stirred at 80 °C under Ar atmosphere for 18 h. Work-up gave product 3-phenyl-1-(4-(trifluoromethyl)phenyl)indolizine-2-carbaldehyde (**110**, 69.4 mg, 0.19 mmol, isolated yield 97%) as a yellow solid. **mp**: 127–128 °C. **FT-IR**: ν (cm^{-1}): 3058, 2848, 2776, 1669, 1613, 1539, 1523, 1445, 1426, 1409, 1383, 1354, 1317, 1261, 1230, 1160, 1107, 1065, 1016, 939, 903, 843, 833, 756, 736, 703, 692, 674, 600, 558, 490, 455, 439. **^1H NMR** (400 MHz, CDCl_3) δ 9.91 (s, 1H), 7.83 (d, $J = 7.2$ Hz, 1H), 7.60 (q, $J = 8.4$ Hz, 4H), 7.52 – 7.45 (m, 5H), 7.40 (d, $J = 9.2$ Hz, 1H), 6.71 (dd, $J = 9.2, 6.4$ Hz, 1H), 6.50 (t, $J = 6.8$ Hz, 1H). **^{13}C NMR** (100 MHz, CDCl_3) δ 188.2, 148.6, 137.2 (q, $^4J_{\text{C-F}} = 1.2$ Hz), 132.2, 131.1, 130.7, 129.4, 129.1, 128.9 (q, $^2J_{\text{C-F}} = 32.2$ Hz), 128.4, 126.7, 125.0 (q, $^3J_{\text{C-F}} = 3.7$ Hz), 124.4 (q, $^1J_{\text{C-F}} = 270.2$ Hz), 122.9, 122.0, 120.5, 119.1, 113.6. **^{19}F NMR** (377 MHz, CDCl_3) δ -62.3. **ESI-HRMS**: m/z calcd. for $\text{C}_{22}\text{H}_{14}\text{F}_3\text{NO}$ $[\text{M}+\text{H}]^+$: 366.1106, found 366.1100.

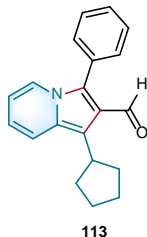


Following the **General procedure D1**, a mixture of α,β -unsaturated aldehyde (0.2 mmol), heteroaryl ketone (2.5 equiv.), catalyst **3b** (0.04 mmol), AcOH (2.0 equiv.) and LiSO_3CF_3 (3.0 equiv.) in the $\text{CF}_3\text{CH}_2\text{OH}$ (0.9 mL) were stirred at 80 °C under Ar atmosphere for 18 h. Work-up gave product 1-(4-bromophenyl)-3-phenylindolizine-2-carbaldehyde (**111**, 67.5 mg, 0.18 mmol, isolated yield 88%) as a yellow solid. **mp**: 150–151 °C. **FT-IR**: ν (cm^{-1}): 3046, 2926, 2846, 2749, 1675, 1597, 1521, 1486, 1447, 1426, 1397, 1379, 1356, 1321, 1259, 1230, 1177, 1123, 1102, 1069, 1039, 1022, 1006, 921, 901, 816, 797, 748, 732, 701, 688, 647, 560, 501, 488, 435, 412. **^1H NMR** (400 MHz, CDCl_3) δ 9.90 (s, 1H), 7.81 (d, J = 7.2 Hz, 1H), 7.50 – 7.45 (m, 7H), 7.38 – 7.32 (m, 3H), 6.69 – 6.65 (m, 1H), 6.47 (t, J = 6.6 Hz, 1H). **^{13}C NMR** (100 MHz, CDCl_3) δ 188.3, 132.1, 131.3, 131.0, 131.0, 130.6, 129.3, 129.1, 128.9, 128.6, 128.2, 122.7, 121.9, 120.8, 120.0, 119.33, 113.5. **ESI-HRMS**: m/z calcd. for $\text{C}_{21}\text{H}_{14}\text{BrNO}$ $[\text{M}+\text{H}]^+$: 376.0337, found 376.0335.

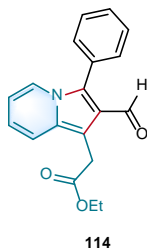


Following the **General procedure D1**, a mixture of α,β -unsaturated aldehyde (0.2 mmol), heteroaryl ketone (2.5 equiv.), catalyst **3b** (0.04 mmol), AcOH (2.0 equiv.) and LiSO_3CF_3 (3.0 equiv.) in the $\text{CF}_3\text{CH}_2\text{OH}$ (0.9 mL) were stirred at 80 °C under Ar atmosphere for 18 h. Work-up gave product 1-(3,5-dibromophenyl)-3-phenylindolizine-2-carbaldehyde (**112**, 86.1 mg, 0.19 mmol, isolated yield 95%) as a yellow solid. **mp**: 57–58 °C. **FT-IR**: ν (cm^{-1}): 3064, 2844, 2747, 1673, 1578, 1543, 1523, 1475, 1453, 1405, 1377, 1356, 1321, 1300, 1280, 1259, 1232, 1156, 1123, 1105, 1074, 1047, 1026, 989, 956, 925, 907, 851, 750, 740, 694, 672, 641, 618, 577, 523, 488, 422. **^1H NMR** (400 MHz, CDCl_3) δ 9.89 (s, 1H), 7.82 (d, J = 7.2 Hz, 1H), 7.55 (d, J = 4.4 Hz, 3H), 7.50 – 7.46 (m, 5H), 7.38 (d, J = 9.2 Hz, 1H), 6.76 – 6.72 (m, 1H), 6.51 (t, J = 6.8 Hz, 1H). **^{13}C NMR** (100 MHz, CDCl_3) δ 187.9, 148.7, 137.8, 137.3, 137.1, 132.1,

131.1, 129.5, 129.1, 128.2, 124.8, 122.9, 122.4, 120.8, 118.9, 113.6, 112.0. **ESI-HRMS:** m/z calcd. for $C_{21}H_{13}Br_2NO$ $[M+H]^+$: 353.9441, found 353.9442.

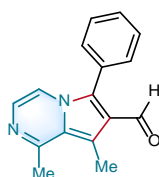


Following the **General procedure D1**, a mixture of α,β -unsaturated aldehyde (0.2 mmol), heteroaryl ketone (2.5 equiv.), catalyst **3b** (0.04 mmol), AcOH (2.0 equiv.) and $LiSO_3CF_3$ (3.0 equiv.) in the CF_3CH_2OH (0.9 mL) were stirred at 80 °C under Ar atmosphere for 18 h. work-up gave product 1-cyclopentyl-3-phenylindolizine-2-carbaldehyde (**113**, 26.6 mg, 0.09 mmol, isolated yield 46%) as a yellow liquid. **FT-IR:** ν (cm^{-1}): 2949, 2864, 1669, 1599, 1515, 1447, 1428, 1395, 1358, 1315, 1241, 1224, 1164, 1074, 1026, 1001, 973, 927, 886, 744, 725, 696, 554, 495, 439, 406. **1H NMR** (400 MHz, $CDCl_3$) δ 9.88 (s, 1H), 7.72 (d, $J = 7.2$ Hz, 1H), 7.48 – 7.39 (m, 6H), 6.57 – 6.53 (m, 1H), 6.38 – 6.34 (m, 1H), 3.84 – 3.76 (m, 1H), 1.96 – 1.89 (m, 6H), 1.69 – 1.67 (m, 2H). **^{13}C NMR** (100 MHz, $CDCl_3$) δ 189.4, 132.5, 131.2, 129.8, 129.0, 129.0, 129.0, 122.9, 122.6, 120.0, 118.3, 117.5, 112.6, 36.2, 33.3, 26.4. **ESI-HRMS:** m/z calcd. for $C_{20}H_{19}NO$ $[M+H]^+$: 290.1547, found 290.1545.



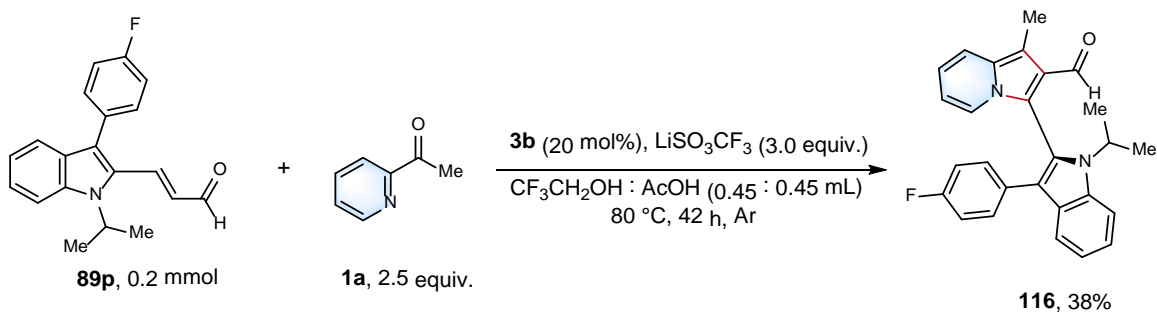
Following the **General procedure D4**, a mixture of α,β -unsaturated aldehyde (0.2 mmol), heteroaryl ketone (2.5 equiv.), catalyst **3b** (0.04 mmol), AcOH (2.0 equiv.) and $LiSO_3CF_3$ (3.0 equiv.) in the CF_3CH_2OH (0.9 mL) were stirred at 80 °C under Ar atmosphere for 36 h. work-up gave product ethyl 2-(2-formyl-3-phenylindolizin-1-yl)acetate (**114**, 19.6 mg, 0.06 mmol, isolated yield 32%) as a yellow liquid. **FT-IR:** ν (cm^{-1}): 3056, 2978, 2926, 2835, 2753, 1731, 1665, 1525, 1445, 1389, 1356, 1321, 1249, 1226, 1212, 1175, 1154, 1105, 1076, 1028, 933, 868, 833, 752, 736, 701, 672, 530, 482, 435. **1H NMR** (400 MHz, $CDCl_3$) δ 9.88 (s, 1H), 7.81 (d, $J = 7.6$ Hz, 1H), 7.50 – 7.43 (m, 5H), 7.33 (d, $J = 9.2$ Hz, 1H),

6.67 (ddd, $J = 9.2, 6.4, 0.8$ Hz, 1H), 6.45 – 6.41 (m, 1H), 4.13 (q, $J = 7.2$ Hz, 2H), 4.06 (s, 2H), 1.23 (t, $J = 7.2$ Hz, 1H). **^{13}C NMR** (100 MHz, CDCl_3) δ 189.3, 171.8, 132.0, 131.6, 131.1, 129.2, 129.1, 128.5, 123.1, 122.7, 119.0, 118.7, 112.9, 105.7, 60.8, 30.1, 14.3. **ESI-HRMS**: m/z calcd. for $\text{C}_{19}\text{H}_{17}\text{NO}_3$ $[\text{M}+\text{H}]^+$: 308.1287, found 308.1281.



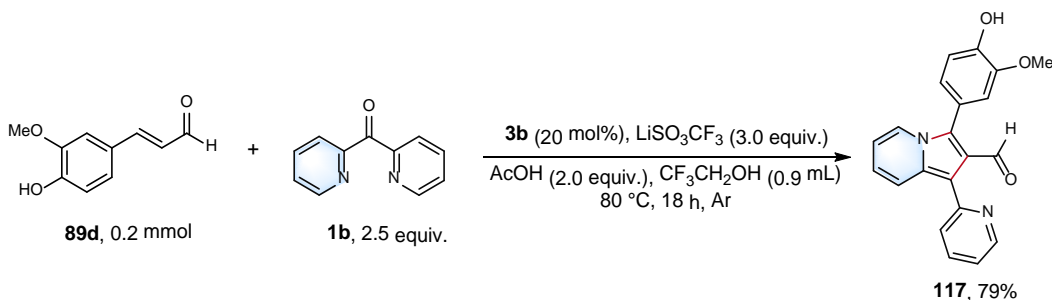
115

Following the **General procedure D4**, a mixture of α,β -unsaturated aldehyde (0.2 mmol), heteroaryl ketone (2.5 equiv.), catalyst **3b** (0.04 mmol), AcOH (2.0 equiv.) and LiSO_3CF_3 (3.0 equiv.) in the $\text{CF}_3\text{CH}_2\text{OH}$ (0.9 mL) were stirred at 80 °C under Ar atmosphere for 36 h. work-up gave product 1,8-dimethyl-6-phenylpyrrolo[1,2-a]pyrazine-7-carbaldehyde (**115**, 21.5 mg, 0.09 mmol, isolated yield 43%) as a yellow liquid. **FT-IR**: ν (cm^{-1}): 3060, 2924, 2852, 2737, 1675, 1607, 1502, 1465, 1453, 1432, 1387, 1372, 1352, 1284, 1208, 1152, 1069, 1024, 956, 826, 760, 705, 593, 556, 488. **^1H NMR** (400 MHz, CDCl_3) δ 9.94 (s, 1H), 7.51 – 7.45 (m, 4H), 7.39 – 7.37 (m, 2H), 7.21 (d, $J = 5.2$ Hz, 1H), 2.84 (s, 3H), 2.77 (s, 3H). **^{13}C NMR** (100 MHz, CDCl_3) δ 189.2, 157.1, 132.7, 130.8, 129.6, 129.2, 127.8, 127.6, 125.7, 123.5, 116.5, 113.7, 24.8, 11.9. **ESI-HRMS**: m/z calcd. for $\text{C}_{16}\text{H}_{14}\text{N}_2\text{O}$ $[\text{M}+\text{H}]^+$: 251.1184, found 251.1178.



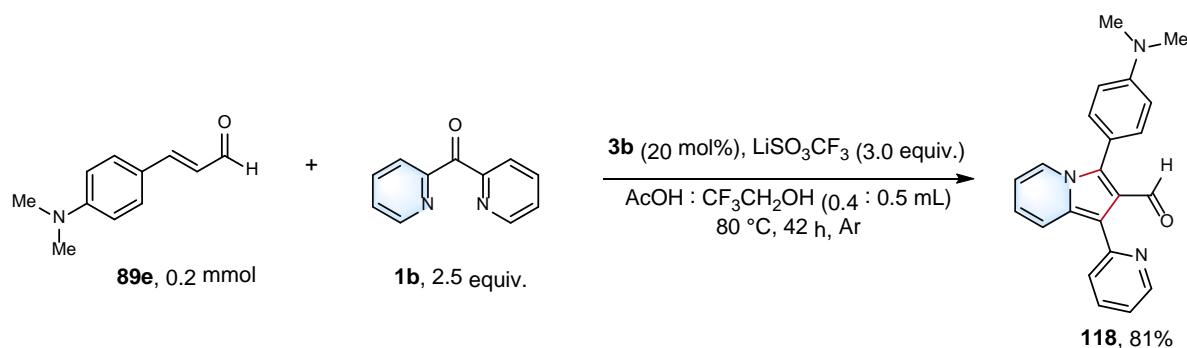
Preparation of 3-(3-(4-fluorophenyl)-1-isopropyl-1H-indol-2-yl)-1-methylindolizine-2-carbaldehyde (**116**): A mixture of (*E*)-3-(3-(4-fluorophenyl)-1-isopropyl-1H-indol-2-yl)acrylaldehyde (0.2 mmol), acetylpyridine (2.5 equiv.), catalyst **3b** (0.04 mmol) and LiSO_3CF_3 (3.0 equiv.) in the $\text{CF}_3\text{CH}_2\text{OH}$: AcOH (0.45 : 0.45 mL) were stirred at 80 °C under Ar atmosphere for 42 h. The reactions were conducted

in a sealed Schlenk tube and heated by an IKA magnetic heating agitator with heating block. The reaction temperature was directly read from temperature detector of IKA apparatus and was calibrated by thermometer. After cooling to room temperature, the reaction mixture was basified up to pH 7 *via* stad. Na₂CO₃ aqueous solution, then extracted by diether (3×3 mL) and dried over anhydrous Na₂SO₄. After filtration and concentrated in rotary evaporator, the crude product was purified with flash chromatography on silica gel (ethyl acetate : *n*-hexane) to give products. Work-up gave product **116** (31.2 mg, 0.08 mmol, isolated yield 38%) as a yellow solid. **mp**: 199–200 °C. **FT-IR**: ν (cm⁻¹): 3039, 2926, 2854, 2743, 2108, 1673, 1599, 1519, 1484, 1475, 1445, 1426, 1395, 1356, 1317, 1259, 1230, 1179, 1123, 1105, 1069, 1039, 1020, 1006, 921, 901, 816, 797, 748, 732, 701, 688, 645, 560, 501, 488, 435, 408. **¹H NMR** (400 MHz, CDCl₃) δ 9.85 (s, 1H), 7.74 (d, *J* = 8.0 Hz, 1H), 7.61 (d, *J* = 8.4 Hz, 1H), 7.40 – 7.35 (m, 2H), 7.27 (t, *J* = 7.7 Hz, 1H), 7.16 (t, *J* = 7.6 Hz, 1H), 7.05 – 7.01 (m, 2H), 6.82 (t, *J* = 8.5 Hz, 2H), 6.65 – 6.62 (m, 1H), 6.39 (t, *J* = 6.7 Hz, 1H), 4.08 (hept, *J* = 6.9 Hz, 1H), 2.53 (s, 3H), 1.48 (d, *J* = 7.2 Hz, 3H), 1.44 (d, *J* = 6.8 Hz, 3H). **¹³C NMR** (100 MHz, CDCl₃) δ 188.6, 161.5 (d, ¹*J*_{C-F} = 243.9 Hz), 135.8, 131.7, 130.1 (d, ⁴*J*_{C-F} = 3.2 Hz), 129.9 (d, ³*J*_{C-F} = 7.8 Hz), 127.7, 126.2, 123.6, 122.9, 122.7, 120.3, 120.2, 119.7, 119.2, 118.1, 115.5 (d, ²*J*_{C-F} = 21.2 Hz), 113.4, 112.4, 110.5, 48.8, 22.0, 21.5, 9.6. **¹⁹F NMR** (377 MHz, CDCl₃) δ -116.1. **ESI-HRMS**: *m/z* calcd. for C₂₇H₂₃FN₂O [M+H]⁺: 411.1873, found 411.1854.



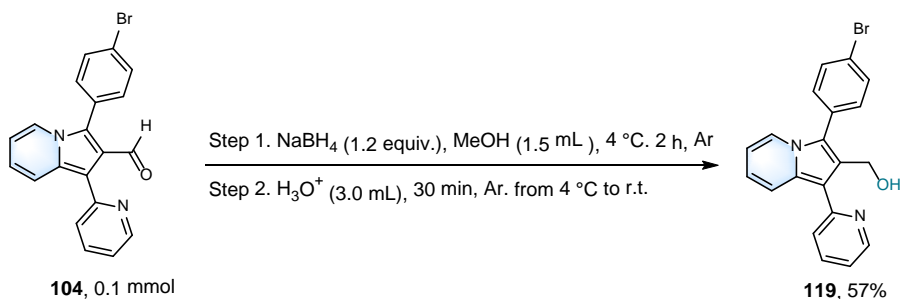
Preparation of 3-(4-hydroxy-3-methoxyphenyl)-1-(pyridin-2-yl)indolizine-2-carbaldehyde (**117**): A mixture of (*E*)-3-(4-hydroxy-3-methoxyphenyl)acrylaldehyde (0.2 mmol), di(pyridin-2-yl)methanone (2.5 equiv.), catalyst **3b** (0.04 mmol), AcOH (2.0 equiv.) and LiSO₃CF₃ (3.0 equiv.) in the CF₃CH₂OH (0.9 mL) were stirred at 80 °C under Ar atmosphere for 18 h. The reactions were conducted in a sealed Schlenk tube and heated by an IKA magnetic heating agitator with heating block. The reaction temperature was directly read from temperature detector of IKA apparatus and was calibrated by thermometer. After cooling to room temperature, the reaction mixture was basified up to pH 7 *via* stad. Na₂CO₃ aqueous solution, then extracted by diether (3×3 mL) and dried over anhydrous Na₂SO₄. After filtration and concentrated in rotary evaporator, the crude product was purified with flash chromatography

on silica gel (ethyl acetate : *n*-hexane) to give products. Work-up gave product **117** (55.0 mg, 0.16 mmol, isolated yield 79 %) as a yellow liquid gel. **FT-IR**: ν (cm⁻¹): 3017, 2955, 2930, 2848, 2755, 1671, 1587, 1533, 1517, 1475, 1422, 1377, 1346, 1265, 1216, 1168, 1121, 1098, 1057, 1026, 960, 925, 876, 810, 785, 742, 696, 664, 626, 598, 556, 437, 408. **¹H NMR** (400 MHz, CDCl₃) δ 10.02 (s, 1H), 8.63 (d, *J* = 4.8 Hz, 1H), 7.85 (d, *J* = 9.2 Hz, 1H), 7.80 (d, *J* = 7.2 Hz, 1H), 7.69 – 7.66 (m, 2H), 7.15 – 7.11 (m, 1H), 7.00 – 6.93 (m, 3H), 6.76 (dd, *J* = 9.0, 6.7 Hz, 1H), 6.50 (t, *J* = 6.8 Hz, 1H), 3.83 (s, 3H). **¹³C NMR** (100 MHz, CDCl₃) δ 188.8, 153.2, 149.1, 147.0, 146.8, 135.9, 132.1, 132.0, 125.6, 124.4, 122.9, 122.2, 121.0, 120.8, 120.4, 120.2, 115.0, 113.7, 113.7, 113.6, 56.1. **ESI-HRMS**: *m/z* calcd. for C₂₁H₁₆N₂O₃ [M+H]⁺: 345.1239, found 345.1234.

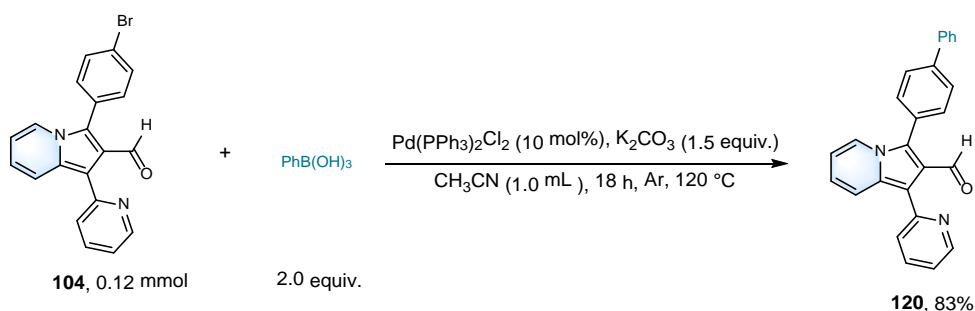


Preparation of 3-(4-(dimethylamino)phenyl)-1-(pyridin-2-yl)indolizine-2-carbaldehyde (**118**): A mixture of (*E*)-3-(4-(dimethylamino)phenyl)acrylaldehyde (0.2 mmol), di(pyridin-2-yl)methanone (2.5 equiv.), catalyst **3b** (0.04 mmol), and LiSO₃CF₃ (3.0 equiv.) in the CF₃CH₂OH : AcOH (0.5 : 0.4 mL) were stirred at 80 °C under Ar atmosphere for 42 h. The reactions were conducted in a sealed Schlenk tube and heated by an IKA magnetic heating agitator with heating block. The reaction temperature was directly read from temperature detector of IKA apparatus and was calibrated by thermometer. After cooling to room temperature, the reaction mixture was basified up to pH 7 *via* stad. Na₂CO₃ aqueous solution, then extracted by diether (3×3 mL) and dried over anhydrous Na₂SO₄. After filtration and concentrated in rotary evaporator, the crude product was purified with flash chromatography on silica gel (ethyl acetate : *n*-hexane) to give products. Work-up gave product **118** (54.6 mg, 0.16 mmol, isolated yield 81%) as a yellow solid. **mp**: 166–167 °C. **FT-IR**: ν (cm⁻¹): 3315, 2899, 2848, 2800, 2774, 2188, 1918, 1665, 1605, 1587, 1535, 1521, 1473, 1440, 1418, 1381, 1354, 1327, 1263, 1228, 1203, 1166, 1117, 1096, 1063, 1053, 1034, 1006, 985, 942, 905, 874, 833, 814, 806, 777, 736, 721, 692, 645, 618, 560, 530, 478, 443, 404. **¹H NMR** (300 MHz, CDCl₃) δ 9.98 (s, 1H), 8.59 (d, *J* = 4.8 Hz, 1H), 7.87 – 7.81 (m, 2H), 7.69 – 7.60 (m, 2H), 7.28 (d, *J* = 8.4 Hz, 2H), 7.09 – 7.04 (m, 1H), 6.76 – 6.67 (m, 3H), 6.43 (t, *J* = 6.8 Hz, 1H). **¹³C**

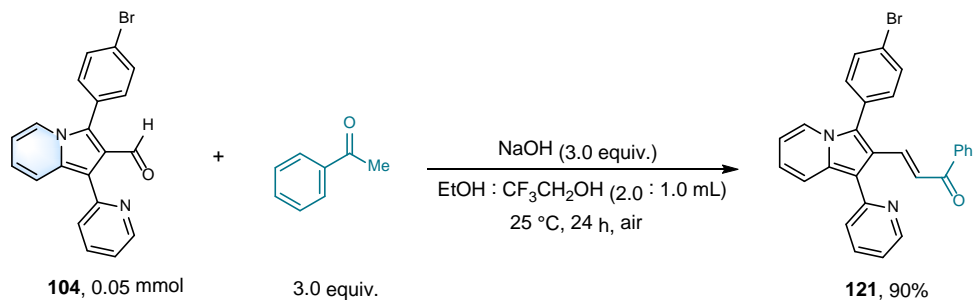
NMR (75 MHz, CDCl₃) δ 188.9, 153.5, 150.7, 148.9, 135.6, 133.7, 131.9, 125.5, 122.9, 121.8, 120.7, 120.6, 120.5, 115.1, 113.2, 113.1, 112.0, 40.1. **ESI-HRMS**: m/z calcd. for C₂₂H₁₉N₃O [M+H]⁺: 342.1606, found 342.1603.



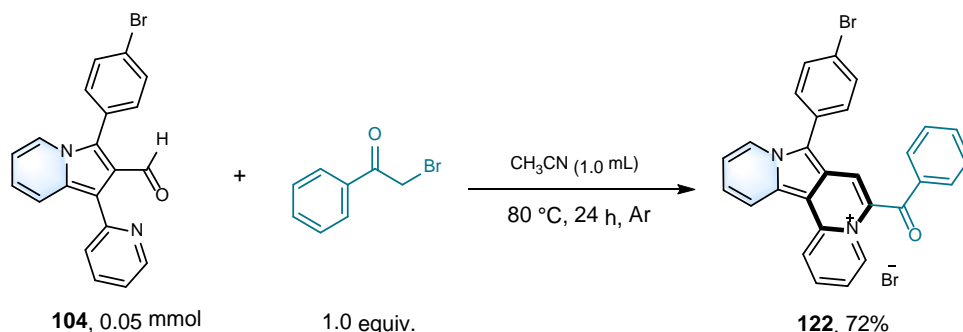
Preparation of (3-(4-bromophenyl)-1-(pyridin-2-yl)indolizin-2-yl)methanol (**119**): A mixture of **104** (0.1 mmol) and NaBH₄ (1.2 equiv.) in anhydrous MeOH solution (1.5 mL) were stirred at 4 °C under Ar atmosphere for 2 h. The reactions were conducted in a sealed Schlenk tube and stirred by an IKA magnetic heating agitator with heating block. The reaction temperature was calibrated by thermometer. After the reaction, the solution was acidified *via* 0.5 N HCl aqueous solution, then stirred for 30 min from 4 °C to room temperature. After completion, the solution was extracted by diether (3×3 mL) and dried over anhydrous Na₂SO₄. After filtration and concentrated in rotary evaporator, the crude product was purified with flash chromatography on silica gel (ethyl acetate : *n*-hexane) to give products. Work-up gave product **119** (22.7 mg, 0.06 mmol, isolated yield 57%) as a yellow solid. **mp**: 99–100 °C. **FT-IR**: ν (cm⁻¹): 2922, 2852, 1587, 1537, 1513, 1473, 1389, 1323, 1273, 1228, 1148, 1123, 1100, 1069, 1055, 1037, 1008, 946, 907, 828, 787, 727, 703, 664, 616, 499, 439, 404. **¹H NMR** (400 MHz, CDCl₃) δ 8.59 (d, J = 4.4 Hz, 1H), 7.95 (d, J = 6.8 Hz, 1H), 7.78 – 7.67 (m, 3H), 7.61 (d, J = 8.0 Hz, 2H), 7.38 (d, J = 8.4 Hz, 2H), 7.12 – 7.09 (m, 1H), 6.86 – 6.82 (m, 1H), 6.50 (t, J = 6.8 Hz, 1H), 4.52 (s, 1H). **¹³C NMR** (100 MHz, CDCl₃) δ 154.3, 148.6, 137.4, 132.4, 132.2, 131.5, 129.0, 127.7, 123.5, 123.3, 122.7, 122.5, 120.8, 120.0, 117.6, 111.8, 55.9. **ESI-HRMS**: m/z calcd. for C₂₀H₁₅N₂OBr [M+H]⁺: 379.0446, found 379.0430.



Preparation of 3-([1,1'-biphenyl]-4-yl)-1-(pyridin-2-yl)indolizine-2-carbaldehyde (**120**): A mixture of 3-(4-bromophenyl)-1-(pyridin-2-yl)indolizine-2-carbaldehyde (0.12 mmol), phenylboronic acid (2.0 equiv.), catalyst $\text{Pd(PPh}_3)_2\text{Cl}_2$ (10% mmol), and K_2CO_3 (1.5 equiv.) in the CH_3CN (1.0 mL) were stirred at 120°C under Ar atmosphere for 18 h. The reactions were conducted in a sealed Schlenk tube and heated by an IKA magnetic heating agitator with heating block. The reaction temperature was directly read from temperature detector of IKA apparatus and was calibrated by thermometer. After cooling to room temperature, the reaction mixture was basified up to pH 7 *via* std. Na_2CO_3 aqueous solution, then extracted by diether (3×3 mL) and dried over anhydrous Na_2SO_4 . After filtration and concentrated in rotary evaporator, the crude product was purified with flash chromatography on silica gel (ethyl acetate : *n*-hexane) to give products. Work-up gave product **33** (37.4 mg, 0.10 mmol, isolated yield 83%) as a yellow solid. **mp**: $146\text{--}147^\circ\text{C}$. **FT-IR**: ν (cm^{-1}): 3058, 3031, 2920, 2854, 2774, 1673, 1630, 1583, 1519, 1475, 1426, 1416, 1356, 1323, 1280, 1269, 1230, 1197, 1181, 1156, 1142, 1100, 1086, 1055, 1034, 1018, 1008, 946, 907, 882, 853, 820, 785, 767, 740, 725, 694, 647, 622, 587, 571, 550, 505, 495, 439, 420, 404. **^1H NMR** (400 MHz, CDCl_3) δ 10.09 (s, 1H), 8.64 (d, $J = 4.8$ Hz, 1H), 7.89 (d, $J = 8.4$ Hz, 2H), 7.71 – 7.67 (m, 4H), 7.59 (d, $J = 7.6$ Hz, 2H), 7.55 – 7.46 (m, 3H), 7.41 (t, $J = 7.6$ Hz, 2H), 7.33 (d, $J = 7.2$ Hz, 1H), 7.15 – 7.11 (m, 1H), 6.78 (dd, $J = 9.6, 6.4$ Hz, 1H), 6.53 (t, $J = 6.0$ Hz, 1H). **^{13}C NMR** (100 MHz, CDCl_3) δ 188.7, 153.1, 149.2, 142.0, 140.2, 135.9, 132.3, 131.5, 131.1, 130.3, 128.9, 127.8, 127.7, 127.6, 127.2, 127.1, 125.6, 122.9, 122.4, 121.0, 120.5. **ESI-HRMS**: m/z calcd. for $\text{C}_{26}\text{H}_{18}\text{N}_2\text{O}$ $[\text{M}+\text{H}]^+$: 375.1497, found 375.1495.

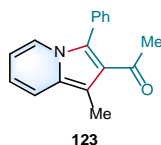


Preparation of (*E*)-3-(3-(4-bromophenyl)-1-(pyridin-2-yl)indolizin-2-yl)-1-phenylprop-2-en-1-one (**121**): A mixture of 3-(4-bromophenyl)-1-(pyridin-2-yl)indolizine-2-carbaldehyde (0.05 mmol), acetophenone (3.0 equiv.) and NaOH (3.0 equiv.) in the EtOH : CF₃CH₂OH (2.0 : 1.0 mL) were stirred at 25 °C under air atmosphere for 24 h. The reactions were conducted in a glass tube and stirred by an IKA magnetic heating agitator. The reaction was monitored by TLC. After filtration and concentrated in rotary evaporator, the crude product was purified with flash chromatography on silica gel (ethyl acetate : *n*-hexane) to give products. Workup gave product **121** (21.5 mg, 0.05 mmol, isolated yield 90%) as a yellow solid. **mp**: 173–174 °C. **FT-IR**: ν (cm⁻¹): 3054, 2920, 1655, 1583, 1504, 1471, 1447, 1387, 1341, 1296, 1212, 1177, 1148, 1102, 1072, 1034, 1008, 991, 853, 830, 789, 742, 690, 647, 583, 548, 509, 441, 408. **¹H NMR** (400 MHz, CDCl₃) δ 8.73 (d, *J* = 4.4 Hz, 1H), 7.91 (d, *J* = 16.0 Hz, 1H), 7.72 – 7.66 (m, 5H), 7.57 (d, *J* = 7.6 Hz, 2H), 7.47 (d, *J* = 7.6 Hz, 1H), 7.42 (t, *J* = 7.1 Hz, 1H), 7.34 – 7.32 (m, 4H), 7.17 (d, *J* = 10.0 Hz, 1H), 6.93 (d, *J* = 15.6 Hz, 1H), 6.77 – 6.73 (m, 1H), 6.48 (t, *J* = 6.8 Hz, 1H). **¹³C NMR** (100 MHz, CDCl₃) δ 189.9, 154.2, 149.9, 138.2, 137.4, 136.4, 132.9, 132.8, 132.5, 132.3, 129.6, 128.4, 128.2, 125.1, 124.7, 124.1, 123.3, 122.3, 121.0, 120.9, 120.1, 119.0, 114.0, 112.8. **ESI-HRMS**: *m/z* calcd. for C₂₈H₁₉N₂OBr [M+H]⁺: 479.0759, found 479.07740.

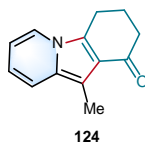


Preparation of 6-benzoyl-8-(4-bromophenyl)indolizino[1,2-a]quinolizine-5-ium bromide (**122**): A mixture of 3-(4-bromophenyl)-1-(pyridin-2-yl)indolizine-2-carbaldehyde (0.05 mmol) and 2-bromo-1-phenylethan-1-one (1.0 equiv.) in the CH₃CN (1.0 mL) were stirred at 80 °C under Ar atmosphere for 24 h. The reactions were conducted in a sealed Schlenk tube and heated by an IKA magnetic heating agitator with heating block. The reaction temperature was directly read from temperature detector of IKA apparatus and was calibrated by thermometer. After cooling to room temperature, resulting precipitate was filtered off, washed with *n*-hexane and washed with ethyl acetate, then gave product **122** (20.0 mg, 0.04 mmol, isolated yield 72%) as a red solid. **mp**: 293–295 °C. **FT-IR**: ν (cm⁻¹): 3392, 3060, 1655, 1630, 1597, 1552, 1533, 1492, 1473, 1436, 1354, 1333, 1302, 1245, 1197, 1164, 1142, 1119, 1076, 1057, 1008, 950, 863, 824, 771, 756, 723, 711, 699, 686, 668, 635, 622, 614, 546, 499, 443, 416. **¹H NMR** (400 MHz,

DMSO- d_6) δ 9.26 (d, J = 8.8 Hz, 1H), 9.19 (d, J = 6.8 Hz, 1H), 9.11 (d, J = 9.2 Hz, 1H), 8.98 (d, J = 6.8 Hz, 1H), 8.49 (t, J = 7.8 Hz, 1H), 8.15 (s, 1H), 8.09 (d, J = 7.6 Hz, 2H), 7.89 – 7.80 (m, 4H), 7.75 – 7.73 (m, 3H), 7.64 (t, J = 7.5 Hz, 2H), 7.58 (t, J = 6.7 Hz, 1H). ^{13}C NMR (100 MHz, DMSO- d_6) δ 188.9, 142.1, 140.3, 137.7, 135.6, 135.2, 135.2, 132.8, 132.1, 130.8, 130.4, 129.1, 126.3, 126.1, 125.6, 122.9, 122.8, 120.5, 120.1, 119.8, 119.5, 118.7, 118.3, 104.2. **ESI-HRMS**: m/z calcd. for $\text{C}_{28}\text{H}_{18}\text{N}_2\text{OBr}^+$ [M]: 477.0603, found 477.0596.



Following the **General procedure D4**, a mixture of α,β -unsaturated ketone (0.2 mmol), heteroaryl ketone (2.5 equiv.), catalyst **3b** (0.04 mmol), AcOH (2.0 equiv.) and LiSO_3CF_3 (3.0 equiv.) in the $\text{CF}_3\text{CH}_2\text{OH}$ (0.9 mL) were stirred at 80 °C under Ar atmosphere for 36 h. Work-up gave product 1-(1-methyl-3-phenylindolizin-2-yl)ethan-1-one (**123**, 9.5 mg, 0.04 mmol, isolated yield 21%) as a yellow liquid. **mp**: 93–94 °C. **FT-IR**: ν (cm^{-1}): 2922, 1655, 1517, 1469, 1445, 1424, 1350, 1323, 1241, 1168, 954, 859, 767, 738, 701, 670, 647, 589, 554, 495, 422. ^1H NMR (400 MHz, CDCl_3) δ 7.49 – 7.40 (m, 4H), 7.35 – 7.31 (m, 3H), 6.5 – 6.53 (m, 1H), 6.32 (t, J = 6.6 Hz, 1H), 2.46 (s, 3H), 1.98 (s, 3H). ^{13}C NMR (100 MHz, CDCl_3) δ 198.3, 131.8, 131.1, 129.9, 129.3, 129.0, 126.1, 126.1, 122.5, 118.7, 116.8, 112.2, 110.1, 31.4, 10.2. **ESI-HRMS**: m/z calcd. for $\text{C}_{17}\text{H}_{15}\text{NO}$ [$\text{M}+\text{H}$] $^+$: 250.1232, found 250.1227.



Following the **General procedure D4**, a mixture of α,β -unsaturated aldehyde (0.2 mmol), heteroaryl ketone (2.5 equiv.), catalyst **3b** (0.04 mmol), AcOH (2.0 equiv.) and LiSO_3CF_3 (3.0 equiv.) in the $\text{CF}_3\text{CH}_2\text{OH}$ (0.9 mL) were stirred at 80 °C under Ar atmosphere for 36 h. Work-up gave product 10-methyl-3,4-dihydropyrido[1,2-a]indol-1(2H)-one (**124**, 15.2 mg, 0.08 mmol, isolated yield 38%) as a yellow solid. **mp**: 111–112 °C. **FT-IR**: ν (cm^{-1}): 2920, 2854, 1649, 1523, 1432, 1414, 1368, 1331, 1265, 1228, 1183, 1140, 1082, 997, 925, 894, 859, 812, 783, 736, 713, 628, 598, 577, 560, 530, 466, 420. ^1H NMR (400 MHz, CDCl_3) δ 7.48 (d, J = 7.2 Hz, 1H), 7.24 (d, J = 9.2 Hz, 1H), 6.49 (dd, J = 9.0, 6.4 Hz, 1H), 6.44 – 6.41 (m, 1H), 2.85 (t, J = 6.2 Hz, 2H), 2.54 – 2.51 (m, 2H), 2.47 (s, 3H), 2.19 (p, J = 6.4 Hz,

2H). **^{13}C NMR** (100 MHz, CDCl_3) δ 197.2, 130.8, 129.9, 121.8, 121.0, 119.1, 116.0, 112.1, 107.8, 39.4, 23.5, 21.1, 9.6. **ESI-HRMS**: m/z calcd. for $\text{C}_{13}\text{H}_{13}\text{NO}$ $[\text{M}+\text{H}]^+$: 200.1075, found 200.1071.

References

- 1 Bar-On, Y. M., Phillips, R. & Milo, R. The biomass distribution on Earth. *Proceedings of the National Academy of Sciences* **115**, 6506-6511 (2018).
- 2 McKendry, P. Energy production from biomass (part 1): overview of biomass. *Bioresource Technology* **83**, 37-46 (2002).
- 3 Alivisatos, A. P. *et al.* Organization of 'nanocrystal molecules' using DNA. *Nature* **382**, 609-611 (1996).
- 4 Calladine, C. R. & Drew, H. *Understanding DNA: the molecule and how it works*. (Academic press, 1997).
- 5 Plomin, R. *Blueprint: How DNA makes us who we are*. (Mit Press, 2019).
- 6 Corma, A., Iborra, S. & Velty, A. Chemical routes for the transformation of biomass into chemicals. *Chemical Reviews* **107**, 2411-2502 (2007).
- 7 Bozell, J. J. & Petersen, G. R. Technology development for the production of biobased products from biorefinery carbohydrates—the US Department of Energy's "Top 10" revisited. *Green Chemistry* **12**, 539-554 (2010).
- 8 Chheda, J. N., Huber, G. W. & Dumesic, J. A. Liquid-phase catalytic processing of biomass-derived oxygenated hydrocarbons to fuels and chemicals. *Angewandte Chemie International Edition* **46**, 7164-7183 (2007).
- 9 Besson, M. I., Gallezot, P. & Pinel, C. Conversion of biomass into chemicals over metal catalysts. *Chemical Reviews* **114**, 1827-1870 (2014).
- 10 Gallezot, P. Conversion of biomass to selected chemical products. *Chemical Society Reviews* **41**, 1538-1558 (2012).
- 11 Schutyser, W. *et al.* Chemicals from lignin: an interplay of lignocellulose fractionation, depolymerisation, and upgrading. *Chemical Society Reviews* **47**, 852-908 (2018).
- 12 Chen, X., Song, S., Li, H., Gözaydın, G. K. & Yan, N. Expanding the Boundary of Biorefinery: Organonitrogen Chemicals from Biomass. *Accounts of Chemical Research* **54**, 1711-1722 (2021).
- 13 Liao, Y. *et al.* A sustainable wood biorefinery for low-carbon footprint chemicals production. *Science* **367**, 1385-1390 (2020).
- 14 Zhang, T. Taking on all of the biomass for conversion. *Science* **367**, 1305-1306 (2020).
- 15 De Schouwer, F., Claes, L., Vandekerckhove, A., Verduyck, J. & De Vos, D. E. Protein-rich biomass waste as a resource for future biorefineries: state of the art, challenges, and opportunities. *ChemSusChem* **12**, 1272-1303 (2019).
- 16 Hodge, R. (New York, 2009).
- 17 Kessel, A. & Ben-Tal, N. *Introduction to proteins: structure, function, and motion*. (Chapman and Hall/CRC, 2018).
- 18 Kim, S.-K. *Chitin, chitosan, oligosaccharides and their derivatives: biological activities and applications*. (CRC Press, 2010).
- 19 Werpy, T. A., Holladay, J. E. & White, J. F. Top value added chemicals from biomass: I. results of screening for potential candidates from sugars and synthesis gas. (Pacific Northwest National Lab.(PNNL), Richland, WA (United States), 2004).
- 20 Zhao, L. *Oligosaccharides of chitin and chitosan: bio-manufacture and applications*. (Springer Nature, 2019).
- 21 Zikakis, J. *Chitin, chitosan, and related enzymes*. (Elsevier, 2012).
- 22 Yang, Q. & Fukamizo, T. *Targeting chitin-containing organisms*. Vol. 1142 (Springer, 2019).

- 23 Nemtsev, S., Zueva, O. Y., Khismatullin, M., Albulov, A. & Varlamov, V. Isolation of chitin and chitosan from honeybees. *Applied Biochemistry and Microbiology* **40**, 39-43 (2004).
- 24 Crestini, C., Kovac, B. & Giovannozzi-Sermanni, G. Production and isolation of chitosan by submerged and solid-state fermentation from *Lentinus edodes*. *Biotechnology and Bioengineering* **50**, 207-210 (1996).
- 25 Hayes, M., Carney, B., Slater, J. & Brück, W. Mining marine shellfish wastes for bioactive molecules: chitin and chitosan—Part B: applications. *Biotechnology Journal: Healthcare Nutrition Technology* **3**, 878-889 (2008).
- 26 Teng, W. L., Khor, E., Tan, T. K., Lim, L. Y. & Tan, S. C. Concurrent production of chitin from shrimp shells and fungi. *Carbohydrate Research* **332**, 305-316 (2001).
- 27 Fernandez-Kim, S.-O. Physicochemical and functional properties of crawfish chitosan as affected by different processing protocols. (2004).
- 28 No, H. K. & Meyers, S. P. Preparation and characterization of chitin and chitosan—a review. *Journal of Aquatic Food Product Technology* **4**, 27-52 (1995).
- 29 Percot, A., Viton, C. & Domard, A. Optimization of chitin extraction from shrimp shells. *Biomacromolecules* **4**, 12-18 (2003).
- 30 Uchida, Y. in *Proc. 4th Int. Conf. Chitin/Chitosan, 1989*. (Elsevier Applied Science).
- 31 Zakaria, Z., Hall, G. & Shama, G. Lactic acid fermentation of scampi waste in a rotating horizontal bioreactor for chitin recovery. *Process Biochemistry* **33**, 1-6 (1998).
- 32 Jung, W., Jo, G., Kuk, J., Kim, K. & Park, R. Extraction of chitin from red crab shell waste by cofermentation with *Lactobacillus paracasei* subsp. *tolerans* KCTC-3074 and *Serratia marcescens* FS-3. *Applied Microbiology and Biotechnology* **71**, 234-237 (2006).
- 33 Jung, W. *et al.* Production of chitin from red crab shell waste by successive fermentation with *Lactobacillus paracasei* KCTC-3074 and *Serratia marcescens* FS-3. *Carbohydrate Polymers* **68**, 746-750 (2007).
- 34 Paulino, A. T., Simionato, J. I., Garcia, J. C. & Nozaki, J. Characterization of chitosan and chitin produced from silkworm crysalides. *Carbohydrate Polymers* **64**, 98-103 (2006).
- 35 Zhang, M., Haga, A., Sekiguchi, H. & Hirano, S. Structure of insect chitin isolated from beetle larva cuticle and silkworm (*Bombyx mori*) pupa exuvia. *International Journal of Biological Macromolecules* **27**, 99-105 (2000).
- 36 Bertuzzi, D. L. *et al.* General Protocol to Obtain D-Glucosamine from Biomass Residues: Shrimp Shells, Cicada Sloughs and Cockroaches. *Global Challenges* **2**, 1800046 (2018).
- 37 Rudall, K. in *Advances in insect physiology* Vol. 1 257-313 (Elsevier, 1963).
- 38 Rudall, K. & Kenchington, W. The chitin system. *Biological Reviews* **48**, 597-633 (1973).
- 39 Sikorski, P., Hori, R. & Wada, M. Revisit of α -chitin crystal structure using high resolution X-ray diffraction data. *Biomacromolecules* **10**, 1100-1105 (2009).
- 40 Yui, T., Taki, N., Sugiyama, J. & Hayashi, S. Exhaustive crystal structure search and crystal modeling of β -chitin. *International Journal of Biological Macromolecules* **40**, 336-344 (2007).
- 41 Ramachandran, G. & Ramakrishnan, C. *Conformation of biopolymers*. Vol. 1 (Elsevier, 1967).
- 42 Muzzarelli, R. A., Jeuniaux, C. & Gooday, G. W. *Chitin in nature and technology*. Vol. 385 (Springer, 1986).
- 43 Giraud-Guille, M.-M., Chanzy, H. & Vuong, R. Chitin crystals in arthropod cuticles revealed by diffraction contrast transmission electron microscopy. *Journal of Structural Biology* **103**, 232-240 (1990).
- 44 Kurita, K. *et al.* Squid chitin as a potential alternative chitin source: deacetylation behavior and characteristic properties. *Journal of Polymer Science Part A: Polymer Chemistry* **31**, 485-491 (1993).
- 45 Smucker, R. A. Chitin primary production. *Biochemical Systematics and Ecology* **19**, 357-369 (1991).
- 46 Blackwell, J. & Walton, A. The polysaccharides. *Biopolymers*. Academic Press, New York/London, 464-513 (1973).

- 47 Robyt, J. F. & John, F. *Essentials of carbohydrate chemistry*. (Springer Science & Business Media, 1998).
- 48 Virués, C. *et al.* Formulation of anomerization and protonation in d-glucosamine, based on ¹H NMR. *Carbohydrate Research* **490**, 107952 (2020).
- 49 Kumar, M. N. R. A review of chitin and chitosan applications. *Reactive and Functional Polymers* **46**, 1-27 (2000).
- 50 Ahmed, S. & Ikram, S. *Chitosan: derivatives, composites and applications*. (John Wiley & Sons, 2017).
- 51 Jain, T., Kumar, H. & Dutta, P. K. D-glucosamine and N-acetyl d-glucosamine: Their potential use as regenerative medicine. *Chitin and Chitosan for Regenerative Medicine*, 279-295 (2016).
- 52 Zeng, K., Groth, T. & Zhang, K. Recent advances in artificially sulfated polysaccharides for applications in cell growth and differentiation, drug delivery, and tissue engineering. *ChemBioChem* **20**, 737-746 (2019).
- 53 No, H., Meyers, S. P., Prinyawiwatkul, W. & Xu, Z. Applications of chitosan for improvement of quality and shelf life of foods: a review. *Journal of Food Science* **72**, R87-R100 (2007).
- 54 Jayakumar, R., Prabakaran, M. & Muzzarelli, R. A. *Chitosan for biomaterials I*. Vol. 243 (Springer, 2011).
- 55 Krajewska, B. Application of chitin-and chitosan-based materials for enzyme immobilizations: a review. *Enzyme and Microbial Technology* **35**, 126-139 (2004).
- 56 Shaunak, S. *et al.* Polyvalent dendrimer glucosamine conjugates prevent scar tissue formation. *Nature Biotechnology* **22**, 977-984 (2004).
- 57 Chopin, N., Guillory, X., Weiss, P., Bideau, J. L. & Collic-Jouault, S. Design polysaccharides of marine origin: chemical modifications to reach advanced versatile compounds. *Current Organic Chemistry* **18**, 867-895 (2014).
- 58 El Knidri, H., Belaabed, R., Addaou, A., Laajeb, A. & Lahsini, A. Extraction, chemical modification and characterization of chitin and chitosan. *International Journal of Biological Macromolecules* **120**, 1181-1189 (2018).
- 59 Rajasree, R. & Rahate, K. An overview on various modifications of chitosan and its applications. *International Journal of Pharmaceutical Sciences and Research* **4**, 4175-4193 (2013).
- 60 Mourya, V. & Inamdar, N. N. Chitosan-modifications and applications: opportunities galore. *Reactive and Functional Polymers* **68**, 1013-1051 (2008).
- 61 Jayakumar, R., Reis, R. & Mano, J. Chemistry and applications of phosphorylated chitin and chitosan. *e-Polymers* **6** (2006).
- 62 Negm, N. A., Hefni, H. H., Abd-Elal, A. A., Badr, E. A. & Abou Kana, M. T. Advancement on modification of chitosan biopolymer and its potential applications. *International Journal of Biological Macromolecules* **152**, 681-702 (2020).
- 63 Jayakumar, R., Selvamurugan, N., Nair, S., Tokura, S. & Tamura, H. Preparative methods of phosphorylated chitin and chitosan—An overview. *International Journal of Biological Macromolecules* **43**, 221-225 (2008).
- 64 Curti, E., de Britto, D. & Campana-Filho, S. P. Methylation of chitosan with iodomethane: effect of reaction conditions on chemoselectivity and degree of substitution. *Macromolecular Bioscience* **3**, 571-576 (2003).
- 65 Wei, L. *et al.* Enhanced antioxidant and antifungal activity of chitosan derivatives bearing 6-O-imidazole-based quaternary ammonium salts. *Carbohydrate Polymers* **206**, 493-503 (2019).
- 66 Chethan, P., Vishalakshi, B., Sathish, L., Ananda, K. & Poojary, B. Preparation of substituted quaternized arylfuran chitosan derivatives and their antimicrobial activity. *International Journal of Biological Macromolecules* **59**, 158-164 (2013).
- 67 Tabriz, A. *et al.* Quaternized trimethyl functionalized chitosan based antifungal membranes for drinking water treatment. *Carbohydrate Polymers* **207**, 17-25 (2019).
- 68 Qin, Y., Li, P. & Guo, Z. Cationic chitosan derivatives as potential antifungals: A review of structural optimization and applications. *Carbohydrate Polymers* **236**, 116002 (2020).

- 69 Jiang, Z. *et al.* Preparation and anti-tumor metastasis of carboxymethyl chitosan. *Carbohydrate Polymers* **125**, 53-60 (2015).
- 70 Zargar, V., Asghari, M. & Dashti, A. A review on chitin and chitosan polymers: structure, chemistry, solubility, derivatives, and applications. *ChemBioEng Reviews* **2**, 204-226 (2015).
- 71 Demina, T. *et al.* The study of the interaction between chitosan and 2, 2-bis (hydroxymethyl) propionic acid during solid-phase synthesis. *Polymer Science Series B* **53**, 358-370 (2011).
- 72 Lee, K. Y., Ha, W. S. & Park, W. H. Blood compatibility and biodegradability of partially N-acetylated chitosan derivatives. *Biomaterials* **16**, 1211-1216 (1995).
- 73 Sashiwa, H. & Aiba, S.-i. Chemically modified chitin and chitosan as biomaterials. *Progress in Polymer Science* **29**, 887-908 (2004).
- 74 Myszk, H., Bednarczyk, D., Najder, M. & Kaca, W. Synthesis and induction of apoptosis in B cell chronic leukemia by diosgenyl 2-amino-2-deoxy- β -D-glucopyranoside hydrochloride and its derivatives. *Carbohydrate Research* **338**, 133-141 (2003).
- 75 Irvine, J. C. & Hynd, A. VIII.—Synthetical aminoglucosides derived from d-glucosamine. *Journal of the Chemical Society, Transactions* **103**, 41-56 (1913).
- 76 Irvine, J. C. & Earl, J. C. CCLXXXVI.—Salicylidene derivatives of d-glucosamine. *Journal of the Chemical Society, Transactions* **121**, 2376-2381 (1922).
- 77 Costamagna, J., Barroso, N., Matsuhira, B. & Villagrán, M. Copper (II) complexes with aminosugar derived Schiff bases as ligands. *Inorganica Chimica Acta* **273**, 191-195 (1998).
- 78 Nguyen, T. D., Van Nguyen, Q., Ho, H. T. & Ngo, D. B. T. in *Proceedings of the 15th international electronic conference on synthetic organic chemistry, sciforum electronic conference series*.
- 79 Wolfrom, M. & Bhat, H. Trifluoroacetyl as N-blocking group in amino-sugar nucleoside synthesis. *Chemical Communications (London)*, 146a-146a (1966).
- 80 Wolfrom, M. L. & Bhat, H. B. Trichloroacetyl and trifluoroacetyl as N-blocking groups in nucleoside synthesis with 2-amino sugars. *The Journal of Organic Chemistry* **32**, 1821-1823 (1967).
- 81 Wolfrom, M. & Conigliaro, P. Trifluoroacetyl as an N-protective group in the synthesis of purine nucleosides of 2-amino-2-deoxy saccharides. *Carbohydrate Research* **11**, 63-76 (1969).
- 82 Laborda, P. *et al.* An enzymatic N-acylation step enables the biocatalytic synthesis of unnatural sialosides. *Angewandte Chemie International Edition* **59**, 5308-5311 (2020).
- 83 Micheel, F. & Wulff, H. Über die Reaktionen des d-Glucosamins (III. Mitteil.); Synthese eines d-Glucosamin-anhydrids. *Chemische Berichte* **89**, 1521-1530 (1956).
- 84 Bertho, A. & Révész, A. Über 1- β , 2-Diamino-3, 4, 6-triacetyl-d-glucose. *Justus Liebigs Annalen der Chemie* **581**, 161-167 (1953).
- 85 Pérez, E. M. *et al.* Schiff bases from d-glucosamine and aliphatic ketones. *Carbohydrate Research* **345**, 23-32 (2010).
- 86 Wolfrom, M. & Shen Han, T. Methyl 2-Deoxy-2-sulfoamino- β -D-glucopyranoside Trisulfate and the Preparation of Tri-O-acetyl-2-amino-2deoxy- α -D-glucopyranosyl Bromide. *The Journal of Organic Chemistry* **26**, 2145-2146 (1961).
- 87 Pajk, S., Garvas, M., Štrancar, J. & Pečar, S. Nitroxide-fluorophore double probes: a potential tool for studying membrane heterogeneity by ESR and fluorescence. *Organic & Biomolecular Chemistry* **9**, 4150-4159 (2011).
- 88 Lin, H. & Walsh, C. T. A chemoenzymatic approach to glycopeptide antibiotics. *Journal of the American Chemical Society* **126**, 13998-14003 (2004).
- 89 Wang, G. *et al.* Convenient synthesis of an N-glycan octasaccharide of the bisecting type. *The Journal of Organic Chemistry* **74**, 2508-2515 (2009).
- 90 Silva, D. J., Wang, H., Allanson, N. M., Jain, R. K. & Sofia, M. J. Stereospecific Solution-and Solid-Phase Glycosylations. Synthesis of β -Linked Saccharides and Construction of Disaccharide Libraries Using Phenylsulfenyl 2-Deoxy-2-Trifluoroacetamido Glycopyranosides as Glycosyl Donors. *The Journal of Organic Chemistry* **64**, 5926-5929 (1999).

- 91 Cheung, A. W.-H., Sidduri, A., Garofalo, L. M. & Goodnow Jr, R. A. An unusual solvent effect on the regiochemical outcome (N-9 versus N-7) of guanine glycosylation using Robins' reagent (2-N-acetyl-6-O-diphenylcarbamoylguanine). *Tetrahedron Letters* **41**, 3303-3307 (2000).
- 92 Goodnow Jr, R. A., Richou, A.-R. & Tam, S. Synthesis of thymine, cytosine, adenine, and guanine containing N-Fmoc protected amino acids: Building blocks for construction of novel oligonucleotide backbone analogs. *Tetrahedron Letters* **38**, 3195-3198 (1997).
- 93 Einbu, A. & Vårum, K. M. Characterization of chitin and its hydrolysis to GlcNAc and GlcN. *Biomacromolecules* **9**, 1870-1875 (2008).
- 94 Einbu, A. & Vårum, K. M. Depolymerization and de-N-acetylation of chitin oligomers in hydrochloric acid. *Biomacromolecules* **8**, 309-314 (2007).
- 95 Einbu, A., Grasdalen, H. & Vårum, K. M. Kinetics of hydrolysis of chitin/chitosan oligomers in concentrated hydrochloric acid. *Carbohydrate Research* **342**, 1055-1062 (2007).
- 96 Yabushita, M., Kobayashi, H., Kuroki, K., Ito, S. & Fukuoka, A. Catalytic depolymerization of chitin with retention of N-acetyl group. *ChemSusChem* **8**, 3760-3763 (2015).
- 97 Chen, X., Yang, H., Zhong, Z. & Yan, N. Base-catalysed, one-step mechanochemical conversion of chitin and shrimp shells into low molecular weight chitosan. *Green Chemistry* **19**, 2783-2792 (2017).
- 98 Margoutidis, G., Parsons, V. H., Bottaro, C. S., Yan, N. & Kerton, F. M. Mechanochemical amorphization of α -chitin and conversion into oligomers of N-acetyl-D-glucosamine. *ACS Sustainable Chemistry & Engineering* **6**, 1662-1669 (2018).
- 99 Kobayashi, H., Techikawara, K. & Fukuoka, A. Hydrolytic hydrogenation of chitin to amino sugar alcohol. *Green Chemistry* **19**, 3350-3356 (2017).
- 100 Pierson, Y., Chen, X., Bobbink, F. D., Zhang, J. & Yan, N. Acid-catalyzed chitin liquefaction in ethylene glycol. *ACS Sustainable Chemistry & Engineering* **2**, 2081-2089 (2014).
- 101 Bobbink, F. D., Zhang, J., Pierson, Y., Chen, X. & Yan, N. Conversion of chitin derived N-acetyl-D-glucosamine (NAG) into polyols over transition metal catalysts and hydrogen in water. *Green Chemistry* **17**, 1024-1031 (2015).
- 102 Kuhn, R. & Krüger, G. 3-Acetamino-furan aus N-Acetyl-D-glicosamin; ein Beitrag Zur Theorie der Morgan-Elson-Reaktion. *Chemische Berichte* **89**, 1473-1486 (1956).
- 103 Kuhn, R. & Krüger, G. Das Chromogen III der Morgan-Elson-Reaktion. *Chemische Berichte* **90**, 264-277 (1957).
- 104 Ohkuma, S. & Shinohara, T. Chromogen Formation and Epimerization of N-Acetylhexosamines on Paper impregnated with Potassium Tetraborate. *Nature* **206**, 513-514 (1965).
- 105 Osada, M. *et al.* Non-catalytic synthesis of Chromogen I and III from N-acetyl-D-glucosamine in high-temperature water. *Green Chemistry* **15**, 2960-2966 (2013).
- 106 Ogata, M., Hattori, T., Takeuchi, R. & Usui, T. Novel and facile synthesis of furanodictines A and B based on transformation of 2-acetamido-2-deoxy-D-glucose into 3, 6-anhydro hexofuranoses. *Carbohydrate Research* **345**, 230-234 (2010).
- 107 Kikuchi, H. *et al.* Furanodictine A and B: Amino Sugar Analogues Produced by Cellular Slime Mold Dictyostelium discoideum Showing Neuronal Differentiation Activity. *The Journal of Organic Chemistry* **66**, 6982-6987 (2001).
- 108 Franich, R. A., Goodin, S. J. & Wilkins, A. L. Acetamidofurans, acetamidopyrones, and acetamidoacetaldehyde from pyrolysis of chitin and n-acetylglucosamine. *Journal of Analytical and Applied Pyrolysis* **7**, 91-100 (1984).
- 109 Kerton, F. M., Omari, K. W. B. & Dodot, L. A simple one-pot dehydration process to convert N-acetyl-D-glucosamine into a nitrogen-containing compound, 3-acetamido-5-acetylfuran. *ChemSusChem* **5**, 1767-1772 (2012).
- 110 Drover, M. W., Omari, K. W., Murphy, J. N. & Kerton, F. M. Formation of a renewable amide, 3-acetamido-5-acetylfuran, via direct conversion of N-acetyl-D-glucosamine. *RSC Advances* **2**, 4642-4644 (2012).

- 111 Chen, X., Chew, S. L., Kerton, F. M. & Yan, N. Direct conversion of chitin into a N-containing furan derivative. *Green Chemistry* **16**, 2204-2212 (2014).
- 112 SUMOTO, K. *et al.* Formation of pyrazine derivatives from D-glucosamine and their deoxyribonucleic acid (DNA) strand breakage activity. *Chemical and Pharmaceutical Bulletin* **39**, 792-794 (1991).
- 113 Wu, M. *et al.* Deep eutectic solvents: Green solvents and catalysts for the preparation of pyrazine derivatives by self-condensation of d-glucosamine. *ACS Sustainable Chemistry & Engineering* **6**, 9434-9441 (2018).
- 114 Ohmi, Y., Nishimura, S. & Ebitani, K. Synthesis of α -Amino Acids from Glucosamine-HCl and its Derivatives by Aerobic Oxidation in Water Catalyzed by Au Nanoparticles on Basic Supports. *ChemSusChem* **6**, 2259-2262 (2013).
- 115 Gao, X. *et al.* Transformation of chitin and waste shrimp shells into acetic acid and pyrrole. *ACS Sustainable Chemistry & Engineering* **4**, 3912-3920 (2016).
- 116 Zhao, H. *et al.* Raw biomass electroreforming coupled to green hydrogen generation. *Nature Communications* **12**, 1-10 (2021).
- 117 Mascal, M. & Nikitin, E. B. Dramatic advancements in the saccharide to 5-(chloromethyl) furfural conversion reaction. *ChemSusChem: Chemistry & Sustainability Energy & Materials* **2**, 859-861 (2009).
- 118 Wang, Y., Pedersen, C. M., Deng, T., Qiao, Y. & Hou, X. Direct conversion of chitin biomass to 5-hydroxymethylfurfural in concentrated ZnCl₂ aqueous solution. *Bioresource Technology* **143**, 384-390 (2013).
- 119 Yu, S. *et al.* Efficient conversion of chitin biomass into 5-hydroxymethylfurfural over metal salts catalysts in dimethyl sulfoxide-water mixture under hydrothermal conditions. *Polymer Degradation and Stability* **134**, 105-114 (2016).
- 120 Zang, H. *et al.* Hydrothermal conversion of N-acetyl-d-glucosamine to 5-hydroxymethylfurfural using ionic liquid as a recycled catalyst in a water-dimethyl sulfoxide mixture. *Carbohydrate Research* **442**, 1-8 (2017).
- 121 Zhang, J. & Yan, N. Formic acid-mediated liquefaction of chitin. *Green Chemistry* **18**, 5050-5058 (2016).
- 122 Xie, S. *et al.* A Shortcut Route to Close Nitrogen Cycle: Bio-Based Amines Production via Selective Deoxygenation of Chitin Monomers over Ru/C in Acidic Solutions. *Iscience* **23**, 101096 (2020).
- 123 Shrotri, A., Kobayashi, H. & Fukuoka, A. Catalytic conversion of structural carbohydrates and lignin to chemicals. *Advances in Catalysis* **60**, 59-123 (2017).
- 124 Wu, J. *et al.* Selectivity-Switchable Conversion of Chitin-Derived N-Acetyl-d-glucosamine into Commodity Organic Acids at Room Temperature. *Industrial & Engineering Chemistry Research* **60**, 3239-3248 (2021).
- 125 Calcaterra, A. & D'Acquarica, I. The market of chiral drugs: Chiral switches versus de novo enantiomerically pure compounds. *Journal of Pharmaceutical and Biomedical Analysis* **147**, 323-340 (2018).
- 126 Santaniello, E., Ferraboschi, P. & Grisenti, P. Lipase-catalyzed transesterification in organic solvents: applications to the preparation of enantiomerically pure compounds. *Enzyme and Microbial Technology* **15**, 367-382 (1993).
- 127 Noyori, R. *Asymmetric Catalysis in Organic Synthesis*. (1994).
- 128 Jacobsen, E. N. & Denmark, S. E. *Catalytic Asymmetric Synthesis*. (2000).
- 129 Ohkuma, T., Noyori, R., Jacobsen, E., Pfaltz, A. & Yamamoto, H. *Comprehensive asymmetric catalysis. by Jacobsen EN, Pfaltz A., Yamamoto H., Springer, Berlin*, 199-246 (1999).
- 130 Xie, J.-H. & Zhou, Q.-L. Chiral diphosphine and monodentate phosphorus ligands on a spiro scaffold for transition-metal-catalyzed asymmetric reactions. *Accounts of Chemical Research* **41**, 581-593 (2008).
- 131 Zhou, Q.-L. *Privileged chiral ligands and catalysts*. (John Wiley & Sons, 2011).

- 132 Xie, J.-H. *et al.* Synthesis of spiro diphosphines and their application in asymmetric
hydrogenation of ketones. *Journal of the American Chemical Society* **125**, 4404-4405 (2003).
- 133 Munson, P. J. & Rodbard, D. Ligand: a versatile computerized approach for characterization of
ligand-binding systems. *Analytical biochemistry* **107**, 220-239 (1980).
- 134 Diéguez, M., Pàmies, O. & Claver, C. Ligands derived from carbohydrates for asymmetric
catalysis. *Chemical Reviews* **104**, 3189-3216 (2004).
- 135 Castellón, S., Claver, C. & Díaz, Y. C 1 and C 2-symmetric carbohydrate phosphorus ligands in
asymmetric catalysis. *Chemical Society Reviews* **34**, 702-713 (2005).
- 136 Woodward, S., Diéguez, M. & Pàmies, O. Use of sugar-based ligands in selective catalysis:
Recent developments. *Coordination Chemistry Reviews* **254**, 2007-2030 (2010).
- 137 Diéguez, M. *et al.* Carbohydrate derivative ligands in asymmetric catalysis. *Coordination
Chemistry Reviews* **248**, 2165-2192 (2004).
- 138 Börner, A. *Phosphorus ligands in asymmetric catalysis: Synthesis and applications*. Vol. 1
(Wiley-VCH, 2008).
- 139 Boysen, M. M. Carbohydrates as synthetic tools in organic chemistry. *Chemistry—A European
Journal* **13**, 8648-8659 (2007).
- 140 Fernandez-Perez, H., Etayo, P., Panossian, A. & Vidal-Ferran, A. Phosphine– phosphinite and
phosphine– phosphite ligands: preparation and applications in asymmetric catalysis. *Chemical
Reviews* **111**, 2119-2176 (2011).
- 141 Plass, W. Chiral and supramolecular model complexes for vanadium haloperoxidases: Host–guest
systems and hydrogen bonding relays for vanadate species. *Coordination Chemistry Reviews* **255**,
2378-2387 (2011).
- 142 Schaper, L. A., Hock, S. J., Herrmann, W. A. & Kühn, F. E. Synthese und Anwendung
wasserlöslicher NHC-Übergangsmetall-Komplexe. *Angewandte Chemie* **125**, 284-304 (2013).
- 143 Kyne, S. H. & Camp, J. E. Use of monosaccharides in metal-catalyzed coupling reactions. *ACS
Sustainable Chemistry & Engineering* **5**, 41-48 (2017).
- 144 Emmerson, D. P. *et al.* Precise structure activity relationships in asymmetric catalysis using
carbohydrate scaffolds to allow ready fine tuning: dialkylzinc–aldehyde additions. *Organic &
Biomolecular Chemistry* **1**, 3826-3838 (2003).
- 145 RajanBabu, T., Ayers, T. A. & Casalnuovo, A. L. Electronic amplification of selectivity in Rh-
catalyzed hydrogenations: D-glucose-derived ligands for the synthesis of D-or L-Amino Acids.
Journal of the American Chemical Society **116**, 4101-4102 (1994).
- 146 Gläser, B. & Kunz, H. Enantioselective allylic substitution using a novel (Phosphino- α -D-
glucopyrano-oxazoline) palladium catalyst. *Synlett* **1998**, 53-54 (1998).
- 147 Hashizume, T., Yonehara, K., Ohe, K. & Uemura, S. A novel amphiphilic chiral ligand derived
from D-glucosamine. Application to palladium-catalyzed asymmetric allylic substitution reaction
in an aqueous or an organic medium, allowing for catalyst recycling. *The Journal of Organic
Chemistry* **65**, 5197-5201 (2000).
- 148 Yonehara, K., Hashizume, T., Mori, K., Ohe, K. & Uemura, S. Palladium-Catalyzed Asymmetric
Allylic Substitution Reactions Using New Chiral Phosphinite– Oxazoline Ligands Derived from
d-Glucosamine. *The Journal of Organic Chemistry* **64**, 9374-9380 (1999).
- 149 Yonehara, K. *et al.* Palladium-catalyzed asymmetric intermolecular arylation of cyclic or acyclic
alkenes using phosphinite-oxazoline ligands derived from d-glucosamine. *Journal of
Organometallic Chemistry* **603**, 40-49 (2000).
- 150 Irmak, M., Groschner, A. & Boysen, M. M. glucoBox ligand—a new carbohydrate-based bis
(oxazoline) ligand. Synthesis and first application. *Chemical communications*, 177-179 (2007).
- 151 Bauer, T., Tarasiuk, J. & Paśniczek, K. Highly enantioselective diethylzinc addition to aldehydes
catalyzed by D-glucosamine derivatives. *Tetrahedron: Asymmetry* **13**, 77-82 (2002).
- 152 Mata, Y., Diéguez, M., Pàmies, O. & Claver, C. New Carbohydrate-Based Phosphite-Oxazoline
Ligands as Highly Versatile Ligands for Palladium-Catalyzed Allylic Substitution Reactions.
Advanced Synthesis & Catalysis **347**, 1943-1947 (2005).

- 153 Emmerson, D. P., Hems, W. P. & Davis, B. G. Carbohydrate-derived amino-alcohol ligands for asymmetric alkynylation of aldehydes. *Organic Letters* **8**, 207-210 (2006).
- 154 Cheng, D., Gan, F., Qian, W. & Bao, W. d-Glucosamine—a natural ligand for the N-arylation of imidazoles with aryl and heteroaryl bromides catalyzed by CuI. *Green Chemistry* **10**, 171-173 (2008).
- 155 Ge, X., Chen, X., Qian, C. & Zhou, S. Combined experimental/theoretical study of d-glucosamine promoted Ullmann-type C–N coupling catalyzed by copper (i): does amino really count? *RSC Advances* **6**, 29638-29645 (2016).
- 156 Thakur, K. G., Srinivas, K. S., Chiranjeevi, K. & Sekar, G. d-Glucosamine as an efficient ligand for the copper-catalyzed selective synthesis of anilines from aryl halides and NaN₃. *Green Chemistry* **13**, 2326-2329 (2011).
- 157 Yang, M., Shen, H., Li, Y., Shen, C. & Zhang, P. D-Glucosamine as a green ligand for copper catalyzed synthesis of aryl sulfones from aryl halides and sodium sulfinates. *RSC Advances* **4**, 26295-26300 (2014).
- 158 Wen, M., Shen, C., Wang, L., Zhang, P. & Jin, J. An efficient d-glucosamine-based copper catalyst for C–X couplings and its application in the synthesis of nilotinib intermediate. *RSC Advances* **5**, 1522-1528 (2015).
- 159 Amini, M. & Etemadi, H. D-Glucosamine as an efficient and green additive for palladium-catalyzed Heck reaction. *Chemical Papers* **67**, 759-763 (2013).
- 160 Sova, M., Frlan, R., Gobec, S., Stavber, G. & Časar, Z. d-Glucosamine in iron-catalysed cross-coupling reactions of Grignards with allylic and vinylic bromides: application to the synthesis of a key sitagliptin precursor. *Applied Organometallic Chemistry* **29**, 528-535 (2015).
- 161 MacMillan, D. W. The advent and development of organocatalysis. *Nature* **455**, 304-308 (2008).
- 162 Eder, U., Sauer, G. & Wiechert, R. New type of asymmetric cyclization to optically active steroid CD partial structures. *Angewandte Chemie International Edition in English* **10**, 496-497 (1971).
- 163 Moyano, A. & Rios, R. Asymmetric organocatalytic cyclization and cycloaddition reactions. *Chemical Reviews* **111**, 4703-4832 (2011).
- 164 Dalko, P. I. *Comprehensive Enantioselective Organocatalysis: Catalysts, Reactions, and Applications, 3 Volume Set*. (John Wiley & Sons, 2013).
- 165 Jones, S. B., Simmons, B., Mastracchio, A. & MacMillan, D. W. Collective synthesis of natural products by means of organocascade catalysis. *Nature* **475**, 183-188 (2011).
- 166 Mukherjee, S., Yang, J. W., Hoffmann, S. & List, B. Asymmetric enamine catalysis. *Chemical Reviews* **107**, 5471-5569 (2007).
- 167 Xiang, S.-H. & Tan, B. Advances in asymmetric organocatalysis over the last 10 years. *Nature Communications* **11**, 1-5 (2020).
- 168 Silvi, M. & Melchiorre, P. Enhancing the potential of enantioselective organocatalysis with light. *Nature* **554**, 41-49 (2018).
- 169 Kaeobamrung, J., Kozłowski, M. C. & Bode, J. W. Chiral N-heterocyclic carbene-catalyzed generation of ester enolate equivalents from α , β -unsaturated aldehydes for enantioselective Diels–Alder reactions. *Proceedings of the National Academy of Sciences* **107**, 20661-20665 (2010).
- 170 Taylor, M. S. & Jacobsen, E. N. Asymmetric catalysis by chiral hydrogen-bond donors. *Angewandte Chemie International Edition* **45**, 1520-1543 (2006).
- 171 Enders, D., Niemeier, O. & Henseler, A. Organocatalysis by N-heterocyclic carbenes. *Chemical Reviews* **107**, 5606-5655 (2007).
- 172 Akiyama, T. Stronger brønsted acids. *Chemical Reviews* **107**, 5744-5758 (2007).
- 173 Parmar, D., Sugiono, E., Raja, S. & Rueping, M. Complete field guide to asymmetric BINOL-phosphate derived Brønsted acid and metal catalysis: history and classification by mode of activation; Brønsted acidity, hydrogen bonding, ion pairing, and metal phosphates. *Chemical Reviews* **114**, 9047-9153 (2014).

- 174 Jensen, K. L., Dickmeiss, G., Jiang, H., Albrecht, L. & Jørgensen, K. A. The diarylprolinol silyl
ether system: a general organocatalyst. *Accounts of Chemical Research* **45**, 248-264 (2012).
- 175 Davie, E. A. C., Mennen, S. M., Xu, Y. & Miller, S. J. Asymmetric catalysis mediated by
synthetic peptides. *Chemical Reviews* **107**, 5759-5812 (2007).
- 176 Shirakawa, S. & Maruoka, K. Recent developments in asymmetric phase-transfer reactions.
Angewandte Chemie International Edition **52**, 4312-4348 (2013).
- 177 Klier, L., Tur, F., Poulsen, P. H. & Jørgensen, K. A. Asymmetric cycloaddition reactions
catalysed by diarylprolinol silyl ethers. *Chemical Society Reviews* **46**, 1080-1102 (2017).
- 178 List, B., Lerner, R. A. & Barbas, C. F. Proline-catalyzed direct asymmetric aldol reactions.
Journal of the American Chemical Society **122**, 2395-2396 (2000).
- 179 Ahrendt, K. A., Borths, C. J. & MacMillan, D. W. New strategies for organic catalysis: the first
highly enantioselective organocatalytic Diels–Alder reaction. *Journal of the American Chemical
Society* **122**, 4243-4244 (2000).
- 180 Erkkilä, A., Majander, I. & Pihko, P. M. Iminium catalysis. *Chemical Reviews* **107**, 5416-5470
(2007).
- 181 Lelais, G. & MacMillan, D. W. Modern strategies in organic catalysis: the advent and
development of iminium activation. *Aldrichimica Acta* **39**, 79-87 (2006).
- 182 Bertelsen, S. & Jørgensen, K. A. Organocatalysis—after the gold rush. *Chemical Society Reviews*
38, 2178-2189 (2009).
- 183 van Der Helm, M. P., Klemm, B. & Eelkema, R. Organocatalysis in aqueous media. *Nature
Reviews Chemistry* **3**, 491-508 (2019).
- 184 Schnitzer, T., Budinska, A. & Wennemers, H. Organocatalysed conjugate addition reactions of
aldehydes to nitroolefins with anti selectivity. *Nature Catalysis* **3**, 143-147 (2020).
- 185 Chauhan, P., Mahajan, S. & Enders, D. Achieving molecular complexity via stereoselective
multiple domino reactions promoted by a secondary amine organocatalyst. *Accounts of Chemical
Research* **50**, 2809-2821 (2017).
- 186 An, F., Maji, B., Min, E., Ofial, A. R. & Mayr, H. Basicities and Nucleophilicities of Pyrrolidines
and Imidazolidinones Used as Organocatalysts. *Journal of the American Chemical Society* **142**,
1526-1547 (2020).
- 187 Sigman, M. S. & Jacobsen, E. N. Schiff base catalysts for the asymmetric Strecker reaction
identified and optimized from parallel synthetic libraries. *Journal of the American Chemical
Society* **120**, 4901-4902 (1998).
- 188 Sigman, M. S., Vachal, P. & Jacobsen, E. N. A general catalyst for the asymmetric Strecker
reaction. *Angewandte Chemie* **112**, 1336-1338 (2000).
- 189 Becker, C., Hoben, C. & Kunz, H. Enantioselective organocatalysis of Strecker and Mannich
reactions based on carbohydrates. *Advanced Synthesis & Catalysis* **349**, 417-424 (2007).
- 190 Kong, S., Fan, W., Wu, G. & Miao, Z. Enantioselective Synthesis of Tertiary α -Hydroxy
Phosphonates Catalyzed by Carbohydrate/Cinchona Alkaloid Thiourea Organocatalysts.
Angewandte Chemie International Edition **51**, 8864-8867 (2012).
- 191 Gao, P., Wang, C., Wu, Y., Zhou, Z. & Tang, C. (Wiley Online Library, 2008).
- 192 Lu, A. *et al.* Highly enantio- and diastereoselective Michael addition of cyclohexanone to
nitroolefins catalyzed by a chiral glucose-based bifunctional secondary amine-thiourea catalyst.
Organic & Biomolecular Chemistry **7**, 3141-3147 (2009).
- 193 Brik, A., Yang, Y.-Y., Ficht, S. & Wong, C.-H. Sugar-assisted glycopeptide ligation. *Journal of
the American Chemical Society* **128**, 5626-5627 (2006).
- 194 Bennett, C. S. *et al.* Sugar-assisted glycopeptide ligation with complex oligosaccharides: scope
and limitations. *Journal of the American Chemical Society* **130**, 11945-11952 (2008).
- 195 Payne, R. J. *et al.* Extended sugar-assisted glycopeptide ligations: development, scope, and
applications. *Journal of the American Chemical Society* **129**, 13527-13536 (2007).
- 196 Yang, Y.-Y., Ficht, S., Brik, A. & Wong, C.-H. Sugar-assisted ligation in glycoprotein synthesis.
Journal of the American Chemical Society **129**, 7690-7701 (2007).

- 197 Brik, A., Ficht, S., Yang, Y.-Y., Bennett, C. S. & Wong, C.-H. Sugar-assisted ligation of N-linked glycopeptides with broad sequence tolerance at the ligation junction. *Journal of the American Chemical Society* **128**, 15026-15033 (2006).
- 198 Ficht, S., Payne, R. J., Brik, A. & Wong, C. H. Second-generation sugar-assisted ligation: a method for the synthesis of cysteine-containing glycopeptides. *Angewandte Chemie International Edition* **46**, 5975-5979 (2007).
- 199 Payne, R. J., Ficht, S., Greenberg, W. A. & Wong, C. H. Cysteine-free peptide and glycopeptide ligation by direct aminolysis. *Angewandte Chemie* **120**, 4483-4487 (2008).
- 200 Bennett, C. S. & Wong, C.-H. Chemoenzymatic approaches to glycoprotein synthesis. *Chemical Society Reviews* **36**, 1227-1238 (2007).
- 201 Chai, H. *et al.* N-Linked Glycosyl Auxiliary-Mediated Native Chemical Ligation on Aspartic Acid: Application towards N-Glycopeptide Synthesis. *Angewandte Chemie* **128**, 10519-10523 (2016).
- 202 Biagetti, M., Accetta, A., Capelli, A. M., Guala, M. & Retini, M. (Google Patents, 2016).
- 203 Burnett, D., A. Compounds, compositions and methods of use. (2018).
- 204 Canfield, D. E., Glazer, A. N. & Falkowski, P. G. The evolution and future of Earth's nitrogen cycle. *Science* **330**, 192-196 (2010).
- 205 Falkowski, P. G. Evolution of the nitrogen cycle and its influence on the biological sequestration of CO₂ in the ocean. *Nature* **387**, 272-275 (1997).
- 206 Smil, V. *Enriching the earth: Fritz Haber, Carl Bosch, and the transformation of world food production*. (MIT press, 2004).
- 207 Hartwig, J. F. Carbon–heteroatom bond-forming reductive eliminations of amines, ethers, and sulfides. *Accounts of Chemical Research* **31**, 852-860 (1998).
- 208 Beccalli, E. M., Broggin, G., Martinelli, M. & Sottocornola, S. C–C, C–O, C–N bond formation on sp² carbon by Pd (II)-catalyzed reactions involving oxidant agents. *Chemical Reviews* **107**, 5318-5365 (2007).
- 209 Song, G., Wang, F. & Li, X. C–C, C–O and C–N bond formation via rhodium (iii)-catalyzed oxidative C–H activation. *Chemical Society Reviews* **41**, 3651-3678 (2012).
- 210 Ouyang, K., Hao, W., Zhang, W.-X. & Xi, Z. Transition-metal-catalyzed cleavage of C–N single bonds. *Chemical Reviews* **115**, 12045-12090 (2015).
- 211 Wu, X.-S., Chen, Y., Li, M.-B., Zhou, M.-G. & Tian, S.-K. Direct substitution of primary allylic amines with sulfinates. *Journal of the American Chemical Society* **134**, 14694-14697 (2012).
- 212 Li, M. B., Wang, Y. & Tian, S. K. Regioselective and Stereospecific Cross-Coupling of Primary Allylic Amines with Boronic Acids and Boronates through Palladium-Catalyzed C–N Bond Cleavage. *Angewandte Chemie* **124**, 3022-3025 (2012).
- 213 Kalutharage, N. & Yi, C. S. Deaminative and Decarboxylative Catalytic Alkylation of Amino Acids with Ketones. *Angewandte Chemie International Edition* **52**, 13651-13655 (2013).
- 214 Huang, H., Ji, X., Wu, W., Huang, L. & Jiang, H. Copper-catalyzed formal C–N bond cleavage of aromatic methylamines: assembly of pyridine derivatives. *The Journal of Organic Chemistry* **78**, 3774-3782 (2013).
- 215 Zhou, M.-G., Zhang, W.-Z. & Tian, S.-K. Direct enantiospecific substitution of primary α -aminoalkylferrocenes via Lewis acid-catalyzed C–N bond cleavage. *Chemical Communications* **50**, 14531-14534 (2014).
- 216 Basch, C. H., Liao, J., Xu, J., Piane, J. J. & Watson, M. P. Harnessing alkyl amines as electrophiles for nickel-catalyzed cross couplings via C–N bond activation. *Journal of the American Chemical Society* **139**, 5313-5316 (2017).
- 217 Klauk, F. J., James, M. J. & Glorius, F. Deaminative Strategy for the Visible-Light-Mediated Generation of Alkyl Radicals. *Angewandte Chemie International Edition in English* **56**, 12336-12339 (2017).

- 218 Wang, J., Hoerrner, M. E., Watson, M. P. & Weix, D. J. Nickel-Catalyzed Synthesis of Dialkyl Ketones from the Coupling of N-Alkyl Pyridinium Salts with Activated Carboxylic Acids. *Angewandte Chemie* **132**, 13586-13591 (2020).
- 219 Plunkett, S., Basch, C. H., Santana, S. O. & Watson, M. P. Harnessing alkylpyridinium salts as electrophiles in deaminative alkyl-alkyl cross-couplings. *Journal of the American Chemical Society* **141**, 2257-2262 (2019).
- 220 Ashley, M. A. & Rovis, T. Photoredox-catalyzed deaminative alkylation via C–N bond activation of primary amines. *Journal of the American Chemical Society* **142**, 18310-18316 (2020).
- 221 Tang, S. *et al.* Catalytic oxidative deamination by water with H₂ liberation. *Journal of the American Chemical Society* **142**, 20875-20882 (2020).
- 222 Hopkinson, M. N., Richter, C., Schedler, M. & Glorius, F. An overview of N-heterocyclic carbenes. *Nature* **510**, 485-496 (2014).
- 223 Alcarazo, M. *et al.* Imidazo [1, 5-a] pyridine: a versatile architecture for stable N-heterocyclic carbenes. *Journal of the American Chemical Society* **127**, 3290-3291 (2005).
- 224 Hahn, F. E. Heterocyclic carbenes. *Angewandte Chemie International Edition in English* **45**, 1348-1352 (2006).
- 225 Kundu, N. *et al.* Reporting a unique example of electronic bistability observed in the form of valence tautomerism with a copper (II) helicate of a redox-active nitrogenous heterocyclic ligand. *Journal of the American Chemical Society* **133**, 20104-20107 (2011).
- 226 Weber, M. D. *et al.* Origin of a counterintuitive yellow light-emitting electrochemical cell based on a blue-emitting heteroleptic copper (I) complex. *Dalton Transactions* **45**, 8984-8993 (2016).
- 227 David Davey, P. W. E. W. C. L., Jr., Jay Wiggins, Mark Sullivan, David Pang, & Cantor, a. E. Cardiotonic Agents. 1. Novel 8-Aryl-Substituted Imidazo[1,2-a]- and -[1,5-a]pyridines and Imidazo[1,5-a]pyridinones as Potential Positive Inotropic Agents. *Journal of Medicinal Chemistry* **30**, 1337-1342 (1987).
- 228 Browne, L., Gude, C., Rodriguez, H., Steele, R. & Bhatnager, A. Fadrozole hydrochloride: a potent, selective, nonsteroidal inhibitor of aromatase for the treatment of estrogen-dependent disease. *Journal of Medicinal Chemistry* **34**, 725-736 (1991).
- 229 Fauber, B. P. *et al.* Discovery of imidazo [1, 5-a] pyridines and-pyrimidines as potent and selective ROR α inverse agonists. *Bioorganic & Medicinal Chemistry Letters* **25**, 2907-2912 (2015).
- 230 Volpi, G. & Rabezzana, R. Imidazo [1, 5-a] pyridine derivatives: useful, luminescent and versatile scaffolds for different applications. *New Journal of Chemistry* **45**, 5737-5743 (2021).
- 231 Volpi, G. *et al.* Blue fluorescent zinc (II) complexes based on tunable imidazo [1, 5-a] pyridines. *Inorganica Chimica Acta* **509**, 119662 (2020).
- 232 Volpi, G., Galliano, S., Buscaino, R., Viscardi, G. & Barolo, C. Fluorescent trifluoromethylated imidazo [1, 5-a] pyridines and their application in luminescent down-shifting conversion. *Journal of Luminescence* **242**, 118529 (2022).
- 233 Priyanga, S. *et al.* Coordination geometry-induced optical imaging of L-cysteine in cancer cells using imidazopyridine-based copper (II) complexes. *Dalton Transactions* **48**, 1489-1503 (2019).
- 234 Wang, J., Dyers, L., Mason, R., Amoyaw, P. & Bu, X. R. Highly efficient and direct heterocyclization of dipyridyl ketone to N, N-bidentate ligands. *The Journal of Organic Chemistry* **70**, 2353-2356 (2005).
- 235 Volpi, G. *et al.* One pot synthesis of low cost emitters with large Stokes' shift. *Dyes and Pigments* **137**, 152-164 (2017).
- 236 Chuprakov, S., Hwang, F. W. & Gevorgyan, V. Rh-Catalyzed Transannulation of Pyridotriazoles with Alkynes and Nitriles. *Angewandte Chemie* **119**, 4841-4843 (2007).
- 237 Shi, Y., Gulevich, A. V. & Gevorgyan, V. Rhodium-Catalyzed NH Insertion of Pyridyl Carbenes Derived from Pyridotriazoles: A General and Efficient Approach to 2-Picolylamines and Imidazo [1, 5-a] pyridines. *Angewandte Chemie International Edition in English* **53**, 14191-14195 (2014).

- 238 Wang, H. *et al.* Synthesis of 1, 3-disubstituted imidazo [1, 5-a] pyridines from amino acids via
catalytic decarboxylative intramolecular cyclization. *The Journal of Organic Chemistry* **81**, 3681-
3687 (2016).
- 239 Li, M. *et al.* Cu (I)-Catalyzed Transannulation of N-Heteroaryl Aldehydes or Ketones with
Alkylamines via C (sp³)-H Amination. *Organic Letter* **16**, 6232-6235 (2014).
- 240 Singh, P. K. & Singh, V. K. Highly Enantioselective Friedel– Crafts Reaction of Indoles with 2-
EnoylPyridine 1-Oxides Catalyzed by Chiral Pyridine 2, 6-Bis (5', 5'-diphenyloxazoline)- Cu (II)
Complexes. *Organic Letter* **10**, 4121-4124 (2008).
- 241 Kunz, H. & Rück, K. Carbohydrates as chiral auxiliaries in stereoselective synthesis. New
synthetic methods (90). *Angewandte Chemie International Edition in English* **32**, 336-358 (1993).
- 242 Zhu, F. & Walczak, M. A. Stereochemistry of transition metal complexes controlled by the
metallo-anomeric effect. *Journal of the American Chemical Society* **142**, 15127-15136 (2020).
- 243 Hong, L., Sun, W., Yang, D., Li, G. & Wang, R. Additive effects on asymmetric catalysis.
Chemical Reviews **116**, 4006-4123 (2016).
- 244 Rayne, S. & Forest, K. Theoretical studies on the pK_a values of perfluoroalkyl carboxylic
acids. *Nature Precedings*, 1-1 (2010).
- 245 Barnum, D. W. Predicting acid-base titration curves without calculations. *Journal of Chemical
Education* **76**, 938 (1999).
- 246 Smith, R. M. & Martell, A. E. *Critical stability constants: second supplement*. (Springer, 1989).
- 247 Goldberg, R. N., Kishore, N. & Lennen, R. M. Thermodynamic quantities for the ionization
reactions of buffers. *Journal of Physical and Chemical Reference Data* **31**, 231-370 (2002).
- 248 Kida, T., Sato, S.-i., Yoshida, H., Teragaki, A. & Akashi, M. 1, 1, 1, 3, 3, 3-Hexafluoro-2-
propanol (HFIP) as a novel and effective solvent to facilely prepare cyclodextrin-assembled
materials. *Chemical Communications* **50**, 14245-14248 (2014).
- 249 Roose, P., Eller, K., Henkes, E., Rossbacher, R. & Höke, H. Amines, aliphatic. *Ullmann's
Encyclopedia of Industrial Chemistry*, 1-55 (2000).
- 250 Olmstead, W. N., Margolin, Z. & Bordwell, F. G. Acidities of water and simple alcohols in
dimethyl sulfoxide solution. *The Journal of Organic Chemistry* **45**, 3295-3299 (1980).
- 251 Büldt, G., Gally, H., Seelig, A., Seelig, J. & Zaccai, G. Neutron diffraction studies on selectively
deuterated phospholipid bilayers. *Nature* **271**, 182-184 (1978).
- 252 Blake, M. I., Crespi, H. L. & Katz, J. J. Studies with deuterated drugs. *Journal of Pharmaceutical
Sciences* **64**, 367-391 (1975).
- 253 Geng, H. *et al.* Practical synthesis of C1 deuterated aldehydes enabled by NHC catalysis. *Nature
Catalysis* **2**, 1071-1077 (2019).
- 254 Su, J., Ma, X., Ou, Z. & Song, Q. Deconstructive Functionalizations of Unstrained Carbon–
Nitrogen Cleavage Enabled by Difluorocarbene. *ACS Central Science* (2020).
- 255 C. Schmidt, *Nature Biotechnology* **35**, 493-495 (2017).
- 256 Monteith, E. R. *et al.* Why we might be misusing process mass intensity (PMI) and a
methodology to apply it effectively as a discovery level metric. *Green Chemistry* **22**, 123-135
(2020).
- 257 Dang, C.-H., Nguyen, C.-H., Nguyen, T.-D. & Im, C. Synthesis and characterization of N-acyl-
tetra-O-acyl glucosamine derivatives. *RSC Advances* **4**, 6239-6245 (2014).
- 258 Kim, M., Kang, S. & Rhee, Y. H. De Novo Synthesis of Furanose Sugars: Catalytic Asymmetric
Synthesis of Apiose and Apiose-Containing Oligosaccharides. *Angewandte Chemie* **128**, 9885-
9889 (2016).
- 259 Omar, C.-A. & Fernando, S.-P. Nucleophilic substitution at the anomeric position of furanose
carbohydrates. The case of the C-allylations. *Carbohydrate Research* **510**, 108441 (2021).
- 260 Dass, A. V. *et al.* Equilibrium and non-equilibrium furanose selection in the ribose isomerisation
network. *Nature Communications* **12**, 1-10 (2021).
- 261 Nester, K., Gaweda, K. & Plazinski, W. A GROMOS force field for furanose-based
carbohydrates. *Journal of Chemical Theory and Computation* **15**, 1168-1186 (2019).

- 262 Hutt, J. T. *et al.* Fluorescence switching of imidazo [1, 5-a] pyridinium ions: pH-sensors with
dual emission pathways. *Organic Letters* **14**, 3162-3165 (2012).
- 263 Zeng, J. *et al.* Hydrogen-bonding-assisted exogenous nucleophilic reagent effect for β -selective
glycosylation of rare 3-amino sugars. *Journal of the American Chemical Society* **141**, 8509-8515
(2019).
- 264 Patyk, E. & Katrusiak, A. Transformable H-bonds and conformation in compressed glucose.
Chemical Science **6**, 1991-1995 (2015).
- 265 Zhu, F. & Walczak, M. A. Stereochemistry of transition metal complexes controlled by the
metallo-anomeric effect. *Journal of the American Chemical Society* **142**, 10 (2020).
- 266 Wang, J., Mason, R., VanDerveer, D., Feng, K. & Bu, X. R. Convenient preparation of a novel
class of imidazo [1, 5-a] pyridines: decisive role by ammonium acetate in chemoselectivity. *The
Journal of Organic Chemistry* **68**, 5415-5418 (2003).
- 267 Meyer, T. H. *et al.* Manganese (I)-Catalyzed Dispersion-Enabled C–H/C–C Activation.
Chemistry—A European Journal **23**, 5443-5447 (2017).
- 268 Mathur, N. K. & Narang, C. K. Chitin and chitosan, versatile polysaccharides from marine
animals. *Journal of Chemical Education* **67**, 938 (1990).
- 269 Croisier, F. & Jérôme, C. Chitosan-based biomaterials for tissue engineering. *European Polymer
Journal* **49**, 780-792 (2013).
- 270 Rinaudo, M. Chitin and chitosan: properties and applications. *Progress in Polymer Science* **31**,
603-632 (2006).
- 271 Kumar, M. R., Muzzarelli, R. A., Muzzarelli, C., Sashiwa, H. & Domb, A. Chitosan chemistry
and pharmaceutical perspectives. *Chemical Reviews* **104**, 6017-6084 (2004).
- 272 Jayakumar, R., Nwe, N., Tokura, S. & Tamura, H. Sulfated chitin and chitosan as novel
biomaterials. *International Journal of Biological Macromolecules* **40**, 175-181 (2007).
- 273 Ferrer, J. *et al.* Acid hydrolysis of shrimp-shell wastes and the production of single cell protein
from the hydrolysate. *Bioresource Technology* **57**, 55-60 (1996).
- 274 Jo, G.-H., Park, R.-D. & Jung, W.-J. Enzymatic production of chitin from crustacean shell waste.
Chitin, Chitosan, Oligosaccharides and Their Derivatives, 37-45 (2010).
- 275 Yang, R., Li, H., Huang, M., Yang, H. & Li, A. A review on chitosan-based flocculants and their
applications in water treatment. *Water Research* **95**, 59-89 (2016).
- 276 Deng, P., Xu, Z. & Kuang, Y. Electrochemical determination of bisphenol A in plastic bottled
drinking water and canned beverages using a molecularly imprinted chitosan–graphene composite
film modified electrode. *Food Chemistry* **157**, 490-497 (2014).
- 277 Rostrup-Nielsen, J. R. Making fuels from biomass. *Science* **308**, 1421-1422 (2005).
- 278 Somerville, C., Youngs, H., Taylor, C., Davis, S. C. & Long, S. P. Feedstocks for lignocellulosic
biofuels. *Science* **329**, 790-792 (2010).
- 279 Huang, N., Wang, P. & Jiang, D. Covalent organic frameworks: a materials platform for
structural and functional designs. *Nature Reviews Materials* **1**, 1-19 (2016).
- 280 Smil, V. Detonator of the population explosion. *Nature* **400**, 415-415 (1999).
- 281 Ferguson, E. & Libby, W. Mechanism for the fixation of nitrogen by lightning. *Nature* **229**, 37-37
(1971).
- 282 Rutledge, H. L. & Tezcan, F. A. Electron Transfer in Nitrogenase. *Chemical Reviews* **120**, 5158-
5193 (2020).
- 283 Chen, H. *et al.* Upgraded bioelectrocatalytic N₂ fixation: from N₂ to chiral amine intermediates.
Journal of the American Chemical Society **141**, 4963-4971 (2019).
- 284 Li, L. *et al.* Reduction of Dinitrogen via 2, 3'-Bipyridine-Mediated Tetraboration. *Journal of the
American Chemical Society* **142**, 6244-6250 (2020).
- 285 Meng, S.-L., Li, X.-B., Tung, C.-H. & Wu, L.-Z. Nitrogenase inspired artificial photosynthetic
nitrogen fixation. *Chem* **7**, 1431-1450 (2020).
- 286 Chen, C. *et al.* Coupling N₂ and CO₂ in H₂O to synthesize urea under ambient conditions.
Nature Chemistry **12**, 717-724 (2020).

- 287 Murray, R. & Smith, D. The activation of molecular nitrogen. *Coordination Chemistry Reviews* **3**,
429-470 (1968).
- 288 Brill, W. J. Biological nitrogen fixation. *Scientific American* **236**, 68-81 (1977).
- 289 MacFarlane, D. R. *et al.* A roadmap to the ammonia economy. *Joule* **4**, 1186-1205 (2020).
- 290 Allan, G. The kinetics of depolymerization of chitosan by nitrous acid. *Chitin and Chitosan*, 443-
466 (1989).
- 291 Muzzarelli, R. A. *Chitin*. (Elsevier, 2013).
- 292 Husteden, C. *et al.* Contact-triggered lipofection from multilayer films designed as surfaces for in
situ transfection strategies in tissue engineering. *ACS Applied Materials & Interfaces* **12**, 8963-
8977 (2020).
- 293 Wu, H., Li, H. & Fang, Z. Hydrothermal amination of biomass to nitrogenous chemicals. *Green
Chemistry* **23**, 6675-6697 (2021).
- 294 Song, S. *et al.* Integrating Biomass into the Organonitrogen Chemical Supply Chain: Production
of Pyrrole and d-Proline from Furfural. *Angewandte Chemie* **132**, 20018-20022 (2020).
- 295 Butola, B. S. *The impact and prospects of green chemistry for textile technology*. (Woodhead
Publishing, 2018).
- 296 Wang, J., Mason, R., VanDerveer, D., Feng, K. & Bu, X. R. Convenient preparation of a novel
class of imidazo [1, 5-a] pyridines: decisive role by ammonium acetate in chemoselectivity. *The
Journal of Organic Chemistry* **68**, 5415-5418 (2003).
- 297 Sharma, P., Kumar, A., Sharma, S. & Rane, N. Studies on synthesis and evaluation of
quantitative structure–activity relationship of 5-[(3'-chloro-4', 4'-disubstituted-2-
oxoazetidinyl)(N-nitro) amino]-6-hydroxy-3-alkyl/aryl [1, 3] azaphospholo [1, 5-a] pyridin-1-yl-
phosphorus dichlorides. *Bioorganic & Medicinal Chemistry Letters* **15**, 937-943 (2005).
- 298 James, D. A. *et al.* Indole-and indolizine-glyoxylamides displaying cytotoxicity against multidrug
resistant cancer cell lines. *Bioorganic & Medicinal Chemistry Letters* **18**, 1784-1787 (2008).
- 299 Gundersen, L. L., Negussie, A. H., Rise, F. & Oestby, O. B. Antimycobacterial Activity of 1-
Substituted Indolizines. *Archiv der Pharmazie: An International Journal Pharmaceutical and
Medicinal Chemistry* **336**, 191-195 (2003).
- 300 Sandeep, C. *et al.* Greener synthesis of indolizine analogues using water as a base and solvent:
study for larvicidal activity against *Anopheles arabiensis*. *Chemical Biology & Drug Design* **88**,
899-904 (2016).
- 301 Harrell, W. B. & Doerge, R. F. Mannich bases from 2-phenylindolizines I. 3-alkyl-1-
dialkylaminomethyl derivatives. *Journal of Pharmaceutical Sciences* **56**, 225-228 (1967).
- 302 Gundersen, L.-L., Charnock, C., Negussie, A. H., Rise, F. & Teklu, S. Synthesis of indolizine
derivatives with selective antibacterial activity against *Mycobacterium tuberculosis*. *European
Journal of Pharmaceutical Sciences* **30**, 26-35 (2007).
- 303 Kemnitzer, W. *et al.* Discovery of 1-benzoyl-3-cyanopyrrolo [1, 2-a] quinolines as a new series of
apoptosis inducers using a cell-and caspase-based high-throughput screening assay. 2: Structure–
activity relationships of the 4-, 5-, 6-, 7-and 8-positions. *Bioorganic & Medicinal Chemistry
Letters* **19**, 3481-3484 (2009).
- 304 Kim, E., Lee, Y., Lee, S. & Park, S. B. Discovery, understanding, and bioapplication of organic
fluorophore: a case study with an indolizine-based novel fluorophore, Seoul-Fluor. *Accounts of
Chemical Research* **48**, 538-547 (2015).
- 305 Correia, J. T. M., List, B. & Coelho, F. Catalytic Asymmetric Conjugate Addition of Indolizines
to α , β -Unsaturated Ketones. *Angewandte Chemie International Edition* **56**, 7967-7970 (2017).
- 306 Zhang, D., Song, H. & Qin, Y. Total synthesis of indoline alkaloids: a cyclopropanation strategy.
Accounts of Chemical Research **44**, 447-457 (2011).
- 307 Boekelheide, V. & Windgassen Jr, R. Syntheses of pyrrocolines unsubstituted in the five-
membered ring1. *Journal of the American Chemical Society* **81**, 1456-1459 (1959).
- 308 Scholtz, M. Die Einwirkung von Essigsäureanhydrid auf α -Picolin. *Berichte der deutschen
chemischen Gesellschaft* **45**, 734-746 (1912).

- 309 Tschitschibabin, A. Tautomerie in der Pyridin-Reihe. *Berichte der deutschen chemischen Gesellschaft (A and B Series)* **60**, 1607-1617 (1927).
- 310 Bora, U., Saikia, A. & Boruah, R. C. A novel microwave-mediated one-pot synthesis of indolizines via a three-component reaction. *Organic Letters* **5**, 435-438 (2003).
- 311 Poissonnet, G., Théret-Bettiol, M.-H. & Dodd, R. H. Preparation and 1, 3-dipolar cycloaddition reactions of β -carboline azomethine ylides: A direct entry into C-1-and/or C-2-functionalized indolizino [8, 7-b] indole derivatives. *The Journal of Organic Chemistry* **61**, 2273-2282 (1996).
- 312 Katritzky, A. R., Qiu, G., Yang, B. & He, H.-Y. Novel syntheses of indolizines and pyrrolo [2, 1-a] isoquinolines via benzotriazole methodology. *The Journal of Organic Chemistry* **64**, 7618-7621 (1999).
- 313 Kel'in, A. V., Sromek, A. W. & Gevorgyan, V. A novel Cu-assisted cycloisomerization of alkynyl imines: efficient synthesis of pyrroles and pyrrole-containing heterocycles. *Journal of the American Chemical Society* **123**, 2074-2075 (2001).
- 314 Seregin, I. V. & Gevorgyan, V. Gold-catalyzed 1, 2-migration of silicon, tin, and germanium en route to C-2 substituted fused pyrrole-containing heterocycles. *Journal of the American Chemical Society* **128**, 12050-12051 (2006).
- 315 Jin, T. *et al.* Iron-Catalyzed Aerobic Oxidation and Annulation Reaction of Pyridine and α -Substituted Allenolate toward Functionalized Indolizine. *Organic Letters* **20**, 413-416 (2018).
- 316 Kalinin, A. A. *et al.* Synthesis and characterization of new second-order NLO chromophores containing the isomeric indolizine moiety for electro-optical materials. *Dyes and Pigments* **147**, 444-454 (2017).
- 317 Colonna, M., Greci, L. & Poloni, M. Electrophilic ipso-substitutions. Part 3. Reactions of 3-substituted indoles, 4-substituted N, N-dimethylanilines, and 1-and 3-substituted indolizines with nitrous acid. *Journal of the Chemical Society, Perkin Transactions 2*, 165-169 (1984).
- 318 Li, H. *et al.* Synthesis of indolizines via reaction of 2-substituted azaarenes with enals by an amine-NHC relay catalysis. *Organic Letters* **19**, 2010-2013 (2017).
- 319 Yuan, Y.-C., Liu, T.-Z. & Zhao, B.-X. Metal-Free Catalyzed Synthesis of Fluorescent Indolizine Derivatives. *The Journal of Organic Chemistry* **86**, 12737-12744 (2021).
- 320 Meazza, M. *et al.* Synergistic catalysis: highly enantioselective acetyl aza-arene addition to enals. *Chemistry-A European Journal* **24**, 13306-13310 (2019).
- 321 Seebach, D. Structure and Reactivity of Lithium Enolates. From Pinacolone to Selective C-Alkylations of Peptides. Difficulties and Opportunities Afforded by Complex Structures. *Angewandte Chemie International Edition in English* **27**, 1624-1654 (1988).
- 322 Anouz, R., Repanas, A., Schwarz, E. & Groth, T. Novel Surface Coatings Using Oxidized Glycosaminoglycans as Delivery Systems of Bone Morphogenetic Protein 2 (BMP-2) for Bone Regeneration. *Macromolecular Bioscience* **18**, 1800283 (2018).
- 323 Glavnik, V., Simonovska, B. & Vovk, I. Densitometric determination of (+)-catechin and (-)-epicatechin by 4-dimethylaminocinnamaldehyde reagent. *Journal of Chromatography A* **1216**, 4485-4491 (2009).
- 324 Reux, B., Nevalainen, T., Raitio, K. H. & Koskinen, A. M. Synthesis of quinolinyl and isoquinolinyl phenyl ketones as novel agonists for the cannabinoid CB2 receptor. *Bioorganic & Medicinal Chemistry* **17**, 4441-4447 (2009).
- 325 Demkiw, K. *et al.* A nitrogen-assisted one-pot heteroaryl ketone synthesis from carboxylic acids and heteroaryl halides. *The Journal of Organic Chemistry* **81**, 3447-3456 (2016).
- 326 Yang, H. *et al.* Rhodium catalyzed asymmetric hydrogenation of 2-pyridine ketones. *Organic Letters* **17**, 4144-4147 (2015).
- 327 Sharma, S., Kumar, M., Vishwakarma, R. A., Verma, M. K. & Singh, P. P. Room temperature metal-catalyzed oxidative acylation of electron-deficient heteroarenes with alkynes, its mechanism, and application studies. *The Journal of Organic Chemistry* **83**, 12420-12431 (2018).

- 328 Reger, D. L., Gardinier, J. R., Smith, M. D. & Pellechia, P. J. Supramolecular assembly and solution properties of bis (bipyridyl) ruthenium (II) coordination complexes of aryl (2-pyridyl) methanones. *Inorganic Chemistry* **42**, 482-491 (2003).
- 329 Patil, S. G., Jadhav, J. S. & Sankpal, S. T. Mg 3 N 2-assisted one-pot synthesis of 1, 3-disubstituted imidazo [1, 5-a] pyridine. *RSC Advances* **10**, 11808-11815 (2020).
- 330 Hu, Z., Hou, J., Liu, J., Yu, W. & Chang, J. Synthesis of imidazo [1, 5-a] pyridines via I 2-mediated sp³ C–H amination. *Organic & Biomolecular Chemistry* **16**, 5653-5660 (2018).
- 331 Volpi, G. *et al.* Halogenated imidazo [1, 5-a] pyridines: chemical structure and optical properties of a promising luminescent scaffold. *Dyes and Pigments* **171**, 107713 (2019).
- 332 Garino, C. *et al.* Spectroscopic and computational study on new blue emitting ReL(CO)₃Cl complexes containing pyridylimidazo [1, 5-a] pyridine ligands. *European Journal of Inorganic Chemistry*, 3587-3591, (2008).